

Kai S Exner

List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/8586310/publications.pdf>

Version: 2024-02-01

56
papers

2,358
citations

218381

26
h-index

214527

47
g-index

57
all docs

57
docs citations

57
times ranked

1684
citing authors

#	ARTICLE	IF	CITATIONS
1	Computational Pourbaix Diagrams for MXenes: A Key Ingredient toward Proper Theoretical Electrocatalytic Studies. <i>Advanced Theory and Simulations</i> , 2023, 6, .	1.3	16
2	Why the microkinetic modeling of experimental tafel plots requires knowledge of the reaction intermediate's binding energy. <i>Electrochemical Science Advances</i> , 2022, 2, e2100037.	1.2	13
3	Method to Determine the Bifunctional Index for the Oxygen Electrocatalysis from Theory. <i>ChemElectroChem</i> , 2022, 9, .	1.7	13
4	Circumventing the OCl versus OOH scaling relation in the chlorine evolution reaction: Beyond dimensionally stable anodes. <i>Current Opinion in Electrochemistry</i> , 2022, 34, 100979.	2.5	12
5	Statistical analysis of breaking scaling relation in the oxygen evolution reaction. <i>Electrochimica Acta</i> , 2022, 412, 140125.	2.6	12
6	A doxorubicinâ€“peptideâ€“gold nanoparticle conjugate as a functionalized drug delivery system: exploring the limits. <i>Physical Chemistry Chemical Physics</i> , 2022, 24, 14985-14992.	1.3	4
7	On the Optimization of Nitrogenâ€“Reduction Electrocatalysts: Breaking Scaling Relation or Catalytic Resonance Theory?. <i>ChemCatChem</i> , 2022, 14, .	1.8	11
8	Computational electrochemistry focusing on nanostructured catalysts: challenges and opportunities. <i>Materials Today Energy</i> , 2022, 28, 101083.	2.5	3
9	Blickpunkt Nachwuchs: Theoretische Elektrokatalyse. <i>Nachrichten Aus Der Chemie</i> , 2022, 70, 82-84.	0.0	0
10	Hydrogen electrocatalysis revisited: Weak bonding of adsorbed hydrogen as the design principle for active electrode materials. <i>Current Opinion in Electrochemistry</i> , 2021, 26, 100673.	2.5	18
11	Boosting the Stability of RuO ₂ in the Acidic Oxygen Evolution Reaction by Tuning Oxygenâ€“Vacancy Formation Energies: A Viable Approach Beyond Nobleâ€“Metal Catalysts?. <i>ChemElectroChem</i> , 2021, 8, 46-48.	1.7	19
12	Method to Construct Volcano Relations by Multiscale Modeling: Building Bridges between the Catalysis and Biosimulation Communities. <i>Journal of Physical Chemistry B</i> , 2021, 125, 2098-2104.	1.2	1
13	Why approximating electrocatalytic activity by a single freeâ€“energy change is insufficient. <i>Electrochimica Acta</i> , 2021, 375, 137975.	2.6	42
14	A Universal Approach to Quantify Overpotential-Dependent Selectivity Trends for the Competing Oxygen Evolution and Peroxide Formation Reactions: A Case Study on Graphene Model Electrodes. <i>Journal of Physical Chemistry C</i> , 2021, 125, 10413-10421.	1.5	9
15	The Sabatier Principle in Electrocatalysis: Basics, Limitations, and Extensions. <i>Frontiers in Energy Research</i> , 2021, 9, .	1.2	175
16	Why the breaking of the OOH versus OH scaling relation might cause decreased electrocatalytic activity. <i>Chem Catalysis</i> , 2021, 1, 258-271.	2.9	30
17	On the Lattice Oxygen Evolution Mechanism: Avoiding Pitfalls. <i>ChemCatChem</i> , 2021, 13, 4066-4074.	1.8	22
18	Why the optimum thermodynamic free-energy landscape of the oxygen evolution reaction reveals an asymmetric shape. <i>Materials Today Energy</i> , 2021, 21, 100831.	2.5	12

