

Mark A Blitz

List of Publications by Year in descending order

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132
papers

4,212
citations

76326

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149698

56
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146
all docs

146
docs citations

146
times ranked

3151
citing authors

#	ARTICLE	IF	CITATIONS
1	Accelerated chemistry in the reaction between the hydroxyl radical and methanol at interstellar temperatures facilitated by tunnelling. <i>Nature Chemistry</i> , 2013, 5, 745-749.	13.6	223
2	Kinetics of CH ₂ OO reactions with SO ₂ , NO ₂ , NO, H ₂ O and CH ₃ CHO as a function of pressure. <i>Physical Chemistry Chemical Physics</i> , 2014, 16, 1139-1149.	2.8	215
3	Direct evidence for a substantive reaction between the Criegee intermediate, CH ₂ OO, and the water vapour dimer. <i>Physical Chemistry Chemical Physics</i> , 2015, 17, 4859-4863.	2.8	155
4	A three-dimensional model study of the effect of new temperature-dependent quantum yields for acetone photolysis. <i>Journal of Geophysical Research</i> , 2005, 110, .	3.3	99
5	Interception of Excited Vibrational Quantum States by O ₂ in Atmospheric Association Reactions. <i>Science</i> , 2012, 337, 1066-1069.	12.6	90
6	Reporting the sensitivity of laser-induced fluorescence instruments used for HO ₂ detection to an interference from RO ₂ radicals and introducing a novel approach that enables HO ₂ and certain RO ₂ types to be selectively measured. <i>Atmospheric Measurement Techniques</i> , 2013, 6, 3425-3440.	3.1	77
7	Gas-phase reactions of copper atoms: formation of copper dicarbonyl, bis(acetylene)copper, and bis(ethylene)copper. <i>The Journal of Physical Chemistry</i> , 1991, 95, 8719-8726.	2.9	71
8	Structure-switching M ₃ L ₂ Ir(III) coordination cages with photo-isomerising azo-aromatic linkers. <i>Chemical Science</i> , 2018, 9, 8150-8159.	7.4	69
9	Low Temperature Kinetics of the CH ₃ OH + OH Reaction. <i>Journal of Physical Chemistry A</i> , 2014, 118, 2693-2701.	2.5	68
10	Determination of the High-Pressure Limiting Rate Coefficient and the Enthalpy of Reaction for OH + SO ₂ . <i>Journal of Physical Chemistry A</i> , 2003, 107, 1971-1978.	2.5	67
11	Study of Acetone Photodissociation over the Wavelength Range 248~330 nm: Evidence of a Mechanism Involving Both the Singlet and Triplet Excited States. <i>Journal of Physical Chemistry A</i> , 2006, 110, 6742-6756.	2.5	66
12	H Atom Yields from the Reactions of CN Radicals with C ₂ H ₂ , C ₂ H ₄ , C ₃ H ₆ , trans-2-C ₄ H ₈ , and iso-C ₄ H ₈ . <i>Journal of Physical Chemistry A</i> , 2007, 111, 6679-6692.	2.5	66
13	Comment on "Atmospheric Hydroxyl Radical Production from Electronically Excited NO ₂ and H ₂ O". <i>Science</i> , 2009, 324, 336-336.	12.6	66
14	Laser induced fluorescence studies of the reactions of O(1D ₂) with N ₂ , O ₂ , N ₂ O, CH ₄ , H ₂ , CO ₂ , Ar, Kr and n-C ₄ H ₁₀ . <i>Physical Chemistry Chemical Physics</i> , 2004, 6, 2162.	2.8	59
15	Pressure and temperature-dependent quantum yields for the photodissociation of acetone between 279 and 327.5 nm. <i>Geophysical Research Letters</i> , 2004, 31, n/a-n/a.	4.0	59
16	Analysis of the Kinetics and Yields of OH Radical Production from the CH ₃ OCH ₂ + O ₂ Reaction in the Temperature Range 195~650 K: An Experimental and Computational study. <i>Journal of Physical Chemistry A</i> , 2014, 118, 6773-6788.	2.5	58
17	OH formation from CH ₃ CO+O ₂ : a convenient experimental marker for the acetyl radical. <i>Chemical Physics Letters</i> , 2002, 365, 374-379.	2.6	57
18	Combined Experimental and Master Equation Investigation of the Multiwell Reaction H + SO ₂ . <i>Journal of Physical Chemistry A</i> , 2006, 110, 2996-3009.	2.5	57

