

Gary T Rochelle

List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/8563653/publications.pdf>

Version: 2024-02-01

214
papers

16,574
citations

29994

54
h-index

16127

124
g-index

217
all docs

217
docs citations

217
times ranked

8924
citing authors

#	ARTICLE	IF	CITATIONS
1	Amine Scrubbing for CO ₂ Capture. <i>Science</i> , 2009, 325, 1652-1654.	6.0	3,490
2	Carbon capture and storage update. <i>Energy and Environmental Science</i> , 2014, 7, 130-189.	15.6	1,765
3	Absorption of carbon dioxide into aqueous piperazine: reaction kinetics, mass transfer and solubility. <i>Chemical Engineering Science</i> , 2000, 55, 5531-5543.	1.9	531
4	Model of vapor-liquid equilibria for aqueous acid gas-alkanolamine systems using the electrolyte-NRTL equation. <i>Industrial & Engineering Chemistry Research</i> , 1989, 28, 1060-1073.	1.8	455
5	Aqueous piperazine as the new standard for CO ₂ capture technology. <i>Chemical Engineering Journal</i> , 2011, 171, 725-733.	6.6	417
6	Carbon dioxide capture with concentrated, aqueous piperazine. <i>International Journal of Greenhouse Gas Control</i> , 2010, 4, 119-124.	2.3	318
7	Modeling of CO ₂ capture by aqueous monoethanolamine. <i>AIChE Journal</i> , 2003, 49, 1676-1686.	1.8	302
8	Alternative stripper configurations for CO ₂ capture by aqueous amines. <i>AIChE Journal</i> , 2007, 53, 3144-3154.	1.8	264
9	Oxidative Degradation of Monoethanolamine. <i>Industrial & Engineering Chemistry Research</i> , 2002, 41, 4178-4186.	1.8	253
10	Rate-Based Process Modeling Study of CO ₂ Capture with Aqueous Monoethanolamine Solution. <i>Industrial & Engineering Chemistry Research</i> , 2009, 48, 9233-9246.	1.8	249
11	Absorption of carbon dioxide in aqueous piperazine/methyldiethanolamine. <i>AIChE Journal</i> , 2002, 48, 2788-2799.	1.8	247
12	Monoethanolamine Degradation: O ₂ Mass Transfer Effects under CO ₂ Capture Conditions. <i>Industrial & Engineering Chemistry Research</i> , 2004, 43, 6400-6408.	1.8	246
13	Energy Performance of Stripper Configurations for CO ₂ Capture by Aqueous Amines. <i>Industrial & Engineering Chemistry Research</i> , 2006, 45, 2457-2464.	1.8	243
14	Thermal degradation of monoethanolamine at stripper conditions. <i>Energy Procedia</i> , 2009, 1, 327-333.	1.8	243
15	Carbon dioxide absorption with aqueous potassium carbonate promoted by piperazine. <i>Chemical Engineering Science</i> , 2004, 59, 3619-3630.	1.9	234
16	Carbon Dioxide Absorption and Desorption in Aqueous Monoethanolamine Solutions in a Rotating Packed Bed. <i>Industrial & Engineering Chemistry Research</i> , 2007, 46, 2823-2833.	1.8	211
17	Innovative Absorber/Stripper Configurations for CO ₂ Capture by Aqueous Monoethanolamine. <i>Industrial & Engineering Chemistry Research</i> , 2006, 45, 2465-2472.	1.8	202
18	Effects of the Temperature Bulge in CO ₂ Absorption from Flue Gas by Aqueous Monoethanolamine. <i>Industrial & Engineering Chemistry Research</i> , 2008, 47, 867-875.	1.8	181