#	ARTICLE	IF	CITATIONS
19	General Efficacy of Atomically Dispersed Pt Catalysts for the Chlorine Evolution Reaction: Potential-Dependent Switching of the Kinetics and Mechanism. <i>ACS Catalysis</i> , 2021, 11, 12232-12246.	5.5	40
20	The electrochemical-step asymmetry index. <i>MethodsX</i> , 2021, 8, 101590.	0.7	4
21	Beyond thermodynamic-based material-screening concepts: Kinetic scaling relations exemplified by the chlorine evolution reaction over transition-metal oxides. <i>Electrochimica Acta</i> , 2020, 334, 135555.	2.6	15
22	Electrolyte Engineering as a Key Strategy Towards a Sustainable Energy Scenario?. <i>ChemElectroChem</i> , 2020, 7, 594-595.	1.7	5
23	Comparison of the Conventional Volcano Analysis with a Unifying Approach: Material Screening Based on a Combination of Experiment and Theory. <i>Journal of Physical Chemistry C</i> , 2020, 124, 822-828.	1.5	22
24	A Universal Descriptor for the Screening of Electrode Materials for Multiple-Electron Processes: Beyond the Thermodynamic Overpotential. <i>ACS Catalysis</i> , 2020, 10, 12607-12617.	5.5	91
25	Recent Progress in the Development of Screening Methods to Identify Electrode Materials for the Oxygen Evolution Reaction. <i>Advanced Functional Materials</i> , 2020, 30, 2005060.	7.8	49
26	Paradigm change in hydrogen electrocatalysis: The volcano's apex is located at weak bonding of the reaction intermediate. <i>International Journal of Hydrogen Energy</i> , 2020, 45, 27221-27229.	3.8	46
27	Design criteria for the competing chlorine and oxygen evolution reactions: avoid the OCl adsorbate to enhance chlorine selectivity. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 22451-22458.	1.3	41
28	Identifying a gold nanoparticle as a proactive carrier for transport of a doxorubicin-peptide complex. <i>Colloids and Surfaces B: Biointerfaces</i> , 2020, 194, 111155.	2.5	5
29	Does a Thermoneutral Electrocatalyst Correspond to the Apex of a Volcano Plot for a Simple Two-Electron Process?. <i>Angewandte Chemie - International Edition</i> , 2020, 59, 10236-10240.	7.2	85
30	Does a Thermoneutral Electrocatalyst Correspond to the Apex of a Volcano Plot for a Simple Two-Electron Process?. <i>Angewandte Chemie</i> , 2020, 132, 10320-10324.	1.6	14
31	Overpotential-Dependent Volcano Plots to Assess Activity Trends in the Competing Chlorine and Oxygen Evolution Reactions. <i>ChemElectroChem</i> , 2020, 7, 1448-1455.	1.7	25
32	Beyond Dimensionally Stable Anodes: Single-Atom Catalysts with Superior Chlorine Selectivity. <i>ChemElectroChem</i> , 2020, 7, 1528-1530.	1.7	12
33	Universality in Oxygen Evolution Electrocatalysis: High-Throughput Screening and a Priori Determination of the Rate-Determining Reaction Step. <i>ChemCatChem</i> , 2020, 12, 2000-2003.	1.8	20
34	Design Criteria for Oxygen Evolution Electrocatalysts from First Principles: Introduction of a Unifying Material-Screening Approach. <i>ACS Applied Energy Materials</i> , 2019, 2, 7991-8001.	2.5	59
35	Beyond the Rate-Determining Step in the Oxygen Evolution Reaction over a Single-Crystalline IrO ₂ (110) Model Electrode: Kinetic Scaling Relations. <i>ACS Catalysis</i> , 2019, 9, 6755-6765.	5.5	117
36	Controlling Stability and Selectivity in the Competing Chlorine and Oxygen Evolution Reaction over Transition Metal Oxide Electrodes. <i>ChemElectroChem</i> , 2019, 6, 3401-3409.	1.7	57

