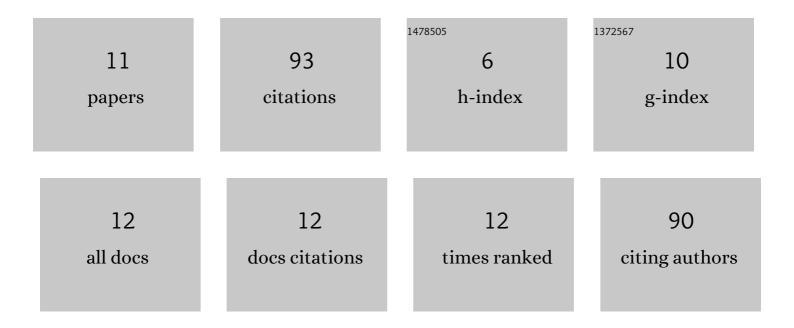
## Ahmad I Alrawashdeh

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Computational study on the deamination reaction of adenine with OH <sup>â^'</sup> / <i>n</i> H <sub>2</sub> O ( <i>n</i> = 0, 1, 2, 3) and 3H <sub>2</sub> O. Canadian Journal of Chemistry, 2013, 91, 518-526.	1.1	27
2	The role of the solvent and the size of the nanotube in the non-covalent dispersion of carbon nanotubes with short organic oligomers $\hat{a} \in $ a DFT study. RSC Advances, 2018, 8, 30520-30529.	3.6	11
3	Dispersion of Single-Walled Carbon Nanotubes with Oligo(p-phenylene ethynylene)s: A DFT Study. Journal of Physical Chemistry C, 2017, 121, 4692-4702.	3.1	9
4	Mechanism for the deamination of ammeline, guanine, and their analogues. Structural Chemistry, 2017, 28, 1467-1477.	2.0	9
5	The reaction of arylmethyl isocyanides and arylmethylamines with xanthate esters: a facile and unexpected synthesis of carbamothioates. Beilstein Journal of Organic Chemistry, 2020, 16, 159-167.	2.2	9
6	DFT investigation of the interaction between single-walled carbon nanotubes and fluorene-based conjugated oligomers. Physical Chemistry Chemical Physics, 2017, 19, 28071-28082.	2.8	7
7	Synthesis, supramolecular complexation and DFTstudies of a bis(pyrene)-appended â€~capped' triazole-linked calix[4]arene as Zn2+ and Cd2+ fluorescent chemosensors. Supramolecular Chemistry, 2020, 32, 325-333.	1.2	6
8	Computational Study on Thermochemical Properties for Perhalogenated Methanols (CX3OH) (X = F, Cl,) Tj ETQq	0 0 0 g rgBT	/Qverlock 1

9	Hydrolytic deamination reactions of amidine and nucleobase derivatives. International Journal of Quantum Chemistry, 2020, 120, e26059.	2.0	4
10	Synthesis, spectroscopic characterization, and theoretical studies on the substitution reaction of chromium(III) picolinate. Journal of Molecular Structure, 2019, 1189, 28-39.	3.6	3
11	Conformational Analysis of the Supramolecular Complexation of Diaryl-Substituted Tetrathiafulvalene Vinylogues with Fullerenes. ACS Omega, 2019, 4, 5630-5639.	3.5	2