

# Curtis Berlinguette

## List of Publications by Year in descending order

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173  
papers

15,256  
citations

17405

63  
h-index

18606

119  
g-index

183  
all docs

183  
docs citations

183  
times ranked

14456  
citing authors

#	ARTICLE	IF	CITATIONS
1	Gas diffusion electrodes and membranes for CO <sub>2</sub> reduction electrolyzers. Nature Reviews Materials, 2022, 7, 55-64.	23.3	265
2	Selective hydrogenation of furfural using a membrane reactor. Energy and Environmental Science, 2022, 15, 215-224.	15.6	37
3	Porous metal electrodes enable efficient electrolysis of carbon capture solutions. Energy and Environmental Science, 2022, 15, 705-713.	15.6	61
4	Quantification of the Effect of an External Magnetic Field on Water Oxidation with Cobalt Oxide Anodes. Journal of the American Chemical Society, 2022, 144, 733-739.	6.6	20
5	Continuum Model to Define the Chemistry and Mass Transfer in a Bicarbonate Electrolyzer. ACS Energy Letters, 2022, 7, 834-842.	8.8	39
6	A self-driving laboratory advances the Pareto front for material properties. Nature Communications, 2022, 13, 995.	5.8	55
7	Electrocatalysts Derived from Copper Complexes Transform CO into C <sub>2+</sub> Products Effectively in a Flow Cell. Chemistry - A European Journal, 2022, 28, e202200340.	1.7	10
8	Electrolytic conversion of carbon capture solutions containing carbonic anhydrase. Journal of Inorganic Biochemistry, 2022, 231, 111782.	1.5	13
9	Flexible automation accelerates materials discovery. Nature Materials, 2022, 21, 722-726.	13.3	33
10	Electrolytic Methane Production from Reactive Carbon Solutions. ACS Energy Letters, 2022, 7, 1712-1718.	8.8	23
11	A magnetic twist on CO <sub>2</sub> electrolysis. Trends in Chemistry, 2022, 4, 465-466.	4.4	0
12	Conversion of Reactive Carbon Solutions into CO at Low Voltage and High Carbon Efficiency. ACS Central Science, 2022, 8, 749-755.	5.3	32
13	Ring walking as a regioselectivity control element in Pd-catalyzed C-N cross-coupling. Nature Communications, 2022, 13, .	5.8	11
14	A self-driving laboratory designed to accelerate the discovery of adhesive materials. , 2022, 1, 382-389.		14
15	Permeability Matters When Reducing CO <sub>2</sub> in an Electrochemical Flow Cell. ACS Energy Letters, 2022, 7, 2382-2387.	8.8	15
16	Designing anion exchange membranes for CO <sub>2</sub> electrolyzers. Nature Energy, 2021, 6, 339-348.	19.8	209
17	Electrolysis Can Be Used to Resolve Hydrogenation Pathways at Palladium Surfaces in a Membrane Reactor. JACS Au, 2021, 1, 336-343.	3.6	11
18	Physical Separation of H <sub>2</sub> Activation from Hydrogenation Chemistry Reveals the Specific Role of Secondary Metal Catalysts. Angewandte Chemie - International Edition, 2021, 60, 11937-11942.	7.2	18

