

Daiki Umeyama

List of Publications by Year in Descending Order

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The third column is the impact factor (IF) of the journal, and the fourth column is the number of citations of the article.

23
papers

2,367
citations

19
h-index

24
g-index

24
ext. papers

2,655
ext. citations

11.5
avg, IF

5.13
L-index

#	Paper	IF	Citations
23	Dynamically Disordered Lattice in a Layered Pb-I-SCN Perovskite Thin Film Probed by Two-Dimensional Infrared Spectroscopy. <i>Journal of the American Chemical Society</i> , 2018 , 140, 9882-9890	16.4	35
22	Carving Out Pores in Redox-Active One-Dimensional Coordination Polymers. <i>Angewandte Chemie</i> , 2018 , 130, 14793-14796	3.6	2
21	Carving Out Pores in Redox-Active One-Dimensional Coordination Polymers. <i>Angewandte Chemie - International Edition</i> , 2018 , 57, 14585-14588	16.4	7
20	Chemical Approaches to Addressing the Instability and Toxicity of Lead-Halide Perovskite Absorbers. <i>Inorganic Chemistry</i> , 2017 , 56, 46-55	5.1	186
19	Encapsulating Mobile Proton Carriers into Structural Defects in Coordination Polymer Crystals: High Anhydrous Proton Conduction and Fuel Cell Application. <i>Journal of the American Chemical Society</i> , 2016 , 138, 8505-11	16.4	116
18	Glass Formation of a Coordination Polymer Crystal for Enhanced Proton Conductivity and Material Flexibility. <i>Angewandte Chemie - International Edition</i> , 2016 , 55, 5195-200	16.4	83
17	Red-to-Black Piezochromism in a Compressible Pb ₃ BCN Layered Perovskite. <i>Chemistry of Materials</i> , 2016 , 28, 3241-3244	9.6	77
16	Glass formation via structural fragmentation of a 2D coordination network. <i>Chemical Communications</i> , 2015 , 51, 12728-31	5.8	25
15	Synthesis of chiral porous coordination polymer that shows structural transformation induced by guest molecules. <i>Inorganica Chimica Acta</i> , 2015 , 424, 221-225	2.7	3
14	Reversible solid-to-liquid phase transition of coordination polymer crystals. <i>Journal of the American Chemical Society</i> , 2015 , 137, 864-70	16.4	124
13	Template-directed proton conduction pathways in a coordination framework. <i>Journal of Materials Chemistry A</i> , 2014 , 2, 10404-10409	13	35
12	Pressure-induced amorphization of a dense coordination polymer and its impact on proton conductivity. <i>APL Materials</i> , 2014 , 2, 124401	5.7	16
11	Order-to-disorder structural transformation of a coordination polymer and its influence on proton conduction. <i>Chemical Communications</i> , 2014 , 50, 10241-3	5.8	69
10	Synthesis and porous properties of chromium azolate porous coordination polymers. <i>Inorganic Chemistry</i> , 2014 , 53, 9870-5	5.1	21
9	Fe ²⁺ -based layered porous coordination polymers and soft encapsulation of guests via redox activity. <i>Journal of Materials Chemistry A</i> , 2013 , 1, 3675	13	29
8	Postsynthesis modification of a porous coordination polymer by LiCl To enhance H ⁺ transport. <i>Journal of the American Chemical Society</i> , 2013 , 135, 4612-5	16.4	67
7	Ion conductivity and transport by porous coordination polymers and metal-organic frameworks. <i>Accounts of Chemical Research</i> , 2013 , 46, 2376-84	24.3	644

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| 6 | Integration of intrinsic proton conduction and guest-accessible nanospace into a coordination polymer. <i>Journal of the American Chemical Society</i> , 2013 , 135, 11345-50 | 16.4 | 99 |
| 5 | Dense coordination network capable of selective CO ₂ capture from C1 and C2 hydrocarbons. <i>Journal of the American Chemical Society</i> , 2012 , 134, 9852-5 | 16.4 | 76 |
| 4 | Investigation of post-grafted groups of a porous coordination polymer and its proton conduction behavior. <i>Dalton Transactions</i> , 2012 , 41, 13261-3 | 4.3 | 28 |
| 3 | Inherent proton conduction in a 2D coordination framework. <i>Journal of the American Chemical Society</i> , 2012 , 134, 12780-5 | 16.4 | 216 |
| 2 | Coordination-network-based ionic plastic crystal for anhydrous proton conductivity. <i>Journal of the American Chemical Society</i> , 2012 , 134, 7612-5 | 16.4 | 198 |
| 1 | Confinement of mobile histamine in coordination nanochannels for fast proton transfer. <i>Angewandte Chemie - International Edition</i> , 2011 , 50, 11706-9 | 16.4 | 211 |