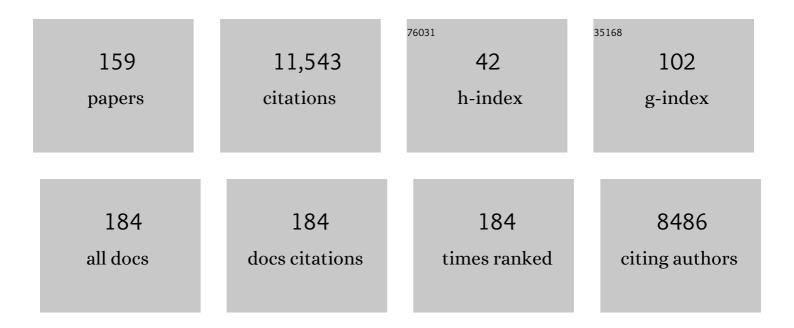
Michel J Rossi

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Coexistence of reactive functional groups at the interface of a powdered activated amorphous carbon: a molecular view. Molecular Physics, 2021, 119, .	0.8	3
2	Formation of highly oxygenated organic molecules from aromatic compounds. Atmospheric Chemistry and Physics, 2018, 18, 1909-1921.	1.9	133
3	The influence of HCl on the evaporation rates of H ₂ O over water ice in the range 188 to 210 K at small average concentrations. Atmospheric Chemistry and Physics, 2018, 18, 15903-15919.	1.9	1
4	Chemical characterization of diesel and hydrotreated vegetable oil (HVO) soot after reactive gas probing using diffuse reflectance FTIR spectroscopy (DRIFTS). Environmental Science and Pollution Research, 2017, 24, 7534-7543.	2.7	8
5	Reactive Uptake of Sulfur Dioxide and Ozone on Volcanic Glass and Ash at Ambient Temperature. Journal of Geophysical Research D: Atmospheres, 2017, 122, 10,077.	1.2	19
6	Chemical characterization of atmospheric ions at the high altitude research station Jungfraujoch (Switzerland). Atmospheric Chemistry and Physics, 2017, 17, 2613-2629.	1.9	24
7	Perspectives on the Future of Ice Nucleation Research: Research Needs and Unanswered Questions Identified from Two International Workshops. Atmosphere, 2017, 8, 138.	1.0	56
8	Frontispiece: Metastable Nitric Acid Trihydrate in Ice Clouds. Angewandte Chemie - International Edition, 2016, 55, .	7.2	1
9	Metastabiles SalpetersÃ ¤ retrihydrat in Eiswolken. Angewandte Chemie, 2016, 128, 3334-3338.	1.6	0
10	Controls on the surface chemical reactivity of volcanic ash investigated with probe gases. Earth and Planetary Science Letters, 2016, 450, 254-262.	1.8	19
11	Heterogeneous kinetics of H ₂ O, HNO ₃ and HCl on HNO ₃ hydrates (<i>α</i> -NAT, <i>β</i> -NAT, NAD) in the range 175–200â€~K. Atmospheric Chemistry and Physics, 2016, 16, 11937-11960.	1.9	3
12	Frontispiz: Metastabiles SalpetersÃ ¤ retrihydrat in Eiswolken. Angewandte Chemie, 2016, 128, .	1.6	0
13	Metastable Nitric Acid Trihydrate in Ice Clouds. Angewandte Chemie - International Edition, 2016, 55, 3276-3280.	7.2	12
14	Molecular Characterization of the Gas–Particle Interface of Soot Sampled from a Diesel Engine Using a Titration Method. Environmental Science & Technology, 2016, 50, 2946-2955.	4.6	15
15	Quantitative Aspects of the Interfacial Catalytic Oxidation of Dithiothreitol by Dissolved Oxygen in the Presence of Carbon Nanoparticles. Environmental Science & (amp; Technology, 2016, 50, 996-1004.	4.6	18
16	The midâ€IR Absorption Cross Sections of <i>α</i> ―and <i>β</i> â€NAT (HNO ₃ • 3H <su the range 170 to 185 K and of metastable NAD (HNO₃ • 2H₂O) in the ra 182 K. Journal of Geophysical Research D: Atmospheres, 2015, 120, 11,707.</su 		
17	The Kinetics of the Reaction C ₂ H ₅ [•] + Hl → C ₂ H ₆ + I [•] over an Extended T (213–623 K): Experiment and Modeling. Zeitschrift Fur Physikalische Chemie, 2015, 229, 1475-1501.	ennperatu	re\$Range
18	The use of heterogeneous chemistry for the characterization of functional groups at the gas/particle interface of soot from a diesel engine at a particular running condition. Environmental Science and Pollution Research, 2015, 22, 4863-4872.	2.7	12

