## Malay K Ghosh

List of Publications by Year in descending order

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430874 289244 1,670 60 18 40 citations h-index g-index papers 61 61 61 1756 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	Waste Printed Circuit Boards recycling: an extensive assessment ofÂcurrent status. Journal of Cleaner Production, 2015, 94, 5-19.	9.3	439
2	Leaching of manganese ores using sawdust as a reductant. Minerals Engineering, 2007, 20, 1293-1295.	4.3	109
3	Defluoridation behavior of nanostructured hydroxyapatite synthesized through an ultrasonic and microwave combined technique. Journal of Hazardous Materials, 2011, 185, 29-37.	12.4	99
4	Efficient Photon Conversion via Double Charge Dynamics CeO <sub>2</sub> â€"BiFeO <sub>3</sub> pâ€"n Heterojunction Photocatalyst Promising toward N <sub>2</sub> Fixation and Phenolâ€"Cr(VI) Detoxification. Inorganic Chemistry, 2020, 59, 3856-3873.	4.0	98
5	Hydrolytically stable citrate capped Fe3O4@UiO-66-NH2 MOF: A hetero-structure composite with enhanced activity towards Cr (VI) adsorption and photocatalytic H2 evolution. Journal of Colloid and Interface Science, 2022, 606, 353-366.	9.4	94
6	Arsenic adsorption on goethite nanoparticles produced through hydrazine sulfate assisted synthesis method. Korean Journal of Chemical Engineering, 2012, 29, 95-102.	2.7	72
7	Visible Light Active Single-Crystal Nanorod/Needle-like $\hat{l}\pm$ -MnO <sub>2</sub> @RGO Nanocomposites for Efficient Photoreduction of Cr(VI). Journal of Physical Chemistry C, 2017, 121, 6039-6049.	3.1	63
8	Leaching of metals from Indian ocean nodules in SO2–H2O–H2SO4–(NH4)2SO4 medium. Hydrometallurgy, 1999, 53, 169-175.	4.3	49
9	Probing Environmental Remediation of RhB Organic Dye Using α-MnO <sub>2</sub> under Visible-Light Irradiation: Structural, Photocatalytic and Mineralization Studies. ChemistrySelect, 2016, 1, 4277-4285.	1.5	49
10	Construing the interactions between MnO <sub>2</sub> nanoparticle and bovine serum albumin: insight into the structure and stability of a proteinâ€"nanoparticle complex. New Journal of Chemistry, 2017, 41, 8130-8139.	2.8	48
11	Leaching kinetics of Cu, Ni and Zn from waste silica rich integrated circuits using mild nitric acid. Hydrometallurgy, 2019, 188, 161-168.	4.3	48
12	Characterization and kinetic study on ammonia leaching of complex copper ore. Transactions of Nonferrous Metals Society of China, 2014, 24, 1587-1595.	4.2	37
13	Oxidative ammonia leaching of sphalerite. International Journal of Mineral Processing, 2002, 66, 241-254.	2.6	34
14	Electrodeposition of manganese metal from sulphate solutions in the presence of sodium octyl sulphate. Hydrometallurgy, 2016, 165, 73-80.	4.3	25
15	Oxidative ammonia leaching of sphalerite. International Journal of Mineral Processing, 2003, 70, 221-234.	2.6	22
16	A Review of Recent Progress on Nano MnO2: Synthesis, Surface Modification and Applications. Journal of Inorganic and Organometallic Polymers and Materials, 2021, 31, 899-922.	3.7	22
17	Dissolution kinetics of nickel from spent catalyst in nitric acid medium. Journal of the Taiwan Institute of Chemical Engineers, 2013, 44, 34-39.	5.3	20
18	Preparation and characterization of EMD from manganese cake â€" A byproduct of manganese nodule processing. Hydrometallurgy, 2011, 110, 44-49.	4.3	19

