

Jonas Elm

List of Publications by Year in descending order

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82
papers

3,088
citations

126708

33
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182168

51
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87
all docs

87
docs citations

87
times ranked

1694
citing authors

#	ARTICLE	IF	CITATIONS
1	Atmospheric Autoxidation of Organophosphate Esters. <i>Environmental Science & Technology</i> , 2022, 56, 6944-6955.	4.6	18
2	Real-time monitoring of aerosol particle formation from sulfuric acid vapor at elevated concentrations and temperatures. <i>Physical Chemistry Chemical Physics</i> , 2022, , .	1.3	0
3	Quantum Machine Learning Approach for Studying Atmospheric Cluster Formation. <i>Environmental Science and Technology Letters</i> , 2022, 9, 239-244.	3.9	18
4	Modeling the Binding Free Energy of Large Atmospheric Sulfuric Acid–Ammonia Clusters. <i>ACS Omega</i> , 2022, 7, 8077-8083.	1.6	6
5	The role of organic acids in new particle formation from methanesulfonic acid and methylamine. <i>Atmospheric Chemistry and Physics</i> , 2022, 22, 2639-2650.	1.9	20
6	Clusteromics III: Acid Synergy in Sulfuric Acid–Methanesulfonic Acid–Base Cluster Formation. <i>ACS Omega</i> , 2022, 7, 15206-15214.	1.6	19
7	Amine-Enhanced Methanesulfonic Acid-Driven Nucleation: Predictive Model and Cluster Formation Mechanism. <i>Environmental Science & Technology</i> , 2022, 56, 7751-7760.	4.6	13
8	Large Discrepancy in the Formation of Secondary Organic Aerosols from Structurally Similar Monoterpenes. <i>ACS Earth and Space Chemistry</i> , 2021, 5, 632-644.	1.2	17
9	Clusteromics I: Principles, Protocols, and Applications to Sulfuric Acid–Base Cluster Formation. <i>ACS Omega</i> , 2021, 6, 7804-7814.	1.6	27
10	Atmospheric Chemistry of Allylic Radicals from Isoprene: A Successive Cyclization-Driven Autoxidation Mechanism. <i>Environmental Science & Technology</i> , 2021, 55, 4399-4409.	4.6	20
11	New Particle Formation and Growth from Dimethyl Sulfide Oxidation by Hydroxyl Radicals. <i>ACS Earth and Space Chemistry</i> , 2021, 5, 801-811.	1.2	15
12	Clusteromics II: Methanesulfonic Acid–Base Cluster Formation. <i>ACS Omega</i> , 2021, 6, 17035-17044.	1.6	28
13	Secondary aerosol formation from dimethyl sulfide – improved mechanistic understanding based on smog chamber experiments and modelling. <i>Atmospheric Chemistry and Physics</i> , 2021, 21, 9955-9976.	1.9	24
14	Tri-Base Synergy in Sulfuric Acid-Base Clusters. <i>Atmosphere</i> , 2021, 12, 1260.	1.0	12
15	Toward a Holistic Understanding of the Formation and Growth of Atmospheric Molecular Clusters: A Quantum Machine Learning Perspective. <i>Journal of Physical Chemistry A</i> , 2021, 125, 895-902.	1.1	20
16	Structural Effects of Amines in Enhancing Methanesulfonic Acid-Driven New Particle Formation. <i>Environmental Science & Technology</i> , 2020, 54, 13498-13508.	4.6	36
17	The reaction of isotope-substituted hydrated iodide I(H ¹⁸ O) ⁺ with ozone: the reactive influence of the solvent water molecule. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 19080-19088.	1.3	2
18	Neutral Sulfuric Acid–Water Clustering Rates: Bridging the Gap between Molecular Simulation and Experiment. <i>Journal of Physical Chemistry Letters</i> , 2020, 11, 4239-4244.	2.1	6

