

Role of O<sub>2</sub> + QOOH in Low-Temperature Ignition  
Pressure Dependent Rate Coefficients

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Citation Report

#	ARTICLE	IF	CITATIONS
1	High-Pressure Rate Rules for Alkyl + O <sub>2</sub> Reactions. 2. The Isomerization, Cyclic Ether Formation, and $\beta$ -Scission Reactions of Hydroperoxy Alkyl Radicals. <i>Journal of Physical Chemistry A</i> , 2012, 116, 5068-5089.	2.5	172
2	MESMER: An Open-Source Master Equation Solver for Multi-Energy Well Reactions. <i>Journal of Physical Chemistry A</i> , 2012, 116, 9545-9560.	2.5	467
3	New Pathways for Formation of Acids and Carbonyl Products in Low-Temperature Oxidation: The Korcek Decomposition of $\beta$ -Keto hydroperoxides. <i>Journal of the American Chemical Society</i> , 2013, 135, 11100-11114.	13.7	153
4	Rate Rules, Branching Ratios, and Pressure Dependence of the HO <sub>2</sub> + Olefin Addition Channels. <i>Journal of Physical Chemistry A</i> , 2013, 117, 6458-6473.	2.5	57
5	Directly measuring reaction kinetics of $\dot{E}^{\text{TM}}\text{QOOH}$ – a crucial but elusive intermediate in hydrocarbon autoignition. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 10753.	2.8	58
6	Uncertainty propagation in the derivation of phenomenological rate coefficients from theory: A case study of n-propyl radical oxidation. <i>Proceedings of the Combustion Institute</i> , 2013, 34, 177-185.	3.9	64
7	A semi-detailed chemical kinetic model of a gasoline surrogate fuel for internal combustion engine applications. <i>Fuel</i> , 2013, 113, 347-356.	6.4	32
8	Reformulation and Solution of the Master Equation for Multiple-Well Chemical Reactions. <i>Journal of Physical Chemistry A</i> , 2013, 117, 12146-12154.	2.5	461
9	Reaction Pathways for the Thermal Decomposition of Methyl Butanoate. <i>Journal of Organic Chemistry</i> , 2013, 78, 5898-5908.	3.2	35
10	Combustion and pyrolysis of iso-butanol: Experimental and chemical kinetic modeling study. <i>Combustion and Flame</i> , 2013, 160, 1907-1929.	5.2	65
11	Synchrotron Photoionization Mass Spectrometry Measurements of Product Formation in Low-Temperature <i>n</i> -Butane Oxidation: Toward a Fundamental Understanding of Autoignition Chemistry and <i>n</i> -C <sub>4</sub> H <sub>9</sub> + O <sub>2</sub> / <i>s</i> -C <sub>4</sub> H <sub>9</sub> + O <sub>2</sub> Reactions. <i>Journal of Physical Chemistry A</i> , 2013, 117, 12216-12235.	2.5	33
12	A coordinated investigation of the combustion chemistry of diisopropyl ketone, a prototype for biofuels produced by endophytic fungi. <i>Combustion and Flame</i> , 2014, 161, 711-724.	5.2	54
13	Experiments and modeling of propane combustion with vitiation. <i>Combustion and Flame</i> , 2014, 161, 2038-2053.	5.2	58
14	Pathways, kinetics and thermochemistry of methyl-ester peroxy radical decomposition in the low-temperature oxidation of methyl butanoate: A computational study of a biodiesel fuel surrogate. <i>Combustion and Flame</i> , 2014, 161, 2270-2287.	5.2	29
15	An improved kinetic mechanism for 3-pentanone pyrolysis and oxidation developed using multispecies time histories in shock-tubes. <i>Combustion and Flame</i> , 2014, 161, 1135-1145.	5.2	23
16	Experimental and ab Initio Investigations of H <sub>2</sub> S-Assisted Propane Oxidative Dehydrogenation Reactions. <i>Journal of Physical Chemistry A</i> , 2014, 118, 1541-1556.	2.5	11
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20	A shock tube laser schlieren study of methyl acetate dissociation in the fall-off regime. <i>Physical Chemistry Chemical Physics</i> , 2014, 16, 7241.	2.8	13
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25	Towards a quantitative understanding of the role of non-Boltzmann reactant distributions in low temperature oxidation. <i>Proceedings of the Combustion Institute</i> , 2015, 35, 205-213.	3.9	48
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34	New reaction classes in the kinetic modeling of low temperature oxidation of <i>n</i> -alkanes. <i>Combustion and Flame</i> , 2015, 162, 1679-1691.	5.2	214
35	Revisiting the Kinetics and Thermodynamics of the Low-Temperature Oxidation Pathways of Alkanes: A Case Study of the Three Pentane Isomers. <i>Journal of Physical Chemistry A</i> , 2015, 119, 7510-7527.	2.5	202
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37	Comparison of Three Isoelectronic Multiple-Well Reaction Systems: $\text{OH} + \text{CH}_2\text{O}$ , $\text{OH} + \text{CH}_2\text{CH}_2$ , and $\text{OH} + \text{CH}_2\text{NH}$ . <i>Journal of Physical Chemistry A</i> , 2015, 119, 7578-7592.	2.5	47

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120	Low to intermediate temperature oxidation studies of dimethoxymethane/n-heptane blends in a jet-stirred reactor. <i>Combustion and Flame</i> , 2019, 207, 20-35.	5.2	23
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