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19	Physical Separation of H <sub>2</sub> Activation from Hydrogenation Chemistry Reveals the Specific Role of Secondary Metal Catalysts. <i>Angewandte Chemie</i> , 2021, 133, 12044-12049.	1.6	0
20	Impact of Alkali Cation Identity on the Conversion of HCO <sub>3</sub> <sup>-</sup> to CO in Bicarbonate Electrolyzers. <i>ChemElectroChem</i> , 2021, 8, 2094-2100.	1.7	29
21	How Catalyst Dispersion Solvents Affect CO <sub>2</sub> Electrolyzer Gas Diffusion Electrodes. <i>Energy &amp; Fuels</i> , 2021, 35, 19178-19184.	2.5	20
22	An industrial perspective on catalysts for low-temperature CO <sub>2</sub> electrolysis. <i>Nature Nanotechnology</i> , 2021, 16, 118-128.	15.6	255
23	A machine vision tool for facilitating the optimization of large-area perovskite photovoltaics. <i>Npj Computational Materials</i> , 2021, 7, .	3.5	11
24	Voltage Matters When Reducing CO <sub>2</sub> in an Electrochemical Flow Cell. <i>ACS Energy Letters</i> , 2020, 5, 215-220.	8.8	123
25	Hydrogenation without H <sub>2</sub> Using a Palladium Membrane Flow Cell. <i>Cell Reports Physical Science</i> , 2020, 1, 100105.	2.8	28
26	Quantification of water transport in a CO <sub>2</sub> electrolyzer. <i>Energy and Environmental Science</i> , 2020, 13, 5126-5134.	15.6	86
27	Sulfuric Acid Electrolyte Impacts Palladium Chemistry at Reductive Potentials. <i>Chemistry of Materials</i> , 2020, 32, 9098-9106.	3.2	5
28	Bioinspiration in light harvesting and catalysis. <i>Nature Reviews Materials</i> , 2020, 5, 828-846.	23.3	136
29	Electrolytic deuteration of unsaturated bonds without using D <sub>2</sub> . <i>Nature Catalysis</i> , 2020, 3, 719-726.	16.1	71
30	Quantifying defects in thin films using machine vision. <i>Npj Computational Materials</i> , 2020, 6, .	3.5	18
31	Conversion of Bicarbonate to Formate in an Electrochemical Flow Reactor. <i>ACS Energy Letters</i> , 2020, 5, 2624-2630.	8.8	84
32	pH Matters When Reducing CO <sub>2</sub> in an Electrochemical Flow Cell. <i>ACS Energy Letters</i> , 2020, 5, 3101-3107.	8.8	131
33	Linking gas diffusion electrode composition to CO <sub>2</sub> reduction in a flow cell. <i>Journal of Materials Chemistry A</i> , 2020, 8, 19493-19501.	5.2	54
34	Photoelectrochemical Decomposition of Lignin Model Compound on a BiVO <sub>4</sub> Photoanode. <i>ChemSusChem</i> , 2020, 13, 3622-3626.	3.6	17
35	Strain Influences the Hydrogen Evolution Activity and Absorption Capacity of Palladium. <i>Angewandte Chemie</i> , 2020, 132, 12290-12296.	1.6	9
36	Self-driving laboratory for accelerated discovery of thin-film materials. <i>Science Advances</i> , 2020, 6, eaaz8867.	4.7	306

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37	Electrodes Designed for Converting Bicarbonate into CO. ACS Energy Letters, 2020, 5, 2165-2173.	8.8	90
38	Rhenium Complexes of Pyridyl-Mesoionic Carbenes: Photochemical Properties and Electrocatalytic CO <sub>2</sub> Reduction. Inorganic Chemistry, 2020, 59, 4215-4227.	1.9	43
39	π-covalency in the halogen bond. Nature Communications, 2020, 11, 3310.	5.8	52
40	Molecular Catalysts Boost the Rate of Electrolytic CO <sub>2</sub> Reduction. ACS Energy Letters, 2020, 5, 1512-1518.	8.8	52
41	Strain Influences the Hydrogen Evolution Activity and Absorption Capacity of Palladium. Angewandte Chemie - International Edition, 2020, 59, 12192-12198.	7.2	28
42	Managing Hydration at the Cathode Enables Efficient CO <sub>2</sub> Electrolysis at Commercially Relevant Current Densities. ACS Energy Letters, 2020, 5, 1612-1618.	8.8	111
43	Defining Direct Orbital Pathways for Intermolecular Electron Transfer Using Sensitized Semiconducting Surfaces. Inorganic Chemistry, 2020, 59, 14696-14705.	1.9	2
44	Chapter 10. Electrochemical Reactors. RSC Energy and Environment Series, 2020, , 408-432.	0.2	1
45	CO <sub>2</sub> electrochemical catalytic reduction with a highly active cobalt phthalocyanine. Nature Communications, 2019, 10, 3602.	5.8	307
46	Design rules for high mobility xanthene-based hole transport materials. Chemical Science, 2019, 10, 8360-8366.	3.7	20
47	Molecular electrocatalysts can mediate fast, selective CO <sub>2</sub> reduction in a flow cell. Science, 2019, 365, 367-369.	6.0	601
48	Ligands Affect Hydrogen Absorption and Desorption by Palladium Nanoparticles. Chemistry of Materials, 2019, 31, 8679-8684.	3.2	18
49	Analytical electrolyzer enabling operando characterization of flow plates. Review of Scientific Instruments, 2019, 90, 074103.	0.6	5
50	Protocol for Quantifying the Doping of Organic Hole-Transport Materials. ACS Energy Letters, 2019, 4, 2547-2551.	8.8	23
51	Entropic Barriers Determine Adiabatic Electron Transfer Equilibrium. Journal of Physical Chemistry C, 2019, 123, 3416-3425.	1.5	8
52	Revisiting the cold case of cold fusion. Nature, 2019, 570, 45-51.	13.7	48
53	Electrolytic Conversion of Bicarbonate into CO in a Flow Cell. Joule, 2019, 3, 1487-1497.	11.7	177
54	Calorimetry under non-ideal conditions using system identification. Journal of Thermal Analysis and Calorimetry, 2019, 138, 3139-3157.	2.0	2

