

# Yuri Perelygin

## List of Publications by Citations

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**Version:** 2024-04-27

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The third column is the impact factor (IF) of the journal, and the fourth column is the number of citations of the article.

12  
papers

46  
citations

4  
h-index

6  
g-index

12  
ext. papers

50  
ext. citations

0.7  
avg, IF

1.32  
L-index

#	Paper	IF	Citations
12	Boric acid. <i>Russian Journal of Applied Chemistry</i> , <b>2006</b> , 79, 2041-2042	0.8	14
11	A new corrosion-resistant multilayer material. <i>Protection of Metals and Physical Chemistry of Surfaces</i> , <b>2014</b> , 50, 856-859	0.9	8
10	Electrodeposition of bismuth from acid lactate electrolyte. <i>Russian Journal of Applied Chemistry</i> , <b>2006</b> , 79, 1200-1201	0.8	5
9	Electrodeposition of Nickel-Tungsten Alloy from Acetate Electrolyte. <i>Russian Journal of Applied Chemistry</i> , <b>2003</b> , 76, 1524-1526	0.8	4
8	Multilayer metallic material with specific properties and the technology of its production. <i>Russian Metallurgy (Metally)</i> , <b>2017</b> , 2017, 426-431	0.5	3
7	Electrodeposition of Palladium-Zinc Alloy from Aminoacetic Electrolyte. <i>Russian Journal of Applied Chemistry</i> , <b>2002</b> , 75, 1713-1714	0.8	3
6	Electrodeposition of Palladium-Zinc Alloy from the Ammonium Chloride-Na <sub>2</sub> EDTA Electrolyte. <i>Russian Journal of Applied Chemistry</i> , <b>2001</b> , 74, 1958-1959	0.8	3
5	Accelerated Corrosion Tests of a New Class of Multilayer Metallic Materials with an Internal Protector. <i>Russian Metallurgy (Metally)</i> , <b>2019</b> , 2019, 247-256	0.5	2
4	Electrodeposition of a lead-indium alloy from an acetic electrolyte. <i>Russian Journal of Applied Chemistry</i> , <b>2010</b> , 83, 165-167	0.8	2
3	Electrolytic Deposition of the Zinc-Cobalt Alloy. <i>Russian Journal of Applied Chemistry</i> , <b>2001</b> , 74, 421-423	0.8	2
2	On the term $\beta H$ of the precipitation onset of heavy metal hydroxides. <i>Russian Journal of Applied Chemistry</i> , <b>2006</b> , 79, 492-493	0.8	0
1	Calculation of a relative fraction of metal ion passed in hydroxy complex depending on pH of a solution. <i>Russian Journal of Applied Chemistry</i> , <b>2011</b> , 84, 1073-1074	0.8	