

# Mark A Blitz

## List of Publications by Year in descending order

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132  
papers

4,212  
citations

76326

40  
h-index

149698

56  
g-index

146  
all docs

146  
docs citations

146  
times ranked

3151  
citing authors

#	ARTICLE	IF	CITATIONS
1	Direct Measurements of Isoprene Autoxidation: Pinpointing Atmospheric Oxidation in Tropical Forests. <i>Jacs Au</i> , 2022, 2, 809-818.	7.9	6
2	The reaction between HgBr and O <sub>3</sub> : kinetic study and atmospheric implications. <i>Physical Chemistry Chemical Physics</i> , 2022, , .	2.8	8
3	Identification, monitoring, and reaction kinetics of reactive trace species using time-resolved mid-infrared quantum cascade laser absorption spectroscopy: development, characterisation, and initial results for the CH <sub>2</sub> OO Criegee intermediate. <i>Atmospheric Measurement Techniques</i> , 2022, 15, 2875-2887.	3.1	2
4	Kinetics of the gas phase reaction of the Criegee intermediate CH <sub>2</sub> OO with SO <sub>2</sub> as a function of temperature. <i>Physical Chemistry Chemical Physics</i> , 2021, 23, 19415-19423.	2.8	10
5	Production of HONO from NO <sub>2</sub> ; uptake on illuminated TiO <sub>2</sub> aerosol particles and following the illumination of mixed TiO <sub>2</sub> •ammonium nitrate particles. <i>Atmospheric Chemistry and Physics</i> , 2021, 21, 5755-5775.	4.9	14
6	Global Master Equation Analysis of Rate Data for the Reaction C <sub>2</sub> H <sub>4</sub> + H•, C <sub>2</sub> H <sub>5</sub> :• + HO• → C <sub>2</sub> H <sub>5</sub> . <i>Journal of Physical Chemistry A</i> , 2021, 125, 9548-9565.	2.5	3
7	OH Kinetics with a Range of Nitrogen-Containing Compounds: N-Methylformamide, t-Butylamine, and N-Methyl-propane Diamine. <i>Journal of Physical Chemistry A</i> , 2021, 125, 10439-10450.	2.5	0
8	Kinetics of the Gas Phase Reactions of the Criegee Intermediate CH <sub>2</sub> OO with O <sub>3</sub> and IO. <i>Journal of Physical Chemistry A</i> , 2020, 124, 6287-6293.	2.5	7
9	Kinetic Study of the Reactions PO + O <sub>2</sub> and PO <sub>2</sub> + O <sub>3</sub> and Spectroscopy of the PO Radical. <i>Journal of Physical Chemistry A</i> , 2020, 124, 7911-7926.	2.5	10
10	A gas-to-particle conversion mechanism helps to explain atmospheric particle formation through clustering of iodine oxides. <i>Nature Communications</i> , 2020, 11, 4521.	12.8	39
11	Kinetics of the Reactions of Hydroxyl Radicals with Furan and Its Alkylated Derivatives 2-Methyl Furan and 2,5-Dimethyl Furan. <i>Journal of Physical Chemistry A</i> , 2020, 124, 7416-7426.	2.5	14
12	Rate coefficients for the reactions of OH with butanols from 298 K to temperatures relevant for low-temperature combustion. <i>International Journal of Chemical Kinetics</i> , 2020, 52, 1046-1059.	1.6	7
13	Direct Trace Fitting of Experimental Data Using the Master Equation: Testing Theory and Experiments on the OH + C <sub>2</sub> H <sub>4</sub> Reaction. <i>Journal of Physical Chemistry A</i> , 2020, 124, 4015-4024.	2.5	12
14	CH <sub>2</sub> OO Criegee intermediate UV absorption cross-sections and kinetics of CH <sub>2</sub> OO + CH <sub>2</sub> OO and CH <sub>2</sub> OO + I as a function of pressure. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 9448-9459.	2.8	25
15	A new instrument for time-resolved measurement of HO <sub>2</sub> radicals. <i>Atmospheric Measurement Techniques</i> , 2020, 13, 839-852.	3.1	6
16	Determination of the absorption cross sections of higher-order iodine oxides at 355 and 532 nm. <i>Atmospheric Chemistry and Physics</i> , 2020, 20, 10865-10887.	4.9	14
17	Time-Resolved Measurements and Master Equation Modelling of the Unimolecular Decomposition of CH <sub>3</sub> OCH <sub>2</sub> . <i>Zeitschrift Fur Physikalische Chemie</i> , 2020, 234, 1233-1250.	2.8	2
18	Experimental Study of the Removal of Ground- and Excited-State Phosphorus Atoms by Atmospherically Relevant Species. <i>Journal of Physical Chemistry A</i> , 2019, 123, 9469-9478.	2.5	19