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55	Efficient Electrocatalytic Hydrogenation with a Palladium Membrane Reactor. <i>Journal of the American Chemical Society</i> , 2019, 141, 7815-7821.	6.6	90
56	Facets and vertices regulate hydrogen uptake and release in palladium nanocrystals. <i>Nature Materials</i> , 2019, 18, 454-458.	13.3	96
57	Strain Engineering Electrocatalysts for Selective CO <sub>2</sub> Reduction. <i>ACS Energy Letters</i> , 2019, 4, 980-986.	8.8	115
58	Supported palladium membrane reactor architecture for electrocatalytic hydrogenation. <i>Journal of Materials Chemistry A</i> , 2019, 7, 26586-26595.	5.2	26
59	Dopant-free molecular hole transport material that mediates a 20% power conversion efficiency in a perovskite solar cell. <i>Energy and Environmental Science</i> , 2019, 12, 3502-3507.	15.6	90
60	Kinetic phases of Ag-Cu alloy films are accessible through photodeposition. <i>Journal of Materials Chemistry A</i> , 2019, 7, 711-715.	5.2	12
61	Spin-coated epoxy resin embedding technique enables facile SEM/FIB thickness determination of porous metal oxide ultrathin films. <i>Journal of Microscopy</i> , 2018, 270, 302-308.	0.8	6
62	Tracking precursor degradation during the photo-induced formation of amorphous metal oxide films. <i>Journal of Materials Chemistry A</i> , 2018, 6, 4544-4549.	5.2	6
63	Photodeposited Amorphous Oxide Films for Electrochromic Windows. <i>CheM</i> , 2018, 4, 821-832.	5.8	95
64	Electrolysis of Gaseous CO <sub>2</sub> to CO in a Flow Cell with a Bipolar Membrane. <i>ACS Energy Letters</i> , 2018, 3, 149-154.	8.8	265
65	Electrolytic CO <sub>2</sub> Reduction in a Flow Cell. <i>Accounts of Chemical Research</i> , 2018, 51, 910-918.	7.6	735
66	Electrocatalytic Alloys for CO <sub>2</sub> Reduction. <i>ChemSusChem</i> , 2018, 11, 48-57.	3.6	249
67	Stabilizing Copper for CO <sub>2</sub> Reduction in Low-Grade Electrolyte. <i>Inorganic Chemistry</i> , 2018, 57, 14624-14631.	1.9	21
68	Resolving orbital pathways for intermolecular electron transfer. <i>Nature Communications</i> , 2018, 9, 4916.	5.8	19
69	Solution-Deposited Solid-State Electrochromic Windows. <i>IScience</i> , 2018, 10, 80-86.	1.9	36
70	Precise Control of Thermal and Redox Properties of Organic Hole-Transport Materials. <i>Angewandte Chemie - International Edition</i> , 2018, 57, 15529-15533.	7.2	41
71	Precise Control of Thermal and Redox Properties of Organic Hole-Transport Materials. <i>Angewandte Chemie</i> , 2018, 130, 15755-15759.	1.6	15
72	Kinetics teach that electronic coupling lowers the free-energy change that accompanies electron transfer. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2018, 115, 7248-7253.	3.3	28