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19	The Measurement of the Rate Parameters for the Reactions i-C3 H7• and n -C3 H7• + HI → C3 H8 + I• over Temperature Range 293-623 K: Implications for the Standard Heat of Formation of the Propyl Radicals. International Journal of Chemical Kinetics, 2014, 46, 305-320.	er the 1.0	6
20	The Reinvestigation of the Kinetics of the Metathesis Reactions t-C ₄ H ₉ [•] + HBr (HI) â†' i-C ₄ H ₁₀ + Br [•] (I [•]) and of the t-C ₄ H ₉ [•] Free Radical Thermochemistry. Journal of Physical Chemistry A, 2014, 118, 5135-5148.	1.1	9
21	H ₂ O and HCl trace gas kinetics on crystalline HCl hydrates and amorphous HCl / H ₂ O in the range 170 to 205 K: the HCl / H ₂ O phase diagram revisited. Atmospheric Chemistry and Physics. 2014. 14. 5183-5204.	1.9	7
22	Î′ 15N measurement of organic and inorganic substances by EA-IRMS: a speciation-dependent procedure. Analytical and Bioanalytical Chemistry, 2013, 405, 159-176.	1.9	30
23	Reinvestigation of the Elementary Chemical Kinetics of the Reaction C2H5• + HBr (HI) → C2H6 + Br• (l•) the Range 293–623 K and Its Implication on the Thermochemical Parameters of C2H5• Free Radical. Journal of Physical Chemistry A, 2013, 117, 11383-11402.	in 1.1	14
24	Transient mid-IR study of electron dynamics in TiO2 conduction band. Analyst, The, 2013, 138, 1966.	1.7	19
25	Comparison of Three Acellular Tests for Assessing the Oxidation Potential of Nanomaterials. Aerosol Science and Technology, 2013, 47, 218-227.	1.5	52
26	Effusive molecular beam-sampled Knudsen flow reactor coupled to vacuum ultraviolet single photon ionization mass spectrometry using an external free radical source. Review of Scientific Instruments, 2013, 84, 114104.	0.6	2
27	Evaluated kinetic and photochemical data for atmospheric chemistry: Volume VI – heterogeneous reactions with liquid substrates. Atmospheric Chemistry and Physics, 2013, 13, 8045-8228.	1.9	167
28	The metastable HCl · 6H ₂ O phase – IR spectroscopy, phase transitions and kinetic/thermodynamic properties in the range 170–205 K. Atmospheric Chemistry and Physics, 2013, 13, 11905-11923.	1.9	8
29	Corrigendum to "Evaluated kinetic and photochemical data for atmospheric chemistry: Volume V – heterogeneous reactions on solid substrates" published in Atmos. Chem. Phys. 10, 9059–9223, 2010. Atmospheric Chemistry and Physics, 2013, 13, 7359-7359.	1.9	9
30	Coating carbon nanotubes with a polystyrene-based polymer protects against pulmonary toxicity. Particle and Fibre Toxicology, 2011, 8, 3.	2.8	74
31	Biomarkers of oxidative stress and its association with the urinary reducing capacity in bus maintenance workers. Journal of Occupational Medicine and Toxicology, 2011, 6, 18.	0.9	39
32	Evaluated kinetic and photochemical data for atmospheric chemistry: Volume V – heterogeneous reactions on solid substrates. Atmospheric Chemistry and Physics, 2010, 10, 9059-9223.	1.9	312
33	Probing Functional Groups at the Gas–Aerosol Interface Using Heterogeneous Titration Reactions: A Tool for Predicting Aerosol Health Effects?. ChemPhysChem, 2010, 11, 3823-3835.	1.0	23
34	Heterogeneous Chemistry of Cl ₂ O and HOCl on Frozen Natural Sea Salt, Recrystallized Sea Salt, KCl and NaCl Solutions at 200 and 215 K. Zeitschrift Fur Physikalische Chemie, 2010, 224, 1119-1150.	1.4	0