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19	Preparation of Sm 2 O 3 and Co 3 O 4 from SmCo magnet swarf by hydrometallurgical processing in chloride media. Journal of Rare Earths, 2018, 36, 725-732.	4.8	19
20	Sulphuric acid leaching of polymetallic nodules using paper as a reductant. Transactions of the Indian Institute of Metals, 2008, 61, 477-481.	1.5	18
21	Nanoscale ZnO-adsorbent carefully designed for the kinetic and thermodynamic studies of Rhodamine B. Inorganic Chemistry Communication, 2022, 138, 109287.	3.9	17
22	Extraction of Cu and Cr from a spent Cuâ€"Cr catalyst: Recovery enhancement through mechanical activation. Hydrometallurgy, 2013, 136, 8-14.	4.3	15
23	Hydrometallurgical application for treating a Nigerian chalcopyrite ore in chloride medium: Part I. Dissolution kinetics assessment. International Journal of Minerals, Metallurgy and Materials, 2013, 20, 1021-1028.	4.9	15
24	Silicate minerals - Potential source of potash - A review. Minerals Engineering, 2022, 179, 107463.	4.3	15
25	Effect of dissolved impurities during ammonia leaching of pure zinc sulphide. Hydrometallurgy, 1989, 22, 207-221.	4.3	14
26	Mineralogical characterization and leaching behavior of Nigerian ilmenite ore. Transactions of Nonferrous Metals Society of China, 2013, 23, 2743-2750.	4.2	14
27	Pathway of Sucrose Oxidation in Manganese (Pyrolusite) Nodule. Industrial & Engineering Chemistry Research, 2015, 54, 12233-12241.	3.7	14
28	Extraction of Copper and Zinc from Waste Printed Circuit Boards. Recycling, 2019, 4, 36.	5.0	13
29	Sustainable Process for the Extraction of Potassium from Feldspar Using Eggshell Powder. ACS Omega, 2020, 5, 14990-14998.	3.5	12
30	Production of Potassium Chloride from K-Feldspar Through Roast–Leach–Solvent Extraction Route. Transactions of the Indian Institute of Metals, 2019, 72, 2613-2622.	1.5	11
31	Leaching of Rare Earth Metals from Phosphor Coating of Waste Fluorescent Lamps. Transactions of the Indian Institute of Metals, 2019, 72, 623-634.	1.5	11
32	Critical analysis of metallic and non-metallic fractions in the flotation of waste printed circuit boards. Powder Technology, 2021, 389, 450-459.	4.2	11
33	Dried leaves â€" Novel reductant for acid leaching of manganese ore. Transactions of the Indian Institute of Metals, 2009, 62, 551-554.	1.5	10
34	Bioleaching of Indian Ocean nodules with in situ iron precipitation by anaerobic Mn reducing consortia. Hydrometallurgy, 2016, 166, 130-135.	4.3	10
35	Reactor and column leaching studies for extraction of copper from two low grade resources: A comparative study. Hydrometallurgy, 2016, 165, 111-117.	4.3	10
36	Extraction of Cu, Zn, and Ni from waste silica-rich integrated circuits by sulfation roasting and water leaching. Chemical Papers, 2020, 74, 663-671.	2.2	10