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73	Correlating cobalt redox couples to photovoltage in the dye-sensitized solar cell. Dalton Transactions, 2018, 47, 11942-11952.	1.6	21
74	Optical Intramolecular Electron Transfer in Opposite Directions through the Same Bridge That Follows Different Pathways. Journal of the American Chemical Society, 2018, 140, 7176-7186.	6.6	27
75	Accurate Coulometric Quantification of Hydrogen Absorption in Palladium Nanoparticles and Thin Films. Chemistry of Materials, 2018, 30, 3963-3970.	3.2	27
76	Complete electron economy by pairing electrolysis with hydrogenation. Nature Catalysis, 2018, 1, 501-507.	16.1	148
77	Organic chemistry at anodes and photoanodes. Sustainable Energy and Fuels, 2018, 2, 1905-1927.	2.5	76
78	High-Throughput Synthesis of Mixed-Metal Electrocatalysts for CO <sub>2</sub> Reduction. Angewandte Chemie - International Edition, 2017, 56, 6068-6072.	7.2	131
79	On how electron density affects the redox stability of phenothiazine sensitizers on semiconducting surfaces. Chemical Communications, 2017, 53, 2547-2550.	2.2	8
80	High-Voltage Dye-Sensitized Solar Cells Mediated by [Co(2,2'-bipyrimidine) <sub>3</sub> ] <sup>3+</sup> . Inorganic Chemistry, 2017, 56, 2383-2386.	1.9	12
81	Comparative analysis of triarylamine and phenothiazine sensitizer donor units in dye-sensitized solar cells. Chemical Communications, 2017, 53, 2367-2370.	2.2	25
82	High-Throughput Synthesis of Mixed-Metal Electrocatalysts for CO <sub>2</sub> Reduction. Angewandte Chemie, 2017, 129, 6164-6168.	1.6	28
83	Frontispiece: High-Throughput Synthesis of Mixed-Metal Electrocatalysts for CO <sub>2</sub> Reduction. Angewandte Chemie - International Edition, 2017, 56, .	7.2	1
84	On the Electrolytic Stability of Iron-Nickel Oxides. Chem, 2017, 2, 590-597.	5.8	104
85	Photodeposited ruthenium dioxide films for oxygen evolution reaction electrocatalysis. Journal of Materials Chemistry A, 2017, 5, 1575-1580.	5.2	24
86	Rapid Quantification of Film Thickness and Metal Loading for Electrocatalytic Metal Oxide Films. Chemistry of Materials, 2017, 29, 7272-7277.	3.2	11
87	High-temperature high-pressure calorimeter for studying gram-scale heterogeneous chemical reactions. Review of Scientific Instruments, 2017, 88, 084101.	0.6	5
88	Frontispiz: High-Throughput Synthesis of Mixed-Metal Electrocatalysts for CO <sub>2</sub> Reduction. Angewandte Chemie, 2017, 129, .	1.6	0
89	Water Oxidation Catalysis: Tuning the Electrocatalytic Properties of Amorphous Lanthanum Cobaltite through Calcium Doping. ACS Catalysis, 2017, 7, 6385-6391.	5.5	18
90	Brass and Bronze as Effective CO <sub>2</sub> Reduction Electrocatalysts. Angewandte Chemie, 2017, 129, 16806-16809.	1.6	15

