Volumetric Properties of Aqueous Sodium Chloride Sol

Journal of Physical and Chemical Reference Data 11, 15-81 DOI: 10.1063/1.555660

Citation Report

#	Article	IF	CITATIONS
2	Survey of selected non-thermodynamic properties and chemical phenomena of fluids and fluid mixtures. Physics and Chemistry of the Earth, 1981, 13-14, 65-88.	0.3	7
3	Thermodynamics of saturated aqueous solutions including mixtures of NaCl, KCl, and CsCl. Journal of Solution Chemistry, 1983, 12, 171-185.	1.2	28
4	Concentrated Aqueous Sodium Chloride Solutions from 200 to 600°C and to 3000 bar Phase Equilibria and PVT-Data. Zeitschrift Fur Elektrotechnik Und Elektrochemie, 1983, 87, 597-600.	0.9	48
5	Thermodynamics of aqueous sodium chloride to 823 K and 1 kilobar (100 MPa). Proceedings of the National Academy of Sciences of the United States of America, 1983, 80, 7689-7693.	7.1	36
6	The enthalpy of dilution of aqueous sodium chloride to 673 K using a new heat-flow and liquid-flow microcalorimeter. Excess thermodynamic properties and their pressure coefficients. Journal of Chemical Thermodynamics, 1984, 16, 343-372.	2.0	96
7	Enthalpies of dilution and heat capacities of aqueous solutions of sodiumn-dodecylsulfonate and sodium 4-(1-methylundecyl)benzenesulfonate from 347 to 451°K. Journal of Colloid and Interface Science, 1984, 100, 68-81.	9.4	26
8	Isentropic compressibilities of univalent electrolytes in methanol at 25�C. Journal of Solution Chemistry, 1984, 13, 699-720.	1.2	32
9	Highâ€precision flow densimeter for fluids at temperatures to 700 K and pressures to 40 MPa. Review of Scientific Instruments, 1984, 55, 589-593.	1.3	83
10	Downhole Density of Heavy Brines. , 1984, , .		9
11	Properties of aqueous solutions near the critical point of water: some remarkable effects. Fluid Phase Equilibria, 1985, 20, 283-296.	2.5	9
12	Estimation of marine sediment bulk physical properties at depth from seafloor geophysical measurements. Journal of Geophysical Research, 1986, 91, 14033-14043.	3.3	62
13	Modeling the PVT properties of concentrated electrolytes in water. AICHE Journal, 1986, 32, 1561-1566.	3.6	3
14	PVT properties of concentrated aqueous electrolytes: V. Densities and apparent molal volumes of the four major sea salts from dilute solution to saturation and from 0 to 100�C. Journal of Solution Chemistry, 1986, 15, 989-1002.	1.2	101
15	Densities and apparent molal volumes of aqueous concentrated calcium chloride solutions from 50 to 200�C at 20.27 bar. Journal of Solution Chemistry, 1986, 15, 409-412.	1.2	26
16	The apparent molar heat capacity of aqueous hydrochloric acid from 10 to 140�C. Journal of Solution Chemistry, 1986, 15, 1-22.	1.2	60
17	Enthalpy of dilution of aqueous sodium chloride from 76 to 225�C and aqueous dodecyltrimethylammonium bromide from 50 to 225�C. Journal of Solution Chemistry, 1986, 15, 727-742.	1.2	16
18	Density Modeling for Pure and Mixed-Salt Brines as a Function of Composition, Temperature, and Pressure. , 1987, , .		13
19	Use of Water Vapor Desorption Data in the Determination of Capillary Pressures. , 1987, , .		5

#	Article	IF	CITATIONS
20	Thermodynamics of NaOH(aq) in hydrothermal solutions. Geochimica Et Cosmochimica Acta, 1987, 51, 829-837.	3.9	91
21	CORRELATION OF PARTIAL MOLAR VOLUMES AT INFINITE DILUTION OF SALTS IN WATER. Chemical Engineering Communications, 1987, 56, 341-349.	2.6	28
22	Quartz emented breccias from the Midâ€Atlantic Ridge: Samples of a highâ€salinity hydrothermal upflow zone. Journal of Geophysical Research, 1987, 92, 9175-9192.	3.3	106
23	Thermodynamics of concentrated electrolyte mixtures and the prediction of mineral solubilities to high temperatures for mixtures in the system Na-K-Mg-Cl-SO4-OH-H2O. Geochimica Et Cosmochimica Acta, 1987, 51, 2429-2443.	3.9	302
24	Determination of the homogenization temperatures and densities of supercritical fluids in the system NaClKClCaCl2H2O using synthetic fluid inclusions. Chemical Geology, 1987, 64, 335-350.	3.3	600
25	Densities and apparent molal volumes of aqueous CaCl2 and MgCl2 solutions. Journal of Solution Chemistry, 1987, 16, 1035-1048.	1.2	33
26	PVT properties of concentrated electrolytes. VI. The speed of sound and apparent molal compressibilities of NaCl, Na2SO4, MgCl2, and MgSO4 solutions from 0 to 100�C. Journal of Solution Chemistry, 1987, 16, 269-284.	1.2	70
27	Heat capacities of aqueous LiCl from 306 to 603 K at 17.5 MPa. Journal of Chemical Thermodynamics, 1987, 19, 1037-1045.	2.0	28
28	Heat capacities of aqueous NaBr from 306 to 603 K at 17.5 MPa. Journal of Chemical Thermodynamics, 1987, 19, 493-503.	2.0	11
29	Method for the representation of thermodynamic properties of surfactant solutions. Journal of Colloid and Interface Science, 1988, 124, 585-590.	9.4	2
30	Mixtures of 1-1 electrolytes: Densities and excess volumes of aqueous NaCl-NaBr solutions at 25�C. Monatshefte Für Chemie, 1988, 119, 1201-1206.	1.8	5
31	Thermodynamics of aqueous association and ionization reactions at high temperatures and pressures. Journal of Solution Chemistry, 1988, 17, 699-718.	1.2	225
32	Heat loss corrections for heat capacity flow calorimeters. Journal of Solution Chemistry, 1988, 17, 733-750.	1.2	9
33	The volume of ions and ion-solvent pair correlation functions. International Journal of Thermophysics, 1988, 9, 689-702.	2.1	6
34	Contribution to the thermodynamic study of electrolyte solutions. Computers and Chemical Engineering, 1988, 12, 461-467.	3.8	1
35	Volumetric properties of aqueous NaCl solutions from 0.0025 to 5.0 mol · kgâ^'1, 323 to 600 K, and 0.1 to 40 MPa. Journal of Chemical Thermodynamics, 1988, 20, 949-968.	2.0	58
36	Apparent molar volumes of aqueous decyl- and dodecyltrimethylammonium bromides from 74 to 176°C at 10 and 320 bar. Journal of Colloid and Interface Science, 1988, 124, 591-605.	9.4	12
37	Heat capacity and other thermodynamic properties of Na2SO4(aq) in hydrothermal solutions and the solubilities of sodium sulfate minerals in the system Na-Cl-SO4-OH-H2O to 300°C. Geochimica Et Cosmochimica Acta, 1988, 52, 2393-2404.	3.9	86

