

# Gaosheng Zhang

## List of Publications by Year in descending order

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52  
papers

4,881  
citations

126708

33  
h-index

174990

52  
g-index

52  
all docs

52  
docs citations

52  
times ranked

4683  
citing authors

#	ARTICLE	IF	CITATIONS
1	Multi-step purification of electrolytic manganese residue leachate using hydroxide sedimentation, struvite precipitation, chlorination and coagulation: Advanced removal of manganese, ammonium, and phosphate. <i>Science of the Total Environment</i> , 2022, 805, 150237.	3.9	32
2	Uptake, organ distribution and health risk assessment of potentially toxic elements in crops in abandoned indigenous smelting region. <i>Chemosphere</i> , 2022, 292, 133321.	4.2	22
3	Highly efficient removal of thallium(I) by facilely fabricated amorphous titanium dioxide from water and wastewater. <i>Scientific Reports</i> , 2022, 12, 72.	1.6	3
4	Magnetite-based Biochar Coupled with Binary Oxidants for the Effective Removal of Mixed Dye from Wastewater. <i>Fibers and Polymers</i> , 2022, 23, 450-462.	1.1	6
5	Facile synthesis of novel tremella-like MnO@Mn <sub>2</sub> O <sub>3</sub> and its exceptional performance on removal of phosphate. <i>Journal of Environmental Chemical Engineering</i> , 2021, 9, 105635.	3.3	4
6	Efficient Sorption of Arsenic on Nanostructured Fe-Cu Binary Oxides: Influence of Structure and Crystallinity. <i>Frontiers in Chemistry</i> , 2021, 9, 840446.	1.8	2
7	Polyvinyl alcohol-stabilized granular Fe-Mn binary oxide as an effective adsorbent for simultaneous removal of arsenate and arsenite. <i>Environmental Technology (United Kingdom)</i> , 2020, 41, 2564-2574.	1.2	6
8	Enhanced thallium(I) removal from wastewater using hypochlorite oxidation coupled with magnetite-based biochar adsorption. <i>Science of the Total Environment</i> , 2020, 698, 134166.	3.9	67
9	Highly efficient removal of thallium(I) from wastewater via hypochlorite catalytic oxidation coupled with adsorption by hydrochar coated nickel ferrite composite. <i>Journal of Hazardous Materials</i> , 2020, 388, 122016.	6.5	27
10	Zero-valent iron-manganese bimetallic nanocomposites catalyze hypochlorite for enhanced thallium(I) oxidation and removal from wastewater: Materials characterization, process optimization and removal mechanisms. <i>Journal of Hazardous Materials</i> , 2020, 386, 121900.	6.5	43
11	Hyperaccumulation and transport mechanism of thallium and arsenic in brake ferns ( <i>Pteris vittata</i> L.): A case study from mining area. <i>Journal of Hazardous Materials</i> , 2020, 388, 121756.	6.5	58
12	Zero-valent manganese nanoparticles coupled with different strong oxidants for thallium removal from wastewater. <i>Frontiers of Environmental Science and Engineering</i> , 2020, 14, 1.	3.3	29
13	Efficient arsenic(III) removal from aqueous solution by a novel nanostructured iron-copper-manganese trimetal oxide. <i>Journal of Molecular Liquids</i> , 2020, 309, 112993.	2.3	23
14	Novel nanostructured Fe-Cu-Al trimetal oxide for enhanced antimony(V) removal: synthesis, characterization and performance. <i>Water Science and Technology</i> , 2019, 79, 1995-2004.	1.2	5
15	Synthesis of manganese dioxide with different morphologies for thallium removal from wastewater. <i>Journal of Environmental Management</i> , 2019, 251, 109563.	3.8	42
16	Biochar derived from watermelon rinds as regenerable adsorbent for efficient removal of thallium(I) from wastewater. <i>Chemical Engineering Research and Design</i> , 2019, 127, 257-266.	2.7	76
17	A novel nanostructured Fe-Ti-Mn composite oxide for highly efficient arsenic removal: Preparation and performance evaluation. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2019, 561, 364-372.	2.3	48
18	Efficient oxidation and sorption of arsenite using a novel titanium(IV)-manganese(IV) binary oxide sorbent. <i>Journal of Hazardous Materials</i> , 2018, 353, 410-420.	6.5	59

