Galina Y Simenyuk

List of Publications by Year
in descending order

31	58	1684188 5	1720034 7
papers	citations	h-index	g-index

31	31	31	70
all docs	docs citations	times ranked	citing authors

#	Article	IF	Citations
1	Morphology and Electrical Capacitance Characteristics of Nanostructured MnxOy/MWCNT Composites. Inorganic Materials, 2021, 57, 487-497.	0.8	1
2	EFFECT OF STRUCTURE AND SURFACE STATE OF NITROGEN DOPED CARBON NANOTUBES ON THEIR FUNCTIONAL AND CATALYTIC PROPERTIES. Journal of Structural Chemistry, 2021, 62, 771-781.	1.0	4
3	Multiwalled Carbon Nanotubes: Matrix Nanostructured Composites as Electrode Materials for Supercapacitors. Energy Technology, 2021, 9, 2100449.	3.8	3
4	ÐÐÐžÐ¡Ð¢ÐÐ£ÐšÐ¢Ð£ÐÐ¯ÐÐžÐ'ÐÐÐ«Ð• ÐšÐžÐœÐŸÐžÐ—Ð¯Ð¢Ð« AU/ ÐœÐ£ÐÐ¢ Ð¯ PTÐœÐ•/ÐœÐ£ÐÐ¢ (ÐœÐ•=FE ÐõCO) Ð"Ð›Ð		
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6	STUDYING THE INFLUENCE OF HEALTH-IMPROVING TRAINING WITH DIFFERENT INTENSITY ON THE PSYCHOPHYSICAL STATE OF MIDDLE-AGED WOMEN OF 35â€“45 YEARS OLD. Siberian Journal of Life Sciences and Agriculture, 2021, 13, 245-265.	0.3	0
7	Nanostructured Composites MWCNT/transition metal oxide obtained by thermal decomposition of hydroxides. Chemistry for Sustainable Development, 2020, , .	0.1	0
8	Investigation of the Structural Features and Capacitive Parameters of Carbon Materials Based on Carbonized Rice Husk. Chemistry for Sustainable Development, 2020, , .	0.1	0
9	Morphology and Electrical Capacity Properties of Nanostructured Composites PtM/Multi-Walled Carbon Nanotubes (M = Fe, Co). Chemistry for Sustainable Development, 2020, , .	0.1	0
10	Electrochemical Properties of Coke-Derived Graphene Oxide Reduced by Ascorbic Acid. Coke and Chemistry, 2019, 62, 353-358.	0.4	0
11	Morphology and Electrochemical Properties of Nanostructured Composite Co2/MWCNT Based on Carbon Nanotubes. Chemistry for Sustainable Development, 2019, , .	0.1	2
12	Electrode Material for Supercapacitors Based on Carbon/Nickel Cobaltate Nanocomposite Synthesized by the Thermal Decomposition of Cobalt and Nickel Azides. Chemistry for Sustainable Development, 2019, , .	0.1	1
13	Influence of the Conditions for Obtaining Nanocomposite Electrode Materials Mny/MWCNT on their electrocapacity characteristics. Chemistry for Sustainable Development, 2019, , .	0.1	1
14	Synthesis of a Carbon/NiCo2O4 Electrode Material for a Supercapacitor by Thermal Decomposition of Mixed Cobaltâ€“Nickel Hydroxides. Chemistry for Sustainable Development, 2018, , .	0.1	1
15	Hybrid Electrode Materials for Supercapacitors Based on Nanostructured Carbon Matrix Composites Filled with Chromium Oxides and Hydroxides. Chemistry for Sustainable Development, 2018, , .	0.1	0
16	Nanostructured Composites Based on Highly Porous Carbon Matrixes Filled with Cobalt and Nickel Hydroxides. Chemistry for Sustainable Development, 2018, , .	0.1	0
17	Development of a Technique and Investigation of Capacitance Characteristics of Electrode Materials for Supercapacitors Based on Nitrogen-Doped Carbon Nanotubes. Eurasian Chemico-Technological Journal, 2017, 19, 201.	0.6	2
18	New Method for Preparation of Nanostructured Composites Based on Porous Carbon Materials to Use as Supercapacitor Electrodes. Chemistry for Sustainable Development, 2017, , .	0.1	0

#	Article	IF	Citations
19	Mesoporous Carbon Matrix-Based MnxOy/C Hybrid Electrode Materials for Asymmetric Supercapitors. Chemistry for Sustainable Development, 2017, , .	0.1	0
20	Nanostructured composites "porous carbon matrices - products of thermolysis Co(N3)2". Chemistry for Sustainable Development, 2017, , .	0.1	0
21	Ultrasonic Assisted Fabrication of Nanocomposite Electrode Materials Au/C for Low-Voltage Electronics. Materials and Manufacturing Processes, 2016, 31, 739-744.	4.7	9
22	Highly porous carbon materials filled with gold and manganese oxide nanoparticles for electrochemical use. Catalysis Today, 2015, 249, 220-227.	4.4	11
23	Nanostructured composites based on highly porous carbon matrices filled with gold. Nanotechnologies in Russia, 2015, 10, 388-399.	0.7	1
24	Bimetallic catalysts for the hydrogenation of aromatic nitro compounds. Solid Fuel Chemistry, 2012, 46, 364-367.	0.7	6
25	Preparation of nanosized copper powders with controlled dispersity. Russian Journal of Applied Chemistry, 2011, 84, 912-915.	0.5	2
26	Effect of stabilizers on the tolerance of copper nanopowders for oxidation by molecular oxygen. Russian Journal of Applied Chemistry, 2010, 83, 345-348.	0.5	0
27	Effect of various factors on the dispersity of copper nanopowders produced by reduction of copper salts with glycerol. Russian Journal of Applied Chemistry, 2009, 82, 981-985.	0.5	6
28	Preparation of ultradisperse copper powders by reduction of copper salts with L-ascorbic acid and electrically conducting formulations based on these powders. Russian Journal of Applied Chemistry, 2006, 79, 707-710.	0.5	1
29	Effect of the nature of a reducing agent on properties of ultradisperse copper powders. Russian Journal of Applied Chemistry, 2006, 79, 1605-1608.	0.5	7
30	Electrically Conducting Formulations Based on Ultradispersed Powders of Copper, Obtained by Reduction of Its Salts with the Hypophosphite Ion. Russian Journal of Applied Chemistry, 2004, 77, 380-384.	0.5	0
31	Title is missing!. Russian Journal of Applied Chemistry, 2002, 75, 1736-1739.	0.5	0