#	Article	IF	CITATIONS
38	Response of twoâ€phase fluids to fracture configurations within submarine hydrothermal systems. Journal of Geophysical Research, 1988, 93, 4585-4594.	3.3	66
39	Use of Water-Vapor Desorption Data in the Determination of Capillary Pressures at Low Water Saturations. SPE Reservoir Engineering, 1988, 3, 913-918.	0.5	15
40	Density Modeling for Brines as a Function of Compositions Temperature, and Pressure. SPE Production Engineering, 1989, 4, 394-400.	0.3	26
41	Extension of Chen's(1982) theory to correlate densities of aqueous electrolytes. AICHE Journal, 1989, 35, 293-299.	3.6	15
42	The enthalpy of dilution and apparent molar heat capacity of NaOH(aq) to 523 K and 40 MPa. Journal of Chemical Thermodynamics, 1989, 21, 561-584.	2.0	84
43	Comparison of calibration methods for flow heat-capacity calorimeters and heat capacities of concentrated NaCl(aq) to 598 K. Journal of Chemical Thermodynamics, 1989, 21, 595-614.	2.0	35
44	Volume changes for mixing the major sea salts: Equations valid to ionic strength 3.0 and temperature 95�C. Journal of Solution Chemistry, 1989, 18, 1007-1017.	1.2	23
45	Solution thermodynamics of first row transition elements. 3. Apparent molal volumes of aqueous ZnCl2 and Zn(ClO4)2 from 15 to 55ïį½2C and an examination of solute-solute and solute-solvent interactions. Journal of Solution Chemistry, 1989, 18, 249-264.	1.2	38
46	An ion interaction model for the volumetric properties of natural waters: Density of the solution and partial molal volumes of electrolytes to high concentrations at 25ŰC. Geochimica Et Cosmochimica Acta, 1989, 53, 1177-1188.	3.9	79
47	Concentrated Mixtures of 1:1 Electrolytes: Densities and Excess Volumes of Aqueous KBr with KCl and NaBr. Bulletin of the Chemical Society of Japan, 1989, 62, 3311-3314.	3.2	3
48	The PVT properties of concentrated aqueous electrolytes IX. The volume properties of KCl and K2SO4 and their mixtures with NaCl and Na2SO4 as a function of temperature. Journal of Solution Chemistry, 1990, 19, 353-374.	1.2	66
49	Measurement of relative quantum efficiencies for the relaxation of conductivity of monovalent electrolytes and polyvalent electrolytes using a modulation technique. Electrochimica Acta, 1990, 35, 861-867.	5.2	0
50	Equation of state of aqueous NaCl solutions over a wide range of temperatures, pressures and concentrations. Fluid Phase Equilibria, 1990, 60, 273-287.	2.5	50
51	Volumetric properties of 1–1 aqueous electrolyte solutions and ionic hydration. Journal of Chemical Physics, 1990, 93, 2939-2945.	3.0	28
52	Effect of a dissolved gas on the solubility of an electrolyte in aqueous solution. Industrial & Engineering Chemistry Research, 1990, 29, 1043-1050.	3.7	18
53	The influence of pressure on the activity coefficients of the solutes and on the solubility of minerals in the system Na-Ca-Cl-SO4-H2O to 200°C and 1 kbar and to high NaCl concentration. Geochimica Et Cosmochimica Acta, 1990, 54, 3265-3282.	3.9	75
54	A new version of vibrating-tube flow densitometer for measurements at temperatures up to 730 K. Journal of Chemical Thermodynamics, 1991, 23, 333-344.	2.0	29
55	Modification of a C-80 Setaram calorimeter for measuring heat capacities of liquids at temperatures up to 548 K and pressures up to 20 MPa. Journal of Chemical Thermodynamics, 1991, 23, 1075-1083.	2.0	27

#	Article	IF	CITATIONS
56	A site–site theory for finite concentration saline solutions. Journal of Chemical Physics, 1992, 97, 7656-7666.	3.0	273
57	Calculation of the thermodynamic properties of aqueous species at high pressures and temperatures. Effective electrostatic radii, dissociation constants and standard partial molal properties to 1000 °C and 5 kbar. Journal of the Chemical Society, Faraday Transactions, 1992, 88, 803-826.	1.7	454
58	Stability of CO2 clathrate hydrate + CO2 liquid + CO2 vapour + aqueous KCl-NaCl solutions: Experimental determination and application to salinity estimates of fluid inclusions â^—. Geochimica Et Cosmochimica Acta, 1992, 56, 273-280.	3.9	182
59	Aqueous partial molar heat capacities and volumes for NaTcO4 and NaReO4. Journal of Solution Chemistry, 1992, 21, 507-523.	1.2	12
60	A simplified ionic hydration model with one parameter for prediction of vapour pressures of aqueous electrolytes at elevated temperatures. Chemical Engineering Science, 1992, 47, 4217-4220.	3.8	3
61	Viscosities, electrolytic conductivities, and volumetric properties of HCl-MClx-H2O as a function of temperature up to high molal ionic strengths. Fluid Phase Equilibria, 1993, 88, 263-275.	2.5	11
62	Heat capacities and densities of electrolyte mixtures in aqueous solution — Application to the determination of apparent molar heat capacities and volumes for potassium triiodide and dioxoneptunium(V) perchlorate. Journal of Nuclear Materials, 1993, 201, 165-175.	2.7	8
63	A simple correlation for estimating viscosities of solutions of salts in aqueous, nonâ€aqueous and mixed solvents applicable to high concentration, temperature and pressure. Canadian Journal of Chemical Engineering, 1993, 71, 948-954.	1.7	11
64	Contributions of Compaction And Aquathermal Pressuring to Geopressure and the Influence of Environmental Conditions: DISCUSSION. AAPG Bulletin, 1993, 77, .	1.5	5
65	An empirical equation for quartz solubility in NaCl solution Journal of Mineralogy, Petrology and Economic Geology, 1994, 89, 203-212.	0.1	3
66	lon interaction approach for volumetric calculations for solutions of single electrolytes at 25�C. Journal of Solution Chemistry, 1994, 23, 849-875.	1.2	35
67	Thermodynamic Data for Ligand Interaction with Protons and Metal Ions in Aqueous Solutions at High Temperatures. Chemical Reviews, 1994, 94, 467-517.	47.7	60
68	Real-Time Pore Pressure and Fracture Gradient Evaluation in All Sedimentary Lithologies. SPE Formation Evaluation, 1995, 10, 215-222.	0.5	55
69	Volumetric properties of aqueous electrolytes at high temperature: Mixtures of LiOH and KOH up to 523 K. Journal of Solution Chemistry, 1995, 24, 121-132.	1.2	4
70	Apparent molar volumes of aqueous calcium chloride to 250�C, 400 bars, and from molalities of 0.242 to 6.150. Journal of Solution Chemistry, 1995, 24, 897-916.	1.2	34
71	Ion interaction approach to calculations of volumetric properties of aqueous multiple-solute electrolyte solutions. Journal of Solution Chemistry, 1995, 24, 1025-1038.	1.2	27
72	Solvation thermodynamic functions in the mean spherical approximation: Behavior near the solvent critical region. Journal of Chemical Physics, 1995, 102, 4217-4226.	3.0	3
73	Hydration of Ni2+ and Clâ^' in a concentrated nickel chloride solution at 100 °C and 300 °C. Journal o Chemical Physics, 1996, 105, 5155-5159.	f _{3.0}	64