35	An overview of current issues in the uptake of atmospheric trace gases by aerosols and clouds. Atmospheric Chemistry and Physics, 2010, 10, 10561-10605.	1.9	352
36	Characterization of surface functional groups present on laboratory-generated and ambient aerosol particles by means of heterogeneous titration reactions. Journal of Aerosol Science, 2009, 40, 534-548.	1.8	12

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37	The use of heterogeneous chemistry for the characterization of functional groups at the gas/particle interface of soot and TiO2 nanoparticles. Physical Chemistry Chemical Physics, 2009, 11, 6205. Thermochemical properties from ab initio calculations: π〕and Ïfã€Free radicals of importance in soot	1.3	31
38	formation: [•] C ₃ H ₃ (propargyl), [•] C ₄ H ₃ , [•] C ₁₃ H ₉ (phenalenyl), [•] C ₆ H ₅ (phenyl), [•] C ₁₀ H ₇ (naphthyl), [•] C ₁₄ H ₉ (anthryl),		12
39	[•] C ₁₄ H ₉ (phenanthryl), [•] C ₁₆ H _{9Adverse Effects of Industrial Multiwalled Carbon Nanotubes on Human Pulmonary Cells. Journal of Toxicology and Environmental Health - Part A: Current Issues, 2008, 72, 60-73.})> 1.1	129
40	Evaluated kinetic and photochemical data for atmospheric chemistry: Volume IV – gas phase reactions of organic halogen species. Atmospheric Chemistry and Physics, 2008, 8, 4141-4496.	1.9	221
41	Heterogeneous Chemistry of the NO3 Free Radical and N2O5 on Decane Flame Soot at Ambient Temperature:  Reaction Products and Kinetics. Journal of Physical Chemistry A, 2007, 111, 1914-1926.	1.1	44
42	The heterogeneous kinetics of HOBr and HOCl on acidified sea salt and model aerosol at 40–90% relative humidity and ambient temperature. Physical Chemistry Chemical Physics, 2006, 8, 3988-4001.	1.3	24
43	The Kinetics of H2O Vapor Condensation and Evaporation on Different Types of Ice in the Range 130â^'210 K. Journal of Physical Chemistry A, 2006, 110, 3042-3058.	1.1	47
44	Uptake of CO2, SO2, HNO3and HCl on Calcite (CaCO3) at 300 K:Â Mechanism and the Role of Adsorbed Waterâ€. Journal of Physical Chemistry A, 2006, 110, 6789-6802.	1.1	64
45	The heterogeneous chemical kinetics of N ₂ O ₅ on CaCO ₃ and other atmospheric mineral dust surrogates. Atmospheric Chemistry and Physics. 2006. 6. 1373-1388.	1.9	70
46	Evaluated kinetic and photochemical data for atmospheric chemistry: Volume II – gas phase reactions of organic species. Atmospheric Chemistry and Physics, 2006, 6, 3625-4055.	1.9	1,508
47	The heterogeneous decomposition of ozone on atmospheric mineral dust surrogates at ambient temperature. International Journal of Chemical Kinetics, 2006, 38, 407-419.	1.0	15
48	IUPAC Critical Evaluation of Thermochemical Properties of Selected Radicals. Part 1 ChemInform, 2005, 36, no.	0.1	1
49	The Heterogeneous Reaction of NO2 with NH4Cl: A Molecular Diffusion Tube Study. Journal of Atmospheric Chemistry, 2005, 50, 171-194.	1.4	5
50	The heterogeneous chemical kinetics of NO3 on atmospheric mineral dust surrogates. Physical Chemistry Chemical Physics, 2005, 7, 3150.	1.3	57
51	The heterogeneous interaction of HOCl with solid KBr substrates: The catalytic role of adsorbed halogens. Physical Chemistry Chemical Physics, 2005, 7, 2599.	1.3	6
52	Influence of Monolayer Amounts of HNO3 on the Evaporation Rate of H2O over Ice in the Range 179 to 208 K:  A Quartz Crystal Microbalance Study. Journal of Physical Chemistry A, 2005, 109, 7151-7165.	1.1	21