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19	Temperature and Pressure Dependent Kinetics of QOOH Decomposition and Reaction with $O_2$ : Experimental and Theoretical Investigations of QOOH Radicals Derived from $Cl + (CH_3)_3COOH$ . <i>Journal of Physical Chemistry A</i> , 2019, 123, 10254-10262.	2.5	11
20	Measurements of Low Temperature Rate Coefficients for the Reaction of CH with $CH_2O$ and Application to Dark Cloud and AGB Stellar Wind Models. <i>Astrophysical Journal</i> , 2019, 885, 134.	4.5	13
21	A generic method for determining $R_{OH} + O_2$ rate parameters via OH regeneration. <i>Chemical Physics Letters</i> , 2019, 730, 213-219.	2.6	4
22	Low temperature gas phase reaction rate coefficient measurements: Toward modeling of stellar winds and the interstellar medium. <i>Proceedings of the International Astronomical Union</i> , 2019, 15, 382-383.	0.0	0
23	Low temperature studies of the rate coefficients and branching ratios of reactive loss vs quenching for the reactions of $1CH_2$ with $C_2H_6$ , $C_2H_4$ , $C_2H_2$ . <i>Icarus</i> , 2019, 321, 752-766.	2.5	8
24	Comment on "Methanol dimer formation drastically enhances hydrogen abstraction from methanol by OH at low temperature" by W. Siebrand, Z. Smedarchina, E. Martínez-Núñez and A. Fernández-Ramos, <i>Phys. Chem. Chem. Phys.</i> , 2016, 18, 22712. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 8349-8354.	2.8	10
25	Low temperature studies of the removal reactions of $1CH_2$ with particular relevance to the atmosphere of Titan. <i>Icarus</i> , 2018, 303, 10-21.	2.5	12
26	A novel multiplex absorption spectrometer for time-resolved studies. <i>Review of Scientific Instruments</i> , 2018, 89, 024101.	1.3	10
27	Exploring the features on the $OH + SO_2$ potential energy surface using theory and testing its accuracy by comparison to experimental data. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 8984-8990.	2.8	5
28	Laser Photolysis Kinetic Study of OH Radical Reactions with Methyl <i>tert</i> -Butyl Ether and Trimethyl Orthoformate under Conditions Relevant to Low Temperature Combustion: Measurements of Rate Coefficients and OH Recycling. <i>Journal of Physical Chemistry A</i> , 2018, 122, 9701-9711.	2.5	10
29	Unimolecular decomposition kinetics of the stabilised Criegee intermediates $CH_2OO$ and $CD_2OO$ . <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 24940-24954.	2.8	41
30	Structure-switching $M_3L_2$ Ir(III) coordination cages with photo-isomerising azo-aromatic linkers. <i>Chemical Science</i> , 2018, 9, 8150-8159.	7.4	69
31	Kinetic studies of $C_1$ and $C_2$ Criegee intermediates with $SO_2$ using laser flash photolysis coupled with photoionization mass spectrometry and time resolved UV absorption spectroscopy. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 22218-22227.	2.8	25
32	Kinetics of the Reaction of OH with Isoprene over a Wide Range of Temperature and Pressure Including Direct Observation of Equilibrium with the OH Adducts. <i>Journal of Physical Chemistry A</i> , 2018, 122, 7239-7255.	2.5	16
33	An Experimental Study of the Kinetics of $OH/OD$ ( $v = 1, 2, 3$ ) + $SO_2$ : The Limiting High-Pressure Rate Coefficients as a Function of Temperature. <i>Journal of Physical Chemistry A</i> , 2017, 121, 3175-3183.	2.5	10
34	An Experimental and Master Equation Study of the Kinetics of $OH/OD + SO_2$ : The Limiting High-Pressure Rate Coefficients. <i>Journal of Physical Chemistry A</i> , 2017, 121, 3184-3191.	2.5	11
35	Obtaining effective rate coefficients to describe the decomposition kinetics of the corannulene oxyradical at high temperatures. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 11064-11074.	2.8	11
36	OH production from the photolysis of isoprene-derived peroxy radicals: cross-sections, quantum yields and atmospheric implications. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 2332-2345.	2.8	16