#	Article	IF	CITATIONS
74	Empirical expressions of quartz solubility in H2O, H2O+CO2, and H2O+NaCl fluids Geochemical Journal, 1996, 30, 339-354.	1.0	18
75	Apparent molar volumes of aqueous solutions of some organic solutes at the pressure 28 MPa and temperatures to 598 K. Journal of Chemical Thermodynamics, 1996, 28, 723-741.	2.0	67
76	Apparent molar heat capacities and volumes for HClO4(aq) to 373 K. Thermochimica Acta, 1996, 286, 225-231.	2.7	7
77	Apparent molar heat capacities of aqueous solutions of acetic, propanoic and succinic acids, sodium acetate and sodium propanoate from 300 to 525 K and a pressure of 28 MPa. Journal of Solution Chemistry, 1996, 25, 849-864.	1.2	20
78	Study of NaCl solutions in a mixed solvent H2O?CH3OH: Experimental densities and comparison with calculated values obtained with a modified Pitzer's model. Journal of Solution Chemistry, 1996, 25, 895-903.	1.2	5
79	Apparent molar volumes and heat capacities of aqueous solutions of sodium benzenesulfonate at 25�C. Journal of Solution Chemistry, 1996, 25, 1083-1088.	1.2	3
80	Mass Diffusion Coefficients and Thermal Diffusivity in Concentrated Hydrothermal NaNO3Solutions. The Journal of Physical Chemistry, 1996, 100, 5982-5992.	2.9	36
81	Hydrogen-bond structure in an aqueous solution of sodium chloride at sub- and supercritical conditions. Journal of Chemical Physics, 1997, 107, 8577-8585.	3.0	18
82	Calculation of densities of aqueous electrolyte solutions at subzero temperatures. Journal of Solution Chemistry, 1997, 26, 433-460.	1.2	17
83	The chemistry of scale prediction. Journal of Petroleum Science and Engineering, 1997, 17, 113-121.	4.2	56
84	Volumes and heat capacities of aqueous solutions of hydrochloric acid at temperatures from 298.15 K to 623 K and pressures to 28 MPa. Journal of Chemical Thermodynamics, 1997, 29, 125-148.	2.0	69
85	Apparent molar heat capacities of aqueous solutions of propylamine, 1,4-butanediamine, 1,6-hexanediamine, propylamine hydrochloride, propionamide, pyridine, and sodium benzenesulfonate at temperatures from 300 K to 525 K and a pressure of 28 MPa. Journal of Chemical Thermodynamics, 1997, 29, 517-531.	2.0	8
86	A Model for Deep Geothermal Brines, II: Thermodynamic Properties – Density. Transport in Porous Media, 1998, 33, 129-154.	2.6	39
87	Application of Pitzer's electrolyte solution theory to frictional model of charged nanofiltration membrane. Thermochimica Acta, 1998, 308, 177-182.	2.7	2
88	Thermodynamic properties of Na2SO4(aq) above 200°C. Geochimica Et Cosmochimica Acta, 1998, 62, 915-916.	3.9	9
89	Theory and experiment on the cuprous–cupric electron transfer rate at a copper electrode. Journal of Chemical Physics, 1999, 110, 6538-6552.	3.0	26
90	High-Pressure Liquid-Liquid Immiscibility in Aqueous Solutions of Tetra-n-butylammonium Bromide Studied by a Diamond Anvil Cell Technique. Journal of Solution Chemistry, 1999, 28, 435-446.	1.2	17
91	Volumetric properties of water, and solutions of sodium chloride and potassium chloride at temperatures fromT= 277.15 K toT= 343.15 K at molalities of (0.1, 0.5, and 1.0)mol·kgâ~'1. Journal of Chemical Thermodynamics, 1999, 31, 869-893.	2.0	67

#	Article	IF	CITATIONS
92	Apparent molar isentropic compressions of electrolytes as a function of the solution molality. Physical Chemistry Chemical Physics, 1999, 1, 1501-1506.	2.8	3
93	Low-temperature thermodynamic model for the system Na2CO3â^'MgCO3â^'CaCO3â^'H2O. Geochimica Et Cosmochimica Acta, 1999, 63, 3105-3119.	3.9	234
94	Excess Partial Molar Enthalpy of 1-Propanol in 1-Propanolâ^'NaClâ^'H2O at 25 °C: The Effect of NaCl on Molecular Organization of H2Oâ€. Journal of Physical Chemistry B, 1999, 103, 2981-2983.	2.6	44
95	The Apparent Molal Volume and Compressibility of Seawater Fit to the Pitzer Equations. Journal of Solution Chemistry, 2000, 29, 719-742.	1.2	26
96	Temperature Dependence of the Densities and Speeds of Sound of the Binary Solutions of LiClO4 with Diethyl Ether, Tetrahydrofuran, Acetone, and Ethyl Acetate. Journal of Chemical & Engineering Data, 2000, 45, 630-635.	1.9	7
97	A Modified Rackett Equation Applied to Water and Aqueous NaCl and KCl Solutions. Journal of Chemical & Engineering Data, 2000, 45, 523-529.	1.9	6
98	Experimental Determination of the Volumetric Properties of NaCl Solutions to 253 K. Journal of Physical Chemistry B, 2001, 105, 9909-9912.	2.6	14
100	Solubility of cesium chloride in water under high pressures. Fluid Phase Equilibria, 2001, 189, 1-11.	2.5	9
101	Thermodynamic Properties of NaCl Solutions at Subzero Temperatures. Journal of Solution Chemistry, 2001, 30, 1065-1080.	1.2	21
102	Aqueous Guanidinium Salts: Part I. Densities, Ultrasonic Velocities, and Apparent Molar Properties. Journal of Solution Chemistry, 2001, 30, 281-290.	1.2	19
103	Molecular Association in Solution:  A Kirkwoodâ^'Buff Analysis of Sodium Chloride, Ammonium Sulfate, Guanidinium Chloride, Urea, and 2,2,2-Trifluoroethanol in Water. Journal of Physical Chemistry B, 2002, 106, 1491-1500.	2.6	120
104	Heat Capacities of Concentrated Aqueous Solutions of Sodium Sulfate, Sodium Carbonate, and Sodium Hydroxide at 25 °C. Journal of Chemical & Engineering Data, 2002, 47, 590-598.	1.9	43
105	Ionic interactions from volumetric investigations of I-alanine in NaBr, KCl, KBr and MgCl2 up to high concentrations. Fluid Phase Equilibria, 2002, 201, 321-333.	2.5	37
106	Thermodynamic solution model for trona brines. AICHE Journal, 2003, 49, 1883-1894.	3.6	22
107	Extension of self-consistent local-composition models for correlation of densities of aqueous electrolyte solutions. Fluid Phase Equilibria, 2003, 209, 1-11.	2.5	6
108	A Kirkwood–Buff derived force field for sodium chloride in water. Journal of Chemical Physics, 2003, 119, 11342-11349.	3.0	199
109	Isopiestic and Volumetric Study of CuCl2+ Ethanol at 25 °C. Journal of Chemical & Engineering Data, 2003, 48, 308-313.	1.9	23
110	Speed of Sound in Concentrated Aqueous KCl Solutions from 278.15 to 338.15 K. Journal of Chemical & Engineering Data, 2003, 48, 388-391.	1.9	5