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19	Experimental and Modeling Studies of the Pressure and Temperature Dependences of the Kinetics and the OH Yields in the Acetyl + O ₂ Reaction. Journal of Physical Chemistry A, 2011, 115, 1069-1085.	2.5	57
20	Experimental and Theoretical Study of the Kinetics and Mechanism of the Reaction of OH Radicals with Dimethyl Ether. Journal of Physical Chemistry A, 2013, 117, 11142-11154.	2.5	55
21	Pulsed Laval nozzle study of the kinetics of OH with unsaturated hydrocarbons at very low temperatures. Physical Chemistry Chemical Physics, 2008, 10, 422-437.	2.8	54
22	A combined experimental and theoretical study of reactions between the hydroxyl radical and oxygenated hydrocarbons relevant to astrochemical environments. Physical Chemistry Chemical Physics, 2014, 16, 3466-3478.	2.8	54
23	Photolysis of methylethyl, diethyl and methylvinyl ketones and their role in the atmospheric HOx budget. Faraday Discussions, 2005, 130, 73.	3.2	52
24	Site-Specific Rate Coefficients for Reaction of OH with Ethanol from 298 to 900 K. Journal of Physical Chemistry A, 2011, 115, 3335-3345.	2.5	52
25	On the mechanism of iodine oxide particle formation. Physical Chemistry Chemical Physics, 2013, 15, 15612.	2.8	52
26	Branching Ratios in Reactions of OH Radicals with Methylamine, Dimethylamine, and Ethylamine. Environmental Science & Technology, 2014, 48, 9935-9942.	10.0	52
27	Observation of a large negative temperature dependence for rate coefficients of reactions of OH with oxygenated volatile organic compounds studied at 86±112 K. Physical Chemistry Chemical Physics, 2010, 12, 13511.	2.8	51
28	Experimental and theoretical study of oxidative addition reaction of nickel atom to O-H bond of water. Journal of Chemical Physics, 1994, 100, 423-433.	3.0	49
29	H Atom Branching Ratios from the Reactions of CH with C ₂ H ₂ , C ₂ H ₄ , C ₂ H ₆ , and neo-C ₅ H ₁₂ at Room Temperature and 25 Torr. Journal of Physical Chemistry A, 2003, 107, 5710-5716.	2.5	48
30	Gas-Phase Reactions of OH with Methyl Amines in the Presence or Absence of Molecular Oxygen. An Experimental and Theoretical Study. Journal of Physical Chemistry A, 2013, 117, 10736-10745.	2.5	48
31	CH ₂ OO Criegee biradical yields following photolysis of CH ₂ I ₂ in O ₂ . Physical Chemistry Chemical Physics, 2013, 15, 19119.	2.8	47
32	Formation of the propargyl radical in the reaction of 1CH ₂ and C ₂ H ₂ : experiment and modelling. Physical Chemistry Chemical Physics, 2000, 2, 805-812.	2.8	46
33	Absolute rate constants for the gas-phase silicon-hydrogen insertion reactions of dimethylsilylene with silane and the methylsilanes in the temperature range 300-600 K. Journal of the American Chemical Society, 1990, 112, 8337-8343.	13.7	45
34	Measurements of Rate Coefficients for Reactions of OH with Ethanol and Propan-2-ol at Very Low Temperatures. Journal of Physical Chemistry A, 2015, 119, 7130-7137.	2.5	45
35	Experimental and Master Equation Study of the Kinetics of OH + C ₂ H ₂ : Temperature Dependence of the Limiting High Pressure and Pressure Dependent Rate Coefficients. Journal of Physical Chemistry A, 2007, 111, 4043-4055.	2.5	44
36	A combined experimental and theoretical study of the reaction between methylglyoxal and OH/OD radical: OH regeneration. Physical Chemistry Chemical Physics, 2007, 9, 4114.	2.8	44

