Ulrich K Krieger

List of Publications by Year in descending order

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80 4,893 35
papers citations h-index

35 65 h-index g-index 114 3159

citing authors

106150

114 all docs 114 docs citations

times ranked

#	Article	IF	CITATIONS
1	New and extended parameterization of the thermodynamic model AIOMFAC: calculation of activity coefficients for organic-inorganic mixtures containing carboxyl, hydroxyl, carbonyl, ether, ester, alkenyl, alkyl, and aromatic functional groups. Atmospheric Chemistry and Physics, 2011, 11, 9155-9206.	1.9	317
2	Exploring the complexity of aerosol particle properties and processes using single particle techniques. Chemical Society Reviews, 2012, 41, 6631.	18.7	294
3	Ultra-slow water diffusion in aqueous sucrose glasses. Physical Chemistry Chemical Physics, 2011, 13, 3514.	1.3	249
4	Hygroscopic growth and water uptake kinetics of two-phase aerosol particles consisting of ammonium sulfate, adipic and humic acid mixtures. Journal of Aerosol Science, 2007, 38, 157-171.	1.8	206
5	Saturation Vapor Pressures and Transition Enthalpies of Low-Volatility Organic Molecules of Atmospheric Relevance: From Dicarboxylic Acids to Complex Mixtures. Chemical Reviews, 2015, 115, 4115-4156.	23.0	196
6	Revising the hygroscopicity of inorganic sea salt particles. Nature Communications, 2017, 8, 15883.	5.8	173
7	Phase Changes during Hygroscopic Cycles of Mixed Organic/Inorganic Model Systems of Tropospheric Aerosols. Journal of Physical Chemistry A, 2006, 110, 1881-1893.	1.1	171
8	Comparing the mechanism of water condensation and evaporation in glassy aerosol. Proceedings of the National Academy of Sciences of the United States of America, 2012, 109, 11613-11618.	3.3	167
9	Liquidâ^'Liquid Phase Separation in Mixed Organic/Inorganic Aerosol Particles. Journal of Physical Chemistry A, 2009, 113, 10966-10978.	1.1	163
10	Liquid-liquid phase separation and morphology of internally mixed dicarboxylic acids/ammonium sulfate/water particles. Atmospheric Chemistry and Physics, 2012, 12, 2691-2712.	1.9	161
11	Measurements of the timescales for the mass transfer of water in glassy aerosol at low relative humidity and ambient temperature. Atmospheric Chemistry and Physics, 2011, 11, 4739-4754.	1.9	149
12	Size-dependent stratospheric droplet composition in Lee wave temperature fluctuations and their potential role in PSC freezing. Geophysical Research Letters, 1995, 22, 3031-3034.	1.5	147
13	A combined particle trap/HTDMA hygroscopicity study of mixed inorganic/organic aerosol particles. Atmospheric Chemistry and Physics, 2008, 8, 5589-5601.	1.9	147
14	Oxalic acid as a heterogeneous ice nucleus in the upper troposphere and its indirect aerosol effect. Atmospheric Chemistry and Physics, 2006, 6, 3115-3129.	1.9	145
15	Ozone uptake on glassy, semi-solid and liquid organic matter and the role of reactive oxygen intermediates in atmospheric aerosol chemistry. Physical Chemistry Chemical Physics, 2016, 18, 12662-12674.	1.3	117
16	Thermodynamic Dissociation Constant of the Bisulfate Ion from Raman and Ion Interaction Modeling Studies of Aqueous Sulfuric Acid at Low Temperatures. Journal of Physical Chemistry A, 2003, 107, 4322-4332.	1.1	114
17	Viscous organic aerosol particles in the upper troposphere: diffusivity-controlled water uptake and ice nucleation?. Atmospheric Chemistry and Physics, 2015, 15, 13599-13613.	1.9	103
18	Liquidâ€liquid phase separation in aerosol particles: Dependence on O:C, organic functionalities, and compositional complexity. Geophysical Research Letters, 2012, 39, .	1.5	95

