List of Publications by Year in descending order

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FIRI KASAI

#	Article	IF	CITATIONS
1	Reduction of CO ₂ Emissions from Blast Furnace Applying Reactive Coke Agglomerate and Hydrogen Reduction. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2022, 108, 335-342.	0.1	2
2	Effects of Iron Ore Type and Gangue Mineral Components on Strength of Sintered Fine Powder Granule. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2021, 107, 463-470.	0.1	3
3	Influence of Oxygen Partial Pressure on Oxidation Reaction of Iron-bearing Materials in Iron Ore Sintering Bed. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2021, 107, 431-438.	0.1	0
4	Effect of Silica in Ash of Coke on Carburization and Melting of Iron. ISIJ International, 2021, 61, 1479-1487.	0.6	1
5	Effect of Ore Type and Gangue Content on Carburization and Melting Behavior of Carbon-Iron Ore Composite. ISIJ International, 2021, 61, 1808-1813.	0.6	5
6	Acceleration of Oxidation Reaction of Iron-bearing Materials Co-existed with Carbonaceous Materials in Iron Ore Sintering Bed. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2021, 107, 439-446.	0.1	0
7	Reduction of Iron Ore by Uncarbonized Biomass in a Rotary Kiln Type Furnace. ISIJ International, 2021, 61, 2971-2978.	0.6	3
8	Effect of Hydrogen Concentration in Reducing Gas on the Changes in Mineral Phases during Reduction of Iron Ore Sinter. ISIJ International, 2020, 60, 2678-2685.	0.6	15
9	Development of High Temperature Oxidation Resistant Iron-Based Heat Storage Materials for Rapid Carbonization and Pulverization Process of Biomass. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2020, 106, 527-533.	0.1	2
10	Forming Behavior of Fine Particulate Matters during Iron Ore Sintering Process. ISIJ International, 2020, 60, 1649-1654.	0.6	0
11	<i>In-situ</i> Evaluation Method for Crack Generation and Propagation Behaviors of Iron Ore Burden during Low Temperature Reduction by Applying Acoustic Emission Method. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2020, 106, 719-726.	0.1	1
12	Influence of Heat Treatment Temperature on Self-healing Effect of Fe Particle/Mullite Ceramic Composites. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2020, 106, 844-850.	0.1	3
13	Simultaneous Carbonization and Pulverization Behaviors of Woody Biomass by a Rapid Carbonization Process Applying Heat Storage Materials. ISIJ International, 2020, 60, 2107-2111.	0.6	2
14	Effect of Types of Carbonaceous Material and CaO Addition on Reduction Behavior of Pre-reduced Iron Ore–Carbon Composite. ISIJ International, 2019, 59, 1011-1017.	0.6	3
15	Intra-Particle Water Migration Dynamics during Iron Ore Granulation Process. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2019, 105, 1033-1041.	0.1	0
16	Morphology Change and Carburization Characteristic of Iron Ore-Coal Composite During Reduction under a Simulated Condition of Blast Furnace. ISIJ International, 2019, 59, 1982-1990.	0.6	10
17	Quantitative Evaluation of Reaction Mode and Reduction Disintegration Behavior of Iron Ore Agglomerates during Low Temperature Reduction. ISIJ International, 2018, 58, 1761-1767.	0.6	5
18	Interâ^'particle water infiltration dynamics of iron ore fines during granulation process. Powder Technology, 2018, 339, 550-559.	2.1	8

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19	A New Approach to Processing Rutile from Ilmenite Ore Utilizing the Instability of Pseudobrookite. Metallurgical and Materials Transactions B: Process Metallurgy and Materials Processing Science, 2018, 49, 2278-2284.	1.0	4
20	<i>In-situ</i> Evaluation Method for Crack Generation and Propagation Behaviors of Iron Ore Burden during Low Temperature Reduction by Applying Acoustic Emission Method. ISIJ International, 2018, 58, 1413-1419.	0.6	4
21	Acceleration of Carburization and Melting of Reduced Iron in Iron Ore–Carbon Composite Using Different Types of Carbonaceous Materials. ISIJ International, 2017, 57, 1928-1936.	0.6	17
22	Influence of Reducing Gas Composition on Disintegration Behavior of Iron Ore Agglomerates. ISIJ International, 2017, 57, 1499-1508.	0.6	11