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37	A Multidimensional Study of the Reaction $\text{CH}_2 + \text{O}_2$: Products and Atmospheric Implications. <i>ChemPhysChem</i> , 2010, 11, 3928-3941.	2.1	43
38	An experimental and theoretical investigation of the competition between chemical reaction and relaxation for the reactions of ICH_2 with acetylene and ethene: implications for the chemistry of the giant planets. <i>Faraday Discussions</i> , 2010, 147, 173.	3.2	43
39	Photodissociation of acetone: Atmospheric implications of temperature-dependent quantum yields. <i>Geophysical Research Letters</i> , 2004, 31, n/a-n/a.	4.0	42
40	Determination of the temperature and pressure dependence of the reaction $\text{OH} + \text{C}_2\text{H}_4$ from 200 to 400 K using experimental and master equation analyses. <i>Physical Chemistry Chemical Physics</i> , 2006, 8, 5633-5642.	2.8	42
41	Reaction of CH radicals with methane isotopomers. <i>Journal of the Chemical Society, Faraday Transactions</i> , 1997, 93, 1473.	1.7	41
42	Unimolecular decomposition kinetics of the stabilised Criegee intermediates CH_2OO and CD_2OO . <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 24940-24954.	2.8	41
43	OH yields from the $\text{CH}_3\text{CO} + \text{O}_2$ reaction using an internal standard. <i>Chemical Physics Letters</i> , 2007, 445, 108-112.	2.6	40
44	A gas-to-particle conversion mechanism helps to explain atmospheric particle formation through clustering of iodine oxides. <i>Nature Communications</i> , 2020, 11, 4521.	12.8	39
45	Direct Determination of the Rate Coefficient for the Reaction of OH Radicals with Monoethanol Amine (MEA) from 296 to 510 K. <i>Journal of Physical Chemistry Letters</i> , 2012, 3, 853-856.	4.6	38
46	Absolute rate measurements for some gas-phase addition reactions of dimethylsilylene. <i>Journal of the Chemical Society, Faraday Transactions 2</i> , 1988, 84, 515.	1.1	37
47	Temperature dependence of the reaction of OH with SO. <i>Proceedings of the Combustion Institute</i> , 2000, 28, 2491-2497.	3.9	37
48	Time resolved kinetic studies of the gas-phase reactions of dimethylsilylene with some O-Donor molecules: Part I. Room temperature studies. <i>International Journal of Chemical Kinetics</i> , 1992, 24, 127-143.	1.6	35
49	The importance of OH radical neutral low temperature tunnelling reactions in interstellar clouds using a new model. <i>Molecular Physics</i> , 2015, 113, 2243-2254.	1.7	35
50	Kinetic Study of the OH + Glyoxal Reaction: Experimental Evidence and Quantification of Direct OH Recycling. <i>Journal of Physical Chemistry A</i> , 2013, 117, 11027-11037.	2.5	34
51	Ketone photolysis in the presence of oxygen: A useful source of OH for flash photolysis kinetics experiments. <i>International Journal of Chemical Kinetics</i> , 2008, 40, 504-514.	1.6	31
52	H-Atom Yields from the Photolysis of Acetylene and from the Reaction of C_2H with H_2 , C_2H_2 , and C_2H_4 . <i>Journal of Physical Chemistry A</i> , 2010, 114, 4735-4741.	2.5	31
53	Temperature Dependent Kinetics (195 to 798 K) and H Atom Yields (298 to 498 K) from Reactions of CH_2 with Acetylene, Ethene, and Propene. <i>Journal of Physical Chemistry A</i> , 2010, 114, 9413-9424.	2.5	30
54	Reanalysis of Rate Data for the Reaction $\text{CH}_3 + \text{C}_2\text{H}_6 \rightarrow \text{C}_2\text{H}_5 + \text{CH}_4$ Using Revised Cross Sections and a Linearized Second-Order Master Equation. <i>Journal of Physical Chemistry A</i> , 2015, 119, 7668-7682.	2.5	28