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19	Morphologies of mixed organic/inorganic/aqueous aerosol droplets. Faraday Discussions, 2013, 165, 289.	1.6	93
20	Field assisted transport of Na+ ions, Ca2+ ions and electrons in commercial soda-lime glass I: Experimental. Journal of Non-Crystalline Solids, 1988, 102, 50-61.	1.5	91
21	Morphological Investigations of Single Levitated H2SO4/NH3/H2O Aerosol Particles during Deliquescence/Efflorescence Experiments. Journal of Physical Chemistry A, 2004, 108, 2700-2709.	1.1	86
22	White light Mie resonance spectroscopy used to measure very low vapor pressures of substances in aqueous solution aerosol particles. Optics Express, 2006, 14, 6951.	1.7	83
23	The vapor pressures and activities of dicarboxylic acids reconsidered: the impact of the physical state of the aerosol. Atmospheric Chemistry and Physics, 2010, 10, 11753-11767.	1.9	67
24	Liquid–Liquid Phase Separation in Mixed Organic/Inorganic Single Aqueous Aerosol Droplets. Journal of Physical Chemistry A, 2015, 119, 4177-4190.	1.1	67
25	Densities and refractive indices of H2SO4/HNO3/H2O solutions to stratospheric temperatures. Geophysical Research Letters, 1996, 23, 3707-3710.	1.5	66
26	Efflorescence of Ammonium Sulfate and Coated Ammonium Sulfate Particles: Evidence for Surface Nucleation. Journal of Physical Chemistry A, 2010, 114, 9486-9495.	1.1	66
27	Retrieving the translational diffusion coefficient of water from experiments on single levitated aerosol droplets. Physical Chemistry Chemical Physics, 2014, 16, 16677.	1.3	64
28	Measurements of Thermodynamic and Optical Properties of Selected Aqueous Organic and Organic–Inorganic Mixtures of Atmospheric Relevance. Journal of Physical Chemistry A, 2012, 116, 9954-9968.	1.1	63
29	Two-dimensional angular light-scattering in aqueous NaCl single aerosol particles during deliquescence and efflorescence. Optics Express, 2001, 8, 314.	1.7	60
30	Evidence of Quantum Correlation Effects of Protons and Deuterons in the Raman Spectra of LiquidH2O-D2O. Physical Review Letters, 1995, 75, 3008-3011.	2.9	47
31	Measurement of the refractive indices of H_2SO_4â€"HNO_3â€"H_2O solutions to stratospheric temperatures. Applied Optics, 2000, 39, 3691.	2.1	45
32	A method for extracting calibrated volatility information from the FIGAERO-HR-ToF-CIMS and its experimental application. Atmospheric Measurement Techniques, 2019, 12, 1429-1439.	1.2	42
33	Shikimic acid ozonolysis kinetics of the transition from liquid aqueous solution to highly viscous glass. Physical Chemistry Chemical Physics, 2015, 17, 31101-31109.	1.3	41
34	A reference data set for validating vapor pressure measurement techniques: homologous series of polyethylene glycols. Atmospheric Measurement Techniques, 2018, 11, 49-63.	1.2	41
35	Climatological and radiative properties of midlatitude cirrus clouds derived by automatic evaluation of lidar measurements. Atmospheric Chemistry and Physics, 2016, 16, 7605-7621.	1.9	40
36	Diffusivity measurements of volatile organics in levitated viscous aerosol particles. Atmospheric Chemistry and Physics, 2017, 17, 8453-8471.	1.9	40

