

The enthalpy of molybdenum disulfide to 1200 K by dro

Journal of Chemical Thermodynamics

3, 693-696

DOI: 10.1016/s0021-9614(71)80091-5

Citation Report

#	ARTICLE	IF	CITATIONS
1	The chemical transport of molybdenum and tungsten and of their dioxides and sulfides. Journal of Solid State Chemistry, 1973, 8, 14-28.	2.9	69
2	The decomposition kinetics of molybdenite in an argon plasma. AIChE Journal, 1975, 21, 1132-1142.	3.6	21
3	Low-temperature heat capacity of anisotropic crystals lamellar molybdenum disulfide. Journal of Chemical Thermodynamics, 1976, 8, 37-44.	2.0	15
4	Enthalpy and heat capacity of molybdenum disulfide. Soviet Powder Metallurgy and Metal Ceramics (English Translation of Poroshkovaya Metallurgiya), 1978, 17, 697-702.	0.1	19
5	Inorganic chalcogenides: high-tech materials, low-tech thermodynamics. Journal of Chemical Thermodynamics, 1987, 19, 675-701.	2.0	35
6	Standard molar enthalpy of formation by fluorine-combustion calorimetry of tungsten diselenide (WSe <sub>2</sub> ). Thermodynamics of the high-temperature vaporization of WSe <sub>2</sub> . Revised value of the standard molar enthalpy of formation of molybdenite (MoS <sub>2</sub> ). Journal of Chemical Thermodynamics, 1988, 20, 681-691.	2.0	29
7	Thermodynamic properties of tungsten ditelluride (WTe <sub>2</sub> ) II. Standard molar enthalpy of formation at the temperature 298.15 K. Journal of Chemical Thermodynamics, 1992, 24, 639-647.	2.0	29
8	Energetics of overall metabolic reactions of thermophilic and hyperthermophilic Archaea and Bacteria. FEMS Microbiology Reviews, 2001, 25, 175-243.	8.6	648
9	Chemical structures of the $\langle \text{mml:math xmlns:mml="http://www.w3.org/1998/Math/MathML" display="inline"} \rangle \langle \text{mml:mrow} \rangle \langle \text{mml:mtext} \rangle \text{Cu} \langle \text{mml:mtext} \rangle \langle \text{mml:mrow} \rangle \langle \text{mml:mo} \rangle \langle \text{mml:mo} \rangle \langle \text{mml:mrow} \rangle \langle \text{mml:mtext} \rangle \text{In} \langle \text{mml:mt} \rangle$ $\langle \text{mml:math xmlns:mml="http://www.w3.org/1998/Math/MathML" display="inline"} \rangle \langle \text{mml:Physical Review B, 2008, 78, 30} \rangle$	3.2	30
10	Thermodynamic assessment of the Mo-S system and its application in thermal decomposition of MoS <sub>2</sub> . Thermochemica Acta, 2018, 660, 44-55.	2.7	10
12	Use of the Bibliography. , 1982, , 337-976.		0
13	Calorimetry of sulfides and sulfide minerals.. Journal of the Mineralogical Society of Japan, 1987, 18, 99-114.	0.2	0