Baile Wu

List of Publications by Year in descending order

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933447 1372567 1,291 11 10 10 h-index citations g-index papers 955 12 12 12 docs citations citing authors all docs times ranked

#	Article	IF	CITATIONS
1	Role of surface functional groups of hydrogels in metal adsorption: From performance to mechanism. Journal of Hazardous Materials, 2021, 408, 124463.	12.4	63
2	Scaled-up development of magnetically recyclable Fe3O4/La(OH)3 composite for river water phosphate removal: From bench-scale to pilot-scale study. Science of the Total Environment, 2021, 791, 148281.	8.0	15
3	Selective Phosphate Removal from Water and Wastewater using Sorption: Process Fundamentals and Removal Mechanisms. Environmental Science & Environment	10.0	437
4	Development of Fe0/Fe3O4 composites with tunable properties facilitated by Fe2+ for phosphate removal from river water. Chemical Engineering Journal, 2020, 388, 124242.	12.7	37
5	Surface Functional Group Engineering of CeO ₂ Particles for Enhanced Phosphate Adsorption. Environmental Science & Enhanced Phosphate Adsorption.	10.0	81
6	Magnetically separable ZrO2@SiO2@Fe3O4 for selective phosphate removal from wastewater. HKIE Transactions, 2019, 26, 88-96.	0.1	0
7	Lanthanum oxide nanorods for enhanced phosphate removal from sewage: A response surface methodology study. Chemosphere, 2018, 192, 209-216.	8.2	95
8	Application of Magnetic Hydrogel for Anionic Pollutants Removal from Wastewater with Adsorbent Regeneration and Reuse. Journal of Hazardous, Toxic, and Radioactive Waste, 2017, 21, .	2.0	12
9	Fabrication of silica-free superparamagnetic ZrO2@Fe3O4 with enhanced phosphate recovery from sewage: Performance and adsorption mechanism. Chemical Engineering Journal, 2017, 319, 258-267.	12.7	130
10	Removal Mechanisms of Phosphate by Lanthanum Hydroxide Nanorods: Investigations using EXAFS, ATR-FTIR, DFT, and Surface Complexation Modeling Approaches. Environmental Science & Emp; Technology, 2017, 51, 12377-12384.	10.0	142
11	Highly efficient and selective phosphate removal from wastewater by magnetically recoverable La(OH)3/Fe3O4 nanocomposites. Water Research, 2017, 126, 179-188.	11.3	279