Evangelos Miliordos

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/8842796/publications.pdf

Version: 2024-02-01

87 papers 1,759 citations

279798 23 h-index 37 g-index

90 all docs 90 docs citations

times ranked

90

1580 citing authors

#	Article	IF	Citations
1	Optimal geometries and harmonic vibrational frequencies of the global minima of water clusters (H2O) $\langle i \rangle n \langle i \rangle n \langle i \rangle = 2 \hat{a} \in 6$, and several hexamer local minima at the CCSD(T) level of theory. Journal of Chemical Physics, 2013, 139, 114302.	3.0	105
2	Benchmark Theoretical Study of the π–π Binding Energy in the Benzene Dimer. Journal of Physical Chemistry A, 2014, 118, 7568-7578.	2.5	77
3	Unusual Inorganic Biradicals: A Theoretical Analysis. Angewandte Chemie - International Edition, 2013, 52, 5736-5739.	13.8	75
4	Electronic Structure and Bonding of the Early 3d-Transition Metal Diatomic Oxides and Their Ions: ScO ^{0,±} , TiO ^{0,±} , CrO ^{0,±} , and MnO ^{0,±} ^{â€} . Journal of Physical Chemistry A, 2010, 114, 8536-8572.	2.5	72
5	Carbonyl Boron and Related Systems:  An ab Initio Study of Bâ^'X and YBâ⟨®BY (1Σg+), Where X = He, Ne, Ar, CO, CS, N2 and Y = Ar, Kr, CO, CS, N2. Journal of Physical Chemistry A, 2004, 108, 4335-4340.	, Kr. 2.5	66
6	On the Bonding Nature of Ozone (O ₃) and Its Sulfur-Substituted Analogues SO ₂ , OS ₂ , and S ₃ : Correlation between Their Biradical Character and Molecular Properties. Journal of the American Chemical Society, 2014, 136, 2808-2817.	13.7	66
7	Electronic Structure of Vanadium Oxide. Neutral and Charged Species, VOO,±. Journal of Physical Chemistry A, 2007, 111, 1953-1965.	2.5	65
8	First principles study of the ground and excited states of FeO, FeO+, and FeOâ^2. Journal of Chemical Physics, 2011, 134, 234308.	3.0	63
9	An accurate and efficient computational protocol for obtaining the complete basis set limits of the binding energies of water clusters at the MP2 and CCSD(T) levels of theory: Application to $(H2O) < i > m < /i > m < /i > = 2-6, 8, 11, 16, and 17. Journal of Chemical Physics, 2015, 142, 234303.$	3.0	54
10	Enantioselective Syntheses of Homopropargylic Alcohols via Asymmetric Allenylboration. Organic Letters, 2018, 20, 3810-3814.	4.6	51
11	Enantioselective Allenylation of Aldehydes via $\mathrm{Br}\tilde{\mathrm{A}}_{,n}$ nsted Acid Catalysis. Advanced Synthesis and Catalysis, 2018, 360, 4634-4639.	4.3	47
12	On the validity of the basis set superposition error and complete basis set limit extrapolations for the binding energy of the formic acid dimer. Journal of Chemical Physics, 2015, 142, 094311.	3.0	40
13	Aufbau Rules for Solvated Electron Precursors: Be(NH ₃) ₄ ^{0,±} Complexes and Beyond. Journal of Physical Chemistry Letters, 2018, 9, 84-88.	4.6	37
14	The Origin of the Reactivity of the Criegee Intermediate: Implications for Atmospheric Particle Growth. Angewandte Chemie - International Edition, 2016, 55, 1015-1019.	13.8	36
15	Ligand field effects on the ground and excited states of reactive FeO ²⁺ species. Physical Chemistry Chemical Physics, 2018, 20, 28786-28795.	2.8	29
16	Molecules mimicking atoms: monomers and dimers of alkali metal solvated electron precursors. Physical Chemistry Chemical Physics, 2018, 20, 24186-24191.	2.8	29
17	<i>Ab initio</i> investigation of the electronic structure and bonding of BH, BHâ^, and HBBH molecules. Journal of Chemical Physics, 2008, 128, 144308.	3.0	28
18	Th(<scp>iv</scp>) and Ce(<scp>iv</scp>) napthylsalophen sandwich complexes: characterization of unusual thorium fluorescence in solution and solid-state. Chemical Communications, 2017, 53, 11984-11987.	4.1	27

