## Malay K Ghosh

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/8965373/publications.pdf

Version: 2024-02-01

430874 289244 1,670 60 18 40 citations h-index g-index papers 61 61 61 1756 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	Hydrolytically stable citrate capped Fe3O4@UiO-66-NH2 MOF: A hetero-structure composite with enhanced activity towards Cr (VI) adsorption and photocatalytic H2 evolution. Journal of Colloid and Interface Science, 2022, 606, 353-366.	9.4	94
2	Nanoscale ZnO-adsorbent carefully designed for the kinetic and thermodynamic studies of Rhodamine B. Inorganic Chemistry Communication, 2022, 138, 109287.	3.9	17
3	Silicate minerals - Potential source of potash - A review. Minerals Engineering, 2022, 179, 107463.	4.3	15
4	Separation and characterization studies of end-of-life mobile printed circuit boards. Particulate Science and Technology, 2021, 39, 467-474.	2.1	5
5	Factorial design for process optimization and generation of kinetic data for yttrium and europium leaching. Mineral Processing and Extractive Metallurgy: Transactions of the Institute of Mining and Metallurgy, 2021, 130, 1-9.	0.2	4
6	Structure and activity of lysozyme on binding to lithium-manganese oxide nanocomposites prepared from seabed nodule. Journal of Physics and Chemistry of Solids, 2021, 151, 109794.	4.0	3
7	A Review of Recent Progress on Nano MnO2: Synthesis, Surface Modification and Applications. Journal of Inorganic and Organometallic Polymers and Materials, 2021, 31, 899-922.	3.7	22
8	Role of Glycerol Oxidation Pathways in the Reductive Acid Leaching Kinetics of Manganese Nodules Using Glycerol. ACS Omega, 2021, 6, 14903-14910.	3.5	6
9	Critical analysis of metallic and non-metallic fractions in the flotation of waste printed circuit boards. Powder Technology, 2021, 389, 450-459.	4.2	11
10	Recovery of Lead as Lead Sulphide from Anode Slime Using Hydrometallurgical Technique. Journal of the Institution of Engineers (India): Series D, 2021, 102, 489-494.	1.0	1
11	Electrodeposition of nanoMnO2 from mineral leach liquor and the investigation on conformational changes of hemoglobin induced by the nanomaterial. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2020, 585, 124102.	4.7	3
12	Molecular interactions of MnO <sub>2</sub> @RGO (manganese dioxide-reduced graphene oxide) nanocomposites with bovine serum albumin. Journal of Biomolecular Structure and Dynamics, 2020, 38, 2038-2046.	3.5	7
13	Quantitative reductive leaching of a low-grade spessartine ore in acidic media. International Journal of Environmental Analytical Chemistry, 2020, 100, 635-646.	3.3	2
14	Extraction of Cu, Zn, and Ni from waste silica-rich integrated circuits by sulfation roasting and water leaching. Chemical Papers, 2020, 74, 663-671.	2.2	10
15	R&D Efforts of CSIR-IMMT Toward Solving Some Issues Related to Aluminum Production. Journal of Sustainable Metallurgy, 2020, 6, 9-17.	2.3	1
16	Efficacy of Dichlorophenolindophenol (DCIP) as Screening Test for Hb E: Revisited. Indian Journal of Hematology and Blood Transfusion, 2020, 36, 535-541.	0.6	1
17	Sustainable Process for the Extraction of Potassium from Feldspar Using Eggshell Powder. ACS Omega, 2020, 5, 14990-14998.	3.5	12
18	Efficient Photon Conversion via Double Charge Dynamics CeO <sub>2</sub> â€"BiFeO <sub>3</sub> pâ€"n Heterojunction Photocatalyst Promising toward N <sub>2</sub> Fixation and Phenolâ€"Cr(VI) Detoxification. Inorganic Chemistry, 2020, 59, 3856-3873.	4.0	98

