

Camã©lia Matei Ghimbeu

List of Publications by Year in descending order

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113
papers

4,691
citations

100601

38
h-index

129628

63
g-index

118
all docs

118
docs citations

118
times ranked

7328
citing authors

#	ARTICLE	IF	CITATIONS
1	Olive Mill by-Products Thermochemical Conversion via Hydrothermal Carbonization and Slow Pyrolysis: Detailed Comparison between the Generated Hydrochars and Biochars Characteristics. Processes, 2022, 10, 231.	1.3	13
2	The role of specific and active surface areas in optimizing hard carbon irreversible capacity loss in sodium ion batteries. Energy Advances, 2022, 1, 185-190.	1.4	11
3	Vegetal-Extracted Polyphenols as a Natural Hard Carbon Anode Source for Na-Ion Batteries. ACS Applied Energy Materials, 2022, 5, 4774-4787.	2.5	6
4	Impact of the biomass precursor composition in the hard carbon properties and performance for application in a Na-ion battery. Fuel Processing Technology, 2022, 231, 107223.	3.7	13
5	Performance evaluation of electrochemical capacitors with activated carbon spheres as electrode material and aqueous electrolyte. Journal of Power Sources, 2022, 542, 231714.	4.0	4
6	Hard carbon derived from coconut shells, walnut shells, and corn silk biomass waste exhibiting high capacity for Na-ion batteries. Journal of Energy Chemistry, 2021, 58, 207-218.	7.1	89
7	Application of olive mill waste-based biochars in agriculture: Impact on soil properties, enzymatic activities and tomato growth. Science of the Total Environment, 2021, 755, 142531.	3.9	58
8	Link between Alkali Metals in Salt Templates and in Electrolytes for Improved Carbon-Based Electrochemical Capacitors. ACS Applied Materials & Interfaces, 2021, 13, 2584-2599.	4.0	20
9	Hard carbon porosity revealed by the adsorption of multiple gas probe molecules (N ₂ , Ar, CO ₂ , CH ₄ , H ₂ , O ₂ , H ₂ O, H ₂ S, SO ₂ , H ₂ CO, H ₂ CO ₃ , H ₂ CO ₃ ·nH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ , H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O·uH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O·uH ₂ O·vH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O·uH ₂ O·vH ₂ O·wH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O·uH ₂ O·vH ₂ O·wH ₂ O·xH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O·uH ₂ O·vH ₂ O·wH ₂ O·xH ₂ O·yH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O·uH ₂ O·vH ₂ O·wH ₂ O·xH ₂ O·yH ₂ O·zH ₂ O, H ₂ CO ₃ ·nH ₂ O·mNH ₃ ·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O·uH 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O·fH ₂ O·gH ₂ O·hH ₂ O·iH ₂ O·jH ₂ O·kH ₂ O·lH ₂ O·mH ₂ O·nH ₂ O·oH ₂ O·pH ₂ O·qH ₂ O·rH ₂ O·sH ₂ O·tH ₂ O·uH ₂ O·vH ₂ O·wH ₂ O·xH ₂ O·yH ₂ O·zH ₂ O·aH ₂ O·b		

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