## Kinetics of the Methanol Reaction with OH at Interstell Temperatures

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**Citation Report** 

#	ARTICLE Reply to the †Comment on "Methanol dimer formation drastically enhances hydrogen abstraction	IF	CITATIONS
1	from methanol by OH at low temperatureâ€â€™ by D. Heard, R. Shannon, J. Gomez Martin, R. Caravan, M. Blitz, J. Plane, M. Antiñolo, M. Agundez, E. Jimenez, B. Ballesteros, A. Canosa, G. El Dib, J. Albaladejo and J. Cernicharo, <i>Phys. Chem. Chem. Phys.</i> , 2018, <b>20</b> , DOI: 10.1039/C7CP04561A. Physical Chemistry Chemical Physics, 2018, 20, 8355-8357.	2.8	5
2	<ul> <li><i>Ab Initio</i>, Transition State Theory, and Kinetic Modeling Study of the HO<sub>2</sub>-Assisted</li> <li>Ketoâ€"Enol Tautomerism Propen-2-ol + HO<sub>2</sub> â‡" Acetone + HO<sub>2</sub> under Combustion,</li> <li>Atmospheric, and Interstellar Conditions. Journal of Physical Chemistry A, 2018, 122, 9792-9805.</li> </ul>	2.5	12
3	Low temperature reaction dynamics for CH <sub>3</sub> OH + OH collisions on a new full dimensional potential energy surface. Physical Chemistry Chemical Physics, 2018, 20, 25951-25958.	2.8	32
4	Pressure-dependent kinetics of methyl formate reactions with OH at combustion, atmospheric and interstellar temperatures. Physical Chemistry Chemical Physics, 2018, 20, 26190-26199.	2.8	40
5	Rapid Acceleration of Hydrogen Atom Abstraction Reactions of OH at Very Low Temperatures through Weakly Bound Complexes and Tunneling. Accounts of Chemical Research, 2018, 51, 2620-2627.	15.6	36
6	A shock tube kinetic study on the branching ratio of methanol +ÂOH reaction. Proceedings of the Combustion Institute, 2019, 37, 153-162.	3.9	14
7	Chemical Kinetics Approves the Occurrence of C ( <sup>3</sup> P <sub><i>j</i></sub> ) Reaction with H <sub>2</sub> O. Journal of Physical Chemistry A, 2019, 123, 5877-5892.	2.5	1
8	Low-Temperature Kinetic Isotope Effects in CH <sub>3</sub> OH + H → CH <sub>2</sub> OH + H <sub>2</sub> Shed Light on the Deuteration of Methanol in Space. Journal of Physical Chemistry A, 2019, 123, 9061-9068.	2.5	13
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10	Multishelled Hollow Structures of Yttrium Oxide for the Highly Selective and Ultrasensitive Detection of Methanol. Small, 2019, 15, e1804688.	10.0	22
11	Experimental Studies of Gas-Phase Reactivity in Relation to Complex Organic Molecules in Star-Forming Regions. ACS Earth and Space Chemistry, 2019, 3, 1109-1134.	2.7	34
12	Zero- and High-Pressure Mechanisms in the Complex Forming Reactions of OH with Methanol and Formaldehyde at Low Temperatures. ACS Earth and Space Chemistry, 2019, 3, 1158-1169.	2.7	14
13	Chemical kinetics of H-abstractions from dimethyl amine by H, CH <sub>3</sub> , OH, and HO <sub>2</sub> radicals with multi-structural torsional anharmonicity. Physical Chemistry Chemical Physics, 2019, 21, 12685-12696.	2.8	21
14	Weak Interactions in Interstellar Chemistry: How Do Open Shell Molecules Interact with Closed Shell Molecules?. ACS Earth and Space Chemistry, 2019, 3, 1080-1095.	2.7	9
15	Gas-phase reactivity of CH <sub>3</sub> OH toward OH at interstellar temperatures (11.7–177.5 K): experimental and theoretical study. Physical Chemistry Chemical Physics, 2019, 21, 6942-6957.	2.8	42
16	Quantum Roaming in the Complex-Forming Mechanism of the Reactions of OH with Formaldehyde and Methanol at Low Temperature and Zero Pressure: A Ring Polymer Molecular Dynamics Approach. Journal of Physical Chemistry Letters, 2019, 10, 1900-1907.	4.6	26