23	Effective Utilization of KR Slag in Iron Ore Sintering Process. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2017, 103, 357-364.	0.1	5
24	Oxidation Characteristics of Metallic Iron and Magnetite Concentrate with Coke in Sintering Bed. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2017, 103, 365-371.	0.1	4
25	Development of Manufacturing Principle of Porous Iron by Carbothermic Reduction of Composite of Hematite and Biomass Char. Materials Transactions, 2017, 58, 1742-1748.	0.4	3
26	Effect of Addition of CaO Component on the Oxidation Reaction of Wüstite Particles in Sintering Bed. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2017, 103, 341-347.	0.1	0
27	Promoting Effect on Oxidation Reaction of Iron-bearing Agglomeration Agent by Melt Formation. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2017, 103, 348-356.	0.1	1
28	Intra–Particle Water Migration Dynamics during Iron Ore Granulation Process. ISIJ International, 2017, 57, 1384-1393.	0.6	12
29	Reduction Mechanism of Fe _x O — Graphite Composite under Elevating Temperature. ISIJ International, 2016, 56, 233-238.	0.6	2
30	Effect of Placement and Reactivity of Iron-ore and Carbon on Iron-ore Softening-melting Properties in Blast Furnace Cohesive Zone. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2016, 102, 475-484.	0.1	4
31	Rapid Carbonization Process Using Heat Storage Materials and Characterization of the Obtained Char. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2016, 102, 730-735.	0.1	5
32	Vapor pressure measurement of lead and lead chlorides in FeOT–CaO–SiO2–Al2O3 system. Transactions of Nonferrous Metals Society of China, 2015, 25, 2772-2780.	1.7	2
33	Reduction and Disintegration Behavior of Sinter under N ₂ –CO–CO ₂ –H ₂ –H _{ Gas at 773 K. ISIJ International, 2015, 55, 1181-1187.}	2&l t;/s ub8	،gt ;D 4
34	Reduction Mechanism of Composite Consisted of Coal and Hematite Ore by Volatile Matter at 700–1100 K. ISIJ International, 2015, 55, 1188-1196.	0.6	17
35	Effect of Addition of CaO Component on the Oxidation Reaction of Wustite Particles in Sintering Bed. ISIJ International, 2015, 55, 940-946.	0.6	4
36	Effect of the Reduction of Calcium Ferrite on Disintegration Behavior of Sinter under High Hydrogen Atmosphere, ISII International, 2015, 55, 1197-1205,	0.6	19

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37	Promotion Effect of Melt-formation in the Sintering Bed on the Oxidation Reaction of Metallic Iron Particle Usedas Agglomeration Agent. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2014, 100, 160-169.	0.1	6
38	Development of Porous Iron based Material by Slag Foaming and its Reduction. , 2014, 4, 27-32.		12
39	Effect of Utilization of Metallic Fe Particles as an Agglomeration Agent on the Permeability of Sintering Bed. ISIJ International, 2013, 53, 1617-1624.	0.6	11
40	Quantitative Analysis on Contribution of Direct Reduction of Iron Oxide in Carbon Composite. ISIJ International, 2013, 53, 1763-1769.	0.6	12
41	Numerical simulation of the ball impact process. Surface and Coatings Technology, 2012, 210, 151-155.	2.2	3
42	Reduction Disintegration Behavior of Iron Ore Sinter under High H2 and H2O Conditions. ISIJ International, 2012, 52, 1447-1453.	0.6	31
43	Fabrication of hydroxyapatite coatings by the ball impact process. Surface and Coatings Technology, 2012, 206, 3949-3954.	2.2	11
44	Gasification and Reduction Behavior of Iron Ore-Carbon Composite under High Pressure. ISIJ International, 2012, 52, 1778-1784.	0.6	8
45	Utilization of Ores with High Combined Water Content for Ore–carbon Composite and Iron Coke. ISIJ International, 2011, 51, 1220-1226.	0.6	27
46	Reduction Mechanism of Iron Oxide–Carbon Composite with Polyethylene at Lower Temperature. ISIJ International, 2011, 51, 9-13.	0.6	58
47	Vitrification Treatment of Asbestos Waste with Incineration Ash of Solid Waste. High Temperature Materials and Processes, 2011, 30, .	0.6	1
48	Investigation of structural formation of Al–SiC surface composite under ball collisions. Materials Science & Engineering A: Structural Materials: Properties, Microstructure and Processing, 2011, 528, 3455-3462.	2.6	25