ARTICLE IF CITATIONS Density. Experimental Thermodynamics, 2003, 6, 125-235. 0.1 8 111 Speciation study on complex formation of uranium(VI) with phosphate and fluoride at high temperatures and pressures by time-resolved laser-induced fluorescence spectroscopy. Radiochimica 1.2 Acta, 2004, 92, 889-896. Extension of some local composition models for correlation of the apparent molal volume of 113 2.59 aqueous electrolytes at 298.15K. Fluid Phase Equilibria, 2004, 226, 91-96. Cosolvent Interactions with Biomolecules:  Relating Computer Simulation Data to Experimental 114 Thermodynamic Data. Journal of Physical Chemistry B, 2004, 108, 18716-18724. Correlation for the Density of Multicomponent Aqueous Electrolytes. Industrial & Amp; Engineering 115 3.7 7 Chemistry Research, 2004, 43, 6247-6252. A new look at homogeneous freezing of water. Geophysical Research Letters, 2004, 31, . 4.0 Standard Partial Molar Volumes of Electrolytes and Ions in Nonaqueous Solvents. Chemical Reviews, 117 47.7 232 2004, 104, 3405-3452. Speciation study on uranium(VI) hydrolysis at high temperatures and pressures. Journal of Alloys and 118 5.5 Compounds, 2004, 374, 277-282. Estimating Density, Formation Volume Factor, Compressibility, Methane Solubility, and Viscosity for Oilfield Brines at Temperatures From 0 to 275 Ű C, Pressures to 200 MPa, and Salinities to 5.7 mole/kg. 119 2.3 52 Journal of Canadian Petroleum Technology, 2004, 43, . Vapor pressures and apparent molal volumes of the solutions of ZnCl2 in ethanol at 298.15K. Fluid 2.5 Phase Equilibria, 2005, 230, 64-71. Apparent molal volumes of the solutions of CaCl2 and Ca(NO3)2 in ethanol at 298.15: experimental data 122 2.5 9 and correlation by local composition models. Fluid Phase Equilibria, 2005, 231, 61-66. Fluctuation Functions in Aqueous NaCl and Urea. Journal of Physical Chemistry B, 2005, 109, 2.6 16886-16890. The Escanaba Trough, Gorda Ridge hydrothermal system: Temporal stability and subseafloor 124 3.9 55 complexity. Geochimica Et Cosmochimica Acta, 2005, 69, 4971-4984. Comprehensive Model of Synthetic Bayer Liquors. Part 1. Overview. Industrial & amp; Engineering Chemistry Research, 2005, 44, 5805-5814. 3.7 Comprehensive Model of Synthetic Bayer Liquors. Part 2. Densities of Alkaline Aluminate Solutions to 126 1.9 17 90 °C. Journal of Chemical & amp; Engineering Data, 2005, 50, 1270-1276. On the dynamics of NaCl-H2O fluid convection in the Earth's crust. Journal of Geophysical Research, 127 2005, 1<u>10, .</u> Temperature Dependence of the Volumetric Parameters of Drug Binding to $Poly[d(A-T)]\hat{A} \cdot Poly[d(A-T)]$ 128 0.5 15 and Poly(dA)·Poly(dT). Biophysical Journal, 2006, 90, 1729-1738. A thermodynamic model for calculating methane solubility, density and gas phase composition of methane-bearing aqueous fluids from 273 to 523K and from 1 to 2000bar. Geochimica Et Cosmochimica 129 Acta, 2006, 70, 3369-3386.

		CITATION REP	PORT	
#	Article		IF	Citations
130	Discrepancies in brine density databases at geothermal conditions. Geothermics, 2006, 35, 600	-606.	3.4	13
131	Volumetric properties of binary mixtures of ionic liquid 1-butyl-3-methylimidazolium octylsulfate water or propanol in the temperature range of 278.15K to 328.15K. Journal of Chemical Thermodynamics, 2006, 38, 1124-1129.	with	2.0	27
132	Multiphase Thermohaline Convection in the Earth's Crust: I. A New Finite Element – Finite Solution Technique Combined With a New Equation of State for NaCl–H2O. Transport in Por Media, 2006, 63, 399-434.		2.6	73
133	lsopiestic Measurement of the Osmotic Coefficients of Aqueous {xH 2 SO 4 + (1â^ x)Fe 2 (SO Solutions at 298.15 and 323.15 K. Journal of Solution Chemistry, 2006, 35, 1699-1730.)3}	1.2	7
134	A thermodynamic model for calculating nitrogen solubility, gas phase composition and density N2–H2O–NaCl system. Fluid Phase Equilibria, 2006, 248, 103-114.	of the	2.5	52
135	The effect of the physical matrix on accurate measurements using fixed volume analytical techr Journal of Separation Science, 2006, 29, 2072-2077.	iques.	2.5	5
136	Interferometrische Messung der Dichte wÄ s sriger LĶsungen unter Ultra-Hochdruck (Interferor Technisches Messen, 2007, 74, 45-50.	netric) Tj ETQq0 0 (0 rgBT /Ov 0.7	verlock 10 Ti 7
137	PVTx properties of the CO2–H2O and CO2–H2O–NaCl systems below 647ÂK: Assessme data and thermodynamic models. Chemical Geology, 2007, 238, 249-267.	nt of experimental	3.3	104
138	The system H2O–NaCl. Part I: Correlation formulae for phase relations in temperature–pressure–composition space from 0 to 1000°C, 0 to 5000bar, and 0 to 1 XI Geochimica Et Cosmochimica Acta, 2007, 71, 4880-4901.	VaCl.	3.9	586
139	In situ measurement of dissolved chloride in high temperature hydrothermal fluids. Geochimica Cosmochimica Acta, 2007, 71, 2510-2523.	Et	3.9	28
140	The system H2O–NaCl. Part II: Correlations for molar volume, enthalpy, and isobaric heat cap from 0 to 1000°C, 1 to 5000bar, and 0 to 1 XNaCl. Geochimica Et Cosmochimica Acta, 2007,	acity 71, 4902-4919.	3.9	305
141	The Marine Inorganic Carbon Cycle. Chemical Reviews, 2007, 107, 308-341.		47.7	410
142	Apparent and Standard Partial Molar Volumes of NaCl, NaOH, and HCl in Water and Heavy Wat 523 K and 573 K atp= 14 MPa. Journal of Physical Chemistry B, 2007, 111, 2015-2024.	er atT=	2.6	27
143	Solubility of sodium chloride in water under high pressure. Fluid Phase Equilibria, 2007, 254, 15	8-162.	2.5	38
144	Volumetric Properties of the Ionic Liquid, 1-Butyl-3-methylimidazolium Tetrafluoroborate, in Org Solvents at T =Â 298.15K. International Journal of Thermophysics, 2008, 29, 534-545.	anic	2.1	36
145	pVTx Properties of Hydrothermal Systems. , 0, , 135-193.			4
146	The P,V,T,x properties of binary aqueous chloride solutions up to T=573K and 100MPa. Journal of Chemical Thermodynamics, 2008, 40, 1046-1063.	of	2.0	71
147	Densities of the CO ₂ –H ₂ O and CO ₂ –H ₂ Systems Up to 647 K and 100 MPa. Energy & Fuels, 2008, 22, 1666-1674.	o>O–NaCl	5.1	111