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55	Evidence for the dominance of collision-induced intersystem crossing in collisions of 1CH ₂ with O ₂ and a determination of the H atom yields from 3CH ₂ +O ₂ , using time-resolved detection of H formation by vuvLIF. <i>Chemical Physics Letters</i> , 2003, 372, 295-299.	2.6	27
56	Kinetic studies of atmospherically relevant silicon chemistry. Part II: Silicon monoxide reactions. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 10945.	2.8	27
57	Global Uncertainty Propagation and Sensitivity Analysis in the CH ₃ OCH ₂ + O ₂ System: Combining Experiment and Theory To Constrain Key Rate Coefficients in DME Combustion. <i>Journal of Physical Chemistry A</i> , 2015, 119, 7430-7438.	2.5	27
58	Kinetic studies of atmospherically relevant silicon chemistry : Part I: Silicon atom reactions. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 671-678.	2.8	26
59	Time-resolved studies of the temperature dependence of gas-phase insertion reactions of phenylsilylene with silicon-hydrogen bonds. <i>The Journal of Physical Chemistry</i> , 1990, 94, 3294-3297.	2.9	25
60	OH formation from the C ₂ H ₅ CO+O ₂ reaction: An experimental marker for the propionyl radical. <i>Chemical Physics Letters</i> , 2005, 408, 232-236.	2.6	25
61	Kinetics study of the reaction of iodine monoxide radicals with dimethyl sulfide. <i>Physical Chemistry Chemical Physics</i> , 2005, 7, 2173.	2.8	25
62	Time-of-flight mass spectrometry for time-resolved measurements. <i>Review of Scientific Instruments</i> , 2007, 78, 034103.	1.3	25
63	Time-of-flight mass spectrometry for time-resolved measurements: Some developments and applications. <i>International Journal of Chemical Kinetics</i> , 2012, 44, 532-545.	1.6	25
64	Kinetic studies of C ₁ and C ₂ Criegee intermediates with SO ₂ using laser flash photolysis coupled with photoionization mass spectrometry and time resolved UV absorption spectroscopy. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 22218-22227.	2.8	25
65	CH ₂ OO Criegee intermediate UV absorption cross-sections and kinetics of CH ₂ OO + CH ₂ OO and CH ₂ OO + I as a function of pressure. <i>Physical Chemistry Chemical Physics</i> , 2020, 22, 9448-9459.	2.8	25
66	Kinetics and Product Branching Ratios of the Reaction of ¹ CH ₂ with H ₂ and D ₂ . <i>Journal of Physical Chemistry A</i> , 2008, 112, 9575-9583.	2.5	23
67	H atom formation from benzene and toluene photoexcitation at 248 nm. <i>Journal of Chemical Physics</i> , 2009, 131, 204304.	3.0	23
68	Laboratory studies of photochemistry and gas phase radical reaction kinetics relevant to planetary atmospheres. <i>Chemical Society Reviews</i> , 2012, 41, 6318.	38.1	23
69	Redetermination of the rate coefficient for the reaction of O(¹ D) with N ₂ . <i>Geophysical Research Letters</i> , 2002, 29, 35-1.	4.0	22
70	Atmospheric Oxidation of Piperazine by OH has a Low Potential To Form Carcinogenic Compounds. <i>Environmental Science and Technology Letters</i> , 2014, 1, 367-371.	8.7	22
71	Reaction of CH with H ₂ O: Temperature Dependence and Isotope Effect. <i>Journal of Physical Chemistry A</i> , 1999, 103, 5699-5704.	2.5	20
72	Wavelength dependent photodissociation of CH ₃ OOH. <i>Journal of Photochemistry and Photobiology A: Chemistry</i> , 2005, 176, 107-113.	3.9	20