49	Process Analysis of the Effective Utilization of Molten Slag Heat by Direct Blast Furnace Cement Production System. ISIJ International, 2010, 50, 1319-1325.	0.6	7
50	Effect of Cr2O3 and WO3 Addition on Pore Formation and Microstructure in Iron Foam. ISIJ International, 2010, 50, 307-313.	0.6	4
51	Nanostructured coatings produced by a novel ultrasonic-assisted method: Coating characterisation and formation mechanism. Surface and Coatings Technology, 2010, 204, 2215-2222.	2.2	25
52	Fabrication of nanostructured Mo coatings on Al and Ti substrates by ball impact cladding. Surface and Coatings Technology, 2010, 205, 2313-2321.	2.2	23
53	Development of dispersed-type sonophotocatalytic process using piezoelectric effect caused by ultrasonic resonance. Ultrasonics Sonochemistry, 2010, 17, 884-891.	3.8	10
54	Effect of annealing treatment on the structure and properties of the nanograined TiN coatings produced by ultrasonic-based coating process. Journal of Alloys and Compounds, 2010, 495, 625-628.	2.8	6

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55	Numerical and Experimental Investigation on Heat Propagation Through Composite Sinter Bed With Non-Uniform Voidage: Part II Prediction of Process Efficiency. Journal of Iron and Steel Research International, 2010, 17, 1-6.	1.4	1
56	Numerical and Experimental Investigation on Heat Propagation Through Composite Sinter Bed With Non-Uniform Voidage: Part I Mathematical Model and Its Experimental Verification. Journal of Iron and Steel Research International, 2010, 17, 1-7.	1.4	9
57	A New Drying Process of Dusts and Sludge by Employing Heat Storage Materials. ISIJ International, 2010, 50, 1282-1290.	0.6	4
58	Heat Transfer Analysis of the Mosaic Embedding Iron Ore Sintering (MEBIOS) Process. ISIJ International, 2009, 49, 681-686.	0.6	19
59	Reduction Behavior of Hematite Composite Containing Polyethylene and Graphite with Different Structures with Increasing Temperature. ISIJ International, 2009, 49, 809-814.	0.6	17
60	Effect of the Sonophotocatalytic Pretreatment on the Volume Reduction of Sewage Sludge and Enhanced Recovery of Methane and Phosphorus. Journal of Environmental Engineering, ASCE, 2009, 135, 1399-1405.	0.7	0
61	Fabrication of TiN coatings using mechanical milling techniques. International Journal of Refractory Metals and Hard Materials, 2009, 27, 492-497.	1.7	47
62	TEM study of TiN coatings fabricated by mechanical milling using vibration technique. Surface and Coatings Technology, 2009, 203, 1879-1884.	2.2	30
63	Ball impact cladding of metals with dissimilar metallic foils. Surface and Coatings Technology, 2009, 204, 125-130.	2.2	19
64	Structural evolution of the Ti–Al coatings produced by mechanical alloying technique. Journal of Alloys and Compounds, 2009, 483, 386-388.	2.8	26
65	Effect of process parameters on the formation of Ti–Al coatings fabricated by mechanical milling. Journal of Alloys and Compounds, 2009, 484, 665-673.	2.8	45
66	Analysis of Granules Behavior in Continuous Drum Mixer by DEM. ISIJ International, 2009, 49, 645-649.	0.6	31
67	Lowering Reduction Temperature of Iron Ore and Carbon Composite by Using Ores with High Combined Water Content. ISIJ International, 2009, 49, 1686-1693.	0.6	29
68	Effect of temperature on deposition of LaPO4 coatings produced by ultrasonic-based coating process on steel substrates. Surface and Coatings Technology, 2008, 202, 4285-4290.	2.2	22
69	Production of LaPO4 coatings using a novel ultrasonically-assisted plating technique. Surface and Coatings Technology, 2008, 202, 5180-5184.	2.2	11
70	Vaporization behavior of lead from the FeO-CaO-SiO2-Al2O3 slag system. International Journal of Minerals, Metallurgy, and Materials, 2008, 15, 671-677.	0.2	4
71	Formation of Hexachlorobenzene from Dusts of an Electric Arc Furnace Used in Steelmaking: Effect of Temperature and Dust Composition. Environmental Science & Technology, 2008, 42, 7459-7463.	4.6	6
72	Reduction in Dioxin Emissions by the Addition of Urea as Aqueous Solution to High-temperature Combustion Gas. ISIJ International, 2008, 48, 1305-1310.	0.6	32