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37	An instrument to measure fast gas phase radical kinetics at high temperatures and pressures. Review of Scientific Instruments, 2016, 87, 054102.	1.3	8
38	Observation of a new channel, the production of CH <sub>3</sub> , in the abstraction reaction of OH radicals with acetaldehyde. Physical Chemistry Chemical Physics, 2016, 18, 26423-26433.	2.8	10
39	Bimolecular reactions of activated species: An analysis of problematic HC(O)C(O) chemistry. Chemical Physics Letters, 2016, 661, 58-64.	2.6	14
40	Temperature and Pressure Studies of the Reactions of CH <sub>3</sub> O <sub>2</sub> , HO <sub>2</sub> , and 1,2-C <sub>4</sub> H <sub>9</sub> O <sub>2</sub> with NO <sub>2</sub> . Journal of Physical Chemistry A, 2016, 120, 1408-1420.	2.5	12
41	Global Uncertainty Propagation and Sensitivity Analysis in the CH <sub>3</sub> OCH <sub>2</sub> + O <sub>2</sub> System: Combining Experiment and Theory To Constrain Key Rate Coefficients in DME Combustion. Journal of Physical Chemistry A, 2015, 119, 7430-7438.	2.5	27
42	Reanalysis of Rate Data for the Reaction CH <sub>3</sub> + CH <sub>3</sub> → C <sub>2</sub> H <sub>6</sub> Using Revised Cross Sections and a Linearized Second-Order Master Equation. Journal of Physical Chemistry A, 2015, 119, 7668-7682.	2.5	28
43	Direct evidence for a substantive reaction between the Criegee intermediate, CH <sub>2</sub> OO, and the water vapour dimer. Physical Chemistry Chemical Physics, 2015, 17, 4859-4863.	2.8	155
44	The importance of OH radical "neutral low temperature tunnelling reactions in interstellar clouds using a new model. Molecular Physics, 2015, 113, 2243-2254.	1.7	35
45	Branching ratios for the reactions of OH with ethanol amines used in carbon capture and the potential impact on carcinogen formation in the emission plume from a carbon capture plant. Physical Chemistry Chemical Physics, 2015, 17, 25342-25353.	2.8	14
46	Measurements of Rate Coefficients for Reactions of OH with Ethanol and Propan-2-ol at Very Low Temperatures. Journal of Physical Chemistry A, 2015, 119, 7130-7137.	2.5	45
47	A combined experimental and theoretical study of reactions between the hydroxyl radical and oxygenated hydrocarbons relevant to astrochemical environments. Physical Chemistry Chemical Physics, 2014, 16, 3466-3478.	2.8	54
48	Kinetics of CH <sub>2</sub> OO reactions with SO <sub>2</sub> , NO <sub>2</sub> , NO, H <sub>2</sub> O and CH <sub>3</sub> CHO as a function of pressure. Physical Chemistry Chemical Physics, 2014, 16, 1139-1149.	2.8	215
49	Analysis of the Kinetics and Yields of OH Radical Production from the CH <sub>3</sub> OCH <sub>2</sub> + O <sub>2</sub> Reaction in the Temperature Range 195-650 K: An Experimental and Computational study. Journal of Physical Chemistry A, 2014, 118, 6773-6788.	2.5	58
50	Atmospheric Oxidation of Piperazine by OH has a Low Potential To Form Carcinogenic Compounds. Environmental Science and Technology Letters, 2014, 1, 367-371.	8.7	22
51	Branching Ratios in Reactions of OH Radicals with Methylamine, Dimethylamine, and Ethylamine. Environmental Science & Technology, 2014, 48, 9935-9942.	10.0	52
52	Low Temperature Kinetics of the CH <sub>3</sub> OH + OH Reaction. Journal of Physical Chemistry A, 2014, 118, 2693-2701.	2.5	68
53	Kinetic Study of the OH + Glyoxal Reaction: Experimental Evidence and Quantification of Direct OH Recycling. Journal of Physical Chemistry A, 2013, 117, 11027-11037.	2.5	34
54	Experimental and Theoretical Study of the Kinetics and Mechanism of the Reaction of OH Radicals with Dimethyl Ether. Journal of Physical Chemistry A, 2013, 117, 11142-11154.	2.5	55

