

Tuan H Vu

List of Publications by Year in descending order

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34
papers

569
citations

567281

15
h-index

642732

23
g-index

35
all docs

35
docs citations

35
times ranked

646
citing authors

#	ARTICLE	IF	CITATIONS
1	Water: A Responsive Small Molecule. <i>Accounts of Chemical Research</i> , 2012, 45, 15-22.	15.6	59
2	Hydrogen Bonding between Water and Tetrahydrofuran Relevant to Clathrate Formation. <i>Journal of Physical Chemistry B</i> , 2015, 119, 9167-9172.	2.6	43
3	Prospects for mineralogy on Titan. <i>American Mineralogist</i> , 2018, 103, 343-349.	1.9	35
4	Composition and Evolution of Frozen Chloride Brines under the Surface Conditions of Europa. <i>ACS Earth and Space Chemistry</i> , 2017, 1, 14-23.	2.7	33
5	Preferential formation of sodium salts from frozen sodium-ammonium-chloride-carbonate brines – Implications for Ceres’s bright spots. <i>Planetary and Space Science</i> , 2017, 141, 73-77.	1.7	31
6	The Acetylene-Ammonia Co-crystal on Titan. <i>ACS Earth and Space Chemistry</i> , 2018, 2, 366-375.	2.7	30
7	CHEMISTRY OF FROZEN SODIUM-MAGNESIUM-SULFATE-CHLORIDE BRINES: IMPLICATIONS FOR SURFACE EXPRESSION OF EUROPA’S OCEAN COMPOSITION. <i>Astrophysical Journal Letters</i> , 2016, 816, L26.	8.3	29
8	A co-crystal between benzene and ethane: a potential evaporite material for Saturn's moon Titan. <i>IUCr</i> , 2016, 3, 192-199.	2.2	26
9	Formation of a New Benzene-Ethane Co-Crystalline Structure Under Cryogenic Conditions. <i>Journal of Physical Chemistry A</i> , 2014, 118, 4087-4094.	2.5	23
10	Experimental determination of the kinetics of formation of the benzene-ethane co-crystal and implications for Titan. <i>Geophysical Research Letters</i> , 2014, 41, 5396-5401.	4.0	21
11	Insights into Europa's ocean composition derived from its surface expression. <i>Icarus</i> , 2019, 321, 857-865.	2.5	21
12	A Co-Crystal between Acetylene and Butane: A Potentially Ubiquitous Molecular Mineral on Titan. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 2808-2815.	2.7	19
13	Rapid Formation of Clathrate Hydrate From Liquid Ethane and Water Ice on Titan. <i>Geophysical Research Letters</i> , 2020, 47, e2019GL086265.	4.0	19
14	Kinetic effect on the freezing of ammonium-sodium-carbonate-chloride brines and implications for the origin of Ceres’s bright spots. <i>Icarus</i> , 2019, 320, 150-158.	2.5	18
15	Titan in a Test Tube: Organic Co-crystals and Implications for Titan Mineralogy. <i>Accounts of Chemical Research</i> , 2021, 54, 3050-3059.	15.6	17
16	Probing Europa's subsurface ocean composition from surface salt minerals using in-situ techniques. <i>Icarus</i> , 2020, 349, 113746.	2.5	15
17	Properties and Behavior of the Acetonitrile-Acetylene Co-Crystal under Titan Surface Conditions. <i>ACS Earth and Space Chemistry</i> , 2020, 4, 1375-1385.	2.7	13
18	Molecular dance: Water’s collective modes. <i>Chemical Physics Letters</i> , 2013, 588, 1-10.	2.6	12

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19	Experimental Study on the Effect of Ammonia on the Phase Behavior of Tetrahydrofuran Clathrates. <i>Journal of Physical Chemistry B</i> , 2014, 118, 13371-13377.	2.6	12
20	Phase Diagram of the Ternary Water–Tetrahydrofuran–Ammonia System at Low Temperatures. Implications for Clathrate Hydrates and Outgassing on Titan. <i>ACS Earth and Space Chemistry</i> , 2018, 2, 135-146.	2.7	12
21	Low-temperature specific heat capacity measurements and application to Mars thermal modeling. <i>Icarus</i> , 2019, 321, 824-840.	2.5	11
22	Vibrating hydroxide in hydrophobic solution: The ion to keep an eye on. <i>Chemical Physics Letters</i> , 2013, 572, 13-15.	2.6	8
23	Competitive Binding of Methanol and Propane for Water Via Matrix-Isolation Spectroscopy: Implications for Inhibition of Clathrate Nucleation. <i>Journal of Physical Chemistry A</i> , 2011, 115, 998-1002.	2.5	7
24	Insights into hydrogen bonding via ice interfaces and isolated water. <i>Journal of Chemical Physics</i> , 2014, 141, 18C521.	3.0	7
25	Raman Signatures and Thermal Expansivity of Acetylene Clathrate Hydrate. <i>Journal of Physical Chemistry A</i> , 2019, 123, 7051-7056.	2.5	7
26	No compelling evidence for clathrate hydrate formation under interstellar medium conditions over laboratory time scales. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2019, 116, 14407-14408.	7.1	7
27	Anisotropic thermal expansion of the acetylene–ammonia co-crystal under Titan's conditions. <i>Journal of Applied Crystallography</i> , 2020, 53, 1524-1530.	4.5	7
28	Specific Heat Capacity Measurements of Selected Meteorites for Planetary Surface Temperature Modeling. <i>Journal of Geophysical Research E: Planets</i> , 2021, 126, .	3.6	7
29	Phase Behavior of Clathrate Hydrates in the Ternary $H_2O-NH_3-Cyclopentane$ System. <i>ACS Earth and Space Chemistry</i> , 2020, 4, 526-534.	2.7	6
30	Cage occupancy of methane clathrate hydrates in the ternary $H_2O-NH_3-CH_4$ system. <i>Chemical Communications</i> , 2020, 56, 12391-12394.	4.1	4
31	Formation of Vitreous Salt Hydrates Under Conditions Relevant to Europa. <i>Planetary Science Journal</i> , 2022, 3, 151.	3.6	4
32	Vertical compositional variations of liquid hydrocarbons in Titan's alkanofers. <i>Astronomy and Astrophysics</i> , 2021, 653, A80.	5.1	3
33	A simple gas introduction system for cryogenic powder X-ray diffraction. <i>Journal of Applied Crystallography</i> , 2021, 54, 1268-1270.	4.5	2
34	Reply to the "Comment on Cage occupancy of methane clathrate hydrates in the ternary $H_2O-NH_3-CH_4$ system" by S. Alavi and J. Ripmeester, <i>Chem. Commun.</i> , 2022, 58, DOI: 10.1039/D1CC06526B. <i>Chemical Communications</i> , 2022, 58, 4099-4102.	4.1	1