#	ARTICLE	IF	CITATIONS
37	Beyond the Traditional Volcano Concept: Overpotential-Dependent Volcano Plots Exemplified by the Chlorine Evolution Reaction over Transition-Metal Oxides. <i>Journal of Physical Chemistry C</i> , 2019, 123, 16921-16928.	1.5	50
38	Activity- σ Stability Volcano Plots for Material Optimization in Electrocatalysis. <i>ChemCatChem</i> , 2019, 11, 3234-3241.	1.8	15
39	Is Thermodynamics a Good Descriptor for the Activity? Re-Investigation of Sabatier's Principle by the Free Energy Diagram in Electrocatalysis. <i>ACS Catalysis</i> , 2019, 9, 5320-5329.	5.5	94
40	Recent Advancements Towards Closing the Gap between Electrocatalysis and Battery Science Communities: The Computational Lithium Electrode and Activity- σ Stability Volcano Plots. <i>ChemSusChem</i> , 2019, 12, 2330-2344.	3.6	33
41	Kinetic study of gold nanoparticles synthesized in the presence of chitosan and citric acid. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2018, 557, 106-115.	2.3	24
42	A Universal Approach To Determine the Free Energy Diagram of an Electrocatalytic Reaction. <i>ACS Catalysis</i> , 2018, 8, 1864-1879.	5.5	128
43	Activity - σ Stability Volcano Plots for the Investigation of Nano-Sized Electrode Materials in Lithium-Ion Batteries. <i>ChemElectroChem</i> , 2018, 5, 3243-3248.	1.7	16
44	A short perspective of modeling electrode materials in lithium-ion batteries by the ab initio atomistic thermodynamics approach. <i>Journal of Solid State Electrochemistry</i> , 2018, 22, 3111-3117.	1.2	14
45	Temperature-Dependent Kinetic Studies of the Chlorine Evolution Reaction over RuO ₂ (110) Model Electrodes. <i>ACS Catalysis</i> , 2017, 7, 2403-2411.	5.5	111
46	Kinetics of Electrocatalytic Reactions from First-Principles: A Critical Comparison with the Ab Initio Thermodynamics Approach. <i>Accounts of Chemical Research</i> , 2017, 50, 1240-1247.	7.6	133
47	Constrained Ab Initio Thermodynamics: Transferring the Concept of Surface Pourbaix Diagrams in Electrocatalysis to Electrode Materials in Lithium-Ion Batteries. <i>ChemElectroChem</i> , 2017, 4, 3231-3237.	1.7	27
48	Full Free Energy Diagram of an Electrocatalytic Reaction over a Single-Crystalline Model Electrode. <i>ChemElectroChem</i> , 2017, 4, 2902-2908.	1.7	27
49	Full Kinetics from First Principles of the Chlorine Evolution Reaction over a RuO ₂ (110) Model Electrode. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 7501-7504.	7.2	120
50	Full Kinetics from First Principles of the Chlorine Evolution Reaction over a RuO ₂ (110) Model Electrode. <i>Angewandte Chemie</i> , 2016, 128, 7627-7630.	1.6	15
51	Combined experiment and theory approach in surface chemistry: Stairway to heaven?. <i>Surface Science</i> , 2015, 640, 165-180.	0.8	22
52	Ligand Effects and Their Impact on Electrocatalytic Processes Exemplified with the Oxygen Evolution Reaction (OER) on RuO ₂ (110). <i>ChemElectroChem</i> , 2015, 2, 707-713.	1.7	64
53	Microscopic Insights into the Chlorine Evolution Reaction on RuO ₂ (110): a Mechanistic Ab Initio Atomistic Thermodynamics Study. <i>Electrocatalysis</i> , 2015, 6, 163-172.	1.5	27
54	Chlorine Evolution Reaction on RuO ₂ (110): Ab initio Atomistic Thermodynamics Study - Pourbaix Diagrams. <i>Electrochimica Acta</i> , 2014, 120, 460-466.	2.6	95

#	ARTICLE	IF	CITATIONS
55	Controlling Selectivity in the Chlorine Evolution Reaction over RuO ₂ -Based Catalysts. <i>Angewandte Chemie - International Edition</i> , 2014, 53, 11032-11035.	7.2	182
56	On the optimum binding energy for the hydrogen evolution reaction: How do experiments contribute?. <i>Electrochemical Science Advances</i> , 0, , e2100101.	1.2	1