#	Article	IF	Citations
19	Selective Activation of the Câ^'H Bond in Methane by Single Platinum Atomic Anions. Angewandte Chemie - International Edition, 2019, 58, 7773-7777.	13.8	27
20	Quantum mechanical calculation of the collision-induced absorption spectra of N2–N2 with anisotropic interactions. Journal of Chemical Physics, 2015, 142, 084306.	3.0	26
21	Simultaneous Functionalization of Methane and Carbon Dioxide Mediated by Single Platinum Atomic Anions. Journal of the American Chemical Society, 2020, 142, 21556-21561.	13.7	24
22	Low energy isomers of (H2O)25 from a hierarchical method based on Monte Carlo temperature basin paving and molecular tailoring approaches benchmarked by MP2 calculations. Journal of Chemical Physics, 2014, 141, 164304.	3.0	23
23	Isotopomer-selective spectra of a single <i>intact</i> H2O molecule in the Cs+(D2O)5H2O isotopologue: Going beyond pattern recognition to harvest the structural information encoded in vibrational spectra. Journal of Chemical Physics, 2016, 144, 074305.	3.0	23
24	Aufbau Principle for Diffuse Electrons of Double-Shell Metal Ammonia Complexes: The Case of $M(NH < sub > 3 < /sub >) < sub > 4 < /sub > @12NH < sub > 3 < /sub >, M = Li, Be < sup > + < /sup >, B < sup > 2 + < /sup >. Journal of Physical Chemistry A, 2020, 124, 505-512.$	2.5	23
25	Highly Diastereo―and Enantioselective Synthesis of 3,6′â€Bisboryl―anti â€1,2â€oxaborinanâ€3â€enes: An I Enantioenriched Homoallylic Alcohols with A Stereodefined Trisubstituted Alkene. Angewandte Chemie - International Edition, 2021, 60, 840-848.	Entry to 13.8	23
26	Ground and Excited States of the [Fe(H ₂ 0) ₆] ²⁺ and [Fe(H ₂ 0) ₆] ³⁺ Clusters: Insight into the Electronic Structure of the [Fe(H ₂ 0) ₆] ³⁺ êf [Fe(H ₂ 0) ₆] ³⁺	5.3	22
27	Complex. Journal of Chemical Theory and Computation, 2015, 11, 1549-1563. Communication: Water activation and splitting by single metal-atom anions. Journal of Chemical Physics, 2018, 149, 221101.	3.0	22
28	Stability and Electronic Features of Calcium Hexa-, Hepta-, and Octa-Coordinated Ammonia Complexes: A First-Principles Study. Journal of Physical Chemistry A, 2019, 123, 6744-6750.	2.5	22
29	Superatomic nature of alkaline earth metal–water complexes: the cases of Be(H ₂ O)0,+4 and Mg(H ₂ O)0,+6. Physical Chemistry Chemical Physics, 2019, 21, 15861-15870.	2.8	20
30	Hýckel versus Möbius aromaticity: The particle in a cylinder versus a Möbius strip. Physical Review A, 2010, 82, .	2.5	19
31	Transition-metal solvated-electron precursors: diffuse and 3d electrons in V(NH ₃)0,±6. Physical Chemistry Chemical Physics, 2019, 21, 7090-7097.	2.8	19
32	Electronic and structural features of octa-coordinated yttrium–ammonia complexes: the first neutral solvated electron precursor with eight ligands and three outer electrons. Physical Chemistry Chemical Physics, 2019, 21, 7098-7104.	2.8	17
33	An accurate first principles study of the geometric and electronic structure of B2, B2â^', B3, B3â^', and B3H: Ground and excited states. Journal of Chemical Physics, 2010, 132, 164307.	3.0	16
34	Elucidating the mechanism behind the stabilization of multi-charged metal cations in water: a case study of the electronic states of microhydrated Mg2+, Ca2+ and Al3+. Physical Chemistry Chemical Physics, 2014, 16, 6886.	2.8	16
35	Dative bonds versus electron solvation in triâ€coordinated beryllium complexes: Be(CX) ₃ [X = O, S, Se, Te, Po] and Be(PH ₃) ₃ versus Be(NH ₃) ₃ . International Journal of Quantum Chemistry, 2018, 118, e25673.	2.0	16
36	Isolating the Contributions of Specific Network Sites to the Diffuse Vibrational Spectrum of Interfacial Water with Isotopomer-Selective Spectroscopy of Cold Clusters. Journal of Physical Chemistry A, 2020, 124, 10393-10406.	2.5	16