#	ARTICLE	IF	CITATIONS
19	Thermal degradation of amines for CO ₂ capture. <i>Current Opinion in Chemical Engineering</i> , 2012, 1, 183-190.	3.8	171
20	Thermodynamics of Piperazine/Methyldiethanolamine/Water/Carbon Dioxide. <i>Industrial & Engineering Chemistry Research</i> , 2002, 41, 604-612.	1.8	162
21	Amine volatility in CO ₂ capture. <i>International Journal of Greenhouse Gas Control</i> , 2010, 4, 707-715.	2.3	159
22	Kinetics of Carbon Dioxide Absorption into Aqueous Potassium Carbonate and Piperazine. <i>Industrial & Engineering Chemistry Research</i> , 2006, 45, 2531-2545.	1.8	155
23	Oxidation Inhibitors for Copper and Iron Catalyzed Degradation of Monoethanolamine in CO ₂ Capture Processes. <i>Industrial & Engineering Chemistry Research</i> , 2006, 45, 2513-2521.	1.8	148
24	Degradation of aqueous piperazine in carbon dioxide capture. <i>International Journal of Greenhouse Gas Control</i> , 2010, 4, 756-761.	2.3	140
25	Thermodynamics of aqueous potassium carbonate, piperazine, and carbon dioxide. <i>Fluid Phase Equilibria</i> , 2005, 227, 197-213.	1.4	129
26	A dimensionless model for predicting the mass transfer area of structured packing. <i>AIChE Journal</i> , 2011, 57, 1173-1184.	1.8	125
27	Absorption and desorption rates of carbon dioxide with monoethanolamine and piperazine. <i>Energy Procedia</i> , 2009, 1, 1163-1169.	1.8	124
28	CO ₂ Absorption Rate and Solubility in Monoethanolamine/Piperazine/Water. <i>Separation Science and Technology</i> , 2003, 38, 337-357.	1.3	114
29	MDEA/Piperazine as a solvent for CO ₂ capture. <i>Energy Procedia</i> , 2009, 1, 1351-1357.	1.8	114
30	A Thermodynamic Model of Methyldiethanolamine-CO ₂ -H ₂ S-Water. <i>Industrial & Engineering Chemistry Research</i> , 1997, 36, 3944-3953.	1.8	113
31	Nitrogen Dioxide Absorption and Sulfite Oxidation in Aqueous Sulfite. <i>Environmental Science & Technology</i> , 1998, 32, 1994-2003.	4.6	113
32	Dynamic Modeling to Minimize Energy Use for CO ₂ Capture in Power Plants by Aqueous Monoethanolamine. <i>Industrial & Engineering Chemistry Research</i> , 2009, 48, 6105-6111.	1.8	110
33	Stripper configurations for CO ₂ capture by aqueous monoethanolamine. <i>Chemical Engineering Research and Design</i> , 2011, 89, 1639-1646.	2.7	110
34	Aqueous Ethylenediamine for CO ₂ Capture. <i>ChemSusChem</i> , 2010, 3, 913-918.	3.6	99
35	Modeling CO ₂ capture with aqueous monoethanolamine. <i>International Journal of Greenhouse Gas Control</i> , 2010, 4, 161-166.	2.3	88
36	Reaction Products from the Oxidative Degradation of Monoethanolamine. <i>Industrial & Engineering Chemistry Research</i> , 2011, 50, 667-673.	1.8	88

#	ARTICLE	IF	CITATIONS
37	Aqueous piperazine derivatives for CO ₂ capture: Accurate screening by a wetted wall column. <i>Chemical Engineering Research and Design</i> , 2011, 89, 1693-1710.	2.7	88
38	Numerical simulation of theories for gas absorption with chemical reaction. <i>AIChE Journal</i> , 1989, 35, 1271-1281.	1.8	80
39	Carbon dioxide capture with concentrated, aqueous piperazine. <i>Energy Procedia</i> , 2009, 1, 1489-1496.	1.8	79
40	Amine blends using concentrated piperazine. <i>Energy Procedia</i> , 2013, 37, 353-369.	1.8	79
41	Rate-Based Modeling of Reactive Absorption of CO ₂ and H ₂ S into Aqueous Methyldiethanolamine. <i>Industrial & Engineering Chemistry Research</i> , 1998, 37, 4107-4117.	1.8	76
42	CO ₂ absorption rate in semi-aqueous monoethanolamine. <i>Chemical Engineering Science</i> , 2018, 182, 56-66.	1.9	73
43	CO ₂ absorption into aqueous mixtures of diglycolamine [®] and methyldiethanolamine. <i>Chemical Engineering Science</i> , 2000, 55, 5125-5140.	1.9	72
44	CO ₂ Absorption Rate into Concentrated Aqueous Monoethanolamine and Piperazine. <i>Journal of Chemical & Engineering Data</i> , 2011, 56, 2187-2195.	1.0	72
45	Catalysts and inhibitors for oxidative degradation of monoethanolamine. <i>International Journal of Greenhouse Gas Control</i> , 2009, 3, 704-711.	2.3	70
46	Optimizing post-combustion CO ₂ capture in response to volatile electricity prices. <i>International Journal of Greenhouse Gas Control</i> , 2012, 8, 180-195.	2.3	68
47	Modeling CO ₂ capture with aqueous monoethanolamine. <i>Energy Procedia</i> , 2009, 1, 1171-1178.	1.8	65
48	Regeneration with Rich Bypass of Aqueous Piperazine and Monoethanolamine for CO ₂ Capture. <i>Industrial & Engineering Chemistry Research</i> , 2014, 53, 4067-4074.	1.8	65
49	Capacity and absorption rate of tertiary and hindered amines blended with piperazine for CO ₂ capture. <i>Chemical Engineering Science</i> , 2016, 155, 397-404.	1.9	65
50	Influence of Surface Tension on Effective Packing Area. <i>Industrial & Engineering Chemistry Research</i> , 2008, 47, 1253-1260.	1.8	60
51	Approaching a reversible stripping process for CO ₂ capture. <i>Chemical Engineering Journal</i> , 2016, 283, 1033-1043.	6.6	60
52	Turning CO ₂ Capture On and Off in Response to Electric Grid Demand: A Baseline Analysis of Emissions and Economics. <i>Journal of Energy Resources Technology, Transactions of the ASME</i> , 2010, 132, .	1.4	58
53	Accurate screening of amines by the Wetted Wall Column. <i>Energy Procedia</i> , 2011, 4, 101-108.	1.8	55
54	Influence of viscosity and surface tension on the effective mass transfer area of structured packing. <i>Energy Procedia</i> , 2009, 1, 1197-1204.	1.8	54