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37	Selective enhancement of Mn bioleaching from ferromanganese ores in presence of electron shuttles using dissimilatory Mn reducing consortia. Hydrometallurgy, 2019, 186, 269-274.	4.3	9
38	Molecular interactions of MnO <sub>2</sub> @RGO (manganese dioxide-reduced graphene oxide) nanocomposites with bovine serum albumin. Journal of Biomolecular Structure and Dynamics, 2020, 38, 2038-2046.	3 <b>.</b> 5	7
39	Oxidative ammonia leaching of pure zinc sulfide in the presence of lead compounds. Metallurgical and Materials Transactions B - Process Metallurgy and Materials Processing Science, 1990, 21, 402-404.	0.4	6
40	Extraction and Purification of Copper from a Nigerian Chalcopyrite Ore Leach Liquor by Dithizone in Kerosene. Solvent Extraction Research and Development, 2015, 22, 135-146.	0.4	6
41	Role of Glycerol Oxidation Pathways in the Reductive Acid Leaching Kinetics of Manganese Nodules Using Glycerol. ACS Omega, 2021, 6, 14903-14910.	3.5	6
42	Assessment of the significant parameters influencing the bio-oxidation and bio-precipitation of iron from industrial leach liquor. Institutions of Mining and Metallurgy Transactions Section C: Mineral Processing and Extractive Metallurgy, 2005, 114, 57-64.	0.6	5
43	Separation and characterization studies of end-of-life mobile printed circuit boards. Particulate Science and Technology, 2021, 39, 467-474.	2.1	5
44	Dissolution Kinetics Potential of a Biotite-Rich Kaolinite Ore for Industrial Applications by Oxalic Acid Solution. Mining, Metallurgy and Exploration, 2019, 36, 1091-1099.	0.8	4
45	Factorial design for process optimization and generation of kinetic data for yttrium and europium leaching. Mineral Processing and Extractive Metallurgy: Transactions of the Institute of Mining and Metallurgy, 2021, 130, 1-9.	0.2	4
46	Behaviour of cobalt during precipitation of manganese from the NH3î—,(NH4)2SO4î—,Mnî—,O2 system. Hydrometallurgy, 1996, 42, 357-366.	4.3	3
47	Preparation of Industrial Manganese Compound from a Low-Grade Spessartine Ore by Hydrometallurgical Process. Transactions of the Indian Institute of Metals, 2018, 71, 2453-2463.	1.5	3
48	Electrodeposition of nanoMnO2 from mineral leach liquor and the investigation on conformational changes of hemoglobin induced by the nanomaterial. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2020, 585, 124102.	4.7	3
49	Structure and activity of lysozyme on binding to lithium-manganese oxide nanocomposites prepared from seabed nodule. Journal of Physics and Chemistry of Solids, 2021, 151, 109794.	4.0	3
50	Aqueous processing of nickel spent catalyst for a value added product. Korean Journal of Chemical Engineering, 2013, 30, 400-404.	2.7	2
51	Extraction equilibrium conditions of beryllium and aluminium from a beryl ore for optimal industrial beryllium compound production. Canadian Metallurgical Quarterly, 2019, 58, 232-240.	1.2	2
52	Quantitative reductive leaching of a low-grade spessartine ore in acidic media. International Journal of Environmental Analytical Chemistry, 2020, 100, 635-646.	3.3	2
53	Leaching of Rare Earth Elements from the Residue Generated by the Lixiviation of Waste Phosphor with Sulphuric Acid. Transactions of the Indian Institute of Metals, 2020, 73, 1081-1091.	1.5	2
54	Processing Nigerian pyrolusite ore, Part I: Characterization and dissolution kinetics analysis. CIM Journal, 2016, 7, 43-51.	0.6	2

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55	Electrodeposition of $\hat{I}^3$ -MnO2 from Manganese Nodule Leach Liquor: Surface Modification and Electrochemical Applications. Minerals, Metals and Materials Series, 2018, , 165-174.	0.4	1
56	Enrichment of a Nigerian chromite ore for metallurgical application by dense medium flotation and magnetic separation. Metallurgical Research and Technology, 2019, 116, 324.	0.7	1
57	R&D Efforts of CSIR-IMMT Toward Solving Some Issues Related to Aluminum Production. Journal of Sustainable Metallurgy, 2020, 6, 9-17.	2.3	1
58	Efficacy of Dichlorophenolindophenol (DCIP) as Screening Test for Hb E: Revisited. Indian Journal of Hematology and Blood Transfusion, 2020, 36, 535-541.	0.6	1
59	Recovery of Lead as Lead Sulphide from Anode Slime Using Hydrometallurgical Technique. Journal of the Institution of Engineers (India): Series D, 2021, 102, 489-494.	1.0	1
60	Effect of Tween 80 on electrochemical deposition of cobalt from sulphate solutions. Russian Journal of Non-Ferrous Metals, 2016, 57, 331-337.	0.6	0