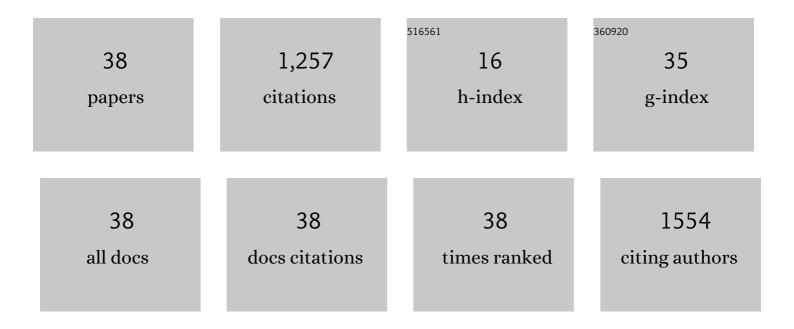
Elisabetta Petrucci

List of Publications by Year in descending order

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FUSABETTA DETDUCCI

#	Article	IF	CITATIONS
1	Hexavalent chromium reduction in contaminated soil: A comparison between ferrous sulphate and nanoscale zero-valent iron. Journal of Hazardous Materials, 2015, 281, 70-76.	6.5	179
2	An experimental comparison of a graphite electrode and a gas diffusion electrode for the cathodic production of hydrogen peroxide. Journal of Applied Electrochemistry, 2005, 35, 413-419.	1.5	139
3	Recycled fibers in reinforced concrete: A systematic literature review. Journal of Cleaner Production, 2020, 248, 119207.	4.6	136
4	On the ability to electrogenerate hydrogen peroxide and to regenerate ferrous ions of three selected carbon-based cathodes for electro-Fenton processes. Chemical Engineering Journal, 2016, 283, 750-758.	6.6	122
5	Treatment of industrial landfill leachate by means of evaporation and reverse osmosis. Waste Management, 2002, 22, 951-955.	3.7	105
6	Pyrolysis wastewater treatment by adsorption on biochars produced by poplar biomass. Journal of Environmental Management, 2017, 197, 231-238.	3.8	66
7	Electrochemical treatment of Remazol Brilliant Blue on a boron-doped diamond electrode. Chemical Engineering Journal, 2009, 153, 138-144.	6.6	53
8	UV-assisted electrochemical degradation of coumarin on boron-doped diamond electrodes. Chemical Engineering Journal, 2017, 323, 512-519.	6.6	48
9	Oxidation efficiency in the electro-Fenton process. Journal of Applied Electrochemistry, 2005, 35, 391-398.	1.5	43
10	Anodic oxidation of a simulated effluent containing Reactive Blue 19 on a boron-doped diamond electrode. Chemical Engineering Journal, 2011, 174, 612-618.	6.6	41
11	Treatment of diazo dye Reactive Green 19 by anodic oxidation on a boron-doped diamond electrode. Journal of Industrial and Engineering Chemistry, 2015, 26, 116-121.	2.9	38
12	Electrogeneration of hydrogen peroxide in seawater and application to disinfection. Journal of Applied Electrochemistry, 2008, 38, 997-1003.	1.5	33
13	Enhanced degradation of paracetamol by combining UV with electrogenerated hydrogen peroxide and ozone. Journal of Water Process Engineering, 2020, 34, 101102.	2.6	28
14	Electrochemically assisted decomposition of ozone for degradation and mineralization of Diuron. Electrochimica Acta, 2020, 331, 135423.	2.6	28
15	Anodic, cathodic and combined treatments for the electrochemical oxidation of an effluent from the flame retardant industry. Journal of Applied Electrochemistry, 2008, 38, 947-954.	1.5	22
16	Influence of surface roughening of Titanium substrate in the electrochemical activity of Manganese oxide thin film electrode in anodic oxidation of dye-containing solutions. Journal of Applied Electrochemistry, 2015, 45, 787-797.	1.5	18
17	FTâ€Raman spectroscopy for quantitative analysis of salt efflorescences. Journal of Raman Spectroscopy, 2012, 43, 1560-1566.	1.2	17
18	Biocides electrogeneration for a zero-reagent on board disinfection of ballast water. Journal of Applied Electrochemistry, 2013, 43, 237-244.	1.5	15