#	ARTICLE	IF	CITATIONS
55	Stripper configurations for CO ₂ capture by aqueous monoethanolamine and piperazine. Energy Procedia, 2011, 4, 1323-1330.	1.8	54
56	Hybrid Membrane-absorption CO ₂ Capture Process. Energy Procedia, 2014, 63, 605-613.	1.8	54
57	Simultaneous Sulfur Dioxide and Nitrogen Dioxide Removal by Calcium Hydroxide and Calcium Silicate Solids. Journal of the Air and Waste Management Association, 1998, 48, 819-828.	0.9	53
58	A simple model for prediction of acid gas solubilities in alkanolamines. Separation and Purification Technology, 1996, 10, 181-186.	0.3	52
59	Physical and chemical solubility of carbon dioxide in aqueous methyldiethanolamine. Fluid Phase Equilibria, 2000, 168, 241-258.	1.4	52
60	Rate modeling of CO ₂ stripping from potassium carbonate promoted by piperazine. International Journal of Greenhouse Gas Control, 2009, 3, 121-132.	2.3	52
61	Total pressure and CO ₂ solubility at high temperature in aqueous amines. Energy Procedia, 2011, 4, 117-124.	1.8	52
62	Volatility of aqueous amines in CO ₂ capture. Energy Procedia, 2011, 4, 1624-1630.	1.8	50
63	Foaming of aqueous piperazine and monoethanolamine for CO ₂ capture. International Journal of Greenhouse Gas Control, 2011, 5, 381-386.	2.3	49
64	Piperazine Degradation in Pilot Plants. Energy Procedia, 2013, 37, 1912-1923.	1.8	49
65	Decomposition of Nitrosamines in CO ₂ Capture by Aqueous Piperazine or Monoethanolamine. Environmental Science & Technology, 2014, 48, 5996-6002.	4.6	47
66	Catalysts and inhibitors for MEA oxidation. Energy Procedia, 2009, 1, 1179-1185.	1.8	44
67	Degradation of aqueous methyldiethanolamine by temperature and oxygen cycling. Energy Procedia, 2011, 4, 23-28.	1.8	43
68	Pilot plant test of the advanced flash stripper for CO ₂ capture. Faraday Discussions, 2016, 192, 37-58.	1.6	43
69	Optimization of Advanced Flash Stripper for CO ₂ Capture using Piperazine. Energy Procedia, 2014, 63, 1504-1513.	1.8	42
70	CO ₂ solubility and mass transfer in water-lean solvents. Chemical Engineering Science, 2019, 202, 403-416.	1.9	42
71	Characterization of Piperazine/2-Aminomethylpropanol for Carbon Dioxide Capture. Energy Procedia, 2013, 37, 340-352.	1.8	41
72	Kinetics of <i>N</i> -Nitrosopiperazine Formation from Nitrite and Piperazine in CO ₂ Capture. Environmental Science & Technology, 2013, 47, 3528-3534.	4.6	41