#	Article	IF	CITATIONS
37	Ab initio investigation of the ground and excited states of MoO+,2+,â° and their catalytic strength on water activation. Physical Chemistry Chemical Physics, 2018, 20, 12278-12287.	2.8	15
38	<i>Ab initio</i> calculations on the ground and excited electronic states of neutral and charged palladium monoxide, PdO ^{0,+,â^'} . Physical Chemistry Chemical Physics, 2018, 20, 14578-14586.	2.8	15
39	The Origin of the Reactivity of the Criegee Intermediate: Implications for Atmospheric Particle Growth. Angewandte Chemie, 2016, 128, 1027-1031.	2.0	14
40	Methane to Methanol Conversion Facilitated by Transition-Metal Methyl and Methoxy Units: The Cases of FeCH ₃ ⁺ . Journal of Physical Chemistry A, 2019, 123, 5590-5599.	2.5	12
41	Geometric and electronic structure analysis of calcium water complexes with one and two solvation shells. Physical Chemistry Chemical Physics, 2020, 22, 22426-22435.	2.8	12
42	Scandium in Neutral and Positively Charged Ammonia Complexes: Balancing between Sc ²⁺ and Sc ³⁺ . Journal of Physical Chemistry A, 2020, 124, 4400-4412.	2.5	12
43	Ab initio investigation of titanium hydroxide isomers and their cations, TiOH0, + and HTiO0, +. Journal of Chemical Physics, 2011, 135, 144111.	3.0	11
44	Interaction-induced dipoles of hydrogen molecules colliding with helium atoms: A new <i>ab initio</i> dipole surface for high-temperature applications. Journal of Chemical Physics, 2012, 136, 044320.	3.0	11
45	Unimolecular and hydrolysis channels for the detachment of water from microsolvated alkaline earth dication (Mg2+, Ca2+, Sr2+, Ba2+) clusters. Theoretical Chemistry Accounts, 2014, 133, 1.	1.4	11
46	A New, Dispersion-Driven Intermolecular Arrangement for the Benzene–Water Octamer Complex: Isomers and Analysis of their Vibrational Spectra. Journal of Chemical Theory and Computation, 2016, 12, 4004-4014.	5. 3	11
47	Selective Activation of the Câ^'H Bond in Methane by Single Platinum Atomic Anions. Angewandte Chemie, 2019, 131, 7855-7859.	2.0	11
48	"Hypervalency―and the chemical bond. Computational and Theoretical Chemistry, 2019, 1153, 65-74.	2.5	11
49	Quantum chemical calculations on NbO and its reaction with methane: ground and excited electronic states. Physical Chemistry Chemical Physics, 2019, 21, 26324-26332.	2.8	11
50	Methane to Methanol Conversion Facilitated by Anionic Transition Metal Centers: The Case of Fe, Ni, Pd, and Pt. Journal of Physical Chemistry A, 2021, 125, 2364-2373.	2.5	11
51	O–H and C–H Bond Activations of Water and Methane by RuO2+ and (NH3)RuO2+: Ground and Excited States. Journal of Physical Chemistry A, 2019, 123, 9336-9344.	2.5	10
52	Addressing the Hypervalent Model: A Straightforward Explanation of Traditionally Hypervalent Molecules. Journal of Chemical Education, 2020, 97, 3638-3646.	2.3	10
53	Weak-field ligands enable inert early transition metal oxides to convert methane to methanol: the case of ZrO. Physical Chemistry Chemical Physics, 2020, 22, 6606-6618.	2.8	10
54	<i>Ab initio</i> investigation of the ground and excited states of RuO ^{+,0,â^'} and their reaction with water. Physical Chemistry Chemical Physics, 2020, 22, 16072-16079.	2.8	10

