

# Martin Mayer

## List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/8235264/publications.pdf>

Version: 2024-02-01

10

papers

241

citations

1307594

7

h-index

1372567

10

g-index

11

all docs

11

docs citations

11

times ranked

199

citing authors

#	ARTICLE	IF	CITATIONS
1	Online Monitoring of Isomeric Reaction Intermediates. <i>Journal of Physical Chemistry A</i> , 2021, 125, 2801-2815.	2.5	9
2	Relevance of $\pi$ -Backbonding for the Reactivity of Electrophilic Anions $[B_{12}X_{11}]^{+}$ ( $X=F, Cl, Br, I, CN$ ). <i>Chemistry - A European Journal</i> , 2021, 27, 10274-10281.	3.3	15
3	Direct functionalization of C-H bonds by electrophilic anions. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2020, 117, 23374-23379.	7.1	21
4	First steps towards a stable neon compound: observation and bonding analysis of $[B_{12}(CN)_{11}Ne]^{+}$ . <i>Chemical Communications</i> , 2020, 56, 4591-4594.	4.1	26
5	Unravelling the configuration of transient <i>ortho</i> -quinone methides by combining microfluidics with gas phase vibrational spectroscopy. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 4610-4616.	2.8	4
6	Rational design of an argon-binding superelectrophilic anion. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2019, 116, 8167-8172.	7.1	69
7	Joining Microfluidics with Infrared Photodissociation: Online Monitoring of Isomeric Flow-Reaction Intermediates. <i>Analytical Chemistry</i> , 2019, 91, 3199-3203.	6.5	18
8	Probing the propensity of perchlorate anions for surface solvation by infrared photodissociation spectroscopy. <i>Journal of Chemical Physics</i> , 2018, 148, 222840.	3.0	7
9	Superelectrophilic Behavior of an Anion Demonstrated by the Spontaneous Binding of Noble Gases to $[B_{12}Cl_{11}]^{+}$ . <i>Angewandte Chemie - International Edition</i> , 2017, 56, 7980-7985.	13.8	55
10	Superelektrrophiles Verhalten eines Anions demonstriert durch spontane Bindung von Edelgasen an $[B_{12}Cl_{11}]^{+}$ . <i>Angewandte Chemie</i> , 2017, 129, 8090-8096.	2.0	17