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37	Electrodynamic balance measurements of thermodynamic, kinetic, and optical aerosol properties inaccessible to bulk methods. Atmospheric Measurement Techniques, 2015, 8, 2397-2408.	1.2	39
38	Photolytic radical persistence due to anoxia in viscous aerosol particles. Nature Communications, 2021, 12, 1769.	5.8	37
39	Supercooling of single H2SO4/H2O aerosols to 158 K: No evidence for the occurrence of the octrahydrate. Geophysical Research Letters, 2000, 27, 2097-2100.	1.5	33
40	Technical note: Monte Carlo genetic algorithm (MCGA) for model analysis of multiphase chemical kinetics to determine transport and reaction rate coefficients using multiple experimental data sets. Atmospheric Chemistry and Physics, 2017, 17, 8021-8029.	1.9	33
41	Vapor pressures of substituted polycarboxylic acids are much lower than previously reported. Atmospheric Chemistry and Physics, 2013, 13, 6647-6662.	1.9	32
42	Balloon-borne match measurements of midlatitude cirrus clouds. Atmospheric Chemistry and Physics, 2014, 14, 7341-7365.	1.9	28
43	Evaporation kinetics of a non-spherical, levitated aerosol particle using optical resonance spectroscopy for precision sizing. Optics Express, 2009, 17, 4659.	1.7	26
44	Redistribution of black carbon in aerosol particles undergoing liquidâ€liquid phase separation. Geophysical Research Letters, 2015, 42, 2532-2539.	1.5	25
45	Ozonolysis of Oleic Acid Aerosol Revisited: Multiphase Chemical Kinetics and Reaction Mechanisms. ACS Earth and Space Chemistry, 2021, 5, 3313-3323.	1.2	25
46	Rutherford Backscattering to Study the Near-Surface Region of Volatile Liquids and Solids. Science, 2002, 295, 1048-1050.	6.0	23
47	Using dynamic light scattering to characterize mixed phase single particles levitated in a quasi-electrostatic balance. Faraday Discussions, 2008, 137, 377-388.	1.6	22
48	Kinetic Limitation to Inorganic Ion Diffusivity and to Coalescence of Inorganic Inclusions in Viscous Liquid–Liquid Phase-Separated Particles. Journal of Physical Chemistry A, 2017, 121, 9284-9296.	1.1	22
49	Electrodynamic balance–mass spectrometry of single particles as a new platform for atmospheric chemistry research. Atmospheric Measurement Techniques, 2018, 11, 33-47.	1.2	22
50	Experimental evidence for excess entropy discontinuities in glass-forming solutions. Journal of Chemical Physics, 2012, 136, 074515.	1.2	21
51	Experimental determination of the temperature dependence of water activities for a selection of aqueous organic solutions. Atmospheric Chemistry and Physics, 2014, 14, 9993-10012.	1.9	20
52	Photochemical degradation of iron(III) citrate/citric acid aerosol quantified with the combination of three complementary experimental techniques and a kinetic process model. Atmospheric Chemistry and Physics, 2021, 21, 315-338.	1.9	20
53	Light-scattering intensity fluctuations in single aerosol particles during deliquescence. Journal of Quantitative Spectroscopy and Radiative Transfer, 2001, 70, 545-554.	1.1	19
54	Bromine Enrichment in the Near-Surface Region of Br-Doped NaCl Single Crystals Diagnosed by Rutherford Backscattering Spectrometry. Journal of Physical Chemistry A, 2007, 111, 4312-4321.	1.1	16