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20	Computational kinetics of the hydrogen abstraction reactions of <i>n</i> -propanol and iso-propanol by OH radical. Physical Chemistry Chemical Physics, 2019, 21, 24458-24468.	2.8	15
21	Kinetics of the Hydrogen Abstraction PAH + <sup>•</sup> OH → PAH Radical + H <sub>2</sub> O Reaction Class: An Application of the Reaction Class Transition State Theory (RC-TST) and Structure–Activity Relationship (SAR). Journal of Physical Chemistry A, 2019, 123, 750-763.	2.5	9
22	Improving the Prediction Accuracy of the Extinction of Stretched Methanol/Air Premixed Flames. Combustion Science and Technology, 2020, 192, 1088-1107.	2.3	4
23	Physisorbed State Regulates the Dissociation Mechanism of H2O on Ni(100). Journal of Physical Chemistry A, 2020, 124, 8724-8732.	2.5	5
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25	Quantum Mechanical Tunneling Is Essential to Understanding Chemical Reactivity. Trends in Chemistry, 2020, 2, 980-989.	8.5	57
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30	Temperature coefficient (Q10) and its applications in biological systems: Beyond the Arrhenius theory. Ecological Modelling, 2020, 431, 109127.	2.5	38
31	Direct dynamics of a large complex hydrocarbon reaction system: The reaction of OH with exo-tricyclodecane (the main component of Jet Propellant-10). Combustion and Flame, 2020, 216, 82-91.	5.2	8
32	DFT Study on the Mechanism of the Water Gas Shift Reaction Over Ni <sub><i>x</i></sub> P <sub><i>y</i></sub> Catalysts: The Role of P. Journal of Physical Chemistry C, 2020, 124, 6598-6610.	3.1	18
33	Pilgrim: A thermal rate constant calculator and a chemical kinetics simulator. Computer Physics Communications, 2020, 256, 107457.	7.5	30
34	Water Catalysis of the Reaction of Methanol with OH Radical in the Atmosphere is Negligible. Angewandte Chemie, 2020, 132, 10918-10922.	2.0	10
35	Water Catalysis of the Reaction of Methanol with OH Radical in the Atmosphere is Negligible. Angewandte Chemie - International Edition, 2020, 59, 10826-10830.	13.8	13
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40	A kinetics study on hydrogen abstraction reactions of cyclopentane by hydrogen, methyl, and ethyl radicals. Physical Chemistry Chemical Physics, 2021, 23, 7333-7342.	2.8	4
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43	Multistructural Variational Reaction Kinetics of the Simplest Unsaturated Methyl Ester: H-Abstraction from Methyl Acrylate by H, OH, CH <sub>3</sub> , and HO <sub>2</sub> Radicals. Journal of Physical Chemistry A, 2021, 125, 5103-5116.	2.5	11
44	Effect of ammonia and water molecule on OH + CH3OH reaction under tropospheric condition. Scientific Reports, 2021, 11, 12185.	3.3	10
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69	A Theoretical Study of H-Abstractions of Benzaldehyde by H, O <sup>3</sup> (P), <sup>3</sup> O <sub>2</sub> , OH, HO <sub>2</sub> , and CH <sub>3</sub> Radicals: Ab Initio Rate Coefficients and Their Uncertainty Quantification. Journal of Physical Chemistry A, 2022, 126, 7523-7533.	2.5	2
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75	<b>Analytical potential energy surface and dynamics for the OH + CH<sub>3</sub>OH reaction</b> . Journal of Chemical Physics, 0, , .	3.0	4
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