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73	Development of a novel method for mechanical plating using ultrasonic vibrations. Surface and Coatings Technology, 2007, 201, 6999-7006.	2.2	36
74	Improvement in sonochemical degradation of 4-chlorophenol by combined use of Fenton-like reagents. Ultrasonics Sonochemistry, 2007, 14, 201-207.	3.8	99
75	Recent trends in the decomposition of chlorinated aromatic hydrocarbons by ultrasound irradiation and Fenton's reagent. Journal of Material Cycles and Waste Management, 2007, 9, 47-55.	1.6	49
76	Effect of Inlet Gas Composition on Dioxins Emission in the Iron Ore Sintering Process. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2006, 92, 417-426.	0.1	3
77	Carbothermic Reduction of the Composite Pellet of Iron Ore and Coal in the Packed Bed with Air Flow. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2006, 92, 809-814.	0.1	1
78	Properties of Dust Particles Sampled from Windboxes of an Iron Ore Sintering Plant: Surface Structures of Unburned Carbon. ISIJ International, 2006, 46, 1020-1026.	0.6	21
79	Vapor Pressure of Zinc and Zinc Chloride in the Fe _t O-CaO-SiO ₂ -Al ₂ O ₃ Slag System. Materials Transactions, 2006, 47, 1341-1346.	0.4	3
80	Design of Bed Structure Aiming the Control of Void Structure Formed in the Sinter Cake. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2006, 92, 788-793.	0.1	4
81	Numerical Simulation Model for Granulation Kinetics of Iron Ores Based on Discrete Element Method. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2006, 92, 742-747.	0.1	9
82	Observation of Behavior of Dioxins and Some Relating Elements in Iron Ore Sintering Bed by Quenching Pot Test. ISIJ International, 2005, 45, 609-617.	0.6	31
83	Vapor Pressure Measurements for Metal Chloride Systems by the Knudsen Effusion Method. Materials Transactions, 2005, 46, 1348-1353.	0.4	5
84	Functional Forms of Carbon and Chlorine in Dust Samples Formed in the Sintering Process of Iron Ores. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2005, 91, 751-756.	0.1	13
85	Vaporization Behavior of Zinc from the FeO-CaO-SiO2-Al2O3 Slag System. ISIJ International, 2005, 45, 1813-1819.	0.6	8
86	Numerical Simulation Model for Granulation Kinetics of Iron Ores. ISIJ International, 2005, 45, 500-505.	0.6	19
87	Design of Bed Structure Aiming the Control of Void Structure Formed in the Sinter Cake. ISIJ International, 2005, 45, 538-543.	0.6	66
88	Preface to the Special Issue on "Recent Progress of the Research on the Iron Ore Agglomeration Process― ISIJ International, 2005, 45, 413-413.	0.6	2
89	Material Flow of Fluorine in Steel and Chemical Industries. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2005, 91, 199-205.	0.1	1
90	Effect of Nitrogen-Containing Compounds on Polychlorinated Dibenzo-p-dioxin/Dibenzofuran Formation through de Novo Synthesis. Environmental Science & Technology, 2005, 39, 795-799.	4.6	42

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91	VAPOR PRESSURES AND ENTHALPIES OF SUBLIMATION OF 17 POLYCHLORINATED DIBENZO-p-DIOXINS AND FIVE POLYCHLORINATED DIBENZOFURANS. Environmental Toxicology and Chemistry, 2004, 23, 348.	2.2	42
92	Effect of Chlorine on the Vaporization Behavior of Zinc and Lead during High Temperature Treatment of Dust and Fly Ash. ISIJ International, 2004, 44, 1457-1468.	0.6	33
93	Influence of Metallic Chlorides on the Formation of PCDD/Fs during Low-Temperature Oxidation of Carbon. Environmental Science & amp; Technology, 2003, 37, 2431-2435.	4.6	74
94	Formation behavior of PCDD/Fs in PVC pyrolysis with copper oxide. Chemosphere, 2003, 50, 1235-1242.	4.2	28
95	Formation of PCDD/Fs during Oxidation of Carbonaceous Materials at Low Temperatures. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2003, 89, 811-818.	0.1	9
96	Vapour Pressure Determination for Dibenzo-p-dioxin, Dibenzofuran, Octachlorodibenzo-p-dioxin and Octachlorodibenzofuran Using a Knudsen Effusion Method. Materials Transactions, 2002, 43, 2903-2907.	0.4	23
97	Behavior of dioxin during thermal remediation in the zone combustion process. Chemosphere, 2002, 47, 687-693.	4.2	22
98	Formation and transport of PCDD/Fs in the packed bed of soil containing organic chloride during a thermal remediation process. Chemosphere, 2002, 49, 217-224.	4.2	7
