## F Marc Michel

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/8530369/publications.pdf

Version: 2024-02-01

623734 1,546 20 14 citations h-index papers

g-index 20 20 20 2142 docs citations times ranked citing authors all docs

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#	Article	IF	CITATIONS
1	Quantitative measurement of corrosion at the nanoscale by in situ spectral modulation interferometry. Materials Characterization, 2022, 189, 111992.	4.4	3
2	Kinetic Effect of Surface Chemisorbed Oxygen on Platinum-Catalyzed Hydrogen Peroxide Decomposition. Catalysis Letters, 2021, 151, 138-146.	2.6	12
3	Mechanism and Kinetics of Methane Oxidation to Methanol Catalyzed by AuPd Nanocatalysts at Low Temperature. ACS Catalysis, 2021, 11, 2837-2845.	11.2	12
4	Ultrahigh tribocorrosion resistance of metals enabled by nano-layering. Acta Materialia, 2021, 206, 116609.	7.9	15
5	Impacts of Initial Ca/P on Amorphous Calcium Phosphate. Crystal Growth and Design, 2021, 21, 3736-3745.	3.0	19
6	Nanocomposite structure of two-line ferrihydrite powder from total scattering. Communications Chemistry, 2020, 3, .	4.5	28
7	Decomposition of Hydrogen Peroxide Catalyzed by AuPd Nanocatalysts during Methane Oxidation to Methanol. ACS Catalysis, 2020, 10, 5115-5123.	11,2	25
8	The origin of passivity in aluminum-manganese solid solutions. Corrosion Science, 2020, 173, 108749.	6.6	22
9	Structure of single sheet iron oxides produced from surfactant interlayered green rusts. Applied Clay Science, 2019, 170, 86-96.	5.2	7
10	Effect of particle size on catalytic decomposition of hydrogen peroxide by platinum nanocatalysts. Journal of Catalysis, 2019, 373, 58-66.	6.2	25
11	Suitability of 3D-Printed devices for low-temperature geochemical experiments. Applied Geochemistry, 2018, 98, 121-126.	3.0	7
12	Mechanism and Kinetics of Hydrogen Peroxide Decomposition on Platinum Nanocatalysts. ACS Applied Materials & Decomposition on Platinum Nanocatalysts. ACS Applied Materials & Decomposition on Platinum Nanocatalysts. ACS Applied Materials & Decomposition on Platinum Nanocatalysts.	8.0	94
13	Abundance and Speciation of Surface Oxygen on Nanosized Platinum Catalysts and Effect on Catalytic Activity. ACS Applied Energy Materials, 2018, 1, 3255-3266.	5.1	12
14	3D printed mixed flow reactor for geochemical rate measurements. Applied Geochemistry, 2018, 89, 86-91.	3.0	16
15	Comparison of isoelectric points of single-crystal and polycrystalline α-Al <sub>2</sub> O <sub>3</sub> American Mineralogist, 2016, 101, 2248-2259.	1.9	19
16	Properties of impurity-bearing ferrihydrite II: Insights into the surface structure and composition of pure, Al- and Si-bearing ferrihydrite from Zn(II) sorption experiments and Zn K-edge X-ray absorption spectroscopy. Geochimica Et Cosmochimica Acta, 2013, 119, 46-60.	3.9	68
17	Reactivity of ferritin and the structure of ferritin-derived ferrihydrite. Biochimica Et Biophysica Acta - General Subjects, 2010, 1800, 871-885.	2.4	43
18	Ordered ferrimagnetic form of ferrihydrite reveals links among structure, composition, and magnetism. Proceedings of the National Academy of Sciences of the United States of America, 2010, 107, 2787-2792.	7.1	312

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#	Article	lF	CITATIONS
19	The Structure of Ferrihydrite, a Nanocrystalline Material. Science, 2007, 316, 1726-1729.	12.6	754
20	Characterization and Surface Reactivity of Ferrihydrite Nanoparticles Assembled in Ferritin. Langmuir, 2006, 22, 9313-9321.	3 <b>.</b> 5	53