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19	Comparing adsorption of arsenic and antimony from single-solute and bi-solute aqueous systems onto ZIF-8. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2018, 538, 164-172.	2.3	50
20	Facile fabrication of nanostructured cerium-manganese binary oxide for enhanced arsenite removal from water. <i>Chemical Engineering Journal</i> , 2018, 334, 1518-1526.	6.6	104
21	Superior adsorption of thallium(I) on titanium peroxide: Performance and mechanism. <i>Chemical Engineering Journal</i> , 2018, 331, 471-479.	6.6	110
22	Enhanced removal of arsenite and arsenate by a multifunctional Fe-Ti-Mn composite oxide: Photooxidation, oxidation and adsorption. <i>Water Research</i> , 2018, 147, 264-275.	5.3	129
23	Removal and recovery of thallium from aqueous solutions via a magnetite-mediated reversible adsorption-desorption process. <i>Journal of Cleaner Production</i> , 2018, 199, 705-715.	4.6	72
24	Efficient removal of thallium(I) from wastewater using flower-like manganese dioxide coated magnetic pyrite cinder. <i>Chemical Engineering Journal</i> , 2018, 353, 867-877.	6.6	90
25	Removal of thallium from wastewater by a combination of persulfate oxidation and iron coagulation. <i>Chemical Engineering Research and Design</i> , 2018, 119, 340-349.	2.7	38
26	Concentrations, spatial distribution, and risk assessment of soil heavy metals in a Zn-Pb mine district in southern China. <i>Environmental Monitoring and Assessment</i> , 2016, 188, 413.	1.3	40
27	Efficient removal of arsenic from water using a granular adsorbent: Fe-Mn binary oxide impregnated chitosan bead. <i>Bioresource Technology</i> , 2015, 193, 243-249.	4.8	135
28	Adsorptive removal of arsenic from aqueous solution by zeolitic imidazolate framework-8 (ZIF-8) nanoparticles. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2015, 465, 67-76.	2.3	429
29	The ammonia effects to the habitat requirements and adaptability of <i>Daphnia magna</i> . <i>Desalination and Water Treatment</i> , 2014, 52, 2695-2699.	1.0	6
30	Enhanced adsorption of phosphate from aqueous solution by nanostructured iron(III)-copper(II) binary oxides. <i>Chemical Engineering Journal</i> , 2014, 235, 124-131.	6.6	164
31	Enhanced arsenate removal by novel Fe-La composite (hydr)oxides synthesized via coprecipitation. <i>Chemical Engineering Journal</i> , 2014, 251, 69-79.	6.6	77
32	Novel Core-Shell Structured Mn-Fe/MnO <sub>2</sub> Magnetic Nanoparticles for Enhanced Pb(II) Removal from Aqueous Solution. <i>Industrial &amp; Engineering Chemistry Research</i> , 2014, 53, 18481-18488.	1.8	33
33	Respective Role of Fe and Mn Oxide Contents for Arsenic Sorption in Iron and Manganese Binary Oxide: An X-ray Absorption Spectroscopy Investigation. <i>Environmental Science &amp; Technology</i> , 2014, 48, 10316-10322.	4.6	200
34	Organochlorine pesticide contamination in marine organisms of Yantai coast, northern Yellow Sea of China. <i>Environmental Monitoring and Assessment</i> , 2014, 186, 1561-1568.	1.3	10
35	Polybrominated Diphenyl Ethers Contamination in Marine Organisms of Yantai Coast, Northern Yellow Sea of China. <i>Bulletin of Environmental Contamination and Toxicology</i> , 2013, 90, 679-683.	1.3	3
36	Simultaneous removal of arsenate and arsenite by a nanostructured zirconium-manganese binary hydrous oxide: Behavior and mechanism. <i>Journal of Colloid and Interface Science</i> , 2013, 397, 137-143.	5.0	68

