Dennis Jung

List of Publications by Year in Descending Order

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Version: 2024-04-28

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The third column is the impact factor (IF) of the journal, and the fourth column is the number of citations of the article.

14 273 9 14 g-index

14 389 4.14 ext. papers ext. citations avg, IF L-index

#	Paper	IF	Citations
14	Metal oxide-doped activated carbons from bakery waste and coffee grounds for application in supercapacitors. <i>Materials Science for Energy Technologies</i> , 2021 , 4, 69-80	5.2	6
13	Structural Effects of Cellulose on Hydrolysis and Carbonization Behavior during Hydrothermal Treatment. <i>ACS Omega</i> , 2020 , 5, 12210-12223	3.9	21
12	Calculating the Reaction Order and Activation Energy for the Hydrothermal Carbonization of Fructose. <i>Chemie-Ingenieur-Technik</i> , 2020 , 92, 692-700	0.8	7
11	Bio-Based Carbon Materials from Potato Waste as Electrode Materials in Supercapacitors. <i>Energies</i> , 2020 , 13, 2406	3.1	7
10	Acid-assisted extraction and hydrolysis of inulin from chicory roots to obtain fructose-enriched extracts. <i>Biomass Conversion and Biorefinery</i> , 2020 , 1	2.3	3
9	Understanding the influence of biomass particle size and reaction medium on the formation pathways of hydrochar. <i>Biomass Conversion and Biorefinery</i> , 2020 , 10, 1357-1380	2.3	16
8	Kinetic study on the impact of acidity and acid concentration on the formation of 5-hydroxymethylfurfural (HMF), humins, and levulinic acid in the hydrothermal conversion of fructose. <i>Biomass Conversion and Biorefinery</i> , 2019 , 11, 1155	2.3	18
7	Hydrothermal carbonization of biogas digestate: Effect of digestate origin and process conditions. <i>Waste Management</i> , 2019 , 100, 138-150	8.6	36
6	Conductive Carbon Materials from the Hydrothermal Carbonization of Vineyard Residues for the Application in Electrochemical Double-Layer Capacitors (EDLCs) and Direct Carbon Fuel Cells (DCFCs). <i>Materials</i> , 2019 , 12,	3.5	18
5	Influence of the pH Value on the Hydrothermal Degradation of Fructose. ChemistryOpen, 2019, 8, 1109	-121320	15
4	The effect of different Brlisted acids on the hydrothermal conversion of fructose to HMF. <i>Green Chemistry</i> , 2018 , 20, 2231-2241	10	54
3	Hydrothermal Carbonization of Fructose: Growth Mechanism and Kinetic Model. <i>ACS Sustainable Chemistry and Engineering</i> , 2018 , 6, 13877-13887	8.3	50
2	Evaluation of Arrhenius-type overall kinetic equations for hydrothermal carbonization. <i>Journal of Analytical and Applied Pyrolysis</i> , 2017 , 127, 286-291	6	18
1	Hydrothermal carbonization of fructosellffect of salts and reactor stirring on the growth and formation of carbon spheres. <i>Biomass Conversion and Biorefinery</i> ,1	2.3	4