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## List of Publications by Year in descending order

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Version: 2024-02-01

11  
papers

112  
citations

1478505

6  
h-index

1281871

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times ranked

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#	ARTICLE	IF	CITATIONS
1	Isopiestic determination of osmotic coefficients in the ionic strength range $I = (0.9670 \pm 2.2160) \text{ mol} \cdot \text{kg}^{-1}$ and activity coefficients determined by electromotive force measurements in the range $I_m = (0.0897 \pm 1.0054) \text{ mol} \cdot \text{kg}^{-1}$ of the $\{\text{yKCl} + (1 - \text{y}) \text{K}_2\text{HPO}_4\}(\text{aq})$ system at $T = 298.15 \text{ K}$ . Journal of Molecular Liquids, 2022, 353, 118767.	4.9	3
2	Isopiestic determination of the osmotic and activity coefficients of $\{\text{yK}_2\text{HPO}_4 + (1 - \text{y})\text{KH}_2\text{PO}_4\}(\text{aq})$ at $T = 298.15 \text{ K}$ . Journal of Chemical Thermodynamics, 2020, 142, 105945.	2.0	4
3	Isopiestic Determination of Osmotic and Activity Coefficients of the $\{\text{yNaH}_2\text{PO}_4 + (1 - \text{y})\text{Na}_2\text{HPO}_4\}(\text{aq})$ System at $T = 298.15 \text{ K}$ . Journal of Chemical & Engineering Data, 2020, 65, 5137-5153.	1.9	5
4	Isopiestic Determination of the Osmotic and Activity Coefficients of the $\{\text{yNaH}_2\text{PO}_4 + (1 - \text{y})\text{KH}_2\text{PO}_4\}(\text{aq})$ System at $T = 298.15 \text{ K}$ . Journal of Solution Chemistry, 2019, 48, 296-328.	1.2	7
5	Isopiestic determination of the osmotic and activity coefficients of the $\{\text{yMg}(\text{NO}_3)_2 + (1 - \text{y})\text{MgSO}_4\}(\text{aq})$ system at $T = 298.15 \text{ K}$ . Journal of Chemical Thermodynamics, 2017, 113, 91-103.	2.0	3
6	Isopiestic Determination of the Osmotic and Activity Coefficients of the $\{\text{yNa}_2\text{HPO}_4 + (1 - \text{y})\text{K}_2\text{HPO}_4\}(\text{aq})$ System at $T = 298.15 \text{ K}$ . Journal of Solution Chemistry, 2016, 45, 1261-1287.	1.2	7
7	Isopiestic determination of the osmotic and activity coefficients of the $\{\text{yK}_2\text{SO}_4 + (1 - \text{y})\text{K}_2\text{HPO}_4\}(\text{aq})$ system at $T = 298.15 \text{ K}$ . Journal of Chemical Thermodynamics, 2014, 79, 84-93.	2.0	11
8	Isopiestic determination of the osmotic and activity coefficients of the $\{\text{yKBr} + (1 - \text{y})\text{K}_2\text{HPO}_4\}(\text{aq})$ system at $T = 298.15 \text{ K}$ . Journal of Chemical Thermodynamics, 2013, 62, 151-161.	2.0	15
9	Isopiestic determination of the osmotic and activity coefficients of the $\{\text{yKNO}_3 + (1 - \text{y})\text{K}_2\text{HPO}_4\}(\text{aq})$ system at $T = 298.15 \text{ K}$ . Journal of Chemical Thermodynamics, 2012, 55, 172-183.	2.0	17
10	Isopiestic determination of the osmotic and activity coefficients of the $\{\text{yKCl} + (1 - \text{y})\text{K}_2\text{HPO}_4\}(\text{aq})$ system at $T = 298.15 \text{ K}$ . Journal of Chemical Thermodynamics, 2011, 43, 1877-1885.	2.0	19
11	Isopiestic Determination of the Osmotic and Activity Coefficients of $\text{K}_2\text{HPO}_4(\text{aq})$ , Including Saturated and Supersaturated Solutions, at $T = 298.15 \text{ K}$ . Journal of Solution Chemistry, 2011, 40, 907-920.	1.2	21