#	Article	IF	CITATIONS
148	Salting out of methane by sodium chloride: A scaled particle theory study. Journal of Chemical Physics, 2008, 129, 084506.	3.0	65
149	Physics of glycine interactions with halide salts of alkali metals (group I) determined with density, viscosity, and activation energy in aqueous solutions at 310.15 K. Physics and Chemistry of Liquids, 2008, 46, 119-139.	1.2	9
150	Concentration fluctuations in fluid mixtures. II Journal of Chemical Physics, 2009, 130, 234508.	3.0	3
151	Role of hydrophobic effect in the saltâ€induced dimerization of bovine βâ€lactoglobulin at pH 3. Biopolymers, 2009, 91, 1182-1188.	2.4	4
152	Thermodynamic and Kinetic Properties of Natural Brines. Aquatic Geochemistry, 2009, 15, 7-41.	1.3	16
153	Effects of density and mutual solubility of a –brine system on storage in geological formations: "Warm―vs. "cold―formations. Advances in Water Resources, 2009, 32, 1685-1702.	3.8	35
154	Interaction of biological buffers with electrolytes: Densities of aqueous solutions of two substituted aminosulfonic acids and ionic salts from T=(298.15 to 328.15)K. Journal of Chemical Thermodynamics, 2009, 41, 705-715.	2.0	7
155	Pressure perturbations from geologic carbon sequestration: Area-of-review boundaries and borehole leakage driving forces. Energy Procedia, 2009, 1, 47-54.	1.8	43
156	Excess Volume of Electrolytes in the Mean Spherical Approximation. Journal of Chemical & Engineering Data, 2009, 54, 548-554.	1.9	0
157	Correction to "Parameters of subsurface brines and hydrothermal processes 12–15 months after the 1999 magmatic event at the Main Endeavor Field as inferred from in situ time series measurements of chloride and temperature†Journal of Geophysical Research, 2009, 114, .	3.3	6
158	Parameters of subsurface brines and hydrothermal processes 12–15 months after the 1999 magmatic event at the Main Endeavor Field as inferred from in situ time series measurements of chloride and temperature. Journal of Geophysical Research, 2009, 114, .	3.3	9
159	Thermodynamic properties of aqueous-alcoholic solutions of sodium chloride. H2O-2-C3H7OH-NaCl. Russian Journal of Physical Chemistry A, 2010, 84, 1877-1885.	0.6	3
160	Thermodynamic model of aqueous CO2–H2O–NaCl solutions from â^'22 to 100°C and from 0.1 to 100MPa. Fluid Phase Equilibria, 2010, 295, 104-124.	2.5	79
161	Measuring the transient diffusion of vapor mixtures through dense membranes. Journal of Membrane Science, 2010, 350, 217-225.	8.2	13
162	Blended silicone–ionic liquid membranes: Transport properties of butan-1-ol vapor. European Polymer Journal, 2010, 46, 123-128.	5.4	21
163	Computer Aided Diagnosis of Parotid Gland Lesions Using Ultrasonic Multi-Feature Tissue Characterization. Ultrasound in Medicine and Biology, 2010, 36, 1525-1534.	1.5	11
165	Density and viscosity of brine: An overview from a process engineers perspective. Chemie Der Erde, 2010, 70, 23-32.	2.0	35
168	Solutes H ₂ SO ₄ , HNO ₃ , HCl, Na ₂ SO ₄ , NaNO ₃ , NaCl, (NH ₄) ₂ SO ₄ , NH ₄ NO ₃ , and NH ₄ Cl from 0 to 50 ŰC, Including Extrapolations to Very Low Temperature and to the Pure Liquid State. and NaHSO ₄ . NaOH. and	2.5	87