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55	Gas-Phase Reactions of OH with Methyl Amines in the Presence or Absence of Molecular Oxygen. An Experimental and Theoretical Study. <i>Journal of Physical Chemistry A</i> , 2013, 117, 10736-10745.	2.5	48
56	On the mechanism of iodine oxide particle formation. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 15612.	2.8	52
57	CH <sub>2</sub> OO Criegee biradical yields following photolysis of CH <sub>2</sub> I <sub>2</sub> in O <sub>2</sub> . <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 19119.	2.8	47
58	Pressure and temperature dependent photolysis of glyoxal in the 355–414 nm region: evidence for dissociation from multiple states. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 6516.	2.8	14
59	Quantum yields for the photolysis of glyoxal below 350 nm and parameterisations for its photolysis rate in the troposphere. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 4984.	2.8	19
60	Mechanism of the Reaction of OH with Alkynes in the Presence of Oxygen. <i>Journal of Physical Chemistry A</i> , 2013, 117, 5407-5418.	2.5	20
61	Accelerated chemistry in the reaction between the hydroxyl radical and methanol at interstellar temperatures facilitated by tunnelling. <i>Nature Chemistry</i> , 2013, 5, 745-749.	13.6	223
62	Reporting the sensitivity of laser-induced fluorescence instruments used for HO <sub>2</sub> detection to an interference from RO <sub>2</sub> radicals and introducing a novel approach that enables HO <sub>2</sub> and certain RO <sub>2</sub> types to be selectively measured. <i>Atmospheric Measurement Techniques</i> , 2013, 6, 3425-3440.	3.1	77
63	Laboratory studies of photochemistry and gas phase radical reaction kinetics relevant to planetary atmospheres. <i>Chemical Society Reviews</i> , 2012, 41, 6318.	38.1	23
64	Direct Determination of the Rate Coefficient for the Reaction of OH Radicals with Monoethanol Amine (MEA) from 296 to 510 K. <i>Journal of Physical Chemistry Letters</i> , 2012, 3, 853-856.	4.6	38
65	Interception of Excited Vibrational Quantum States by O <sub>2</sub> in Atmospheric Association Reactions. <i>Science</i> , 2012, 337, 1066-1069.	12.6	90
66	Rate Constants and Branching Ratios for the Reaction of CH <sub>3</sub> Radicals with NH <sub>3</sub> : A Combined Experimental and Theoretical Study. <i>Journal of Physical Chemistry A</i> , 2012, 116, 5877-5885.	2.5	20
67	Time-of-flight mass spectrometry for time-resolved measurements: Some developments and applications. <i>International Journal of Chemical Kinetics</i> , 2012, 44, 532-545.	1.6	25
68	Site-Specific Rate Coefficients for Reaction of OH with Ethanol from 298 to 900 K. <i>Journal of Physical Chemistry A</i> , 2011, 115, 3335-3345.	2.5	52
69	Experimental and Modeling Studies of the Pressure and Temperature Dependences of the Kinetics and the OH Yields in the Acetyl + O <sub>2</sub> Reaction. <i>Journal of Physical Chemistry A</i> , 2011, 115, 1069-1085.	2.5	57
70	Developments in Laboratory Studies of Gas-Phase Reactions for Atmospheric Chemistry with Applications to Isoprene Oxidation and Carbonyl Chemistry. <i>Annual Review of Physical Chemistry</i> , 2011, 62, 351-373.	10.8	6
71	<sup>3</sup> CH <sub>2</sub> + O <sub>2</sub> : Kinetics and Product Channel Branching Ratios. <i>Zeitschrift Fur Physikalische Chemie</i> , 2011, 225, 957-967.	2.8	17
72	Product branching fractions for the reaction of O(3P) atoms with methanol and ethanol. <i>Chemical Physics Letters</i> , 2011, 511, 207-212.	2.6	4

