Hao Yi

List of Publications by Year in descending order

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346980 355658 1,525 45 22 38 citations h-index g-index papers 45 45 45 1236 citing authors all docs docs citations times ranked

#	Article	IF	Citations
1	Montmorillonite facilitated Pb(II) biomineralization by Chlorella sorokiniana FK in soil. Journal of Hazardous Materials, 2022, 423, 127007.	6.5	21
2	Three-dimensional montmorillonite/Ag nanowire aerogel supported stearic acid as composite phase change materials for superior solar-thermal energy harvesting and storage. Composites Science and Technology, 2022, 217, 109121.	3.8	49
3	3D-printed montmorillonite nanosheets based hydrogel with biocompatible polymers as excellent adsorbent for Pb(â;) removal. Separation and Purification Technology, 2022, 283, 120176.	3.9	34
4	Double-layered montmorillonite/MoS2 aerogel with vertical channel for efficient and stable solar interfacial desalination. Applied Clay Science, 2022, 217, 106389.	2.6	17
5	Selective flotation separation of bastnaesite from dolomite using \hat{l}^2 -naphthyl sulfonate formaldehyde condensate as depressant: Experimental and calculational studies. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2022, 639, 128380.	2.3	9
6	Simultaneous Recovery of Niobium and Sulfur from Carbonate Niobite Ore with Flotation. Minerals (Basel, Switzerland), 2022, 12, 432.	0.8	1
7	Mechanically strong hectorite aerogel encapsulated octadecane as shape-stabilized phase change materials for thermal energy storage and management. Applied Clay Science, 2022, 223, 106511.	2.6	2
8	Development of superior stable two-dimensional montmorillonite nanosheet based working nanofluids for direct solar energy harvesting and utilization. Applied Clay Science, 2021, 200, 105886.	2.6	16
9	Activation of Fenton reaction by controllable oxygen incorporation in MoS2-Fe under visible light irradiation. Applied Surface Science, 2021, 566, 150674.	3.1	15
10	Rich Se nanoparticles modified cobalt carbonate hydroxide as an efficient electrocatalyst for boosted hydrogen evolution in alkaline conditions. Applied Surface Science, 2021, 565, 150505.	3.1	13
11	Recent advances in engineering cobalt carbonate hydroxide for enhanced alkaline water splitting. Journal of Alloys and Compounds, 2021, 887, 161405.	2.8	23
12	Design of 3D-network montmorillonite nanosheet/stearic acid shape-stabilized phase change materials for solar energy storage. Solar Energy Materials and Solar Cells, 2020, 204, 110233.	3.0	78
13	Superior arsenate adsorption and comprehensive investigation of adsorption mechanism on novel Mn-doped La2O2CO3 composites. Chemical Engineering Journal, 2020, 391, 123623.	6.6	26
14	Effect of magnesium ion on sylvite flotation: An experiment and molecular dynamic simulation study. Chemical Physics Letters, 2020, 752, 137586.	1.2	11
15	Atomic insights into flotation separation of KCl and NaCl from a new viewpoint of hydration layer: A molecular dynamic study. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2020, 602, 125071.	2.3	6
16	Hydrophobic agglomeration of montmorillonite fines in aqueous solutions induced by dodecyl trimethyl ammonium bromides. Chemical Physics Letters, 2020, 739, 136999.	1.2	6
17	Development of 2D-Mt/SA/AgNPs microencapsulation phase change materials for solar energy storage with enhancement of thermal conductivity and latent heat capacity. Solar Energy Materials and Solar Cells, 2019, 201, 110090.	3.0	37
18	Synthesis of chitosan cross-linked 3D network-structured hydrogel for methylene blue removal. International Journal of Biological Macromolecules, 2019, 141, 98-107.	3.6	55