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73	Rate Constants and Branching Ratios for the Reaction of CH Radicals with NH ₃ : A Combined Experimental and Theoretical Study. <i>Journal of Physical Chemistry A</i> , 2012, 116, 5877-5885.	2.5	20
74	Mechanism of the Reaction of OH with Alkynes in the Presence of Oxygen. <i>Journal of Physical Chemistry A</i> , 2013, 117, 5407-5418.	2.5	20
75	Quantum yields for the photolysis of glyoxal below 350 nm and parameterisations for its photolysis rate in the troposphere. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 4984.	2.8	19
76	Experimental Study of the Removal of Ground- and Excited-State Phosphorus Atoms by Atmospherically Relevant Species. <i>Journal of Physical Chemistry A</i> , 2019, 123, 9469-9478.	2.5	19
77	The reaction of methylidene (CH) with methanol isotopomers. <i>Physical Chemistry Chemical Physics</i> , 2000, 2, 2549-2553.	2.8	18
78	State resolved measurements of α -C ₁ H ₂ removal confirm predictions of the gateway model for electronic quenching. <i>Journal of Chemical Physics</i> , 2010, 132, 024302.	3.0	18
79	³ CH ₂ + O ₂ : Kinetics and Product Channel Branching Ratios. <i>Zeitschrift Fur Physikalische Chemie</i> , 2011, 225, 957-967.	2.8	17
80	OH production from the photolysis of isoprene-derived peroxy radicals: cross-sections, quantum yields and atmospheric implications. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 2332-2345.	2.8	16
81	Kinetics of the Reaction of OH with Isoprene over a Wide Range of Temperature and Pressure Including Direct Observation of Equilibrium with the OH Adducts. <i>Journal of Physical Chemistry A</i> , 2018, 122, 7239-7255.	2.5	16
82	Determination of the Rate Coefficients for the Reactions IO + NO ₂ + M (Air) → IONO ₂ + M and O(3P) + NO ₂ → O ₂ + NO Using Laser-Induced Fluorescence Spectroscopy. <i>Journal of Physical Chemistry A</i> , 2006, 110, 6995-7002.	2.5	15
83	New Chemical Source of the HCO Radical Following Photoexcitation of Glyoxal, (HCO) ₂ . <i>Journal of Physical Chemistry A</i> , 2009, 113, 8278-8285.	2.5	14
84	Pressure and temperature dependent photolysis of glyoxal in the 355–414 nm region: evidence for dissociation from multiple states. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 6516.	2.8	14
85	Branching ratios for the reactions of OH with ethanol amines used in carbon capture and the potential impact on carcinogen formation in the emission plume from a carbon capture plant. <i>Physical Chemistry Chemical Physics</i> , 2015, 17, 25342-25353.	2.8	14
86	Bimolecular reactions of activated species: An analysis of problematic HC(O)C(O) chemistry. <i>Chemical Physics Letters</i> , 2016, 661, 58-64.	2.6	14
87	Kinetics of the Reactions of Hydroxyl Radicals with Furan and Its Alkylated Derivatives 2-Methyl Furan and 2,5-Dimethyl Furan. <i>Journal of Physical Chemistry A</i> , 2020, 124, 7416-7426.	2.5	14
88	Production of HONO from NO ₂ ; uptake on illuminated TiO ₂ aerosol particles and following the illumination of mixed TiO ₂ •ammonium nitrate particles. <i>Atmospheric Chemistry and Physics</i> , 2021, 21, 5755-5775.	4.9	14
89	Determination of the absorption cross sections of higher-order iodine oxides at 355 and 532 nm. <i>Atmospheric Chemistry and Physics</i> , 2020, 20, 10865-10887.	4.9	14
90	Kinetics and yields of OH radical from the CH ₃ OCH ₂ +O ₂ reaction using a new photolytic source. <i>Chemical Physics Letters</i> , 2010, 487, 45-50.	2.6	13