#	Article	IF	CITATIONS
169	Molecular Simulation of Aqueous Electrolyte Solubility. 2. Osmotic Ensemble Monte Carlo Methodology for Free Energy and Solubility Calculations and Application to NaCl. Journal of Physical Chemistry B, 2011, 115, 7849-7861.	2.6	102
170	A method for calculating the liquid density for the CO2–H2O–NaCl system under CO2 storage condition. Energy Procedia, 2011, 4, 3817-3824.	1.8	28
171	Local fluctuations in solution mixtures. Journal of Chemical Physics, 2011, 135, 044506.	3.0	23
172	Competition Reactions of H ₂ O ^{•+} Radical in Concentrated Cl [–] Aqueous Solutions: Picosecond Pulse Radiolysis Study. Journal of Physical Chemistry A, 2012, 116, 11509-11518.	2.5	33
173	IUPAC-NIST Solubility Data Series. 95. Alkaline Earth Carbonates in Aqueous Systems. Part 1. Introduction, Be and Mg. Journal of Physical and Chemical Reference Data, 2012, 41, 013105-013105-67.	4.2	37
174	Volumes of aqueous hydrogen and hydroxide ions at 0 to 200 °C. Journal of Chemical Physics, 2012, 137, 154501.	3.0	111
175	Concentration dependence of NaCl in a wide range of temperatures and pressures. Journal of Structural Chemistry, 2013, 54, 341-348.	1.0	0
176	Volumetrics of CO ₂ Storage in Deep Saline Formations. Environmental Science & Technology, 2013, 47, 79-86.	10.0	19
177	Modeling Multiphase, Multicomponent Processes at Oceanic Spreading Centers. Geophysical Monograph Series, 0, , 15-44.	0.1	9
178	Structure and Thermodynamics of Subduction Zone Fluids from Spectroscopic Studies. Reviews in Mineralogy and Geochemistry, 2013, 76, 265-309.	4.8	27
179	Computationally efficient Monte Carlo simulations for polarisable models: multi-particle move method for water and aqueous electrolytes. Molecular Simulation, 2013, 39, 1125-1134.	2.0	27
180	1-Propanol probing methodology: two-dimensional characterization of the effect of solute on H2O. Physical Chemistry Chemical Physics, 2013, 15, 14548.	2.8	34
181	A Pitzer-based characterization of aqueous magnesium chloride, calcium chloride and potassium iodide solution densities to high temperature and pressure. Fluid Phase Equilibria, 2013, 338, 54-62.	2.5	17
182	On the signature of the hydrophobic effect at a single molecule level. Physical Chemistry Chemical Physics, 2013, 15, 7389.	2.8	2
183	Molecular Force Field Development for Aqueous Electrolytes: 1. Incorporating Appropriate Experimental Data and the Inadequacy of Simple Electrolyte Force Fields Based on Lennard-Jones and Point Charge Interactions with Lorentz–Berthelot Rules. Journal of Chemical Theory and Computation, 2013, 9, 5076-5085.	5.3	48
184	PVTX Properties of H2O-CO2-"salt" at PTX Conditions Applicable to Carbon Sequestration in Saline Formations. Reviews in Mineralogy and Geochemistry, 2013, 77, 123-152.	4.8	17
185	Modeling Reveals Hidden Conditions That Can Impair Wellbore Stability and Integrity. , 2013, , .		4
186	Wellbore Stability and Integrity Contributors Revealed by Thermal Modeling and Fluid Analysis. , 2013,		2

#	Article	IF	CITATIONS
188	Evolution of Titan× ³ s major atmospheric gases and cooling since accretion. Planetary and Space Science, 2014, 93-94, 41-53.	1.7	5
189	Equations for calculating hydrogeochemical reactions of minerals and gases such as CO2 at high pressures and temperatures. Geochimica Et Cosmochimica Acta, 2014, 125, 49-67.	3.9	134
190	A new model for the density of saturated solutions of CO2–H2O–NaCl in saline aquifers. International Journal of Greenhouse Gas Control, 2014, 31, 192-204.	4.6	20
191	Infinitely Dilute Partial Molar Properties of Proteins from Computer Simulation. Journal of Physical Chemistry B, 2014, 118, 12844-12854.	2.6	11
192	Sorption of vapours and liquids in PDMS: novel data and analysis with the GAB model of multilayer adsorption. European Polymer Journal, 2014, 60, 49-57.	5.4	26
193	Volumetric Properties of Mixed Electrolyte Aqueous Solutions at Elevated Temperatures and Pressures. The System KCl–NaCl–H ₂ O to 523.15 K, 40 MPa, and Ionic Strength from (0.1 to) Tj	ETiQ9q11(0. 7 84314 rg
194	A new aqueous activity model for geothermal brines in the system Na-K-Ca-Mg-H-Cl-SO4-H2O from 25 to 300°C. Chemical Geology, 2014, 381, 78-93.	3.3	24
195	Physicothermal Properties of Aqueous Sodium Chloride Solutions. Journal of Food Process Engineering, 2015, 38, 234-242.	2.9	23
196	Modeling Complex Reservoir and Mud Filtrate Brines for Petrophysical Interpretation Parameters: Why Equivalent NaCl Salinity can Produce Misleading Results. , 2015, , .		1
197	The Development and Application of Aqueous Thermodynamic Models: The Specific Ion-Interaction Approach. SSSA Special Publication Series, 2015, , 31-52.	0.2	2
198	Mean ionic activity coefficients in aqueous NaCl solutions from molecular dynamics simulations. Journal of Chemical Physics, 2015, 142, 044507.	3.0	113
199	Geophysical Properties of the Near Surface Earth: Seismic Properties. , 2015, , 43-87.		18
200	Evaluation of Different Artificial Intelligent Models to Predict Reservoir Formation Water Density. Petroleum Science and Technology, 2015, 33, 1749-1756.	1.5	8
201	Influence of He and Ar Flow Rates and NaCl Concentration on the Size Distribution of Bubbles Generated by Power Ultrasound. Journal of Physical Chemistry B, 2015, 119, 12682-12688.	2.6	26
202	Hydraulic conductivity of fractured upper crust: insights from hydraulic tests in boreholes and fluidâ€rock interaction in crystalline basement rocks. Geofluids, 2015, 15, 161-178.	0.7	53
203	A predictive model for the PVTx properties of CO2–H2O–NaCl fluid mixture up to high temperature and high pressure. Applied Geochemistry, 2015, 54, 54-64.	3.0	25
204	Thermodynamic Modeling of Apparent Molal Volumes of Metal Nitrate Salts with Pitzer Model. Journal of Chemical & Engineering Data, 2015, 60, 856-859.	1.9	3
205	Estimating the Density and Compressibility of Natural Hypersaline Brines Using the Pitzer Ionic Interaction Model. Aquatic Geochemistry, 2015, 21, 11-29.	1.3	4