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91	Photodecomposition of Metal Nitrate and Chloride Compounds Yields Amorphous Metal Oxide Films. <i>Journal of the American Chemical Society</i> , 2017, 139, 18174-18177.	6.6	17
92	Brass and Bronze as Effective CO <sub>2</sub> Reduction Electrocatalysts. <i>Angewandte Chemie - International Edition</i> , 2017, 56, 16579-16582.	7.2	43
93	Electrolytic CO <sub>2</sub> Reduction in Tandem with Oxidative Organic Chemistry. <i>ACS Central Science</i> , 2017, 3, 778-783.	5.3	93
94	Spectroscopic detection of halogen bonding resolves dye regeneration in the dye-sensitized solar cell. <i>Nature Communications</i> , 2017, 8, 1761.	5.8	35
95	Photoelectrochemical oxidation of organic substrates in organic media. <i>Nature Communications</i> , 2017, 8, 390.	5.8	123
96	Evidence for Interfacial Halogen Bonding. <i>Angewandte Chemie</i> , 2016, 128, 6060-6064.	1.6	11
97	Evidence for Interfacial Halogen Bonding. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 5956-5960.	7.2	40
98	Curing BiVO <sub>4</sub> Photoanodes with Ultraviolet Light Enhances Photoelectrocatalysis. <i>Angewandte Chemie</i> , 2016, 128, 1801-1804.	1.6	94
99	Rapid prototyping of electrolyzer flow field plates. <i>Energy and Environmental Science</i> , 2016, 9, 3417-3423.	15.6	49
100	On How Experimental Conditions Affect the Electrochemical Response of Disordered Nickel Oxyhydroxide Films. <i>Chemistry of Materials</i> , 2016, 28, 5635-5642.	3.2	22
101	Halogen Bonding Promotes Higher Dye-Sensitized Solar Cell Photovoltages. <i>Journal of the American Chemical Society</i> , 2016, 138, 10406-10409.	6.6	65
102	Exposure of WO <sub>3</sub> Photoanodes to Ultraviolet Light Enhances Photoelectrochemical Water Oxidation. <i>ACS Applied Materials &amp; Interfaces</i> , 2016, 8, 25010-25013.	4.0	26
103	Electrolysis of CO <sub>2</sub> to Syngas in Bipolar Membrane-Based Electrochemical Cells. <i>ACS Energy Letters</i> , 2016, 1, 1149-1153.	8.8	235
104	Curing BiVO <sub>4</sub> Photoanodes with Ultraviolet Light Enhances Photoelectrocatalysis. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 1769-1772.	7.2	138
105	Kinetic pathway for interfacial electron transfer from a semiconductor to a molecule. <i>Nature Chemistry</i> , 2016, 8, 853-859.	6.6	96
106	Water oxidation catalysis: an amorphous quaternary Ba-Sr-Co-Fe oxide as a promising electrocatalyst for the oxygen-evolution reaction. <i>Chemical Communications</i> , 2016, 52, 1513-1516.	2.2	63
107	Accounting for the Dynamic Oxidative Behavior of Nickel Anodes. <i>Journal of the American Chemical Society</i> , 2016, 138, 1561-1567.	6.6	91
108	Near-infrared-driven decomposition of metal precursors yields amorphous electrocatalytic films. <i>Science Advances</i> , 2015, 1, e1400215.	4.7	48