#	Article	IF	CITATIONS
19	Leaching of Rare Earth Elements from the Residue Generated by the Lixiviation of Waste Phosphor with Sulphuric Acid. Transactions of the Indian Institute of Metals, 2020, 73, 1081-1091.	1.5	2
20	Dissolution Kinetics Potential of a Biotite-Rich Kaolinite Ore for Industrial Applications by Oxalic Acid Solution. Mining, Metallurgy and Exploration, 2019, 36, 1091-1099.	0.8	4
21	Leaching kinetics of Cu, Ni and Zn from waste silica rich integrated circuits using mild nitric acid. Hydrometallurgy, 2019, 188, 161-168.	4.3	48
22	Extraction of Copper and Zinc from Waste Printed Circuit Boards. Recycling, 2019, 4, 36.	5.0	13
23	Production of Potassium Chloride from K-Feldspar Through Roast–Leach–Solvent Extraction Route. Transactions of the Indian Institute of Metals, 2019, 72, 2613-2622.	1.5	11
24	Selective enhancement of Mn bioleaching from ferromanganese ores in presence of electron shuttles using dissimilatory Mn reducing consortia. Hydrometallurgy, 2019, 186, 269-274.	4.3	9
25	Enrichment of a Nigerian chromite ore for metallurgical application by dense medium flotation and magnetic separation. Metallurgical Research and Technology, 2019, 116, 324.	0.7	1
26	Leaching of Rare Earth Metals from Phosphor Coating of Waste Fluorescent Lamps. Transactions of the Indian Institute of Metals, 2019, 72, 623-634.	1.5	11
27	Extraction equilibrium conditions of beryllium and aluminium from a beryl ore for optimal industrial beryllium compound production. Canadian Metallurgical Quarterly, 2019, 58, 232-240.	1.2	2
28	Electrodeposition of $\hat{I}^3$ -MnO2 from Manganese Nodule Leach Liquor: Surface Modification and Electrochemical Applications. Minerals, Metals and Materials Series, 2018, , 165-174.	0.4	1
29	Preparation of Sm 2 O 3 and Co 3 O 4 from SmCo magnet swarf by hydrometallurgical processing in chloride media. Journal of Rare Earths, 2018, 36, 725-732.	4.8	19
30	Preparation of Industrial Manganese Compound from a Low-Grade Spessartine Ore by Hydrometallurgical Process. Transactions of the Indian Institute of Metals, 2018, 71, 2453-2463.	1.5	3
31	Visible Light Active Single-Crystal Nanorod/Needle-like α-MnO <sub>2</sub> @RGO Nanocomposites for Efficient Photoreduction of Cr(VI). Journal of Physical Chemistry C, 2017, 121, 6039-6049.	3.1	63
32	Construing the interactions between MnO <sub>2</sub> nanoparticle and bovine serum albumin: insight into the structure and stability of a proteinâ€"nanoparticle complex. New Journal of Chemistry, 2017, 41, 8130-8139.	2.8	48
33	Bioleaching of Indian Ocean nodules with in situ iron precipitation by anaerobic Mn reducing consortia. Hydrometallurgy, 2016, 166, 130-135.	4.3	10
34	Probing Environmental Remediation of RhB Organic Dye Using $\hat{l}\pm$ -MnO <sub>2</sub> under Visible- Light Irradiation: Structural, Photocatalytic and Mineralization Studies. ChemistrySelect, 2016, 1, 4277-4285.	1.5	49
35	Effect of Tween 80 on electrochemical deposition of cobalt from sulphate solutions. Russian Journal of Non-Ferrous Metals, 2016, 57, 331-337.	0.6	0
36	Electrodeposition of manganese metal from sulphate solutions in the presence of sodium octyl sulphate. Hydrometallurgy, 2016, 165, 73-80.	4.3	25