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91	Measurements of Low Temperature Rate Coefficients for the Reaction of CH with CH ₂ O and Application to Dark Cloud and AGB Stellar Wind Models. <i>Astrophysical Journal</i> , 2019, 885, 134.	4.5	13
92	Temperature and Pressure Studies of the Reactions of CH ₃ O ₂ , HO ₂ , and 1,2-C ₄ H ₉ O ₂ with NO ₂ . <i>Journal of Physical Chemistry A</i> , 2016, 120, 1408-1420.	2.5	12
93	Low temperature studies of the removal reactions of 1CH ₂ with particular relevance to the atmosphere of Titan. <i>Icarus</i> , 2018, 303, 10-21.	2.5	12
94	Direct Trace Fitting of Experimental Data Using the Master Equation: Testing Theory and Experiments on the OH + C ₂ H ₄ Reaction. <i>Journal of Physical Chemistry A</i> , 2020, 124, 4015-4024.	2.5	12
95	A Kinetic and Spectroscopic Study of the CH ₃ ~Cl and ICH ₂ ~Cl Adducts. <i>Journal of Physical Chemistry A</i> , 2008, 112, 9544-9554.	2.5	11
96	An experimental confirmation of the products of the reaction between CN radicals and NH ₃ . <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 10824.	2.8	11
97	An Experimental and Master Equation Study of the Kinetics of OH/OD + SO ₂ : The Limiting High-Pressure Rate Coefficients. <i>Journal of Physical Chemistry A</i> , 2017, 121, 3184-3191.	2.5	11
98	Obtaining effective rate coefficients to describe the decomposition kinetics of the corannulene oxyradical at high temperatures. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 11064-11074.	2.8	11
99	Temperature and Pressure Dependent Kinetics of QOOH Decomposition and Reaction with O ₂ : Experimental and Theoretical Investigations of QOOH Radicals Derived from Cl + (CH ₃) ₃ COOH. <i>Journal of Physical Chemistry A</i> , 2019, 123, 10254-10262.	2.5	11
100	Collision induced intersystem crossing in methylene on reactive surfaces: application of a new technique to CH ₂ (a 1A ₁) + H ₂ . <i>Physical Chemistry Chemical Physics</i> , 2001, 3, 2241-2244.	2.8	10
101	Observation of a new channel, the production of CH ₃ , in the abstraction reaction of OH radicals with acetaldehyde. <i>Physical Chemistry Chemical Physics</i> , 2016, 18, 26423-26433.	2.8	10
102	An Experimental Study of the Kinetics of OH/OD(<i>v</i> = 1,2,3) + SO ₂ : The Limiting High-Pressure Rate Coefficients as a Function of Temperature. <i>Journal of Physical Chemistry A</i> , 2017, 121, 3175-3183.	2.5	10
103	Comment on "Methanol dimer formation drastically enhances hydrogen abstraction from methanol by OH at low temperature" by W. Siebrand, Z. Smedarchina, E. Martnez-Nez and A. Fernandez-Ramos, <i>Phys. Chem. Chem. Phys.</i> , 2016, 18, 22712. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 8349-8354.	2.8	10
104	A novel multiplex absorption spectrometer for time-resolved studies. <i>Review of Scientific Instruments</i> , 2018, 89, 024101.	1.3	10
105	Laser Photolysis Kinetic Study of OH Radical Reactions with Methyl <i>tert</i> -Butyl Ether and Trimethyl Orthoformate under Conditions Relevant to Low Temperature Combustion: Measurements of Rate Coefficients and OH Recycling. <i>Journal of Physical Chemistry A</i> , 2018, 122, 9701-9711.	2.5	10
106	Kinetic Study of the Reactions PO + O ₂ and PO ₂ + O ₃ and Spectroscopy of the PO Radical. <i>Journal of Physical Chemistry A</i> , 2020, 124, 7911-7926.	2.5	10
107	Kinetics of the gas phase reaction of the Criegee intermediate CH ₂ OO with SO ₂ as a function of temperature. <i>Physical Chemistry Chemical Physics</i> , 2021, 23, 19415-19423.	2.8	10
108	The effect of temperature on collision induced intersystem crossing in the reaction of 1CH ₂ with H ₂ . <i>Proceedings of the Combustion Institute</i> , 2005, 30, 927-933.	3.9	8