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19	Formation of Low-Volatile Products and Unexpected High Formaldehyde Yield from the Atmospheric Oxidation of Methylsiloxanes. <i>Environmental Science & Technology</i> , 2020, 54, 7136-7145.	4.6	27
20	Hydration of Atmospheric Molecular Clusters III: Procedure for Efficient Free Energy Surface Exploration of Large Hydrated Clusters. <i>Journal of Physical Chemistry A</i> , 2020, 124, 5253-5261.	1.1	16
21	Assessment of the DLPNO Binding Energies of Strongly Noncovalent Bonded Atmospheric Molecular Clusters. <i>ACS Omega</i> , 2020, 5, 7601-7612.	1.6	38
22	Thermodynamic properties of isoprene- and monoterpene-derived organosulfates estimated with COSMO<i>+</i>therm<i>+</i>. <i>Atmospheric Chemistry and Physics</i> , 2020, 20, 5679-5696.	1.9	25
23	Modeling the formation and growth of atmospheric molecular clusters: A review. <i>Journal of Aerosol Science</i> , 2020, 149, 105621.	1.8	98
24	The Aarhus Chamber Campaign on Highly Oxygenated Organic Molecules and Aerosols (ACCHA): particle formation, organic acids, and dimer esters from <i>+</i>pinene ozonolysis at different temperatures. <i>Atmospheric Chemistry and Physics</i> , 2020, 20, 12549-12567.	1.9	21
25	Technical note: Estimating aqueous solubilities and activity coefficients of mono- and <i>+</i>-dicarboxylic acids using COSMO<i>+</i>therm<i>+</i>. <i>Atmospheric Chemistry and Physics</i> , 2020, 20, 13131-13143.	1.9	6
26	Mechanism and predictive model development of reaction rate constants for N-center radicals with O ₂ . <i>Chemosphere</i> , 2019, 237, 124411.	4.2	8
27	Benchmarking sampling methodology for calculations of Rayleigh light scattering properties of atmospheric molecular clusters. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 17274-17287.	1.3	4
28	Piperazine Enhancing Sulfuric Acid-Based New Particle Formation: Implications for the Atmospheric Fate of Piperazine. <i>Environmental Science & Technology</i> , 2019, 53, 8785-8795.	4.6	41
29	Methanesulfonic Acid-driven New Particle Formation Enhanced by Monoethanolamine: A Computational Study. <i>Environmental Science & Technology</i> , 2019, 53, 14387-14397.	4.6	50
30	Strong Even/Odd Pattern in the Computed Gas-Phase Stability of Dicarboxylic Acid Dimers: Implications for Condensation Thermodynamics. <i>Journal of Physical Chemistry A</i> , 2019, 123, 9594-9599.	1.1	10
31	An Atmospheric Cluster Database Consisting of Sulfuric Acid, Bases, Organics, and Water. <i>ACS Omega</i> , 2019, 4, 10965-10974.	1.6	58
32	The role of highly oxygenated organic molecules in the Boreal aerosol-cloud-climate system. <i>Nature Communications</i> , 2019, 10, 4370.	5.8	91
33	The reaction of hydrated iodide I(H₂O)<sup>+</sup> with ozone: a new route to IO₂<sup>+</sup> products. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 17546-17554.	1.3	19
34	Unexpected Growth Coordinate in Large Clusters Consisting of Sulfuric Acid and C₈H₁₂O₆ Tricarboxylic Acid. <i>Journal of Physical Chemistry A</i> , 2019, 123, 3170-3175.	1.1	15
35	Reply to the "Comment on "Atmospheric chemistry of iodine anions: elementary reactions of I<sup>+</sup>, IO<sup>+</sup>, and IO&sup>2 with ozone studied in the gas-phase at 300 K using an ion trap" by D. Britz, <i>Phys. Chem. Chem. Phys.</i>, 2019, 21, C9CP03851E. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 22656-22656.	1.3	0
36	Guanidine: A Highly Efficient Stabilizer in Atmospheric New-Particle Formation. <i>Journal of Physical Chemistry A</i> , 2018, 122, 4717-4729.	1.1	32