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37	Modeling macrozooplankton and water quality relationships after wetland construction in the Wenyuhe River Basin, China. <i>Ecological Modelling</i> , 2013, 252, 97-105.	1.2	7
38	Nanostructured iron(III)-copper(II) binary oxide: A novel adsorbent for enhanced arsenic removal from aqueous solutions. <i>Water Research</i> , 2013, 47, 4022-4031.	5.3	290
39	Adsorption of Phosphate from Aqueous Solution Using an Iron-Zirconium Binary Oxide Sorbent. <i>Water, Air, and Soil Pollution</i> , 2012, 223, 4221-4231.	1.1	101
40	Heavy metal contamination in the marine organisms in Yantai coast, northern Yellow Sea of China. <i>Ecotoxicology</i> , 2012, 21, 1726-1733.	1.1	54
41	Evidence for the Stepwise Behavioral Response Model (SBRM): The effects of Carbamate Pesticides on medaka ( <i>Oryzias latipes</i> ) in an online monitoring system. <i>Chemosphere</i> , 2012, 87, 734-741.	4.2	27
42	Arsenate uptake and arsenite simultaneous sorption and oxidation by Fe-Mn binary oxides: Influence of Mn/Fe ratio, pH, Ca <sup>2+</sup> , and humic acid. <i>Journal of Colloid and Interface Science</i> , 2012, 366, 141-146.	5.0	108
43	Facile synthesis, characterization of a MnFe <sub>2</sub> O <sub>4</sub> /activated carbon magnetic composite and its effectiveness in tetracycline removal. <i>Materials Chemistry and Physics</i> , 2012, 135, 16-24.	2.0	175
44	Improvement of Biological Early Warning System Based on Medaka ( <i>Oryzias latipes</i> ) Behavioral Responses to Physiochemical Factors. <i>Journal of Biobased Materials and Bioenergy</i> , 2012, 6, 678-681.	0.1	5
45	Adsorptive removal of arsenic from water by an iron-zirconium binary oxide adsorbent. <i>Journal of Colloid and Interface Science</i> , 2011, 358, 230-237.	5.0	236
46	A new online monitoring and management system for accidental pollution events developed for the regional water basin in Ningbo, China. <i>Water Science and Technology</i> , 2011, 64, 1828-1834.	1.2	9
47	Adsorption behavior and mechanism of arsenate at Fe-Mn binary oxide/water interface. <i>Journal of Hazardous Materials</i> , 2009, 168, 820-825.	6.5	194
48	Removal of phosphate from water by a Fe-Mn binary oxide adsorbent. <i>Journal of Colloid and Interface Science</i> , 2009, 335, 168-174.	5.0	356
49	Preparation and evaluation of a novel Fe-Mn binary oxide adsorbent for effective arsenite removal. <i>Water Research</i> , 2007, 41, 1921-1928.	5.3	538
50	CuFe <sub>2</sub> O <sub>4</sub> /activated carbon composite: A novel magnetic adsorbent for the removal of acid orange II and catalytic regeneration. <i>Chemosphere</i> , 2007, 68, 1058-1066.	4.2	270
51	Silicate Hinder In Situ Formed Ferric Hydroxide Precipitation: Inhibiting Arsenic Removal from Water. <i>Environmental Engineering Science</i> , 2007, 24, 707-715.	0.8	20
52	Optimization of initial substrate and pH levels for germination of sporing hydrogen-producing anaerobes in cow dung compost. <i>Bioresource Technology</i> , 2004, 91, 189-193.	4.8	181