#	Article	IF	CITATIONS
206	Competitive Adsorption of Monoclonal Antibodies and Nonionic Surfactants at Solid Hydrophobic Surfaces. Journal of Pharmaceutical Sciences, 2015, 104, 593-601.	3.3	37
207	Electric Interfacial Layer of Modified Cellulose Nanocrystals in Aqueous Electrolyte Solution: Predictions by the Molecular Theory of Solvation. Langmuir, 2015, 31, 7106-7116.	3.5	15
208	pH- and Ionic-Strength-Induced Contraction of Polybasic Micelles in Buffered Aqueous Solutions. Macromolecules, 2015, 48, 2677-2685.	4.8	47
209	Chemical Potentials, Activity Coefficients, and Solubility in Aqueous NaCl Solutions: Prediction by Polarizable Force Fields. Journal of Chemical Theory and Computation, 2015, 11, 1756-1764.	5.3	60
210	Density measurement and equal density temperature of CO2+brine from Dagang — formation from 313 to 363 K. Korean Journal of Chemical Engineering, 2015, 32, 141-148.	2.7	2
211	Aqueous electrolyte solution modelling: Some limitations of the Pitzer equations. Applied Geochemistry, 2015, 55, 170-183.	3.0	52
212	Decomposition Kinetics of Glucose and Fructose in Subcritical Water Containing Sodium Chloride. Journal of Applied Glycoscience (1999), 2016, 63, 99-104.	0.7	7
213	Consensus on the solubility of NaCl in water from computer simulations using the chemical potential route. Journal of Chemical Physics, 2016, 144, 124504.	3.0	108
214	Thermophysical properties of seawater: A review and new correlations that include pressure dependence. Desalination, 2016, 390, 1-24.	8.2	370
215	Measurements of the density, speed of sound, viscosity and derived thermodynamic properties of geothermal fluids from south Russia Geothermal Field. Part II. Applied Geochemistry, 2016, 69, 28-41.	3.0	7
216	Injection of CO2-saturated brine in geological reservoir: A way to enhanced storage safety. International Journal of Greenhouse Gas Control, 2016, 54, 129-144.	4.6	11
217	Osmotic pressure of aqueous electrolyte solutions via molecular simulations of chemical potentials: Application to NaCl. Fluid Phase Equilibria, 2016, 407, 76-83.	2.5	24
218	Determination of boron in produced water using the carminic acid assay. Talanta, 2016, 150, 240-252.	5.5	11
219	Influences of fluid properties on the hydrothermal fluid flow and alteration halos at the Dajishan tungsten deposit, China. Journal of Geochemical Exploration, 2016, 163, 53-69.	3.2	20
220	Thermodynamic description of H2S–H2O–NaCl solutions at temperatures to 573 K and pressures to 40 MPa. Chemical Geology, 2016, 424, 1-11.	3.3	10
221	Measurements of the Density, Speed of Sound, Viscosity, and Derived Thermodynamic Properties of Geothermal Fluids. Journal of Chemical & Engineering Data, 2016, 61, 234-246.	1.9	12
222	Coarse-Grained Model for Water Involving a Virtual Site. Journal of Physical Chemistry B, 2016, 120, 733-739.	2.6	19
223	Theoretical framework for energy analysis of hypersaline pressure retarded osmosis. Chemical Engineering Science, 2016, 139, 211-220.	3.8	13

#	Article	IF	CITATIONS
224	Calcite and Barite Solubility Measurements in Mixed Electrolyte Solutions and Development of a Comprehensive Model for Water-Mineral-Gas Equilibrium of the Na-K-Mg-Ca-Ba-Sr-Cl-SO ₄ -CO ₃ -HCO ₃ -CO ₂ (aq)-H _{2System up to 250 ŰC and 1500 bar. Industrial & amp; Engineering Chemistry Research, 2017, 56, 6548-6561.}	»°₫	19
225	The PVTx properties of aqueous electrolyte solutions containing Li + , Na + , K + , Mg 2+ , Ca 2+ , Cl â^' and SO 4 2â^' under conditions of CO 2 capture and sequestration. Applied Geochemistry, 2017, 86, 105-120.	3.0	4
226	A potential model for sodium chloride solutions based on the TIP4P/2005 water model. Journal of Chemical Physics, 2017, 147, 104501.	3.0	82
227	Experimental Data and Modeling of Solution Density and Heat Capacity in the Na–K–Ca–Mg–Cl–H ₂ O System up to 353.15 K and 5 mol·kg ^{–1} Ionic Stre Journal of Chemical & Engineering Data, 2017, 62, 3561-3576.	eng⊈h.	8
228	Ionic Solution: What Goes Right and Wrong with Continuum Solvation Modeling. Journal of Physical Chemistry B, 2017, 121, 11169-11179.	2.6	9
229	A Pitzer Parametrization To Predict Solution Properties and Salt Solubility in the H–Na–K–Ca–Mg–NO ₃ –H ₂ O System at 298.15 K. Journal of Chemical &am Engineering Data, 2018, 63, 787-800.	ו ם: ,9	26
230	Electrochemical aspects of copper atmospheric corrosion in the presence of sodium chloride. Electrochimica Acta, 2018, 276, 194-206.	5.2	44
231	3D multiphysics model for the simulation of electrochemical machining of stainless steel (SS316). International Journal of Advanced Manufacturing Technology, 2018, 95, 2959-2972.	3.0	24
232	Thermodynamic, Exergy, and Thermoeconomic analysis of Multiple Effect Distillation Processes. , 2018, , 445-489.		5
233	Fluid inclusions in the system H2O-NaCl-CO2: An algorithm to determine composition, density and isochore. Chemical Geology, 2018, 498, 31-44.	3.3	49
234	Density variations of TMAO solutions in the kilobar range: Experiments, PC-SAFT predictions, and molecular dynamics simulations. Biophysical Chemistry, 2019, 253, 106222.	2.8	8
235	Thermodynamic properties of aqueous osmolyte solutions at high-pressure conditions. Biophysical Chemistry, 2019, 253, 106211.	2.8	7
236	A force field of Li+, Na+, K+, Mg2+, Ca2+, Clâ^', and SO42â^' in aqueous solution based on the TIP4P/2005 water model and scaled charges for the ions. Journal of Chemical Physics, 2019, 151, 134504.	3.0	166
237	Reverse electrodialysis heat engine with multi-effect distillation: Exergy analysis and perspectives. Energy Conversion and Management, 2019, 194, 140-159.	9.2	48
238	Thermodynamic Properties of Aqueous Sodium Nitrate Solutions Under Superambient Conditions. Journal of Solution Chemistry, 2019, 48, 167-179.	1.2	4
239	Effect of surfactants on CO2 solubility and reaction in CO2-brine-clay mineral systems during CO2-enhanced fossil fuel recovery. Chemical Engineering Journal, 2020, 382, 123014.	12.7	15
240	On the influence of salt concentration on the transport properties of reverse osmosis membranes in high recovery desalination. Journal of Membrane Science, 2020, 594, 117339.	8.2	14
241	A method to predict the homogenization temperatures of easily decrepitated fluid inclusions. Ore Geology Reviews, 2020, 117, 103311.	2.7	1