#	Article	IF	CITATIONS
55	Efficient Procedure for the Numerical Calculation of Harmonic Vibrational Frequencies Based on Internal Coordinates. Journal of Physical Chemistry A, 2013, 117, 7019-7029.	2.5	9
56	A new variation of the Buckingham exponential-6 potential with a tunable, singularity-free short-range repulsion and an adjustable long-range attraction. Chemical Physics Letters, 2015, 619, 133-138.	2.6	9
57	Dependence of the multipole moments, static polarizabilities, and static hyperpolarizabilities of the hydrogen molecule on the H–H separation in the ground singlet state. Journal of Chemical Physics, 2018, 149, 234103.	3.0	9
58	Electronic and geometric structure of cationic and neutral chromium and molybdenum ammonia complexes. Journal of Chemical Physics, 2021, 155, 014303.	3.0	9
59	All-Electron First Principles Calculations of the Ground and Some Low-Lying Excited States of Bal. Journal of Physical Chemistry A, 2007, 111, 10002-10009.	2.5	8
60	Ortho-para interconversion in cation-water complexes: The case of V+(H2O) and Nb+(H2O) clusters. Journal of Chemical Physics, 2017, 146, 224305.	3.0	8
61	Metal-Free Activation of N ₂ by Persistent Carbene Pairs: An Ab Initio Investigation. Journal of Physical Chemistry C, 2019, 123, 21548-21553.	3.1	8
62	Be–Be Bond in Action: Lessons from the Beryllium–Ammonia Complexes [Be(NH ₃) _{0–4}] ₂ <o,2+< sup="">. Journal of Physical Chemistry A, 2020, 124, 9783-9792.</o,2+<>	2.5	8
63	Particle in a Möbius wire and half-integer orbital angular momentum. Physical Review A, 2011, 83, .	2.5	7
64	First-Principles Calculations of the Electronic and Geometrical Structures of Neutral [Sc,O,H] Molecules and the Monocations, ScOH ^{0,+} and HScO ^{0,+} . Journal of Physical Chemistry A, 2011, 115, 4436-4447.	2.5	7
65	Infrared spectroscopy of RG–Co+(H2O) complexes (RG = Ar, Ne, He): The role of rare gas "tag―atoms. Journal of Chemical Physics, 2021, 154, 064306.	3.0	7
66	The Versatile Personality of Beryllium: Be(O ₂) _{1â€"2} vs Be(CO) _{1â€"2} . Journal of Physical Chemistry A, 2017, 121, 7051-7058.	2.5	6
67	A periodic-DFT study of retro-aldol fragmentation of fuctose on MoO3. Applied Catalysis A: General, 2017, 530, 75-82.	4.3	6
68	Highly Diastereo―and Enantioselective Synthesis of 3,6′â€Bisboryl―anti â€1,2â€oxaborinanâ€3â€enes: An Enantioenriched Homoallylic Alcohols with A Stereodefined Trisubstituted Alkene. Angewandte Chemie, 2021, 133, 853-861.	Entry to 2.0	6
69	Ground and excited states analysis of alkali metal ethylenediamine and crown ether complexes. Physical Chemistry Chemical Physics, 2021, 23, 20298-20306.	2.8	6
70	Hirshfeld density partitioning technique: A first application to the transition metal compounds, HScO, TiO, VO. Journal of Chemical Physics, 2013, 138, 184305.	3.0	5
71	Electronic and geometric structure analysis of neutral and anionic alkali metal complexes of the CX series (X = O, S, Se, Te, Po): The case of M(CX) _{<i>n</i>i>a€‰= 1–4} (M = Li, Journal of Computational Chemistry, 2019, 40, 1344-1351.	N a) and tl	heħ dimers.
72	Ab initio investigation of the ground and excited states of ZrO+ and NbO+. Journal of Quantitative Spectroscopy and Radiative Transfer, 2020, 255, 107265.	2.3	5

