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## List of Publications by Year in descending order

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11  
papers

112  
citations

1478505

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1281871

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g-index

11  
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11  
docs citations

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times ranked

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citing authors

#	ARTICLE	IF	CITATIONS
1	Isopiestic Determination of the Osmotic and Activity Coefficients of $K_2HPO_4(aq)$ , Including Saturated and Supersaturated Solutions, at $T=298.15\text{ÅK}$ . Journal of Solution Chemistry, 2011, 40, 907-920.	1.2	21
2	Isopiestic determination of the osmotic and activity coefficients of the $\{yKCl+(1\hat{\sim}y)K_2HPO_4\}(aq)$ system at $T=298.15\text{K}$ . Journal of Chemical Thermodynamics, 2011, 43, 1877-1885.	2.0	19
3	Isopiestic determination of the osmotic and activity coefficients of the $\{yKNO_3+(1\hat{\sim}y)K_2HPO_4\}(aq)$ system at $T=298.15\text{K}$ . Journal of Chemical Thermodynamics, 2012, 55, 172-183.	2.0	17
4	Isopiestic determination of the osmotic and activity coefficients of the $\{yKBr+(1\hat{\sim}y)K_2HPO_4\}(aq)$ system at $T=298.15\text{K}$ . Journal of Chemical Thermodynamics, 2013, 62, 151-161.	2.0	15
5	Isopiestic determination of the osmotic and activity coefficients of the $\{yK_2SO_4+(1\hat{\sim}y)K_2HPO_4\}(aq)$ system at $T=298.15\text{K}$ . Journal of Chemical Thermodynamics, 2014, 79, 84-93.	2.0	11
6	Isopiestic Determination of the Osmotic and Activity Coefficients of the $\{yNa_2HPO_4\hat{+}\hat{+}(1\hat{\sim}y)K_2HPO_4\}(aq)$ System at $T\hat{=}\hat{+}298.15\text{ÅK}$ . Journal of Solution Chemistry, 2016, 45, 1261-1287.	1.2	7
7	Isopiestic Determination of the Osmotic and Activity Coefficients of the $\{yNaH_2PO_4\hat{+}\hat{+}(1\hat{\sim}y)KH_2PO_4\}(aq)$ System at $T\hat{=}\hat{+}298.15\text{ÅK}$ . Journal of Solution Chemistry, 2019, 48, 296-328.	1.2	7
8	Isopiestic Determination of Osmotic and Activity Coefficients of the $\{yNaH_2PO_4+(1\hat{\sim}y)Na_2HPO_4\}(aq)$ System at $T=298.15\text{K}$ . Journal of Chemical & Engineering Data, 2020, 65, 5137-5153.	1.9	5
9	Isopiestic determination of the osmotic and activity coefficients of $\{yK_2HPO_4\hat{+}\hat{+}(1\hat{\sim}y)KH_2PO_4\}(aq)$ at $T\hat{=}\hat{+}298.15\hat{\sim}\text{K}$ . Journal of Chemical Thermodynamics, 2020, 142, 105945.	2.0	4
10	Isopiestic determination of the osmotic and activity coefficients of the $\{yMg(NO_3)_2+(1\hat{\sim}y)MgSO_4\}(aq)$ system at $T=298.15\text{K}$ . Journal of Chemical Thermodynamics, 2017, 113, 91-103.	2.0	3
11	Isopiestic determination of osmotic coefficients in the ionic strength range $I = (0.9670\hat{\sim}2.2160)\text{mol}\hat{\sim}\text{kg}\hat{\sim}^{-1}$ and activity coefficients determined by electromotive force measurements in the range $I_m = (0.0897\hat{\sim}1.0054)\text{mol}\hat{\sim}\text{kg}\hat{\sim}^{-1}$ of the $\{yKCl+(1\hat{\sim}y)K_2HPO_4\}(aq)$ system at $T\hat{=}\hat{+}298.15\text{ÅK}$ . Journal of Molecular Liquids, 2022, 353, 118767.	4.9	3