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73	An Experimental and Theoretical Study of the Reaction Between $\text{NH}(\text{X}3\hat{\text{I}}\hat{\text{a}}\hat{\text{~}}) + \text{SO}(\text{X}3\hat{\text{I}}\hat{\text{a}}\hat{\text{~}})$ . Zeitschrift Fur Physikalische Chemie, 2010, 224, 1009-1024.	2.8	1
74	A Multidimensional Study of the Reaction $\text{CH}_{2} + \text{O}_{2}$ : Products and Atmospheric Implications. ChemPhysChem, 2010, 11, 3928-3941.	2.1	43
75	Kinetics and yields of OH radical from the $\text{CH}_3\text{OCH}_2 + \text{O}_2$ reaction using a new photolytic source. Chemical Physics Letters, 2010, 487, 45-50.	2.6	13
76	State resolved measurements of $\text{C}_2\text{H}_2$ removal confirm predictions of the gateway model for electronic quenching. Journal of Chemical Physics, 2010, 132, 024302.	3.0	18
77	A laser induced fluorescence study relating to physical properties of the iodine monoxide radical. Physical Chemistry Chemical Physics, 2010, 12, 823-834.	2.8	7
78	H-Atom Yields from the Photolysis of Acetylene and from the Reaction of $\text{C}_2\text{H}$ with $\text{H}_2$ , $\text{C}_2\text{H}_2$ , and $\text{C}_2\text{H}_4$ . Journal of Physical Chemistry A, 2010, 114, 4735-4741.	2.5	31
79	Temperature Dependent Kinetics (195~798 K) and H Atom Yields (298~498 K) from Reactions of $\text{C}_2\text{H}$ with Acetylene, Ethene, and Propene. Journal of Physical Chemistry A, 2010, 114, 9413-9424.	2.5	30
80	Comment on "The Conical Intersection Dominates the Generation of Tropospheric Hydroxyl Radicals from $\text{NO}_2$ and $\text{H}_2\text{O}$ ". Journal of Physical Chemistry A, 2010, 114, 8016-8016.	2.5	4
81	Observation of a large negative temperature dependence for rate coefficients of reactions of OH with oxygenated volatile organic compounds studied at 86~112 K. Physical Chemistry Chemical Physics, 2010, 12, 13511.	2.8	51
82	An experimental and theoretical investigation of the competition between chemical reaction and relaxation for the reactions of $\text{CH}_2$ with acetylene and ethene: implications for the chemistry of the giant planets. Faraday Discussions, 2010, 147, 173.	3.2	43
83	H atom formation from benzene and toluene photoexcitation at 248 nm. Journal of Chemical Physics, 2009, 131, 204304.	3.0	23
84	Comment on "Atmospheric Hydroxyl Radical Production from Electronically Excited $\text{NO}_2$ and $\text{H}_2\text{O}$ ". Science, 2009, 324, 336-336.	12.6	66
85	Kinetic studies of atmospherically relevant silicon chemistry : Part I: Silicon atom reactions. Physical Chemistry Chemical Physics, 2009, 11, 671-678.	2.8	26
86	An experimental confirmation of the products of the reaction between CN radicals and $\text{NH}_3$ . Physical Chemistry Chemical Physics, 2009, 11, 10824.	2.8	11
87	New Chemical Source of the HCO Radical Following Photoexcitation of Glyoxal, $(\text{HCO})_2$ . Journal of Physical Chemistry A, 2009, 113, 8278-8285.	2.5	14
88	Kinetic studies of atmospherically relevant silicon chemistry. Part II: Silicon monoxide reactions. Physical Chemistry Chemical Physics, 2009, 11, 10945.	2.8	27
89	Studies on the $\text{Cl} + \text{C}_2\text{H}_5\text{I}$ reaction; site specific abstraction reactions and thermodynamics of adduct formation studied by observation of HCl product. Physical Chemistry Chemical Physics, 2009, 11, 10417.	2.8	6
90	Ketone photolysis in the presence of oxygen: A useful source of OH for flash photolysis kinetics experiments. International Journal of Chemical Kinetics, 2008, 40, 504-514.	1.6	31

