

Riccardo Pellegrini

List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/9576138/publications.pdf>

Version: 2024-02-01

34
papers

1,224
citations

361413

20
h-index

395702

33
g-index

34
all docs

34
docs citations

34
times ranked

1485
citing authors

#	ARTICLE	IF	CITATIONS
1	Determination of the Particle Size, Available Surface Area, and Nature of Exposed Sites for Silica-Alumina-Supported Pd Nanoparticles: A Multitechnical Approach. <i>Journal of Physical Chemistry C</i> , 2009, 113, 10485-10492.	3.1	124
2	In situ formation of hydrides and carbides in palladium catalyst: When XANES is better than EXAFS and XRD. <i>Catalysis Today</i> , 2017, 283, 119-126.	4.4	103
3	A comprehensive approach to investigate the structural and surface properties of activated carbons and related Pd-based catalysts. <i>Catalysis Science and Technology</i> , 2016, 6, 4910-4922.	4.1	96
4	Graphitization of Activated Carbons: A Molecular-level Investigation by INS, DRIFT, XRD and Raman Techniques. <i>Physics Procedia</i> , 2016, 85, 20-26.	1.2	68
5	Core-Shell Structure of Palladium Hydride Nanoparticles Revealed by Combined X-ray Absorption Spectroscopy and X-ray Diffraction. <i>Journal of Physical Chemistry C</i> , 2017, 121, 18202-18213.	3.1	67
6	Effect of reduction in liquid phase on the properties and the catalytic activity of Pd/Al ₂ O ₃ catalysts. <i>Journal of Catalysis</i> , 2012, 287, 44-54.	6.2	62
7	Effect of Pre-Reduction on the Properties and the Catalytic Activity of Pd/Carbon Catalysts: A Comparison with Pd/Al ₂ O ₃ . <i>ACS Catalysis</i> , 2014, 4, 187-194.	11.2	62
8	Preparation of Supported Pd Catalysts: From the Pd Precursor Solution to the Deposited Pd ²⁺ Phase. <i>Langmuir</i> , 2010, 26, 11204-11211.	3.5	61
9	Palladium Carbide and Hydride Formation in the Bulk and at the Surface of Palladium Nanoparticles. <i>Journal of Physical Chemistry C</i> , 2018, 122, 12029-12037.	3.1	61
10	0.5wt.% Pd/C catalyst for purification of terephthalic acid: Irreversible deactivation in industrial plants. <i>Journal of Catalysis</i> , 2011, 280, 150-160.	6.2	57
11	Role of the Support in Determining the Vibrational Properties of Carbonyls Formed on Pd Supported on SiO ₂ -Al ₂ O ₃ , Al ₂ O ₃ , and MgO. <i>Journal of Physical Chemistry C</i> , 2007, 111, 7021-7028.	3.1	54
12	Time-resolved operando studies of carbon supported Pd nanoparticles under hydrogenation reactions by X-ray diffraction and absorption. <i>Faraday Discussions</i> , 2018, 208, 187-205.	3.2	47
13	Effect of Different Face Centered Cubic Nanoparticle Distributions on Particle Size and Surface Area Determination: A Theoretical Study. <i>Journal of Physical Chemistry C</i> , 2014, 118, 4085-4094.	3.1	45
14	Influence of K-doping on a Pd/SiO ₂ -Al ₂ O ₃ catalyst. <i>Journal of Catalysis</i> , 2009, 267, 40-49.	6.2	44
15	Pd-Supported Catalysts: Evolution of Support Porous Texture along Pd Deposition and Alkali-Metal Doping. <i>Langmuir</i> , 2009, 25, 6476-6485.	3.5	34
16	Dynamics of Reactive Species and Reactant-Induced Reconstruction of Pt Clusters in Pt/Al ₂ O ₃ Catalysts. <i>ACS Catalysis</i> , 2019, 9, 7124-7136.	11.2	31
17	The role of palladium carbides in the catalytic hydrogenation of ethylene over supported palladium nanoparticles. <i>Catalysis Today</i> , 2019, 336, 40-44.	4.4	29
18	Direct IR observation of vibrational properties of carbonyl species formed on Pd nano-particles supported on amorphous carbon: comparison with Pd/SiO ₂ -Al ₂ O ₃ . <i>Physical Chemistry Chemical Physics</i> , 2006, 8, 3676-3681.	2.8	28

#	ARTICLE	IF	CITATIONS
19	Progress in the Characterization of the Surface Species in Activated Carbons by means of INS Spectroscopy Coupled with Detailed DFT Calculations. <i>Advances in Condensed Matter Physics</i> , 2015, 2015, 1-8.	1.1	22
20	The effect of surface chemistry on the performances of Pd-based catalysts supported on activated carbons. <i>Catalysis Science and Technology</i> , 2017, 7, 4162-4172.	4.1	21
21	Looking for the active hydrogen species in a 5Åwt% Pt/C catalyst: a challenge for inelastic neutron scattering. <i>Faraday Discussions</i> , 2018, 208, 227-242.	3.2	20
22	Dynamic Behavior of Pd/P4VP Catalyst during the Aerobic Oxidation of 2-Propanol: A Simultaneous SAXS/XAS/MS Operando Study. <i>ACS Catalysis</i> , 2018, 8, 6870-6881.	11.2	13
23	How do the graphenic domains terminate in activated carbons and carbon-supported metal catalysts?. <i>Carbon</i> , 2020, 169, 357-369.	10.3	9
24	Hydrogenation of ethylene over palladium: evolution of the catalyst structure by operando synchrotron-based techniques. <i>Faraday Discussions</i> , 2021, 229, 197-207.	3.2	9
25	Evidence for H ₂ -Induced Ductility in a Pt/Al ₂ O ₃ Catalyst. <i>ACS Catalysis</i> , 2022, 12, 5979-5989.	11.2	9
26	Formation and growth of palladium nanoparticles inside porous poly(4-vinyl-pyridine) monitored by operando techniques: The role of different reducing agents. <i>Catalysis Today</i> , 2017, 283, 144-150.	4.4	8
27	Operando X-ray absorption spectra and mass spectrometry data during hydrogenation of ethylene over palladium nanoparticles. <i>Data in Brief</i> , 2019, 24, 103954.	1.0	8
28	Preparation of Pd/C catalysts: from the Pd-precursor solution to the final systems. <i>Studies in Surface Science and Catalysis</i> , 2006, 162, 721-728.	1.5	7
29	Assessing the functional groups in activated carbons through a multi-technique approach. <i>Catalysis Science and Technology</i> , 2022, 12, 1271-1288.	4.1	7
30	Changes of Pd Oxidation State in Pd/Al ₂ O ₃ Catalysts Using Modulated Excitation DRIFTS. <i>Catalysts</i> , 2021, 11, 116.	3.5	6
31	Deactivation of Industrial Pd/Al ₂ O ₃ Catalysts by Ethanol: A Spectroscopic Study. <i>ChemCatChem</i> , 2021, 13, 900-908.	3.7	5
32	Gas phase vs. liquid phase: monitoring H ₂ and CO adsorption phenomena on Pt/Al ₂ O ₃ by IR spectroscopy. <i>Catalysis Science and Technology</i> , 2022, 12, 1359-1367.	4.1	5
33	Investigation of carbon and alumina supported Pd catalysts during catalyst preparation. <i>Studies in Surface Science and Catalysis</i> , 2010, , 437-440.	1.5	2
34	Pd supported catalysts: Evolution of the support during Pd deposition and K doping. <i>Studies in Surface Science and Catalysis</i> , 2010, , 433-436.	1.5	0