53	IUPAC Critical Evaluation of Thermochemical Properties of Selected Radicals. Part I. Journal of Physical and Chemical Reference Data, 2005, 34, 573-656.	1.9	283
54	Thermochemical properties from G3MP2B3 calculations, Set-2: Free radicals with special consideration of CH2?CH?C??CH2, cyclo??C5H5,?CH2OOH, HO??CO, and HC(O)O?. International Journal of Chemical Kinetics, 2004, 36, 661-686.	1.0	35

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55	Heterogeneous interaction of Br2, Cl2 and Cl2O with solid KBr and NaCl substrates: The role of adsorbed H2O and halogens. Physical Chemistry Chemical Physics, 2004, 6, 3447.	1.3	6
56	The kinetics of condensation and evaporation of H2O from pure ice in the range 173–223 K: a quartz crystal microbalance study. Physical Chemistry Chemical Physics, 2004, 6, 4665-4676.	1.3	33
57	Chemical Kinetics of the Interaction of H2O Vapor with Soot in the Range 190 K ≤ ≤300 K:  A Diffusion Tube Study. Journal of Physical Chemistry A, 2004, 108, 10667-10680.	n 1.1	23
58	Evaluated kinetic and photochemical data for atmospheric chemistry: Volume I - gas phase reactions of O _x , HO _x , NO _x and SO _x species. Atmospheric Chemistry and Physics, 2004, 4, 1461-1738.	1.9	1,597
59	Kinetic model for non-sticky collisions in pulsed molecular diffusion tube experiments. Surface Science, 2003, 542, 150-159.	0.8	0
60	Heterogeneous Reactions on Salts. Chemical Reviews, 2003, 103, 4823-4882.	23.0	194
61	Common Precursor Mechanism for the Heterogeneous Reaction of D2O, HCl, HBr, and HOBr with Water Ice in the Range 170â^'230 K:  Mass Accommodation Coefficients on Ice. Journal of Physical Chemistry A, 2003, 107, 4103-4115.	1.1	28
62	The nature of the interface and the diffusion coefficient of HCl/ice and HBr/ice in the temperature range 190–205 K. Physical Chemistry Chemical Physics, 2003, 5, 4157-4169.	1.3	10
63	The rate of water vapor evaporation from ice substrates in the presence of HCl and HBr: implications for the lifetime of atmospheric ice particles. Atmospheric Chemistry and Physics, 2003, 3, 1131-1145.	1.9	35
64	Heterogeneous Hydrolysis and Reaction of BrONO2 and Br2O on Pure Ice and Ice Doped with HBr. Journal of Physical Chemistry A, 2002, 106, 5891-5901.	1.1	12
65	Heterogeneous reactions of HNO3with flame soot generated under different combustion conditions. Reaction mechanism and kinetics. Physical Chemistry Chemical Physics, 2002, 4, 5110-5118.	1.3	32
66	Can soot particles emitted by airplane exhaust contribute to the formation of aviation contrails and cirrus clouds?. Geophysical Research Letters, 2002, 29, 1-1-1-4.	1.5	24
67	Thermochemical properties of free radicals from G3MP2B3 calculations. International Journal of Chemical Kinetics, 2002, 34, 550-560.	1.0	53
68	Flame soot generated under controlled combustion conditions: Heterogeneous reaction of NO2 on hexane soot. International Journal of Chemical Kinetics, 2002, 34, 620-631.	1.0	41
69	The kinetics of the uptake of HNO3 on ice, solid H2SO4–H2O and solid ternary solutions of H2SO4–HNO3–H2O in the temperature range 180–211 K. Physical Chemistry Chemical Physics, 2001, 3, 3707-3716.	1.3	23
70	Evaluated Kinetic and Photochemical Data for Atmospheric Chemistry: Supplement VIII, Halogen Species Evaluation for Atmospheric Chemistry. Journal of Physical and Chemical Reference Data, 2000, 29, 167-266.	1.9	183