99	Effect of Properties of Solid Fuel on Dioxin Concentration of the Exhaust Gas in the Iron Ore Sintering Process. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2002, 88, 378-385.	0.1	8
100	Promoter Material and Inhibitor Material for Dioxins Formation in Sintering Process. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2002, 88, 370-377.	0.1	7
101	Behavior of Trace Chlorine in Sintering Bed and Its Effect on Dioxins Concentration in Exhaust Gas of Iron Ore Sintering. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2002, 88, 59-65.	0.1	10
102	Behavior of Dioxins in the Sintering Process of Iron Ores. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2001, 87, 228-237.	0.1	10
103	Measurement of Thermodynamic Functions of Solid Phase for DD, DF, OCDD and OCDF, and Estimation of Thermodynamic Functions of Gas Phase for PCDD/Fs Using Molecular Orbital Method with Density Functional Theory. Materials Transactions, 2001, 42, 2531-2536.	0.4	4
104	Effects of Slag Compositions on the Rate of Methane-Steam Reaction ISIJ International, 2001, 41, 111-115.	0.6	65
105	Feasibility of Rotary Cup Atomizer for Slag Granulation ISIJ International, 2001, 41, 1423-1428.	0.6	118
106	Effect of Additives on the Dioxins Emissions in the Iron Ore Sintering Process ISIJ International, 2001, 41, 93-97.	0.6	37
107	Macroscopic Behaviors of Dioxins in the Iron Ore Sintering Plants ISIJ International, 2001, 41, 86-92.	0.6	37
108	Influence of Properties of Fluxing Materials on the Flow of Melt Formed in the Sintering Process ISIJ International, 2000, 40, 857-862.	0.6	54

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109	Thermodynamic Analysis of Thermochemical Recovery of High Temperature Wastes ISIJ International, 2000, 40, 286-291.	0.6	69
110	Carbothermic Reduction in the Combustion Bed Packed with Composite Pellets of Iron Oxide and Coal ISIJ International, 2000, 40, 842-849.	0.6	21
111	Numerical Simulation Model of the Iron Ore Sintering Process Directly Describing the Agglomeration Phenomenon of Granules in the Packed Bed ISIJ International, 2000, 40, 448-454.	0.6	65
112	Observation of Molten Slag Surface under Gas Impingement by X-ray Computed Tomography ISIJ International, 2000, 40, 958-963.	0.6	16
113	Thermal remediation of PCDD/Fs contaminated soil by zone combustion process. Chemosphere, 2000, 41, 857-864.	4.2	31
114	Influence of Iron Ore Properties on the Flow of Melt Formed in the Sintering Process. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 2000, 86, 139-145.	0.1	11
115	Remediation Technologies of Ash and Soil Contaminated by Dioxins and Relating Hazardous Compounds ISIJ International, 2000, 40, 266-274.	0.6	1
116	Effect of Raw Materials Bed Segregation on the Structural Change of Iron Ore Sintering Bed ISIJ International, 1999, 39, 396-398.	0.6	12
117	Thermodynamic properties of oxygen in RE–O (RE=Gd, Tb, Dy, Er) solid solutions. Journal of Alloys and Compounds, 1998, 279, 184-191.	2.8	58
118	Rate of Methane-steam Reforming Reaction on the Surface of Molten BF Slag. For Heat Recovery from Molten Slag by Using a Chemical Reaction ISIJ International, 1997, 37, 1031-1036.	0.6	92
119	Direct Evidence of Electronically Mediated Reaction during TiCl ₄ Reduction by Magnesium. Nippon Kinzoku Gakkaishi/Journal of the Japan Institute of Metals, 1997, 61, 602-609.	0.2	13
120	Phase Equilibria and Reaction Pathways during TiCl ₄ Reduction by Magnesium and Sodium Involving Electronically Mediated Reaction. Nippon Kinzoku Gakkaishi/Journal of the Japan Institute of Metals, 1997, 61, 610-618.	0.2	11
121	Thermal Analyses of the Sintering Reactions of Iron Ores. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1997, 83, 539-544.	0.1	10
122	Mechanochemical changes in gypsum when dry ground with hydrated minerals. Powder Technology, 1996, 87, 67-71.	2.1	27
123	Differential Thermal Analysis of Assimilation and Melt-formation Phenomena in the Sintering Process of Iron Ores ISIJ International, 1996, 36, 1109-1111.	0.6	26
124	Suppression of the Formation of Large Pores in the Assimilated Parts of Sinter Produced Using Pisolitic Ores ISIJ International, 1996, 36, 1338-1343.	0.6	19