#	ARTICLE	IF	CITATIONS
242	Methane dissolution inside bulk or porous-medium-confined water at near-hydrate equilibrium conditions. Molecular Physics, 2020, 118, e1699186.	1.7	1
243	CFD-DEM Modeling of soluble NaCl particles conveyed in brine. Powder Technology, 2020, 360, 1278-1294.	4.2	6
244	Consistent description of ion-specificity in bulk and at interfaces by solvent implicit simulations and mean-field theory. Journal of Chemical Physics, 2020, 153, 034103.	3.0	11
245	Contribution of Secondary Structure Changes to the Surface Activity of Proteins. Journal of Biotechnology, 2020, 323, 208-220.	3.8	5
246	Smallâ€Scale Capillary Heterogeneity Linked to Rapid Plume Migration During CO ₂ Storage. Geophysical Research Letters, 2020, 47, e2020GL088616.	4.0	45
247	Molecular Force Field Development for Aqueous Electrolytes: 2. Polarizable Models Incorporating Crystalline Chemical Potential and Their Accurate Simulations of Halite, Hydrohalite, Aqueous Solutions of NaCl, and Solubility. Journal of Chemical Theory and Computation, 2020, 16, 3677-3688.	5.3	13
248	Correlation of the Volumetric Properties of Uni-Univalent Electrolytes in Methanol–Water Mixed Solvent Media: A Pitzer Ion-Interaction Approach. Journal of Solution Chemistry, 2020, 49, 825-835.	1.2	1
249	Thermodynamic Properties of Geothermal Fluids from South Russia: Kayakent and Kizlyar Hot Sources. Innovation and Discovery in Russian Science and Engineering, 2021, , 275-301.	0.2	2
250	The solubility behavior of NaCl in water at high pressure studied by neutron diffraction and Raman scattering. High Pressure Research, 2021, 41, 39-51.	1.2	1
251	Composition dependent density of ternary aqueous solutions of ionic surfactants and salts. Journal of Atmospheric Chemistry, 2021, 78, 99-123.	3.2	4
252	Experimental Measurement of Physical, Transport, and Optical Properties of Binary Mixtures of N-Hexyl Pyridinium Nitrate [HPy][NO3] Ionic Liquid with Water, Ethanol, and Acetonitrile at 298.15ÂK and 101ÂkPa. Journal of Solution Chemistry, 2021, 50, 576-590.	1.2	8
253	Thermophysical Properties of Protic Ionic Liquids Monoethanolamine, Diethanolamine, and Triethanolamine Lactate in Water. Journal of Chemical & Engineering Data, 2021, 66, 1890-1899.	1.9	10
254	Simultaneous Crystallization of Cyclopentane Hydrate and Sodium Chloride for Desalination and Salt Manufacture. ACS Sustainable Chemistry and Engineering, 2021, 9, 9078-9084.	6.7	17
255	Thermodynamic Properties of Geothermal Fluids from South Russia: Izberbash and Ternair Hot Sources. Innovation and Discovery in Russian Science and Engineering, 2021, , 303-333.	0.2	2
256	The effects of poreâ€fluid salinity on ultrasonic wave propagation in sandstones. Geophysics, 1998, 63, 928-934.	2.6	13
257	Ultrasonic velocity and attenuation in water and brine $\hat{a} {\in} \mathbf{s}$ aturated sandstones. , 1995, , .		2
262	Coupling of Rigorous Multiphase Flash with Advanced Linearization Schemes for Accurate Compositional Simulation. , 2021, , .		0
263	On the Estimation of the Density of Brine with an Extensive Range of Different Salts Compositions and Concentrations. Journal of Thermodynamics & Catalysis, 2016, 7, .	0.2	0

			_
#	ARTICLE	IF	CITATIONS
265	Carbon Dioxide Sequestration and Enhanced Recovery Techniques. , 2019, , 681-750.		0
266	Cloud drop nucleation. , 2022, , 171-207.		0
267	Volumetric properties for the ternary system (CaCl2-SrCl2-H2O) and binary sub-systems at temperatures from 278.15ÂK to 323.15ÂK. Journal of Chemical Thermodynamics, 2022, 166, 106690.	2.0	0
268	Small-to-large length scale transition of TMAO interaction with hydrophobic solutes. Physical Chemistry Chemical Physics, 2022, 24, 2080-2087.	2.8	5
269	Conclusions About Osmotically Inactive Volume and Osmotic Fragility from a Detailed Erythrocyte Model. Journal of Theoretical Biology, 2022, , 110982.	1.7	0
270	The effects of temperature, pressure and concentration on the hydraulic conductivity of deep groundwater-bearing layers. Hydrogeology Journal, 0, , 1.	2.1	1
271	Wetting and evaporation of multicomponent droplets. Physics Reports, 2022, 960, 1-37.	25.6	56
272	Fragmentation of drops falling through a miscible liquid with and without drop–drop interactions. International Journal of Multiphase Flow, 2022, 153, 104134.	3.4	0
273	Development of dual functional methodology for seawater desalination and salt manufacture by carbon dioxide hydrate formation. Desalination, 2022, 539, 115937.	8.2	13
274	Effects of compressibility and wetting on the liquid–vapor transition in a confined fluid. Journal of Chemical Physics, 2022, 157, 054506.	3.0	3
275	Phase equilibrium conditions in cyclopentane hydrate forming systems coexisting with sodium chloride aqueous solution under atmospheric pressure and vacuum condition. Journal of Chemical Thermodynamics, 2022, 175, 106886.	2.0	3
276	Osmotic and Activity Coefficients of Lithium Nitrate in Ethanol Under High Pressures. Journal of Solution Chemistry, 2022, 51, 1589-1602.	1.2	3
277	Zero emissions, low-energy water production system using clathrate hydrate: Engineering design and techno-economic assessment. Journal of Cleaner Production, 2023, 383, 135425.	9.3	7
279	Controls on the Salinity of Sedimentary Basinal Fluids Under Constant Chemogravitational Potential Conditions. Geochemistry, Geophysics, Geosystems, 2023, 24, .	2.5	0
280	Hydrogen and air storage in salt caverns: a thermodynamic model for phase equilibrium calculations. , 2023, 78, 10.		1
281	Investigating the effect of preformed particle gel on conformance control of naturally fractured tight sandstone reservoirs: An experimental study using pressure transfer test method. Fuel, 2023, 347, 128479.	6.4	2
282	High-temperature and high-pressure thermodynamic properties of aqueous zinc sulfate solutions. Fluid Phase Equilibria, 2023, 572, 113850.	2.5	1
283	Tables for Chapter 4. , 2001, , .		О

#	Article	IF	CITATIONS
284	Tables for Chapter 6. , 2001, , .		0
285	Tables for Chapter 5. , 2001, , .		0
286	1 Methods of Calculation of Volumetric Properties. , 2001, , .		0
288	Tables for Chapter 2. , 2001, , .		0
289	5 Properties of Solutions. , 2001, , .		0
290	Tables for Chapter 3. , 2001, , .		0
291	4 Thermal Properties. , 2001, , .		0
292	3 Phase Equilibria. , 2001, , .		0
295	2 Equilibria in Solutions. , 2001, , .		0
296	6 Transport Phenomena. , 2001, , .		0
297	Tables for Chapter 1., 2001, , .		0
299	Minimizing exposure to legacy wells and avoiding conflict between storage projects: Exploring area of review as a screening tool. International Journal of Greenhouse Gas Control, 2023, 129, 103967.	4.6	0
300	1. Methods for Calculation of Thermal Properties. , 1996, , .		0
301	2. Experimental Values of Thermal Properties. , 1996, , .		0
304	On the Elusive Links between Solution Microstructure, Dynamics, and Solvation Thermodynamics: Demystifying the Path through a Bridge over Troubled Conjectures and Misinterpretations. Journal of Physical Chemistry B, 0, , .	2.6	1
305	A critical review of models for density, viscosity, and diffusivity in aqueous sodium chloride solutions. Electrochimica Acta, 2024, 477, 143766.	5.2	0