

Jurgens H De Lange

List of Publications by Year in descending order

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#	ARTICLE	IF	CITATIONS
1	Physical Nature of Interactions in Zn ^{II} Complexes with 2,2'-Bipyridyl: Quantum Theory of Atoms in Molecules (QTAIM), Interacting Quantum Atoms (IQA), Noncovalent Interactions (NCI), and Extended Transition State Coupled with Natural Orbitals for Chemical Valence (ETS-NOCV) Comparative Studies. <i>Journal of Physical Chemistry A</i> , 2014, 118, 623-637.	2.5	81
2	Evaluating common QTAIM and NCI interpretations of the electron density concentration through IQA interaction energies and 1D cross-sections of the electron and deformation density distributions. <i>Computational and Theoretical Chemistry</i> , 2015, 1053, 60-76.	2.5	50
3	Adsorption of antiretroviral drugs, efavirenz and nevirapine from aqueous solution by graphene wool: Kinetic, equilibrium, thermodynamic and computational studies. <i>Applied Surface Science Advances</i> , 2021, 6, 100157.	6.8	30
4	Synthesis, structure and DFT study of asymmetrical NHC complexes of cymantrene derivatives and their application in the dehydrogenative dimerization reaction of thiols. <i>Journal of Organometallic Chemistry</i> , 2017, 840, 11-22.	1.8	16
5	Origin of Hydrocarbons Stability from a Computational Perspective: A Case Study of Ortho-Xylene Isomers. <i>ChemPhysChem</i> , 2020, 21, 494-502.	2.1	15
6	Exploring fundamental differences between red- and blue-shifted intramolecular hydrogen bonds using FAMSEC, FALDI, IQA and QTAIM. <i>Structural Chemistry</i> , 2017, 28, 1429-1444.	2.0	13
7	Toward deformation densities for intramolecular interactions without radical reference states using the fragment, atom, localized, delocalized, and interatomic (FALDI) charge density decomposition scheme. <i>Journal of Computational Chemistry</i> , 2017, 38, 981-997.	3.3	12
8	FALDI-based decomposition of an atomic interaction line leads to 3D representation of the multicenter nature of interactions. <i>Journal of Computational Chemistry</i> , 2018, 39, 973-985.	3.3	10
9	Quantifying individual (anti)bonding molecular orbitals' contributions to chemical bonding. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 20988-20998.	2.8	10
10	FALDI-based criterion for and the origin of an electron density bridge with an associated (3, -1) critical point on Bader's molecular graph. <i>Journal of Computational Chemistry</i> , 2018, 39, 2283-2299.	3.3	8
11	Exact and exclusive electron localization indices within QTAIM atomic basins. <i>Journal of Computational Chemistry</i> , 2018, 39, 1517-1530.	3.3	7
12	The CH...HC interaction in biphenyl is a delocalized, molecular-wide and entirely nonclassical interaction: Results from FALDI analysis. <i>Journal of Computational Chemistry</i> , 2021, 42, 706-718.	3.3	7
13	Gold(I) Hydrides as Proton Acceptors in Dihydrogen Bond Formation. <i>ChemPhysChem</i> , 2017, 18, 2288-2294.	2.1	3
14	Molecular Orbitals Support Energy-Stabilizing σ -Bonding Nature of Bader's Bond Paths. <i>Journal of Physical Chemistry A</i> , 2020, 124, 5523-5533.	2.5	2
15	Comparison of DAFH and FALDI-like approaches. <i>Theoretical Chemistry Accounts</i> , 2020, 139, 1.	1.4	1
16	Characterization of bonding modes in metal complexes through electron density cross-sections. <i>Journal of Computational Chemistry</i> , 2020, 41, 2695-2706.	3.3	1