Yong Zhang

List of Publications by Year in descending order

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471061 580395 1,967 25 17 25 h-index citations g-index papers 27 27 27 1416 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	Refined Classical Force Field for Choline Chloride and Ethylene Glycol Mixtures over Wide Composition Range. Journal of Chemical & Engineering Data, 2022, 67, 1864-1871.	1.0	19
2	Sigma profiles in deep learning: towards a universal molecular descriptor. Chemical Communications, 2022, 58, 5630-5633.	2.2	20
3	Structure of water-in-salt and water-in-bisalt electrolytes. Physical Chemistry Chemical Physics, 2022, 24, 10727-10736.	1.3	5
4	Exchange-Mediated Transport in Battery Electrolytes: Ultrafast or Ultraslow?. Journal of the American Chemical Society, 2022, 144, 8591-8604.	6.6	18
5	From Networked to Isolated: Observing Water Hydrogen Bonds in Concentrated Electrolytes with Two-Dimensional Infrared Spectroscopy. Journal of Physical Chemistry B, 2022, 126, 5305-5319.	1.2	9
6	Deep Eutectic Solvents: A Review of Fundamentals and Applications. Chemical Reviews, 2021, 121, 1232-1285.	23.0	1,334
7	Water-in-Salt LiTFSI Aqueous Electrolytes. 1. Liquid Structure from Combined Molecular Dynamics Simulation and Experimental Studies. Journal of Physical Chemistry B, 2021, 125, 4501-4513.	1.2	52
8	Solvation Dynamics of Wet Ethaline: Water is the Magic Component. Journal of Physical Chemistry B, 2021, 125, 8888-8901.	1.2	32
9	Water or Anion? Uncovering the Zn ²⁺ Solvation Environment in Mixed Zn(TFSI) ₂ and LiTFSI Water-in-Salt Electrolytes. ACS Energy Letters, 2021, 6, 3458-3463.	8.8	45
10	Functionalized Phosphonium Cations Enable Zn Metal Reversibility in Aqueous Electrolytes. ECS Meeting Abstracts, 2021, MA2021-02, 14-14.	0.0	0
11	Water-In-Salt LiTFSI Aqueous Electrolytes (2): Transport Properties and Li ⁺ Dynamics Based on Molecular Dynamics Simulations. Journal of Physical Chemistry B, 2021, 125, 13246-13254.	1.2	26
12	Effect of alkyl-group flexibility on the melting point of imidazolium-based ionic liquids. Journal of Chemical Physics, 2020, 153, 044504.	1.2	20
13	Comparison of fixed charge and polarizable models for predicting the structural, thermodynamic, and transport properties of molten alkali chlorides. Journal of Chemical Physics, 2020, 153, 214502.	1.2	19
14	Impact of anion shape on Li ⁺ solvation and on transport properties for lithium–air batteries: a molecular dynamics study. Physical Chemistry Chemical Physics, 2020, 22, 15842-15852.	1.3	19
15	Liquid Structure and Transport Properties of the Deep Eutectic Solvent Ethaline. Journal of Physical Chemistry B, 2020, 124, 5251-5264.	1.2	84
16	Layer-based thermal migration of an ionic liquid nano-droplet on a graphene surface: a molecular dynamics study. Molecular Simulation, 2020, 46, 829-836.	0.9	3
17	Melting points of alkali chlorides evaluated for a polarizable and non-polarizable model. Journal of Chemical Physics, 2020, 153, 011101.	1.2	15
18	Signatures of Ion Pairing and Aggregation in the Vibrational Spectroscopy of Super-Concentrated Aqueous Lithium Bistriflimide Solutions. Journal of Physical Chemistry C, 2020, 124, 3470-3481.	1.5	44

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19	Liquid Structure and Transport Properties of the Deep Eutectic Solvent Ethaline. ECS Meeting Abstracts, 2020, MA2020-02, 2910-2910.	0.0	0
20	PyLAT: Python LAMMPS Analysis Tools. Journal of Chemical Information and Modeling, 2019, 59, 1301-1305.	2.5	49
21	Solvation Structure and Dynamics of Li ⁺ in Ternary Ionic Liquid–Lithium Salt Electrolytes. Journal of Physical Chemistry B, 2019, 123, 516-527.	1.2	62
22	Temperature Dependence of Volumetric and Dynamic Properties of Imidazolium-Based Ionic Liquids. Journal of Physical Chemistry B, 2018, 122, 2414-2424.	1.2	19
23	Anion Enhancement at the Liquid–Vacuum Interface of an Ionic Liquid Mixture. Journal of Physical Chemistry C, 2018, 122, 27392-27401.	1.5	19
24	Evaluation and Refinement of the General AMBER Force Field for Nineteen Pure Organic Electrolyte Solvents. Journal of Chemical & Engineering Data, 2018, 63, 3488-3502.	1.0	23
25	Computational Design of New Magnesium Electrolytes with Improved Properties. Journal of Physical Chemistry C, 2017, 121, 16126-16136.	1.5	26