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91	Pulsed Laval nozzle study of the kinetics of OH with unsaturated hydrocarbons at very low temperatures. <i>Physical Chemistry Chemical Physics</i> , 2008, 10, 422-437.	2.8	54
92	Kinetics and Product Branching Ratios of the Reaction of ${}^1\text{CH}_2$ with $\text{H}_2$ and $\text{D}_2$ . <i>Journal of Physical Chemistry A</i> , 2008, 112, 9575-9583.	2.5	23
93	A Kinetic and Spectroscopic Study of the $\text{CH}_3\text{I}^+\text{Cl}$ and $\text{ICH}_2^+\text{Cl}$ Adducts. <i>Journal of Physical Chemistry A</i> , 2008, 112, 9544-9554.	2.5	11
94	Time-of-flight mass spectrometry for time-resolved measurements. <i>Review of Scientific Instruments</i> , 2007, 78, 034103.	1.3	25
95	Experimental and Master Equation Study of the Kinetics of $\text{OH} + \text{C}_2\text{H}_2$ : Temperature Dependence of the Limiting High Pressure and Pressure Dependent Rate Coefficients. <i>Journal of Physical Chemistry A</i> , 2007, 111, 4043-4055.	2.5	44
96	A combined experimental and theoretical study of the reaction between methylglyoxal and OH/OD radical: OH regeneration. <i>Physical Chemistry Chemical Physics</i> , 2007, 9, 4114.	2.8	44
97	H Atom Yields from the Reactions of CN Radicals with $\text{C}_2\text{H}_2$ , $\text{C}_2\text{H}_4$ , $\text{C}_3\text{H}_6$ , <i>trans</i> - $\text{C}_4\text{H}_8$ , and <i>cis</i> - $\text{C}_4\text{H}_8$ . <i>Journal of Physical Chemistry A</i> , 2007, 111, 6679-6692.	2.5	66
98	OH yields from the $\text{CH}_3\text{CO} + \text{O}_2$ reaction using an internal standard. <i>Chemical Physics Letters</i> , 2007, 445, 108-112.	2.6	40
99	Determination of the temperature and pressure dependence of the reaction $\text{OH} + \text{C}_2\text{H}_4$ from 200 to 400 K using experimental and master equation analyses. <i>Physical Chemistry Chemical Physics</i> , 2006, 8, 5633-5642.	2.8	42
100	Combined Experimental and Master Equation Investigation of the Multiwell Reaction $\text{H} + \text{SO}_2$ . <i>Journal of Physical Chemistry A</i> , 2006, 110, 2996-3009.	2.5	57
101	Study of Acetone Photodissociation over the Wavelength Range 248 to 330 nm: Evidence of a Mechanism Involving Both the Singlet and Triplet Excited States. <i>Journal of Physical Chemistry A</i> , 2006, 110, 6742-6756.	2.5	66
102	Determination of the Rate Coefficients for the Reactions $\text{IO} + \text{NO}_2 + \text{M}$ (Air) $\rightarrow$ $\text{IONO}_2 + \text{M}$ and $\text{O}(^3\text{P}) + \text{NO}_2 \rightarrow \text{O}_2 + \text{NO}$ Using Laser-Induced Fluorescence Spectroscopy. <i>Journal of Physical Chemistry A</i> , 2006, 110, 6995-7002.	2.5	15
103	Wavelength dependent photodissociation of $\text{CH}_3\text{OOH}$ . <i>Journal of Photochemistry and Photobiology A: Chemistry</i> , 2005, 176, 107-113.	3.9	20
104	OH formation from the $\text{C}_2\text{H}_5\text{CO} + \text{O}_2$ reaction: An experimental marker for the propionyl radical. <i>Chemical Physics Letters</i> , 2005, 408, 232-236.	2.6	25
105	The effect of temperature on collision induced intersystem crossing in the reaction of $^1\text{CH}_2$ with $\text{H}_2$ . <i>Proceedings of the Combustion Institute</i> , 2005, 30, 927-933.	3.9	8
106	Photolysis of methylethyl, diethyl and methylvinyl ketones and their role in the atmospheric HOx budget. <i>Faraday Discussions</i> , 2005, 130, 73.	3.2	52
107	Kinetics study of the reaction of iodine monoxide radicals with dimethyl sulfide. <i>Physical Chemistry Chemical Physics</i> , 2005, 7, 2173.	2.8	25
108	A three-dimensional model study of the effect of new temperature-dependent quantum yields for acetone photolysis. <i>Journal of Geophysical Research</i> , 2005, 110, .	3.3	99