3

#	Article	IF	CITATIONS
37	Reactor and column leaching studies for extraction of copper from two low grade resources: A comparative study. Hydrometallurgy, 2016, 165, 111-117.	4.3	10
38	Processing Nigerian pyrolusite ore, Part I: Characterization and dissolution kinetics analysis. CIM Journal, 2016, 7, 43-51.	0.6	2
39	Extraction and Purification of Copper from a Nigerian Chalcopyrite Ore Leach Liquor by Dithizone in Kerosene. Solvent Extraction Research and Development, 2015, 22, 135-146.	0.4	6
40	Pathway of Sucrose Oxidation in Manganese (Pyrolusite) Nodule. Industrial & Engineering Chemistry Research, 2015, 54, 12233-12241.	3.7	14
41	Waste Printed Circuit Boards recycling: an extensive assessment ofÂcurrent status. Journal of Cleaner Production, 2015, 94, 5-19.	9.3	439
42	Characterization and kinetic study on ammonia leaching of complex copper ore. Transactions of Nonferrous Metals Society of China, 2014, 24, 1587-1595.	4.2	37
43	Aqueous processing of nickel spent catalyst for a value added product. Korean Journal of Chemical Engineering, 2013, 30, 400-404.	2.7	2
44	Mineralogical characterization and leaching behavior of Nigerian ilmenite ore. Transactions of Nonferrous Metals Society of China, 2013, 23, 2743-2750.	4.2	14
45	Dissolution kinetics of nickel from spent catalyst in nitric acid medium. Journal of the Taiwan Institute of Chemical Engineers, 2013, 44, 34-39.	5.3	20
46	Extraction of Cu and Cr from a spent Cu–Cr catalyst: Recovery enhancement through mechanical activation. Hydrometallurgy, 2013, 136, 8-14.	4.3	15
47	Hydrometallurgical application for treating a Nigerian chalcopyrite ore in chloride medium: Part I. Dissolution kinetics assessment. International Journal of Minerals, Metallurgy and Materials, 2013, 20, 1021-1028.	4.9	15
48	Arsenic adsorption on goethite nanoparticles produced through hydrazine sulfate assisted synthesis method. Korean Journal of Chemical Engineering, 2012, 29, 95-102.	2.7	72
49	Preparation and characterization of EMD from manganese cake — A byproduct of manganese nodule processing. Hydrometallurgy, 2011, 110, 44-49.	4.3	19
50	Defluoridation behavior of nanostructured hydroxyapatite synthesized through an ultrasonic and microwave combined technique. Journal of Hazardous Materials, 2011, 185, 29-37.	12.4	99
51	Dried leaves â€" Novel reductant for acid leaching of manganese ore. Transactions of the Indian Institute of Metals, 2009, 62, 551-554.	1.5	10
52	Sulphuric acid leaching of polymetallic nodules using paper as a reductant. Transactions of the Indian Institute of Metals, 2008, 61, 477-481.	1.5	18
53	Leaching of manganese ores using sawdust as a reductant. Minerals Engineering, 2007, 20, 1293-1295.	4.3	109
54	Assessment of the significant parameters influencing the bio-oxidation and bio-precipitation of iron from industrial leach liquor. Institutions of Mining and Metallurgy Transactions Section C: Mineral Processing and Extractive Metallurgy, 2005, 114, 57-64.	0.6	5

#	Article	IF	CITATION
55	Oxidative ammonia leaching of sphalerite. International Journal of Mineral Processing, 2003, 70, 221-234.	2.6	22
56	Oxidative ammonia leaching of sphalerite. International Journal of Mineral Processing, 2002, 66, 241-254.	2.6	34
57	Leaching of metals from Indian ocean nodules in SO2–H2O–H2SO4–(NH4)2SO4 medium. Hydrometallurgy, 1999, 53, 169-175.	4.3	49
58	Behaviour of cobalt during precipitation of manganese from the NH3î—,(NH4)2SO4î—,Mnî—,O2 system. Hydrometallurgy, 1996, 42, 357-366.	4.3	3
59	Oxidative ammonia leaching of pure zinc sulfide in the presence of lead compounds. Metallurgical and Materials Transactions B - Process Metallurgy and Materials Processing Science, 1990, 21, 402-404.	0.4	6
60	Effect of dissolved impurities during ammonia leaching of pure zinc sulphide. Hydrometallurgy, 1989, 22, 207-221.	4.3	14