#	ARTICLE	IF	CITATIONS
73	Mass Transfer Parameters for Packings: Effect of Viscosity. Industrial & Engineering Chemistry Research, 2018, 57, 718-729.	1.8	41
74	Thermal degradation of piperazine and its structural analogs. Energy Procedia, 2011, 4, 43-50.	1.8	40
75	Absorption of CO ₂ in Aqueous Diglycolamine. Industrial & Engineering Chemistry Research, 2006, 45, 2473-2482.	1.8	38
76	Density and Viscosity of Aqueous (Piperazine + Carbon Dioxide) Solutions. Journal of Chemical & Engineering Data, 2011, 56, 574-581.	1.0	38
77	Modeling pilot plant results for CO ₂ capture by aqueous piperazine. Energy Procedia, 2011, 4, 1593-1600.	1.8	38
78	Limestone dissolution in stack gas desulfurization. A mass-transfer model is shown to predict the measured dissolution rates with less than 30% error. Environmental Progress, 1982, 1, 52-58.	0.8	37
79	Pilot plant demonstration of piperazine with the advanced flash stripper. International Journal of Greenhouse Gas Control, 2019, 84, 72-81.	2.3	37
80	Nitrosamine Formation in Amine Scrubbing at Desorber Temperatures. Environmental Science & Technology, 2014, 48, 8777-8783.	4.6	36
81	Thermodynamic modeling of piperazine/2-aminomethylpropanol/CO ₂ /water. Chemical Engineering Science, 2014, 117, 331-341.	1.9	36
82	Effectiveness of absorber intercooling for CO ₂ absorption from natural gas fired flue gases using monoethanolamine solvent. International Journal of Greenhouse Gas Control, 2017, 58, 246-255.	2.3	36
83	CO ₂ absorption rate in biphasic solvent of aminoethylethanolamine and diethylethanolamine. Chemical Engineering Journal, 2021, 404, 126503.	6.6	36
84	Dry Absorption of HCL and SO ₂ with Hydrated Lime from Humidified Flue Gas. Industrial & Engineering Chemistry Research, 1999, 38, 4068-4080.	1.8	35
85	Absorber Intercooling Configurations using Aqueous Piperazine for Capture from Sources with 4 to 27% CO ₂ . Energy Procedia, 2014, 63, 1637-1656.	1.8	35
86	Piperazine/4-hydroxy-1-methylpiperidine for CO ₂ capture. Chemical Engineering Journal, 2017, 307, 258-263.	6.6	35
87	Oxidative Degradation of Amines With High-Temperature Cycling. Energy Procedia, 2013, 37, 2118-2132.	1.8	34
88	Liquid-phase mass transfer in spray contactors. AIChE Journal, 2003, 49, 2363-2373.	1.8	33
89	Regulatory Control of Amine Scrubbing for CO ₂ Capture from Power Plants. Industrial & Engineering Chemistry Research, 2016, 55, 4646-4657.	1.8	33
90	Products and process variables in oxidation of monoethanolamine for CO ₂ capture. International Journal of Greenhouse Gas Control, 2013, 12, 472-477.	2.3	32

#	ARTICLE	IF	CITATIONS
91	Effect of Liquid Viscosity on the Liquid Phase Mass Transfer Coefficient of Packing. Energy Procedia, 2014, 63, 1268-1286.	1.8	32
92	CO ₂ absorption rate and capacity of semi-aqueous piperazine for CO ₂ capture. International Journal of Greenhouse Gas Control, 2019, 85, 182-186.	2.3	32
93	Demonstration of 99% CO ₂ removal from coal flue gas by amine scrubbing. International Journal of Greenhouse Gas Control, 2019, 83, 236-244.	2.3	32
94	Oxidation of amines at absorber conditions for CO ₂ capture from flue gas. Energy Procedia, 2011, 4, 171-178.	1.8	30
95	Volatility of amines for CO ₂ capture. International Journal of Greenhouse Gas Control, 2017, 58, 1-9.	2.3	30
96	Lost work: A comparison of water-lean solvent to a second generation aqueous amine process for CO ₂ capture. International Journal of Greenhouse Gas Control, 2019, 84, 82-90.	2.3	29
97	Aqueous Piperazine/N-(2-Aminoethyl) Piperazine for CO ₂ Capture. Energy Procedia, 2013, 37, 1621-1638.	1.8	28
98	Energy Performance of Advanced Reboiled and Flash Stripper Configurations for CO ₂ Capture Using Monoethanolamine. Industrial & Engineering Chemistry Research, 2016, 55, 4622-4631.	1.8	28
99	Mercury Absorption in Aqueous Oxidants Catalyzed by Mercury(II). Industrial & Engineering Chemistry Research, 1998, 37, 380-387.	1.8	27
100	Rate-based modeling and economic optimization of next-generation amine-based carbon capture plants. Applied Energy, 2019, 252, 113379.	5.1	27
101	Removal of SO ₂ and NO _x from Stack Gas by Reaction with Calcium Hydroxide Solids. Japca, 1989, 39, 175-179.	0.3	26
102	Limestone Dissolution in Flue Gas Scrubbing: Effect of Sulfite. Journal of the Air and Waste Management Association, 1992, 42, 926-935.	0.2	26
103	Modeling piperazine thermodynamics. Energy Procedia, 2011, 4, 35-42.	1.8	26
104	Aqueous 3-(methylamino)propylamine for CO ₂ capture. International Journal of Greenhouse Gas Control, 2013, 15, 70-77.	2.3	26
105	Modeling Aerosols in Amine-based CO ₂ Capture. Energy Procedia, 2013, 37, 1706-1719.	1.8	26
106	Packing Characterization: Mass Transfer Properties. Energy Procedia, 2012, 23, 23-32.	1.8	25
107	Thermal degradation of novel piperazine-based amine blends for CO ₂ capture. International Journal of Greenhouse Gas Control, 2016, 49, 239-249.	2.3	25
108	Inhibitors of Monoethanolamine Oxidation in CO ₂ Capture Processes. Industrial & Engineering Chemistry Research, 2014, 53, 16222-16228.	1.8	24