ELISABETTA PETRUCCI

#	Article	IF	CITATIONS
19	Degradation of chloramphenicol in water by oxidation on a boron-doped diamond electrode under UV irradiation. Journal of Water Process Engineering, 2021, 41, 101995.	2.6	14
20	Treatment of the solution extracted from metal contaminated soils by reverse osmosis and chemical precipitation. Annali Di Chimica, 2003, 93, 1005-11.	0.6	14
21	Experimental study of the remediation of atrazine contaminated soils through soil extraction and subsequent peroxidation. Journal of Hazardous Materials, 2003, 99, 265-276.	6.5	11
22	Practical Aspects on Electrochemical Disinfection of Urban and Domestic Wastewater. , 2018, , 421-447.		11
23	Sequential extraction analysis provides decision-making tools for the use of contaminated sediments. Chemistry and Ecology, 2011, 27, 107-118.	0.6	10
24	Modeling and optimization of Reactive Green 19 oxidation on a BDD thin-film electrode. Journal of the Taiwan Institute of Chemical Engineers, 2015, 51, 152-158.	2.7	10
25	Micro- and nanostructured TiO2 substrate: Influence on the electrocatalytic properties of manganese oxide based electrodes. Journal of Electroanalytical Chemistry, 2018, 808, 380-386.	1.9	8
26	Effect of Spin Coating Parameters on the Electrochemical Properties of Ruthenium Oxide Thin Films. Electrochem, 2021, 2, 83-94.	1.7	8
27	Shortcut Biological Nitrogen Removal (SBNR) in an MFC Anode Chamber Under Microaerobic Conditions: The Effect of C/N Ratio and Kinetic Study. Sustainability, 2018, 10, 1062.	1.6	7
28	Fenton-type treatment: state of the art. Annali Di Chimica, 2003, 93, 761-70.	0.6	6
29	Mixed Oxide Electrodes Based on Ruthenium and Copper: Electrochemical Properties as a Function of the Composition and Method of Manufacture. Metals, 2022, 12, 316.	1.0	6
30	Environmental Effects of Using Chelating Agents in Polluted Sediment Remediation. Bulletin of Environmental Contamination and Toxicology, 2015, 94, 340-344.	1.3	5
31	Ozone-based electrochemical advanced oxidation processes. Current Opinion in Electrochemistry, 2022, 34, 101017.	2.5	5
32	Oxidation of phosphorus compounds by Fenton's reagent. Annali Di Chimica, 2003, 93, 935-45.	0.6	3
33	Sequential use of Fenton and electro-Fenton process for the oxidation of an effluent-containing hypophosphite and phosphite. Desalination and Water Treatment, 2013, , 1-9.	1.0	2
34	Treatment and recovery of contaminated railway ballast. Turkish Journal of Engineering and Environmental Sciences, 2014, 38, 248-255.	0.1	2
35	Tetrachloroethene recovery and hazard reduction of spent powders from dry cleaning process. Waste Management and Research, 2015, 33, 339-344.	2.2	2
36	Polymerisation occurrence in the anodic oxidation of phosphite on a boron-doped diamond electrode. Electrochimica Acta, 2008, 53, 4952-4957.	2.6	1

#	Article	IF	CITATIONS
37	Use of a standard system to evaluate the matrix effect on the treatment of a solution from atrazine contaminated soils. Annali Di Chimica, 2003, 93, 997-1004.	0.6	1
38	Experimental assessment of electrochemical processes in the remediation of atrazine contaminated soils. Annali Di Chimica, 2002, 92, 1007-13.	0.6	0