71	Properties of the HCl/Ice, HBr/Ice, and H2O/Ice Interface at Stratospheric Temperatures (200 K) and Its Importance for Atmospheric Heterogeneous Reactions. Journal of Physical Chemistry A, 2000, 104, 11739-11750.	1.1	32
72	Heterogeneous Chemistry of HOBr on Different Types of Ice and on Ice Doped with HCl, HBr, and HNO3at 175 K <t< 104,="" 2000,="" 215="" 7268-7277.<="" a,="" chemistry="" journal="" k.="" of="" physical="" td=""><td>1.1</td><td>12</td></t<>	1.1	12

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73	Reactivity of NO2 and H2O on soot generated in the laboratory: a diffusion tube study at ambient temperature. Physical Chemistry Chemical Physics, 2000, 2, 5584-5593.	1.3	41
74	The reactivity of NO2 and HONO on flame soot at ambient temperature: The influence of combustion conditions. Physical Chemistry Chemical Physics, 2000, 2, 5420-5429.	1.3	87
75	Evaluated Kinetic and Photochemical Data for Atmospheric Chemistry, Organic Species: Supplement VII. Journal of Physical and Chemical Reference Data, 1999, 28, 191-393.	1.9	338
76	Heterogeneous reactivity of the nitrate radical: reactions on halogen salt at ambient temperature and on ice in the presence of HX (X=Cl, Br, I) at 190 K. Physical Chemistry Chemical Physics, 1999, 1, 2257-2266.	1.3	21
77	The kinetics of the heterogeneous reaction of BrONO2 with solid alkali halides at ambient temperature. A comparison with the interaction of ClONO2 on NaCl and KBr. Physical Chemistry Chemical Physics, 1999, 1, 4337-4346.	1.3	22
78	A molecular diffusion tube study of N2O5 and HONO2 interacting with NaCl and KBr at ambient temperature. Physical Chemistry Chemical Physics, 1999, 1, 2687-2694.	1.3	19
79	The heterogeneous reaction of N ₂ O ₅ with HBr on Ice comparison with N ₂ O ₅ +HCl. Zeitschrift Fur Elektrotechnik Und Elektrochemie, 1998, 102, 811-820.	0.9	16
80	Real time kinetics and thermochemistry of the uptake of HCl, HBr and HI on water ice in the temperature range 190 to 210 K. Zeitschrift Fur Elektrotechnik Und Elektrochemie, 1998, 102, 915-928.	0.9	26
81	Real-Time Kinetic Measurements of the Condensation and Evaporation of D2O Molecules on Ice at 140 K <t< 102,="" 10300-10309.<="" 1998,="" 220="" a,="" chemistry="" journal="" k.="" of="" physical="" td=""><td>1.1</td><td>41</td></t<>	1.1	41
82	Reactivity of BrNO2and ClNO2with Solid Alkali Salt Substrates. Journal of Physical Chemistry A, 1998, 102, 7470-7479.	1.1	12
83	Heterogeneous Kinetics of the Uptake of HOBr on Solid Alkali Metal Halides at Ambient Temperatureâ€. Journal of Physical Chemistry A, 1998, 102, 4819-4828.	1.1	38
84	Direct Measurement of Surface Residence Times:Â Nitryl Chloride and Chlorine Nitrate on Alkali Halides at Room Temperature. Journal of Physical Chemistry A, 1998, 102, 9193-9201.	1.1	14
85	The chemical kinetics of HONO formation resulting from heterogeneous interaction of NO2with flame soot. Geophysical Research Letters, 1998, 25, 2453-2456.	1.5	108
86	Paper II: Simulation of flow conditions in low-pressure flow reactors (Knudsen cells) using a Monte Carlo technique. Review of Scientific Instruments, 1997, 68, 3180-3186.	0.6	29
87	Paper I: Design and construction of a Knudsen-cell reactor for the study of heterogeneous reactions over the temperature range 130–750 K: Performances and limitations. Review of Scientific Instruments, 1997, 68, 3172-3179.	0.6	84