#	ARTICLE	IF	CITATIONS
109	Pilot-scale parametric evaluation of concentrated piperazine for CO ₂ capture at an Australian coal-fired power station. , 2015, 5, 7-16.		24
110	Dynamic modeling and control of an intercooled absorber for post-combustion CO ₂ capture. Chemical Engineering and Processing: Process Intensification, 2016, 107, 1-10.	1.8	24
111	Absorber modeling for NGCC carbon capture with aqueous piperazine. Faraday Discussions, 2016, 192, 459-477.	1.6	24
112	Nitrogen Dioxide Reaction with Alkaline Solids. Industrial & Engineering Chemistry Research, 1996, 35, 999-1005.	1.8	23
113	Modeling of CO ₂ Absorption Kinetics in Aqueous 2-Methylpiperazine. Industrial & Engineering Chemistry Research, 2013, 52, 4239-4248.	1.8	23
114	Reaction kinetics of carbon dioxide and hydroxide in aqueous glycerol. Chemical Engineering Science, 2017, 161, 151-158.	1.9	23
115	Absorption of Nitrogen Oxides in Aqueous Amines. Energy Procedia, 2014, 63, 830-847.	1.8	22
116	Modeling pilot plant results for CO ₂ stripping using piperazine in two stage flash. Energy Procedia, 2013, 37, 386-399.	1.8	21
117	Dimensionless Models for Predicting the Effective Area, Liquid-Film, and Gas-Film Mass-Transfer Coefficients of Packing. Industrial & Engineering Chemistry Research, 2016, 55, 5373-5384.	1.8	21
118	Optimum heat of absorption for CO ₂ capture using the advanced flash stripper. International Journal of Greenhouse Gas Control, 2016, 53, 169-177.	2.3	21
119	Control Relevant Model of Amine Scrubbing for CO ₂ Capture from Power Plants. Industrial & Engineering Chemistry Research, 2016, 55, 1690-1700.	1.8	21
120	Modeling of absorber pilot plant performance for CO ₂ capture with aqueous piperazine. International Journal of Greenhouse Gas Control, 2017, 64, 300-313.	2.3	21
121	Zero- and negative-emissions fossil-fired power plants using CO ₂ capture by conventional aqueous amines. International Journal of Greenhouse Gas Control, 2021, 111, 103473.	2.3	21
122	Thermodynamics and Equilibrium Solubility of Carbon Dioxide in Diglycolamine/Morpholine/Water. Journal of Chemical & Engineering Data, 2006, 51, 708-717.	1.0	20
123	Absorption of CO ₂ in aqueous blends of diglycolamine and morpholine. Chemical Engineering Science, 2006, 61, 3830-3837.	1.9	20
124	Thermodynamics of CO ₂ /2-Methylpiperazine/Water. Industrial & Engineering Chemistry Research, 2013, 52, 4229-4238.	1.8	20
125	Pilot-scale evaluation of concentrated piperazine for CO ₂ capture at an Australian coal-fired power station: Nitrosamine measurements. International Journal of Greenhouse Gas Control, 2015, 37, 256-263.	2.3	20
126	Hg absorption in aqueous permanganate. AIChE Journal, 1996, 42, 3559-3562.	1.8	19