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109	Laser induced fluorescence studies of the reactions of O(1D <sub>2</sub> ) with N <sub>2</sub> , O <sub>2</sub> , N <sub>2</sub> O, CH <sub>4</sub> , H <sub>2</sub> , CO <sub>2</sub> , Ar, Kr and n-C <sub>4</sub> H <sub>10</sub> . <i>Physical Chemistry Chemical Physics</i> , 2004, 6, 2162.	2.8	59
110	Pressure and temperature-dependent quantum yields for the photodissociation of acetone between 279 and 327.5 nm. <i>Geophysical Research Letters</i> , 2004, 31, n/a-n/a.	4.0	59
111	Photodissociation of acetone: Atmospheric implications of temperature-dependent quantum yields. <i>Geophysical Research Letters</i> , 2004, 31, n/a-n/a.	4.0	42
112	Correction to "Pressure and temperature-dependent quantum yields for the photodissociation of acetone between 279 and 327.5 nm". <i>Geophysical Research Letters</i> , 2004, 31, n/a-n/a.	4.0	3
113	Evidence for the dominance of collision-induced intersystem crossing in collisions of 1CH <sub>2</sub> with O <sub>2</sub> and a determination of the H atom yields from 3CH <sub>2</sub> +O <sub>2</sub> , using time-resolved detection of H formation by vuvLIF. <i>Chemical Physics Letters</i> , 2003, 372, 295-299.	2.6	27
114	H Atom Branching Ratios from the Reactions of CH with C <sub>2</sub> H <sub>2</sub> , C <sub>2</sub> H <sub>4</sub> , C <sub>2</sub> H <sub>6</sub> , and neo-C <sub>5</sub> H <sub>12</sub> at Room Temperature and 25 Torr. <i>Journal of Physical Chemistry A</i> , 2003, 107, 5710-5716.	2.5	48
115	Determination of the High-Pressure Limiting Rate Coefficient and the Enthalpy of Reaction for OH + SO <sub>2</sub> . <i>Journal of Physical Chemistry A</i> , 2003, 107, 1971-1978.	2.5	67
116	Experimental Rate Measurements for NS + NO, O <sub>2</sub> and NO <sub>2</sub> , and Electronic Structure Calculations of the Reaction Paths for NS + NO <sub>2</sub> . <i>Journal of Physical Chemistry A</i> , 2002, 106, 8406-8410.	2.5	4
117	Redetermination of the rate coefficient for the reaction of O( <sup>1</sup> D) with N <sub>2</sub> . <i>Geophysical Research Letters</i> , 2002, 29, 35-1.	4.0	22
118	OH formation from CH <sub>3</sub> CO+O <sub>2</sub> : a convenient experimental marker for the acetyl radical. <i>Chemical Physics Letters</i> , 2002, 365, 374-379.	2.6	57
119	Collision induced intersystem crossing in methylene on reactive surfaces: application of a new technique to CH <sub>2</sub> (a <sup>1</sup> A <sub>1</sub> ) + H <sub>2</sub> . <i>Physical Chemistry Chemical Physics</i> , 2001, 3, 2241-2244.	2.8	10
120	Temperature dependence of the reaction of OH with SO. <i>Proceedings of the Combustion Institute</i> , 2000, 28, 2491-2497.	3.9	37
121	Formation of the propargyl radical in the reaction of 1CH <sub>2</sub> and C <sub>2</sub> H <sub>2</sub> : experiment and modelling. <i>Physical Chemistry Chemical Physics</i> , 2000, 2, 805-812.	2.8	46
122	The reaction of methyldiene (CH) with methanol isotopomers. <i>Physical Chemistry Chemical Physics</i> , 2000, 2, 2549-2553.	2.8	18
123	Reaction of CH with H <sub>2</sub> O: Temperature Dependence and Isotope Effect. <i>Journal of Physical Chemistry A</i> , 1999, 103, 5699-5704.	2.5	20
124	Reaction of CH radicals with methane isotopomers. <i>Journal of the Chemical Society, Faraday Transactions</i> , 1997, 93, 1473.	1.7	41
125	Experimental and theoretical study of oxidative addition reaction of nickel atom to O-H bond of water. <i>Journal of Chemical Physics</i> , 1994, 100, 423-433.	3.0	49
126	Gas-phase reactions of copper atoms with alkynes: sequential ligand addition via steady-state kinetics. <i>The Journal of Physical Chemistry</i> , 1993, 97, 5298-5304.	2.9	3