88	Evaluated Kinetic and Photochemical Data for Atmospheric Chemistry: Supplement VI. IUPAC Subcommittee on Gas Kinetic Data Evaluation for Atmospheric Chemistry. Journal of Physical and Chemical Reference Data, 1997, 26, 1329-1499.	1.9	661
89	Real-Time Kinetics of the Uptake of ClONO2on Ice and in the Presence of HCl in the Temperature Range 160 K ≤≤200 K. Journal of Physical Chemistry A, 1997, 101, 1903-1911.	1.1	69
90	Heterogeneous Reaction of NO3with Ice and Sulfuric Acid Solutions:Â Upper Limits for the Uptake Coefficients. Journal of Physical Chemistry A, 1997, 101, 4110-4113.	1.1	16

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91	Evaluated Kinetic, Photochemical and Heterogeneous Data for Atmospheric Chemistry: Supplement V. IUPAC Subcommittee on Gas Kinetic Data Evaluation for Atmospheric Chemistry. Journal of Physical and Chemical Reference Data, 1997, 26, 521-1011.	1.9	903
92	The heterogeneous reaction of NO3with NaCl and KBr: A nonphotolytic source of halogen atoms. Geophysical Research Letters, 1997, 24, 2757-2760.	1.5	37
93	The heterogeneous reaction of HONO and HBr on ice and on sulfuric acid. Zeitschrift Fur Elektrotechnik Und Elektrochemie, 1997, 101, 943-955.	0.9	12
94	Gas-phase UV spectroscopy of anthracene, xanthone, pyrene, 1-bromopyrene and 1,2,4-trichlorobenzene at elevated temperatures. Journal of Photochemistry and Photobiology A: Chemistry, 1997, 104, 25-33.	2.0	43
95	UV photon-assisted incineration of polycyclic aromatic hydrocarbons at elevated temperatures between 150 and 800 °C. Journal of Photochemistry and Photobiology A: Chemistry, 1997, 109, 267-280.	2.0	4
96	The heterogeneous formation of N2O in the presence of acidic solutions: Experiments and modeling. International Journal of Chemical Kinetics, 1997, 29, 869-891.	1.0	17
97	Real-time measurement of residence times of gas molecules on solid surfaces. Chemical Physics Letters, 1997, 275, 253-260.	1.2	15
98	Enhanced Thermal Destruction of Anthracene Vapor upon Laser Irradiation at 248 nm in the 150â^'800 °C Range. Environmental Science & Technology, 1996, 30, 1789-1793.	4.6	3
99	The heterogeneous formation of N2O over bulk condensed phases in the presence of SO2 at high humidities. Journal of Atmospheric Chemistry, 1996, 25, 229-250.	1.4	12
100	Heterogeneous Kinetics of N2O5Uptake on Salt, with a Systematic Study of the Role of Surface Presentation (for N2O5and HNO3). The Journal of Physical Chemistry, 1996, 100, 1008-1019.	2.9	97
101	Heterogeneous Kinetics of the Uptake of ClONO2on NaCl and KBr. The Journal of Physical Chemistry, 1996, 100, 7494-7501.	2.9	37
102	Heterogeneous Kinetics of HONO on H2SO4Solutions and on Ice:Â Activation of HCl. The Journal of Physical Chemistry, 1996, 100, 13765-13775.	2.9	36
103	The heterogeneous interaction of Br(2P3/2) and Br(2P1/2) with surfaces of Teflonâ,,¢ and polycrystalline nickel. International Journal of Chemical Kinetics, 1995, 27, 403-418.	1.0	8
104	Experimental evidence for the efficient "dry deposition―of nitric acid on calcite. Atmospheric Environment, 1995, 29, 3365-3372.	1.9	57
105	The heterogeneous generation of N2O from exhaust gases of combustion: A laboratory study. Geophysical Research Letters, 1995, 22, 3509-3512.	1.5	2