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109	How a [Co <sup>IV</sup> ]-O <sub>2</sub> Fragment Oxidizes Water: Involvement of a Biradicaloid [Co <sup>II</sup> ]-O <sub>2</sub> Species in Forming the O-O Bond. <i>ChemSusChem</i> , 2015, 8, 844-852.	3.6	46
110	Mapping the performance of amorphous ternary metal oxide water oxidation catalysts containing aluminium. <i>Journal of Materials Chemistry A</i> , 2015, 3, 756-761.	5.2	48
111	Structural Characteristics and Eutaxy in the Photo-Deposited Amorphous Iron Oxide Oxygen Evolution Catalyst. <i>Chemistry of Materials</i> , 2015, 27, 3462-3470.	3.2	28
112	Editorial for the ACS Select Virtual Issue on Inorganic Chemistry Driving the Energy Sciences. <i>Inorganic Chemistry</i> , 2015, 54, 3079-3083.	1.9	5
113	Tris-Heteroleptic Ruthenium-Dipyrrinate Chromophores in a Dye-Sensitized Solar Cell. <i>Chemistry - A European Journal</i> , 2015, 21, 2173-2181.	1.7	23
114	Substitution Effects on the Water Oxidation of Ruthenium Catalysts: A Quantum-Chemical Look. <i>Journal of Physical Chemistry C</i> , 2015, 119, 242-250.	1.5	15
115	Water Oxidation Catalysis: Survey of Amorphous Binary Metal Oxide Films Containing Lanthanum and Late 3d Transition Metals. <i>European Journal of Inorganic Chemistry</i> , 2014, 2014, 660-664.	1.0	17
116	Near-IR Photoresponse of Ruthenium Dipyrrinate Terpyridine Sensitizers in the Dye-Sensitized Solar Cells. <i>Inorganic Chemistry</i> , 2014, 53, 5417-5419.	1.9	37
117	Novel triphenylamine-modified ruthenium(ii) terpyridine complexes for nickel oxide-based cathodic dye-sensitized solar cells. <i>RSC Advances</i> , 2014, 4, 5782.	1.7	37
118	Direct Spectroscopic Evidence for Constituent Heteroatoms Enhancing Charge Recombination at a TiO <sub>2</sub> -Ruthenium Dye Interface. <i>Journal of Physical Chemistry C</i> , 2014, 118, 17079-17089.	1.5	20
119	Physicochemical Analysis of Ruthenium(II) Sensitizers of 1,2,3-Triazole-Derived Mesoionic Carbene and Cyclometalating Ligands. <i>Inorganic Chemistry</i> , 2014, 53, 2083-2095.	1.9	81
120	Intramolecular and Lateral Intermolecular Hole Transfer at the Sensitized TiO <sub>2</sub> Interface. <i>Journal of the American Chemical Society</i> , 2014, 136, 1034-1046.	6.6	54
121	Facile Photochemical Preparation of Amorphous Iridium Oxide Films for Water Oxidation Catalysis. <i>Chemistry of Materials</i> , 2014, 26, 1654-1659.	3.2	201
122	A Heteroleptic Bis(tridentate) Ruthenium(II) Platform Featuring an Anionic 1,2,3-Triazololate-Based Ligand for Application in the Dye-Sensitized Solar Cell. <i>Inorganic Chemistry</i> , 2014, 53, 1637-1645.	1.9	65
123	Donor-acceptor organic hybrid TiO <sub>2</sub> interfaces for solar energy conversion. <i>Thin Solid Films</i> , 2014, 560, 49-54.	0.8	7
124	Water Oxidation Catalysis: Electrocatalytic Response to Metal Stoichiometry in Amorphous Metal Oxide Films Containing Iron, Cobalt, and Nickel. <i>Journal of the American Chemical Society</i> , 2013, 135, 11580-11586.	6.6	817
125	Cyclometalated Ruthenium(II) Complexes Featuring Tridentate Click-Derived Ligands for Dye-Sensitized Solar Cell Applications. <i>Chemistry - A European Journal</i> , 2013, 19, 14171-14180.	1.7	35
126	Stabilization of Ruthenium Sensitizers to TiO <sub>2</sub> Surfaces through Cooperative Anchoring Groups. <i>Journal of the American Chemical Society</i> , 2013, 135, 1692-1695.	6.6	123



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127	Proton-coupled electron transfer at a [Co-OHx] <sub>z</sub> unit in aqueous media: evidence for a concerted mechanism. <i>Chemical Science</i> , 2013, 4, 734-738.	3.7	19
128	Homogeneous water oxidation catalysts containing a single metal site. <i>Chemical Communications</i> , 2013, 49, 218-227.	2.2	184
129	Atomic Level Resolution of Dye Regeneration in the Dye-Sensitized Solar Cell. <i>Journal of the American Chemical Society</i> , 2013, 135, 1961-1971.	6.6	133
130	Ruthenium(II) Complexes Bearing a Naphthalimide Fragment: A Modular Dye Platform for the Dye-Sensitized Solar Cell. <i>Inorganic Chemistry</i> , 2013, 52, 3001-3006.	1.9	47
131	Photochemical Route for Accessing Amorphous Metal Oxide Materials for Water Oxidation Catalysis. <i>Science</i> , 2013, 340, 60-63.	6.0	1,321
132	Interrogation of electrocatalytic water oxidation mediated by a cobalt complex. <i>Chemical Communications</i> , 2012, 48, 2107.	2.2	127
133	Cycloruthenated sensitizers: improving the dye-sensitized solar cell with classical inorganic chemistry principles. <i>Dalton Transactions</i> , 2012, 41, 7814.	1.6	101
134	Cyclometalated ruthenium chromophores for the dye-sensitized solar cell. <i>Coordination Chemistry Reviews</i> , 2012, 256, 1438-1450.	9.5	275
135	Bis(tridentate) Ruthenium-terpyridine Complexes Featuring Microsecond Excited-State Lifetimes. <i>Journal of the American Chemical Society</i> , 2012, 134, 12354-12357.	6.6	206
136	Derivatization of Bichromic Cyclometalated Ru(II) Complexes with Hydrophobic Substituents. <i>Inorganic Chemistry</i> , 2012, 51, 1501-1507.	1.9	25
137	Intramolecular Hole Transfer at Sensitized TiO <sub>2</sub> Interfaces. <i>Journal of the American Chemical Society</i> , 2012, 134, 8352-8355.	6.6	40
138	Ru complexes of thienyl-functionalized dipyrins as NCS-free sensitizers for the dye-sensitized solar cell. <i>Chemical Communications</i> , 2012, 48, 8790.	2.2	41
139	Three is not a crowd: efficient sensitization of TiO <sub>2</sub> by a bulky trichromic trisheteroleptic cycloruthenated dye. <i>Chemical Communications</i> , 2012, 48, 5599.	2.2	35
140	Systematic Modulation of a Bichromic Cyclometalated Ruthenium(II) Scaffold Bearing a Redox-Active Triphenylamine Constituent. <i>Inorganic Chemistry</i> , 2011, 50, 6019-6028.	1.9	59
141	Electrochemical evidence for catalytic water oxidation mediated by a high-valent cobalt complex. <i>Chemical Communications</i> , 2011, 47, 4249.	2.2	343
142	Regioselective C-H Activation of Cyclometalated Bis-Tridentate Ruthenium Complexes. <i>Organometallics</i> , 2011, 30, 6628-6635.	1.1	10
143	Unraveling the Roles of the Acid Medium, Experimental Probes, and Terminal Oxidant, (NH <sub>4</sub> ) <sub>2</sub> [Ce(NO <sub>3</sub> ) <sub>6</sub> ], in the Study of a Homogeneous Water Oxidation Catalyst. <i>Inorganic Chemistry</i> , 2011, 50, 3662-3672.	1.9	107
144	External-Stimuli Responsive Photophysics and Liquid Crystal Properties of Self-Assembled $\alpha$ -Phosphole-Lipids. <i>Journal of the American Chemical Society</i> , 2011, 133, 17014-17026.	6.6	146