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37	Closed-Shell Organic Compounds Might Form Dimers at the Surface of Molecular Clusters. <i>Journal of Physical Chemistry A</i> , 2018, 122, 1771-1780.	1.1	16
38	Atmospheric chemistry of iodine anions: elementary reactions of I^{-} , IO^{-} , and IO_2^{-} with ozone studied in the gas-phase at 300 K using an ion trap. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 28606-28615.	1.3	24
39	Hydration of Atmospheric Molecular Clusters II: Organic Acid-Water Clusters. <i>Journal of Physical Chemistry A</i> , 2018, 122, 8549-8556.	1.1	36
40	New Particle Formation and Growth. , 2018, , 315-352.		12
41	Atmospheric Oxidation of Piperazine Initiated by $\dot{A}Cl$: Unexpected High Nitrosamine Yield. <i>Environmental Science & Technology</i> , 2018, 52, 9801-9809.	4.6	45
42	Hydration of Atmospheric Molecular Clusters: A New Method for Systematic Configurational Sampling. <i>Journal of Physical Chemistry A</i> , 2018, 122, 5026-5036.	1.1	53
43	Effect of Bisulfate, Ammonia, and Ammonium on the Clustering of Organic Acids and Sulfuric Acid. <i>Journal of Physical Chemistry A</i> , 2017, 121, 4812-4824.	1.1	35
44	What Is Required for Highly Oxidized Molecules To Form Clusters with Sulfuric Acid?. <i>Journal of Physical Chemistry A</i> , 2017, 121, 4578-4587.	1.1	56
45	Benchmark Study of the Structural and Thermochemical Properties of a Dihydroazulene/Vinylheptafulvene Photoswitch. <i>Journal of Physical Chemistry A</i> , 2017, 121, 3148-3154.	1.1	23
46	Formation of atmospheric molecular clusters consisting of sulfuric acid and $C_8H_{12}O_6$ tricarboxylic acid. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 4877-4886.	1.3	47
47	Phosphoric acid – a potentially elusive participant in atmospheric new particle formation. <i>Molecular Physics</i> , 2017, 115, 2168-2179.	0.8	15
48	Basis set convergence of the binding energies of strongly hydrogen-bonded atmospheric clusters. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 1122-1133.	1.3	82
49	Towards Storage of Solar Energy in Photochromic Molecules: Benzannulation of the Dihydroazulene/Vinylheptafulvene Couple. <i>ChemPhotoChem</i> , 2017, 1, 206-212.	1.5	29
50	Elucidating the Limiting Steps in Sulfuric Acid-Base New Particle Formation. <i>Journal of Physical Chemistry A</i> , 2017, 121, 8288-8295.	1.1	60
51	Diamines Can Initiate New Particle Formation in the Atmosphere. <i>Journal of Physical Chemistry A</i> , 2017, 121, 6155-6164.	1.1	72
52	Can COSMOTherm Predict a Salting in Effect?. <i>Journal of Physical Chemistry A</i> , 2017, 121, 6288-6295.	1.1	17
53	Atmospheric Fate of Monoethanolamine: Enhancing New Particle Formation of Sulfuric Acid as an Important Removal Process. <i>Environmental Science & Technology</i> , 2017, 51, 8422-8431.	4.6	95
54	The Effect of Water and Bases on the Clustering of a Cyclohexene Autoxidation Product $C_6H_8O_7$ with Sulfuric Acid. <i>Journal of Physical Chemistry A</i> , 2016, 120, 2240-2249.	1.1	30