106	Kinetics of Nitric Acid Uptake by Salt. The Journal of Physical Chemistry, 1994, 98, 9801-9810.	2.9	97
107	Relationship between bond dissociation energies and activation energies for bond scission reactions. International Journal of Chemical Kinetics, 1994, 26, 211-217.	1.0	20
108	Kinetic and mechanistic aspects of the NO oxidation by O2 in aqueous phase. International Journal of Chemical Kinetics, 1994, 26, 1207-1227.	1.0	98

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109	The thermal decomposition of the new energetic material ammoniumdinitramide (NH4N(NO2)2) in relation to Nitramide (NH2NO2) and NH4NO3. International Journal of Chemical Kinetics, 1993, 25, 549-570.	1.0	78
110	The heterogeneous interaction of NO ₂ with amorphous carbon. Geophysical Research Letters, 1993, 20, 1431-1434.	1.5	18
111	Pump-and-probe lidar for in-situ probing of atmospheric chemistry. , 1992, 1714, 291.		2
112	Interaction of chlorine atom(2P3/2) and chlorine atom(2P1/2) with polycrystalline nickel surfaces. The Journal of Physical Chemistry, 1991, 95, 825-834.	2.9	10
113	The interaction of Cl(2P3/2) and Cl(2P1/2) with nâ€Si(100): Spontaneous etching. Journal of Vacuum Science and Technology A: Vacuum, Surfaces and Films, 1991, 9, 217-222.	0.9	9
114	A pulsed halogen atom source for kinetic measurements in a Knudsen cell reactor. Review of Scientific Instruments, 1990, 61, 1217-1222.	0.6	12
115	Sticking coefficient of the SiH2 free radical on a hydrogenated silicon arbon surface. Applied Physics Letters, 1989, 54, 185-187.	1.5	23
116	Atom―and radicalâ€surface sticking coefficients measured using resonanceâ€enhanced multiphoton ionization. Journal of Chemical Physics, 1989, 91, 5037-5049.	1.2	27
117	Rate Constants for the Reactions t-C4H9+DXâ†'i-C4H9D+X(X = Br,I), 295T(K) < 384: Heat of Formation of the tert-Butyl Radical. Journal of the American Chemical Society, 1989, 111, 956-962.	6.6	21
118	High density chemiluminescence studies: explosive decomposition of solid phase tetramethyldioxetane. Journal of Photochemistry and Photobiology A: Chemistry, 1988, 42, 73-85.	2.0	3
119	Heterogeneous interactions of chlorine nitrate, hydrogen chloride, and nitric acid with sulfuric acid surfaces at stratospheric temperatures. Geophysical Research Letters, 1988, 15, 847-850.	1.5	161
120	Summary Abstract: Spontaneous thermal etching of silicon by CF3 radicals. Journal of Vacuum Science and Technology A: Vacuum, Surfaces and Films, 1988, 6, 1407-1408.	0.9	3
121	Optical switching of a dc discharge using an excimer laser. Journal of Applied Physics, 1988, 63, 4849-4853.	1.1	2
122	Reaction probability for the spontaneous etching of silicon by CF3 free radicals. Journal of Vacuum Science & Technology an Official Journal of the American Vacuum Society B, Microelectronics Processing and Phenomena, 1988, 6, 1632.	1.6	23
123	[3+2] resonance enhanced multiphoton ionization of I and Br formed from the infrared multiphoton decomposition of CF3I and CF3Br. Journal of Chemical Physics, 1988, 89, 2925-2931.	1.2	11
124	Antarctic Ozone Depletion Chemistry: Reactions of N2O5 with H2O and HCl on Ice Surfaces. Science, 1988, 240, 1018-1021.	6.0	143
125	Atom- and Radical-Surface Sticking Coefficients Measured Using Resonance Enhanced Multiphoton Ionization (REMPI). Materials Research Society Symposia Proceedings, 1988, 131, 251.	0.1	0