#	ARTICLE	IF	CITATIONS
127	Gas-phase reaction of copper atoms with tetramethylethylene: sequential ligand addition via non-steady-state kinetics. <i>The Journal of Physical Chemistry</i> , 1993, 97, 5305-5312.	2.9	2
128	Time resolved kinetic studies of the gas-phase reactions of dimethylsilylene with some O-Donor molecules: Part I. Room temperature studies. <i>International Journal of Chemical Kinetics</i> , 1992, 24, 127-143.	1.6	35
129	Gas-phase reactions of copper atoms: formation of copper dicarbonyl, bis(acetylene)copper, and bis(ethylene)copper. <i>The Journal of Physical Chemistry</i> , 1991, 95, 8719-8726.	2.9	71
130	Time-resolved studies of the temperature dependence of gas-phase insertion reactions of phenylsilylene with silicon-hydrogen bonds. <i>The Journal of Physical Chemistry</i> , 1990, 94, 3294-3297.	2.9	25
131	Absolute rate constants for the gas-phase silicon-hydrogen insertion reactions of dimethylsilylene with silane and the methylsilanes in the temperature range 300-600 K. <i>Journal of the American Chemical Society</i> , 1990, 112, 8337-8343.	13.7	45
132	Absolute rate measurements for some gas-phase addition reactions of dimethylsilylene. <i>Journal of the Chemical Society, Faraday Transactions 2</i> , 1988, 84, 515.	1.1	37