#	ARTICLE	IF	CITATIONS
127	Modeling CO ₂ absorption into concentrated aqueous monoethanolamine and piperazine. <i>Chemical Engineering Science</i> , 2011, 66, 5212-5218.	1.9	19
128	Optimal CO ₂ Capture Operation in an Advanced Electric Grid. <i>Energy Procedia</i> , 2013, 37, 2585-2594.	1.8	19
129	Modeling of pilot stripper results for CO ₂ capture by aqueous piperazine. <i>International Journal of Greenhouse Gas Control</i> , 2013, 12, 280-287.	2.3	19
130	Thermal Degradation of Linear Amines for CO ₂ Capture. <i>Energy Procedia</i> , 2014, 63, 1558-1568.	1.8	19
131	Maximizing coal-fired power plant efficiency with integration of amine-based CO ₂ capture in greenfield and retrofit scenarios. <i>Energy</i> , 2014, 72, 824-831.	4.5	19
132	Calcium sulfite hemihydrate: Crystal growth rate and crystal habit. <i>Environmental Progress</i> , 1986, 5, 5-11.	0.8	18
133	Preparation of Calcium Silicate Absorbent from Iron Blast Furnace Slag. <i>Journal of the Air and Waste Management Association</i> , 2000, 50, 1655-1662.	0.9	18
134	CO ₂ Absorption from Gas Turbine Flue Gas by Aqueous Piperazine with Intercooling. <i>Industrial & Engineering Chemistry Research</i> , 2020, 59, 7174-7181.	1.8	18
135	Approximate simulation of CO ₂ and H ₂ s absorption into aqueous alkanolamines. <i>AIChE Journal</i> , 1993, 39, 1389-1397.	1.8	17
136	Thermal Decomposition of N-nitrosopiperazine. <i>Energy Procedia</i> , 2013, 37, 1678-1686.	1.8	17
137	Pilot Plant Activities with Concentrated Piperazine. <i>Energy Procedia</i> , 2014, 63, 1376-1391.	1.8	17
138	NO ₂ -Catalyzed Sulfite Oxidation. <i>Industrial & Engineering Chemistry Research</i> , 2015, 54, 4815-4822.	1.8	17
139	Thermodynamic and Mass-Transfer Modeling of Carbon Dioxide Absorption into Aqueous 2-Amino-2-Methyl-1-Propanol. <i>Industrial & Engineering Chemistry Research</i> , 2017, 56, 319-330.	1.8	17
140	Pilot testing of a heat integrated 0.7 MWe CO ₂ capture system with two-stage air-stripping: Emission. <i>International Journal of Greenhouse Gas Control</i> , 2017, 64, 267-275.	2.3	17
141	Effect of Liquid Viscosity on Mass Transfer Area and Liquid Film Mass Transfer Coefficient for GT-OPTIMPAK 250Y. <i>Energy Procedia</i> , 2017, 114, 2713-2727.	1.8	17
142	Thermal Degradation of Piperazine Blends with Diamines. <i>Energy Procedia</i> , 2013, 37, 1904-1911.	1.8	16
143	Pilot Plant Results with Piperazine. <i>Energy Procedia</i> , 2013, 37, 1572-1583.	1.8	16
144	Nitrosamine formation and mitigation in blended amines for CO ₂ capture. <i>International Journal of Greenhouse Gas Control</i> , 2015, 39, 329-334.	2.3	16

#	ARTICLE	IF	CITATIONS
145	Thermodynamic and mass transfer modeling of carbon dioxide absorption into aqueous 2-piperidineethanol. Chemical Engineering Science, 2016, 153, 295-307.	1.9	16
146	Characterization of Novel Structured Packings for CO2 Capture. Energy Procedia, 2013, 37, 2145-2153.	1.8	15
147	Energy Performance of Advanced Stripper Configurations. Energy Procedia, 2013, 37, 1696-1705.	1.8	15
148	Two-Stage Flash for CO2 Regeneration: Dynamic Modeling and Pilot Plant Validation. Energy Procedia, 2013, 37, 2133-2144.	1.8	15
149	Absorber intercooling in CO ₂ absorption by piperazine-promoted potassium carbonate. AIChE Journal, 2010, 56, 905-914.	1.8	14
150	Modeling Pilot Plant Performance of an Absorber with Aqueous Piperazine. Energy Procedia, 2013, 37, 1987-2001.	1.8	14
151	Carbon Capture with 4 m Piperazine/4 m 2-Methylpiperazine. Energy Procedia, 2013, 37, 436-447.	1.8	14
152	Absorption rates and CO2 solubility in new piperazine blends. Energy Procedia, 2013, 37, 370-385.	1.8	14
153	Managing n-nitrosopiperazine and dinitrosopiperazine. Energy Procedia, 2013, 37, 273-284.	1.8	14
154	CO2 Mass Transfer and Solubility in Aqueous Primary and Secondary Amine. Energy Procedia, 2014, 63, 1487-1496.	1.8	14
155	MEA and Piperazine Corrosion of Carbon Steel and Stainless Steel. Energy Procedia, 2017, 114, 1751-1764.	1.8	14
156	Effects of Viscosity on CO2 Absorption in Aqueous Piperazine/2-methylpiperazine. Energy Procedia, 2017, 114, 2103-2120.	1.8	14
157	Nitrogen Dioxide Absorption and Sulfide Oxidation in Aqueous Sulfide. Journal of the Air and Waste Management Association, 1999, 49, 332-338.	0.9	13
158	Absorption of HCl and SO2 from Humidified Flue Gas with Calcium Silicate Solids. Industrial & Engineering Chemistry Research, 2000, 39, 1048-1060.	1.8	13
159	Oxidative Degradation of Amine Solvents for CO2 Capture. Energy Procedia, 2014, 63, 1546-1557.	1.8	13
160	Pilot plant results with the piperazine advanced stripper at NGCC conditions. International Journal of Greenhouse Gas Control, 2022, 113, 103551.	2.3	13
161	Packing characterization: Absorber economic analysis. International Journal of Greenhouse Gas Control, 2015, 42, 124-131.	2.3	12
162	Modeling Amine Aerosol Growth at Realistic Pilot Plant Conditions. Energy Procedia, 2017, 114, 1045-1060.	1.8	12