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19	A novel core-shell structural montmorillonite nanosheets/stearic acid composite PCM for great promotion of thermal energy storage properties. Solar Energy Materials and Solar Cells, 2019, 192, 57-64.	3.0	91
20	$Pb(\widehat{l}^{\text{TM}}\widehat{l}^{\text{TM}})$ removal from water using porous hydrogel of chitosan-2D montmorillonite. International Journal of Biological Macromolecules, 2019, 128, 85-93.	3.6	70
21	Colloidal stability of silica and graphite in aqueous suspensions. Chemical Physics, 2019, 525, 110405.	0.9	7
22	Effect of interlayer cations on exfoliating 2D montmorillonite nanosheets with high aspect ratio: From experiment to molecular calculation. Ceramics International, 2019, 45, 17054-17063.	2.3	16
23	Design of MtNS/SA microencapsulated phase change materials for enhancement of thermal energy storage performances: Effect of shell thickness. Solar Energy Materials and Solar Cells, 2019, 200, 109935.	3.0	31
24	A novel method for surface wettability modification of talc through thermal treatment. Applied Clay Science, 2019, 176, 21-28.	2.6	24
25	Competition of Hg2+ adsorption and surface oxidation on MoS2 surface as affected by sulfur vacancy defects. Applied Surface Science, 2019, 483, 521-528.	3.1	47
26	Hydrophobic agglomeration behaviors of clay minerals as affected by siloxane structure. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2019, 568, 36-42.	2.3	24
27	Correlation of exfoliation performance with interlayer cations of montmorillonite in the preparation of twoâ€dimensional nanosheets. Journal of the American Ceramic Society, 2019, 102, 3908-3922.	1.9	29
28	Driving force for the swelling of montmorillonite as affected by surface charge and exchangeable cations: A molecular dynamic study. Results in Physics, 2019, 12, 113-117.	2.0	36
29	Removal of methylene blue from water with montmorillonite nanosheets/chitosan hydrogels as adsorbent. Applied Surface Science, 2018, 448, 203-211.	3.1	208
30	Removal of graphene oxide from water by floc-flotation. Separation and Purification Technology, 2018, 202, 27-33.	3.9	21
31	AFM study on the wettability of mica and graphite modified with surfactant DTAB. Journal of Dispersion Science and Technology, 2018, 39, 1060-1064.	1.3	10
32	Molecular dynamics simulations study for the effect of cations hydration on the surface tension of the electrolyte solutions. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2018, 539, 80-84.	2.3	35
33	Preparation and characterization of self-assembly hydrogels with exfoliated montmorillonite nanosheets and chitosan. Nanotechnology, 2018, 29, 025605.	1.3	27
34	Hydrophobic agglomeration of talc fines in aqueous suspensions. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2018, 538, 327-332.	2.3	28
35	Thermal Modification of the Molybdenum Disulfide Surface for Tremendous Improvement of Hg ²⁺ Adsorption from Aqueous Solution. ACS Sustainable Chemistry and Engineering, 2018, 6, 9065-9073.	3.2	48
36	Surface wettability of montmorillonite (0†0†1) surface as affected by surface charge and exchangeable cations: A molecular dynamic study. Applied Surface Science, 2018, 459, 148-154.	3.1	113

#	Article	IF	Citations
37	Vanadium Transitions during Roasting-Leaching Process of Vanadium Extraction from Stone Coal. Minerals (Basel, Switzerland), 2018, 8, 63.	0.8	17
38	Can carboxymethyl cellulose molecules bind swelling montmorillonite layers in Water?. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2018, 553, 515-519.	2.3	11
39	A novel method for the quantitative determination of defects on graphene surfaces. Journal of Colloid and Interface Science, 2017, 499, 62-66.	5.0	10
40	Correlation of montmorillonite exfoliation with interlayer cations in the preparation of two-dimensional nanosheets. RSC Advances, 2017, 7, 41471-41478.	1.7	49
41	A novel method for determining the thickness of hydration shells on nanosheets: A case of montmorillonite in water. Powder Technology, 2017, 306, 74-79.	2.1	49
42	Hydrophobic agglomeration kinetics of fine kaolinite particles in aqueous suspensions. Journal of Dispersion Science and Technology, 2017, 38, 1336-1341.	1.3	7
43	A Novel Model of Aggregate Gradation for Autoclaved Bricks from Tailings. Minerals (Basel,) Tj ETQq1 1 0.78431	4 rgBT /Ov	verlock 10 Tf
44	Study on the differences of Na- and Ca-montmorillonites in crystalline swelling regime through molecular dynamics simulation. Advanced Powder Technology, 2016, 27, 779-785.	2.0	55
45	Molecular dynamics simulations of hydration shell on montmorillonite (001) in water. Surface and Interface Analysis, 2016, 48, 976-980.	0.8	37