126 Chemiluminescence From Laser-Heated Solid And Gaseous Tetramethyldioxetane., 1988,,.

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127	In situ radical detection under very low pressure photolysis conditions using resonance-enhanced multiphoton ionization. Kinetics of trifluoromethyl radicals produced from IR multiphoton dissociation of hexafluoroacetone. The Journal of Physical Chemistry, 1988, 92, 5338-5347.	2.9	17
128	Reaction of Chlorine Nitrate with Hydrogen Chloride and Water at Antarctic Stratospheric Temperatures. Science, 1987, 238, 1258-1260.	6.0	315
129	Multiphoton ionization of vinylchloride, trifluoroethylene, and benzene at 193 nm. Journal of Chemical Physics, 1987, 87, 902-909.	1.2	17
130	Kinetics of surface reactions of CF3 radicals. Journal of Vacuum Science and Technology A: Vacuum, Surfaces and Films, 1987, 5, 3351-3358.	0.9	10
131	Heterogeneous chemical reaction of chlorine nitrate and water on sulfuricâ€acid surfaces at room temperature. Geophysical Research Letters, 1987, 14, 127-130.	1.5	33
132	Absolute rate of recombination of CF3 radicals. Chemical Physics Letters, 1986, 124, 68-72.	1.2	34
133	The Heterogeneous Reaction of Ozone on Carbonaceous Surfaces. International Journal of Chemical Kinetics, 1986, 18, 1133-1149.	1.0	89
134	Reaction of CF3 radicals on fused silica between 320 and 530 K. Journal of Chemical Physics, 1986, 84, 2400-2407.	1.2	10
135	Quantitative aspects of benzene photoionization at 248 nm. Chemical Physics Letters, 1985, 120, 118-123.	1.2	14
136	N2O5 photolysis products investigated by fluorescence and optoacoustic techniques. International Journal of Chemical Kinetics, 1985, 17, 991-1006.	1.0	12
137	Photoenhanced electron attachment of vinylchloride and trifluoroethylene at 193 nm. Applied Physics Letters, 1985, 47, 576-578.	1.5	30
138	Aliphatic carbon-hydrogen bond scission processes in diphenylmethane and 2-benzyl- and 4-benzylpyridine. The heat of formation of the diphenylmethyl and .alphaphenylethyl radical in the gas phase. The Journal of Physical Chemistry, 1984, 88, 5031-5039.	2.9	42
139	Laser-induced kinetics: Arrhenius parameters fort-C4H9 + XI ?i-C4H9X + I (X = H,D) and the Heat of Formation of thet-butyl radical. International Journal of Chemical Kinetics, 1983, 15, 1283-1300.	1.0	18
140	Pulsed field desorption mass spectrometry: A technique for investigating field desorption ion formation mechanisms. International Journal of Mass Spectrometry and Ion Physics, 1983, 49, 319-335.	1.3	8
141	Energyâ€dependent energy transfer: Deactivation of azulene (S0, Evib) by 17 collider gases. Journal of Chemical Physics, 1983, 78, 6695-6708.	1.2	159
142	Infrared multiphoton photophysics: Decomposition of CnF2n+1I (n = 1, 2, 3). Journal of Chemical Physics, 1982, 76, 406-416.	1.2	18
143	Laser-induced chemical kinetics: Absolute rate constants for the reactions ?2F2 + Br2 ? C2F5Br + ??r andn-?3F7 + Br2 ?n-C3F7Br + ??r. International Journal of Chemical Kinetics, 1982, 14, 499-506.	1.0	7
144	Infrared fluorescence and collisional energy transfer parameters for vibrationally excited azulene*(So): dependence on internal energy (Evib). Chemical Physics Letters, 1982, 85, 21-26.	1.2	45

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