#	ARTICLE	IF	CITATIONS
163	Review of Recent Pilot Plant Activities with Concentrated Piperazine. Energy Procedia, 2017, 114, 1110-1127.	1.8	12
164	Cold Rich Bypass to Strippers for CO ₂ Capture by Concentrated Piperazine. Chemical Engineering and Technology, 2014, 37, 149-156.	0.9	11
165	Pilot-scale evaluation of concentrated piperazine for CO ₂ capture at an Australian coal-fired power station: duration experiments. , 2015, 5, 363-373.		11
166	Regeneration Design for NGCC CO ₂ Capture with Amine-only and Hybrid Amine/Membrane. Energy Procedia, 2017, 114, 1394-1408.	1.8	11
167	Effect of limestone type and grind on SO ₂ scrubber performance. The cost-reduction effect of finer limestone grinding on SO ₂ scrubber efficiency can be very considerable. Environmental Progress, 1982, 1, 59-65.	0.8	9
168	Thermodynamic Modeling of Aqueous Piperazine/N-(2-Aminoethyl) Piperazine for CO ₂ Capture. Energy Procedia, 2014, 63, 997-1017.	1.8	9
169	Piperazine aerosol mitigation for post-combustion carbon capture. International Journal of Greenhouse Gas Control, 2019, 91, 102845.	2.3	9
170	Volatility of 2-(diethylamino)-ethanol and 2-((2-aminoethyl) amino) ethanol, a biphasic solvent for CO ₂ capture. International Journal of Greenhouse Gas Control, 2021, 106, 103257.	2.3	9
171	Preparation of calcium silicate absorbent from recycled glass. Environmental Progress, 1998, 17, 86-91.	0.8	8
172	Effect of mixing on efficiencies for reactive tray contactors. AIChE Journal, 2002, 48, 2537-2544.	1.8	7
173	Chlorine Absorption in Sulfite Solutions. Separation Science and Technology, 2004, 39, 3057-3077.	1.3	7
174	Piperazine/N-methylpiperazine/N,N-dimethylpiperazine as an Aqueous Solvent for Carbon Dioxide Capture. Oil and Gas Science and Technology, 2014, 69, 903-914.	1.4	7
175	Quantification of Gas and Aerosol-phase Piperazine Emissions by FTIR Under Variable Bench-scale Absorber Conditions. Energy Procedia, 2014, 63, 871-883.	1.8	7
176	Modeling Amine Aerosol Growth in the Absorber and Water Wash. Energy Procedia, 2017, 114, 959-976.	1.8	7
177	Effect of Deliquescent Salt Additives on the Reaction of SO ₂ with Ca(OH) ₂ . ACS Symposium Series, 1986, , 208-222.	0.5	6
178	Optimization of Stripping Piperazine with Variable Rich Loading. Energy Procedia, 2014, 63, 1842-1853.	1.8	6
179	Absorber Performance with High CO ₂ . Energy Procedia, 2014, 63, 1329-1338.	1.8	6
180	Corrosion by Aqueous Piperazine at 40–150 °C in Pilot Testing of CO ₂ Capture. Industrial & Engineering Chemistry Research, 2020, 59, 7189-7197.	1.8	6