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109	An instrument to measure fast gas phase radical kinetics at high temperatures and pressures. Review of Scientific Instruments, 2016, 87, 054102.	1.3	8
110	Low temperature studies of the rate coefficients and branching ratios of reactive loss vs quenching for the reactions of 1CH2 with C2H6, C2H4, C2H2. Icarus, 2019, 321, 752-766.	2.5	8
111	The reaction between HgBr and O₃: kinetic study and atmospheric implications. Physical Chemistry Chemical Physics, 2022, , .	2.8	8
112	A laser induced fluorescence study relating to physical properties of the iodine monoxide radical. Physical Chemistry Chemical Physics, 2010, 12, 823-834.	2.8	7
113	Kinetics of the Gas Phase Reactions of the Criegee Intermediate CH2OO with O3 and IO. Journal of Physical Chemistry A, 2020, 124, 6287-6293.	2.5	7
114	Rate coefficients for the reactions of OH with butanols from 298 K to temperatures relevant for low-temperature combustion. International Journal of Chemical Kinetics, 2020, 52, 1046-1059.	1.6	7
115	Studies on the Cl + C2H5I reaction; site specific abstraction reactions and thermodynamics of adduct formation studied by observation of HCL product. Physical Chemistry Chemical Physics, 2009, 11, 10417.	2.8	6
116	Developments in Laboratory Studies of Gas-Phase Reactions for Atmospheric Chemistry with Applications to Isoprene Oxidation and Carbonyl Chemistry. Annual Review of Physical Chemistry, 2011, 62, 351-373.	10.8	6
117	A new instrument for time-resolved measurement of HO<sub>2</sub> radicals. Atmospheric Measurement Techniques, 2020, 13, 839-852.	3.1	6
118	Direct Measurements of Isoprene Autoxidation: Pinpointing Atmospheric Oxidation in Tropical Forests. JACS, 2022, 144, 809-818.	7.9	6
119	Exploring the features on the OH + SO₂ potential energy surface using theory and testing its accuracy by comparison to experimental data. Physical Chemistry Chemical Physics, 2018, 20, 8984-8990.	2.8	5
120	Experimental Rate Measurements for NS + NO, O2 and NO2, and Electronic Structure Calculations of the Reaction Paths for NS + NO2. Journal of Physical Chemistry A, 2002, 106, 8406-8410.	2.5	4
121	Comment on "The Conical Intersection Dominates the Generation of Tropospheric Hydroxyl Radicals from NO₂ and H₂O". Journal of Physical Chemistry A, 2010, 114, 8016-8016.	2.5	4
122	Product branching fractions for the reaction of O(3P) atoms with methanol and ethanol. Chemical Physics Letters, 2011, 511, 207-212.	2.6	4
123	A generic method for determining R<sub>OH</sub> + O2 rate parameters via OH regeneration. Chemical Physics Letters, 2019, 730, 213-219.	2.6	4
124	Gas-phase reactions of copper atoms with alkynes: sequential ligand addition via steady-state kinetics. The Journal of Physical Chemistry, 1993, 97, 5298-5304.	2.9	3
125	Correction to "Pressure and temperature-dependent quantum yields for the photodissociation of acetone between 279 and 327.5 nm". Geophysical Research Letters, 2004, 31, n/a-n/a.	4.0	3
126	Global Master Equation Analysis of Rate Data for the Reaction C2H4 + H, C2H5: fH-C2H5. Journal of Physical Chemistry A, 2021, 125, 9548-9565.	2.5	3

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127	Gas-phase reaction of copper atoms with tetramethylethylene: sequential ligand addition via non-steady-state kinetics. <i>The Journal of Physical Chemistry</i> , 1993, 97, 5305-5312.	2.9	2
128	Time-Resolved Measurements and Master Equation Modelling of the Unimolecular Decomposition of CH_3OCH_2 . <i>Zeitschrift Fur Physikalische Chemie</i> , 2020, 234, 1233-1250.	2.8	2
129	Identification, monitoring, and reaction kinetics of reactive trace species using time-resolved mid-infrared quantum cascade laser absorption spectroscopy: development, characterisation, and initial results for the CH_2OO Criegee intermediate. <i>Atmospheric Measurement Techniques</i> , 2022, 15, 2875-2887.	3.1	2
130	An Experimental and Theoretical Study of the Reaction Between $\text{NH}_3 + \text{SO}$. <i>Zeitschrift Fur Physikalische Chemie</i> , 2010, 224, 1009-1024.	2.8	1
131	Low temperature gas phase reaction rate coefficient measurements: Toward modeling of stellar winds and the interstellar medium. <i>Proceedings of the International Astronomical Union</i> , 2019, 15, 382-383.	0.0	0
132	OH Kinetics with a Range of Nitrogen-Containing Compounds: N-Methylformamide, t-Butylamine, and N-Methyl-propane Diamine. <i>Journal of Physical Chemistry A</i> , 2021, 125, 10439-10450.	2.5	0