#	Article	IF	CITATIONS
73	Electronic structure of the dicationic first row transition metal oxides. Physical Chemistry Chemical Physics, 2021, 23, 21172-21182.	2.8	5
74	Can boron form coordination complexes with diffuse electrons? Evidence for linked solvated electron precursors. Electronic Structure, 2022, 4, 015001.	2.8	5
75	Simultaneous CO ₂ capture and functionalization: solvated electron precursors as novel catalysts. Chemical Communications, 2022, 58, 1310-1313.	4.1	5
76	The nature of supermolecular bonds: Investigating hydrocarbon linked beryllium solvated electron precursors. Journal of Chemical Physics, 2022, 156, .	3.0	5
77	The role of O(¹ D) in the oxidation mechanism of ethylene by iodosobenzene and other hypervalent molecules. Physical Chemistry Chemical Physics, 2017, 19, 18152-18155.	2.8	3
78	Electronic and geometric structure analysis of neutral and anionic metal nitric chalcogens: The case of MNX series (M=Li, Na, Be and X=O, S, Se, Te). Journal of Computational Chemistry, 2019, 40, 1740-1751.	3.3	3
79	Carbon monoxide activation by atomic thorium: ground and excited state reaction pathways. Physical Chemistry Chemical Physics, 2019, 21, 24469-24477.	2.8	3
80	Radical abstraction vs. oxidative addition mechanisms for the activation of the S–H, O–H, and C–H bonds using early transition metal oxides. Physical Chemistry Chemical Physics, 2021, 23, 1437-1442.	2.8	3
81	Electronic Structure of RhO ²⁺ , Its Ammoniated Complexes (NH ₃) _{1–5} RhO ²⁺ , and Mechanistic Exploration of CH ₄ Activation by Them. Inorganic Chemistry, 2021, 60, 16111-16119.	4.0	3
82	Molecular-level electrocatalytic CO ₂ reduction reaction mediated by single platinum atoms. Physical Chemistry Chemical Physics, 2022, 24, 4226-4231.	2.8	3
83	Ground and excited states of vanadium hydroxide isomers and their cations, VOH0,+ and HVO0,+. Journal of Chemical Physics, 2013, 138, 114305.	3.0	2
84	Comparing coordination uranyl(<scp>vi</scp>) complexes with 2-(1 <i>H</i> -imidazo[4,5- <i>b</i> -jphenazin-2-yl)phenol and derivatives. Dalton Transactions, 2021, 50, 11113-11122.	3.3	2
85	The interaction-induced dipole of H2–H: Newab initioresults and spherical tensor analysis. Journal of Chemical Physics, 2019, 150, 204307.	3.0	1
86	Ab initio investigation of the ground and excited states of TcO+ and RhO+. Journal of Quantitative Spectroscopy and Radiative Transfer, 2022, 280, 108074.	2.3	0
87	Electronic structure of the ground and excited states of neutral and charged silicon hydrides, SiH _{<i>x</i><csup>0/+/â°', <i>x</i><csup>0/+/â°', <i>x</i></csup>0/+/â°', <i>x</i></csup>0/+/â°', <i>x</i>0/+/â°', <i>x0/+/â°', <i>x0/+/â°',</i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i></i>}	2.8	0