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55	Tracking Water Sorption in Glassy Aerosol Particles using Morphology-Dependent Resonances. Journal of Physical Chemistry A, 2017, 121, 8176-8184.	1.1	16
56	Relevance of Particle Morphology for Atmospheric Aerosol Processing. Trends in Chemistry, 2020, 2, 1-3.	4.4	16
57	Shortwave radiative impact of liquid–liquid phase separation in brown carbon aerosols. Atmospheric Chemistry and Physics, 2018, 18, 13511-13530.	1.9	15
58	Visualizing reaction and diffusion in xanthan gum aerosol particles exposed to ozone. Physical Chemistry Chemical Physics, 2019, 21, 20613-20627.	1.3	15
59	Reply to "Comment on the  Thermodynamic Dissociation Constant of the Bisulfate Ion from Raman and Ion Interaction Modeling Studies of Aqueous Sulfuric Acid at Low Temperatures'― Journal of Physical Chemistry A, 2005, 109, 2707-2709.	1.1	12
60	Technical Note: Organics-Induced Fluorescence in Raman Studies of Sulfuric Acid Aerosols. Aerosol Science and Technology, 2002, 36, 510-512.	1.5	11
61	Using photon-counting histograms to characterize levitated liquid aerosol particles with a single, solid inclusion. Journal of Quantitative Spectroscopy and Radiative Transfer, 2004, 89, 191-200.	1.1	10
62	Time evolution of steep diffusion fronts in highly viscous aerosol particles measured with Mie resonance spectroscopy. Journal of Chemical Physics, 2018, 149, 244506.	1.2	10
63	Extension of the AIOMFAC model by iodine and carbonate species: applications for aerosol acidity and cloud droplet activation. Atmospheric Chemistry and Physics, 2022, 22, 973-1013.	1.9	8
64	Improved inverted bubble method for measuring small contact angles at crystal-solution-vapor interfaces. Applied Optics, 2007, 46, 5835.	2.1	6
65	Observations and calculations of two-dimensional angular optical scattering (TAOS) patterns of a single levitated cluster of two and four microspheres. Journal of Quantitative Spectroscopy and Radiative Transfer, 2011, 112, 1761-1765.	1.1	6
66	Measured solid state and subcooled liquid vapour pressures of nitroaromatics using Knudsen effusion mass spectrometry. Atmospheric Chemistry and Physics, 2020, 20, 8293-8314.	1.9	6
67	An experimental examination of intensity fluctuations of a host droplet containing an inclusion. Journal of Quantitative Spectroscopy and Radiative Transfer, 2003, 79-80, 873-880.	1.1	5
68	Simultaneous Measurements of PM ₁₀ and PM ₁ using a single TEOM [#] . Aerosol Science and Technology, 2007, 41, 975-980.	1.5	5
69	Carbon Dioxide Diffusivity in Single, Levitated Organic Aerosol Particles. Journal of Physical Chemistry Letters, 2019, 10, 4484-4489.	2.1	5
70	Experimental evidence for nonclassical fourth-order interferences in the quasielastic light scattering of water. Physical Review A, 1995, 52, R1827-R1830.	1.0	4
71	Photophoretic spectroscopy in atmospheric chemistry – high-sensitivity measurements of light absorption by a single particle. Atmospheric Measurement Techniques, 2020, 13, 3191-3203.	1.2	4
72	RBS analysis of trace gas uptake on ice. Nuclear Instruments & Methods in Physics Research B, 2002, 190, 47-53.	0.6	3

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73	Evaporation kinetics of a non-spherical, levitated aerosol particle using optical resonance spectroscopy for precision sizing: Errata. Optics Express, 2010, 18, 10760.	1.7	2
74	Ion Depletion Near a Solution Surface: Is Image-Charge Repulsion Sufficient?. Physical Review Letters, 2013, 111, 266102.	2.9	2
75	Grazing angle 2MeV RBS on the surface of a liquid with atomic layer depth resolution. Nuclear Instruments & Methods in Physics Research B, 2010, 268, 1711-1713.	0.6	1
76	Uptake of nitric acid on NaCl single crystals measured by backscattering spectrometry. Nuclear Instruments & Methods in Physics Research B, 2010, 268, 2202-2204.	0.6	1
77	The Application of RBS To Investigate The Diffusion of HCl Into The Near Surface Region Of Ice. AIP Conference Proceedings, 2003, , .	0.3	0
78	Diffusion constants of Br in NaCl measured by Rutherford backscattering spectroscopy. Journal of Applied Physics, 2009, 105, 124910.	1.1	0
79	Response to "Comment on â€Experimental evidence for excess entropy discontinuities in glass-forming solutionsâ€â€™ [J. Chem. Phys. 139, 047101 (2013)]. Journal of Chemical Physics, 2013, 139, 047102.	1.2	0
80	Imaging Molecular Reaction and Diffusion in Organic Aerosol Particles. Microscopy and Microanalysis, 2018, 24, 496-497.	0.2	0