

Kinetics of the reduction of ferrous chloride with hydro

Canadian Journal of Chemical Engineering

42, 247-253

DOI: 10.1002/cjce.5450420603

Citation Report

#	ARTICLE	IF	CITATIONS
1	A rate equation for the reduction of ferrous chloride by hydrogen. Canadian Journal of Chemical Engineering, 1965, 43, 173-177.	1.7	7
2	Hydrogen reduction of ferrous chloride. Canadian Journal of Chemical Engineering, 1966, 44, 117-118.	1.7	5
3	Thermodynamics of the Chlorination of Iron, Cobalt, Nickel and Copper Sulphides. Canadian Metallurgical Quarterly, 1967, 6, 39-54.	1.2	7
4	Hydrogen reduction of chromium chlorides: A kinetic investigation. Canadian Journal of Chemical Engineering, 1968, 46, 450-456.	1.7	1
5	THE PEACE RIVER PROCESS FOR THE PRODUCTION OF IRON POWDER. Powder Metallurgy, 1968, 11, 224-232.	1.7	18
6	Hydrogen reduction of the chlorides of bivalent chromium and iron. Canadian Journal of Chemical Engineering, 1970, 48, 84-90.	1.7	4
7	Criterion of applicability of the moving boundary model. Journal of Catalysis, 1970, 17, 143-150.	6.2	13
8	The reduction of ferrous chloride with hydrogen. Canadian Journal of Chemical Engineering, 1971, 49, 804-809.	1.7	3
9	Experimental Techniques for the Study of Gas-Solid Reactions. , 1976, , 205-247.		3
10	Preparation of High-Purity Iron by Means of Ammonium Chloride Complexes. Russian Journal of Inorganic Chemistry, 2018, 63, 736-741.	1.3	4
11	Experimental techniques in fluid-solid reactions. , 2020, , 479-514.		0
12	Synthesis and structural characterizations of CrCoFeNiMnx (0 ≤ x ≤ 1) high-entropy-alloy thin films by thermal reduction in hydrogen. Journal of Materials Science, 0, , .	3.7	