## Pasquale Iovino

List of Publications by Year in descending order

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		361413	414414	
38	1,028	20	32	
papers	citations	h-index	g-index	
20	20	20	1177	
38	38	38	1177	
all docs	docs citations	times ranked	citing authors	

#	Article	IF	CITATIONS
1	Considerations about the correct evaluation of sorption thermodynamic parameters from equilibrium isotherms. Journal of Chemical Thermodynamics, 2014, 68, 310-316.	2.0	143
2	Atrazine adsorption by acid-activated zeolite-rich tuffs. Applied Clay Science, 2010, 49, 330-335.	5.2	87
3	Degradation of ibuprofen by hydrodynamic cavitation: Reaction pathways and effect of operational parameters. Ultrasonics Sonochemistry, 2016, 29, 76-83.	8.2	84
4	Sorption of Organic Pollutants by Humic Acids: A Review. Molecules, 2020, 25, 918.	3.8	84
5	Triclosan photolysis: operating condition study and photo-oxidation pathway. Chemical Engineering Journal, 2019, 377, 121045.	12.7	40
6	Use and Misuse of Sorption Kinetic Data: A Common Mistake That Should Be Avoided. Adsorption Science and Technology, 2012, 30, 217-225.	3.2	39
7	Cr(VI) Sorption from Aqueous Solution: A Review. Applied Sciences (Switzerland), 2020, 10, 6477.	2.5	38
8	Sorption of humic acids on zeolitic tuffs. Microporous and Mesoporous Materials, 2007, 105, 324-328.	4.4	37
9	Ibuprofen photodegradation in aqueous solutions. Environmental Science and Pollution Research, 2016, 23, 22993-23004.	5.3	37
10	Diclofenac sorption from synthetic water: Kinetic and thermodynamic analysis. Journal of Environmental Chemical Engineering, 2020, 8, 104105.	6.7	35
11	Comment on "Removal of anionic dye Congo red from aqueous solution by raw pine and acid-treated pine cone powder as adsorbent: Equilibrium, thermodynamic, kinetics, mechanism and process design― Water Research, 2012, 46, 4314-4315.	11.3	34
12	Sorption of non-ionic organic pollutants onto a humic acids-zeolitic tuff adduct: Thermodynamic aspects. Chemosphere, 2014, 95, 75-80.	8.2	33
13	Temporal and spatial distribution of BTEX pollutants in the atmosphere of metropolitan areas and neighbouring towns. Environmental Monitoring and Assessment, 2009, 150, 437-44.	2.7	32
14	Degradation of Ibuprofen in Aqueous Solution with UV Light: the Effect of Reactor Volume and pH. Water, Air, and Soil Pollution, 2016, 227, 1.	2.4	31
15	Remediation of Groundwater Polluted by Aromatic Compounds by Means of Adsorption. Sustainability, 2014, 6, 4807-4822.	3.2	29
16	Photodegradation of Diclofenac Sodium Salt in Water Solution: Effect of HA, NO3 â^ and TiO2 on Photolysis Performance. Water, Air, and Soil Pollution, 2017, 228, 1.	2.4	27
17	Electro-Oxidation of Humic Acids Using Platinum Electrodes: An Experimental Approach and Kinetic Modelling. Water (Switzerland), 2020, 12, 2250.	2.7	26
18	Ibuprofen degradation in aqueous solution by using UV light. Desalination and Water Treatment, 2016, 57, 22878-22886.	1.0	24

#	Article	IF	CITATIONS
19	Modelling the biphasic sorption of simazine, imidacloprid, and boscalid in water/soil systems. Journal of Environmental Science and Health - Part B Pesticides, Food Contaminants, and Agricultural Wastes, 2014, 49, 578-590.	1.5	21
20	Some remarks on "A critical review of the estimation of the thermodynamic parameters on adsorption equilibria. Wrong use of equilibrium constant in the Van't Hoof equation for calculation of thermodynamic parameters of adsorptionâ€⊷ Journal of Molecular Liquids 273 (2019) 425–434. Journal of Molecular Liquids, 2019, 276, 529-530.	4.9	20
21	Sorption of humic acids by a zeolite-feldspar-bearing tuff in batch and fixed-bed column. Journal of Porous Materials, 2012, 19, 449-453.	2.6	14
22	Sorption of non-ionic organic pollutants onto immobilized humic acid. Desalination and Water Treatment, 2015, 56, 55-62.	1.0	14
23	Identification of stationary sources of air pollutants by concentration statistical analysis. Chemosphere, 2008, 73, 614-618.	8.2	13
24	ADSORPTION OF SIMAZINE AND BOSCALID ONTO ACID-ACTIVATED NATURAL CLINOPTILOLITE. Environmental Engineering and Management Journal, 2015, 14, 1705-1712.	0.6	13
25	Impact assessment of PM10 cement plants emissions on urban air quality using the SCIPUFF dispersion model. Environmental Monitoring and Assessment, 2016, 188, 499.	2.7	10
26	Sorption of benzene derivatives onto insolubilized humic acids. Chemical Papers, 2018, 72, 929-935.	2.2	8
27	Catalytic effect of dissolved humic acids on the chemical degradation of phenylurea herbicides. Pest Management Science, 2008, 64, 768-774.	3.4	7
28	Sorption Equilibrium of Aromatic Pollutants onto Dissolved Humic Acids. Water, Air, and Soil Pollution, 2017, 228, 1.	2.4	7
29	Electrochemical Removal of Humic Acids from Water Using Aluminum Anode: Influence of Chloride Ion and Current Parameters. Journal of Chemistry, 2019, 2019, 1-6.	1.9	7
30	Macromolecular Structure of a Commercial Humic Acid Sample. Environments - MDPI, 2020, 7, 32.	3.3	7
31	Contribution of vehicular traffic and industrial facilities to PM10 concentrations in a suburban area of Caserta (Italy). Environmental Science and Pollution Research, 2014, 21, 13169-13174.	<b>5.</b> 3	5
32	Sorption of a Cationic Surfactant Benzyldimethyldodecyl Ammonium Chloride onto a Natural Zeolite. Water, Air, and Soil Pollution, 2016, 227, 1.	2.4	5
33	Comments on "Re-evaluation of the century-old Langmuir isotherm for modeling adsorption phenomena in solution― Chemical Physics, 2019, 517, 270-271.	1.9	5
34	Background Atmospheric Levels of Aldehydes, BTEX and PM10 Pollutants in a Medium‧ized City of Southern Italy. Annali Di Chimica, 2007, 97, 597-604.	0.6	4
35	A Phenomenological Interpretation of Two-Step Adsorption Kinetics of Humic Acids on Zeolitic Tuff. Adsorption Science and Technology, 2013, 31, 373-384.	3.2	3
36	Experimental analysis of benzene derivative adsorption in single and binary systems using activated carbon. International Journal of Environment and Waste Management, 2015, 16, 336.	0.3	2

#	Article	lF	CITATIONS
37	Thermodynamics of Clay Minerals-Humic Acids Interaction. Advanced Science Letters, 2017, 23, 5859-5861.	0.2	2
38	Sorption of benzene derivatives onto a humic acid-zeolitic tuff adduct. Environmental Science and Pollution Research, 2018, 25, 26831-26836.	5.3	1