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145	Design and Development of Functionalized Cyclometalated Ruthenium Chromophores for Light-Harvesting Applications. <i>Inorganic Chemistry</i> , 2011, 50, 5494-5508.	1.9	180
146	Strategies for Optimizing the Performance of Cyclometalated Ruthenium Sensitizers for Dye-Sensitized Solar Cells. <i>European Journal of Inorganic Chemistry</i> , 2011, 2011, 1806-1814.	1.0	84
147	A Trisheteroleptic Cyclometalated Ru(II) Sensitizer that Enables High Power Output in a Dye-Sensitized Solar Cell. <i>Angewandte Chemie - International Edition</i> , 2011, 50, 10682-10685.	7.2	127
148	Inside Cover: A Trisheteroleptic Cyclometalated Ru(II) Sensitizer that Enables High Power Output in a Dye-Sensitized Solar Cell ( <i>Angew. Chem. Int. Ed.</i> 45/2011). <i>Angewandte Chemie - International Edition</i> , 2011, 50, 10464-10464.	7.2	4
149	Structure-property relationships of acylated asymmetric dithienophospholes. <i>Comptes Rendus Chimie</i> , 2010, 13, 971-979.	0.2	11
150	Examination of Water Oxidation by Catalysts Containing Cofacial Metal Sites. <i>European Journal of Inorganic Chemistry</i> , 2010, 2010, 3135-3142.	1.0	36
151	Electronic Modification of the [Ru(II)(tpy)(bpy)(OH) <sub>2</sub> ] <sup>2+</sup> Scaffold: Effects on Catalytic Water Oxidation. <i>Journal of the American Chemical Society</i> , 2010, 132, 16094-16106.	6.6	299
152	Insight into Water Oxidation by Mononuclear Polypyridyl Ru Catalysts. <i>Inorganic Chemistry</i> , 2010, 49, 2202-2209.	1.9	256
153	Cyclometalated Ru Complexes of Type [Ru(II)(N <sup>+</sup> ) <sub>2</sub> (C <sup>-</sup> N)] <sup>z</sup> : Physicochemical Response to Substituents Installed on the Anionic Ligand. <i>Inorganic Chemistry</i> , 2010, 49, 4960-4971.	1.9	127
154	Sol-gel synthesis of linear Sn-doped TiO <sub>2</sub> nanostructures. <i>Journal of Materials Chemistry</i> , 2010, 20, 498-503.	6.7	50
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