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55	Strong Hydrogen Bonded Molecular Interactions between Atmospheric Diamines and Sulfuric Acid. <i>Journal of Physical Chemistry A</i> , 2016, 120, 3693-3700.	1.1	70
56	Fine-tuning the lifetimes and energy storage capacities of meta-stable vinylheptafulvenes via substitution at the vinyl position. <i>RSC Advances</i> , 2016, 6, 49003-49010.	1.7	23
57	Effect of Conformers on Free Energies of Atmospheric Complexes. <i>Journal of Physical Chemistry A</i> , 2016, 120, 8613-8624.	1.1	36
58	Aromaticityâ€‘Controlled Energy Storage Capacity of the Dihydroazuleneâ€‘Vinylheptafulvene Photochromic System. <i>Chemistry - A European Journal</i> , 2016, 22, 14567-14575.	1.7	55
59	Theoretical Investigation of Substituent Effects on the Dihydroazulene/Vinylheptafulvene Photoswitch: Increasing the Energy Storage Capacity. <i>Journal of Physical Chemistry A</i> , 2016, 120, 9782-9793.	1.1	39
60	Density functional theory basis set convergence of sulfuric acid-containing molecular clusters. <i>Computational and Theoretical Chemistry</i> , 2016, 1098, 1-12.	1.1	53
61	Coupled Cluster Evaluation of the Stability of Atmospheric Acidâ€‘Base Clusters with up to 10 Molecules. <i>Journal of Physical Chemistry A</i> , 2016, 120, 621-630.	1.1	83
62	Azulenium chemistry: towards new derivatives of photochromic dihydroazulenes. <i>Organic and Biomolecular Chemistry</i> , 2016, 14, 2403-2412.	1.5	14
63	Gas-Phase Spectroscopy of a Vinylheptafulvene Chromophore. <i>European Journal of Mass Spectrometry</i> , 2015, 21, 569-577.	0.5	3
64	Towards Solar Energy Storage in the Photochromic Dihydroazuleneâ€‘Vinylheptafulvene System. <i>Chemistry - A European Journal</i> , 2015, 21, 7454-7461.	1.7	79
65	Computational Study of the Effect of Glyoxalâ€‘Sulfate Clustering on the Henryâ€‘s Law Coefficient of Glyoxal. <i>Journal of Physical Chemistry A</i> , 2015, 119, 4509-4514.	1.1	35
66	Computational Methodology Study of the Optical and Thermochemical Properties of a Molecular Photoswitch. <i>Journal of Physical Chemistry A</i> , 2015, 119, 896-904.	1.1	57
67	Computational Study of the Clustering of a Cyclohexene Autoxidation Product $C_6H_8O_7$ with Itself and Sulfuric Acid. <i>Journal of Physical Chemistry A</i> , 2015, 119, 8414-8421.	1.1	45
68	Rayleigh light scattering properties of atmospheric molecular clusters consisting of sulfuric acid and bases. <i>Physical Chemistry Chemical Physics</i> , 2015, 17, 15701-15709.	1.3	14
69	Glyoxal and Methylglyoxal Setschenow Salting Constants in Sulfate, Nitrate, and Chloride Solutions: Measurements and Gibbs Energies. <i>Environmental Science & Technology</i> , 2015, 49, 11500-11508.	4.6	64
70	Computational approaches for efficiently modelling of small atmospheric clusters. <i>Chemical Physics Letters</i> , 2014, 615, 26-29.	1.2	75
71	Computational study of the Rayleigh light scattering properties of atmospheric pre-nucleation clusters. <i>Physical Chemistry Chemical Physics</i> , 2014, 16, 10883-10890.	1.3	37
72	Molecular Interaction of Pinic Acid with Sulfuric Acid: Exploring the Thermodynamic Landscape of Cluster Growth. <i>Journal of Physical Chemistry A</i> , 2014, 118, 7892-7900.	1.1	64

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73	Methane sulfonic acid-enhanced formation of molecular clusters of sulfuric acid and dimethyl amine. <i>Atmospheric Chemistry and Physics</i> , 2014, 14, 12023-12030.	1.9	110
74	Assessment of binding energies of atmospherically relevant clusters. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 16442.	1.3	130
75	Ambient reaction kinetics of atmospheric oxygenated organics with the OH radical: a computational methodology study. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 9636.	1.3	36
76	Interaction of Glycine with Common Atmospheric Nucleation Precursors. <i>Journal of Physical Chemistry A</i> , 2013, 117, 12990-12997.	1.1	55
77	Influence of Nucleation Precursors on the Reaction Kinetics of Methanol with the OH Radical. <i>Journal of Physical Chemistry A</i> , 2013, 117, 6695-6701.	1.1	51
78	Large area, soft crystalline thin films of N,N'-trialkyltriaza-triangulenium salts with homeotropic alignment of the discotic cores in a lamellar lattice. <i>Journal of Materials Chemistry</i> , 2012, 22, 4797.	6.7	26
79	Assessment of Density Functional Theory in Predicting Structures and Free Energies of Reaction of Atmospheric Prenucleation Clusters. <i>Journal of Chemical Theory and Computation</i> , 2012, 8, 2071-2077.	2.3	168
80	Obtaining Enhanced Circular Dichroism in [4]Heterohelicenium Analogues. <i>Journal of Physical Chemistry A</i> , 2012, 116, 8744-8752.	1.1	14
81	Racemization Mechanisms and Electronic Circular Dichroism of [4]Heterohelicenium Dyes: A Theoretical Study. <i>Journal of Physical Chemistry A</i> , 2011, 115, 12025-12033.	1.1	18
82	Direct probing of ion pair formation using a symmetric triangulenium dye. <i>Photochemical and Photobiological Sciences</i> , 2011, 10, 1963-1973.	1.6	26