125	Effect of Mixed-grinding on Reduction Process of Carbonaceous Material and Iron Oxide Composite ISIJ International, 1995, 35, 1444-1451.	0.6	44
126	Elimination Reaction of NO Gas Generated from Coke Combustion in Iron Ore Sinter Bed. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1994, 80, 276-281.	0.1	23

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127	Mechano-chemical changes in natural and synthetic zeolites by dry grinding using a planetary ball mill. Advanced Powder Technology, 1994, 5, 189-203.	2.0	8
128	Effect of dry grinding on ion-exchange characteristics of synthetic mordenite. Advanced Powder Technology, 1994, 5, 289-296.	2.0	6
129	Grinding of ep dust and its effect on solubility of metal compounds in water Journal of Chemical Engineering of Japan, 1994, 27, 492-497.	0.3	13
130	Effect of water content on grindability of dolomite and its structural change Journal of Chemical Engineering of Japan, 1994, 27, 279-283.	0.3	3
131	Fossil Energy. Reduction of Nitrogen Oxides Emission from the Iron Ore Sintering Process by Optimizing the Structure of Carbonaceous Fuels Kagaku Kogaku Ronbunshu, 1994, 20, 857-864.	0.1	6
132	Effects of moisture on grinding of natural calcite by a tumbling ball mill. Advanced Powder Technology, 1993, 4, 311-319.	2.0	9
133	Effect of Dry Mixed Grinding of Talc, Kaolinite and Gibbsite on Preparation of Cordierite Ceramics Journal of Chemical Engineering of Japan, 1993, 26, 565-569.	0.3	8
134	Effect of Mixed Grinding of Powders on Superconducting Properties of YBa2Cu3O7- Ceramics Journal of Chemical Engineering of Japan, 1993, 26, 627-632.	0.3	1
135	FORMATION OF MULLITE FROM GROUND PRODUCT OF A KAOLINITE-ALUMINUM TRIHYDROXIDE MIXTURE BY SOLID PHASE REACTION. Particulate Science and Technology, 1993, 11, 157-164.	1.1	0
136	Subjects on the Evaluation of the Process and Products in the Sintering of Iron Ores. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1993, 79, 1217-1223.	0.1	9
137	Mechanism of the Formation of Large Pore in the Assimilated Part in the Sintering Process of Pisolite Ore. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1993, 79, 1145-1150.	0.1	4
138	Combustion Rate and NO Emission during Combustion of Coke Granules in Packed Beds. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1992, 78, 1005-1012.	0.1	22
139	Effect of Mixed Grinding of Kaolinite-Gibbsite Mixture on Formation of Mullite Shigen-to-Sozai, 1992, 108, 221-226.	0.1	9
140	Factors governing the strength of agglomerated granules after sintering ISIJ International, 1991, 31, 17-23.	0.6	35
141	Permeation Characteristics and Void Structure of Iron Ore Sinter Cake ISIJ International, 1991, 31, 1286-1291.	0.6	13
142	Influence of Property of Iron Ores on the Coalescing Phenomenon of Granules during Sintering. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1991, 77, 56-62.	0.1	15
143	Properties of Sinter Produced from Mini-Pellets Consisting of Coarse Iron Ore Particles and Adhering Fine Mixtures Having Quaternaly Calcium Ferrite Composition. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1990, 76, 683-690.	0.1	0
144	The effect of raw mixture properties on bed permeability during sintering ISIJ International, 1989, 29, 33-42.	0.6	40

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145	Structural Analysis of the Void in Iron Ore Sinter Cake. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1989, 75, 228-234.	0.1	1
146	An analysis of the structure of iron ore sinter cake ISIJ International, 1989, 29, 635-641.	0.6	18
147	Combustion Rate of Coke at Different Existing States Prepared by Fine Alumina. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1986, 72, 1537-1544.	0.1	14
148	Fundamental Study on the Sintering Process Using Duplex Mini-pellets. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1984, 70, 520-526.	0.1	10
149	Mathematical Modeling of Sintering Process Considering Influence of Changes in Void Fraction and Apparent Particle Size in the Bed on Pressure Drop. Tetsu-To-Hagane/Journal of the Iron and Steel Institute of Japan, 1984, 70, 1567-1574.	0.1	13