#	ARTICLE	IF	CITATIONS
181	Creative absorber design and optimization for CO ₂ capture with aqueous piperazine. <i>International Journal of Greenhouse Gas Control</i> , 2022, 113, 103534.	2.3	6
182	CCS "A technology for now: general discussion. <i>Faraday Discussions</i> , 2016, 192, 125-151.	1.6	5
183	Comment on "Reassessing the Efficiency Penalty from Carbon Capture in Coal-Fired Power Plants". <i>Environmental Science & Technology</i> , 2016, 50, 6112-6113.	4.6	5
184	Amine Aerosol Characterization by Phase Doppler Interferometry. <i>Energy Procedia</i> , 2017, 114, 939-951.	1.8	5
185	Heat Transfer Enhancement and Optimization of Lean/Rich Solvent Cross Exchanger for Amine Scrubbing. <i>Energy Procedia</i> , 2017, 114, 1890-1903.	1.8	5
186	Corrosion of carbon steel by aqueous piperazine protected by FeCO ₃ . <i>International Journal of Greenhouse Gas Control</i> , 2019, 85, 23-29.	2.3	5
187	FEED for Piperazine with the Advanced Stripperr, on NGCC at Denver City, Texas. <i>SSRN Electronic Journal</i> , 0, , .	0.4	5
188	Dynamic Modeling, Validation, and Time Scale Decomposition of an Advanced Post-combustion Amine Scrubbing Process. <i>Energy Procedia</i> , 2014, 63, 1296-1307.	1.8	4
189	Effects of Catalysts, Inhibitors, and Contaminants on Piperazine Oxidation. <i>Energy Procedia</i> , 2017, 114, 1919-1929.	1.8	4
190	Demonstrating solvent management technologies for an aqueous AMP/PZ solvent. <i>SSRN Electronic Journal</i> , 0, , .	0.4	4
191	Process design of the piperazine advanced stripper for a 460 MW NGCC. <i>International Journal of Greenhouse Gas Control</i> , 2022, 115, 103631.	2.3	4
192	Oxidation of Aqueous Piperazine: Oxidation Rates, Products, and High-Temperature Oxidation. <i>ACS Symposium Series</i> , 2012, , 219-237.	0.5	3
193	Thermally Degraded Diglycolamine®/Dimethylaminoethoxyethanol for CO ₂ Capture. <i>Energy Procedia</i> , 2017, 114, 1737-1750.	1.8	3
194	CO ₂ Absorption Rate and Solubility in Monoethanolamine/Piperazine/Water. , 0, .		3
195	Effects of carbon treating on piperazine oxidation in pilot plant testing of PZAS, . <i>International Journal of Greenhouse Gas Control</i> , 2021, 112, 103502.	2.3	3
196	Sulfur Dioxide Vapor Pressure and pH of Sodium Citrate Buffer Solutions with Dissolved Sulfur Dioxide. <i>ACS Symposium Series</i> , 1980, , 269-291.	0.5	2
197	The Impact of Electricity Market Conditions on the Value of Flexible CO ₂ Capture. , 2012, , .		2
198	Thermodynamic and Kinetic Modeling of Piperazine/2-Methylpiperazine. <i>Energy Procedia</i> , 2014, 63, 1243-1255.	1.8	2

#	ARTICLE	IF	CITATIONS
199	Thermal Degradation of Piperazine/4-Hydroxy-1-methylpiperidine for CO ₂ Capture. Industrial & Engineering Chemistry Research, 2016, 55, 10004-10010.	1.8	2
200	Thermal Degradation of Aminosilicone Carbamates. Energy & Fuels, 2016, 30, 10671-10678.	2.5	2
201	Process control of the advanced flash stripper for CO ₂ solvent regeneration. Chemical Engineering and Processing: Process Intensification, 2016, 107, 21-28.	1.8	2
202	Field Measurement of Amine Aerosol by FTIR and Phase Doppler Interferometry. Energy Procedia, 2017, 114, 906-929.	1.8	2
203	Pilot Plant Results With the Piperazine Advanced Stripper at NGCC Conditions. SSRN Electronic Journal, 0, , .	0.4	2
204	Activity Coefficients Predicted by the Local Composition Model for Aqueous Solutions Used in Flue Gas Desulfurization. ACS Symposium Series, 1986, , 223-233.	0.5	1
205	Subspace system identification for CO ₂ recovery processes. , 2011, , .		1
206	Heat Loss and Energy Use in Pilot Plant Testing of Piperazine With the Advanced Stripper. SSRN Electronic Journal, 0, , .	0.4	1
207	CO ₂ Absorption Rate and Capacity of Semi-Aqueous Piperazine for CO ₂ Capture. SSRN Electronic Journal, 0, , .	0.4	1
208	Fe ²⁺ Solubility and Siderite Formation in Monoethanolamine and Piperazine Solvents. SSRN Electronic Journal, 0, , .	0.4	1
209	Energy use of piperazine with the advanced stripper from pilot plant testing. International Journal of Greenhouse Gas Control, 2022, 113, 103531.	2.3	1
210	Diethylenetriamine solutions for stack gas desulfurization by absorption/stripping. Drastic reductions in steam consumption can be realized in a simple absorption/stripping process. Environmental Progress, 1982, 1, 160-167.	0.8	0
211	Modeling CO ₂ recovery for optimal dynamic operations. , 2011, , .		0
212	Control of carbon dioxide solubility in aqueous piperazine. Computers and Chemical Engineering, 2013, 54, 122-124.	2.0	0
213	An effective multi-loop control system to improve control performance of CO ₂ capture. , 2013, , .		0
214	Advancing CO ₂ Capture from Natural Gas Combined Cycle Power Plants with Piperazine Scrubbing. SSRN Electronic Journal, 0, , .	0.4	0