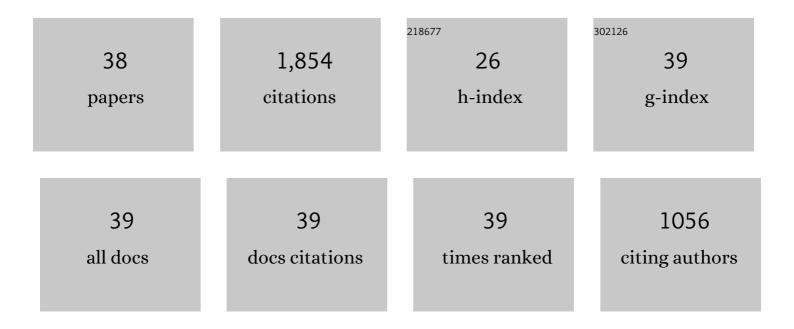
## Laurence Romsted

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Simultaneous determination of interfacial molarities of an alcohol, bromide ion, and water during an alcohol induced microstructural transition: the difference between medium and long chain alcohols. Soft Matter, 2020, 16, 5148-5156.	2.7	7
2	Effects of interfacial specific cations and water molarities on AOT micelle-to-vesicle transitions by chemical trapping: the specific ion-pair/hydration model. Physical Chemistry Chemical Physics, 2019, 21, 8633-8644.	2.8	18
3	Evidence of coexisting microemulsion droplets in oil-in-water emulsions revealed by 2D DOSY 1H NMR. Journal of Colloid and Interface Science, 2018, 514, 83-92.	9.4	35
4	A novel combined chemical kinetic and trapping method for probing the relationships between chemical reactivity and interfacial H <sub>2</sub> 0, Br <sup>â^'</sup> and H <sup>+</sup> ion molarities in CTAB/C <sub>12</sub> E <sub>6</sub> mixed micelles. Physical Chemistry Chemical Physics, 2017, 19, 23747-23761.	2.8	12
5	Using a pseudophase model to determine AO distributions in emulsions: Why dynamic equilibrium matters. European Journal of Lipid Science and Technology, 2017, 119, 1600277.	1.5	12
6	Interfacial Concentrations of Hydroxytyrosol and Its Lipophilic Esters in Intact Olive Oil-in-Water Emulsions: Effects of Antioxidant Hydrophobicity, Surfactant Concentration, and the Oil-to-Water Ratio on the Oxidative Stability of the Emulsions. Journal of Agricultural and Food Chemistry, 2016, 64, 5274-5283.	5.2	63
7	Distributions of phenolic acid antioxidants between the interfacial and aqueous regions of corn oil emulsions: Effects of pH and emulsifier concentration. European Journal of Lipid Science and Technology, 2015, 117, 1801-1813.	1.5	26
8	A direct correlation between the antioxidant efficiencies of caffeic acid and its alkyl esters and their concentrations in the interfacial region of olive oil emulsions. The pseudophase model interpretation of the "cut-off―effect. Food Chemistry, 2015, 175, 233-242.	8.2	79
9	Influence of Temperature on the Distribution of Catechin in Corn Oil-in-Water Emulsions and Some Relevant Thermodynamic Parameters. Food Biophysics, 2014, 9, 380-388.	3.0	17
10	Modeling chemical reactivity in emulsions. Current Opinion in Colloid and Interface Science, 2013, 18, 3-14.	7.4	77
11	Simultaneous Determination of Interfacial Molarities of Amide Bonds, Carboxylate Groups, and Water by Chemical Trapping in Micelles of Amphiphiles Containing Peptide Bond Models. Langmuir, 2013, 29, 534-544.	3.5	12
12	Using the pseudophase kinetic model to interpret chemical reactivity in ionic emulsions: Determining antioxidant partition constants and interfacial rate constants. Journal of Colloid and Interface Science, 2013, 400, 41-48.	9.4	25
13	Maxima in Antioxidant Distributions and Efficiencies with Increasing Hydrophobicity of Gallic Acid and Its Alkyl Esters. The Pseudophase Model Interpretation of the "Cutoff Effect― Journal of Agricultural and Food Chemistry, 2013, 61, 6533-6543.	5.2	89
14	Specific Ion-Pair/Hydration Model for the Sphere-To-Rod Transitions of Aqueous Cationic Micelles. The Evidence from Chemical Trapping. Statistical Science and Interdisciplinary Research, 2012, , 171-198.	0.0	1
15	Do Amphiphile Aggregate Morphologies and Interfacial Compositions Depend Primarily on Interfacial Hydration and Ion-Specific Interactions? The Evidence from Chemical Trapping. Langmuir, 2007, 23, 414-424.	3.5	117
16	Structural, Infrared, and Density Functional Theory Studies of N,N,Nâ€~,Nâ€~-Tetramethylimidazolidinium Dichloride:  A Model for CationⰒAnion Association of Headgroups and Counterions in the Interfacial Regions of Gemini Micelles. Journal of Physical Chemistry B, 2007, 111, 13668-13674.	2.6	2
17	Specific Ion Pairing and Interfacial Hydration as Controlling Factors in Gemini Micelle Morphology. Chemical Trapping Studies. Journal of the American Chemical Society, 2006, 128, 492-501.	13.7	101
18	Determining α-tocopherol distributions between the oil, water, and interfacial regions of macroemulsions: Novel applications of electroanalytical chemistry and the pseudophase kinetic model. Advances in Colloid and Interface Science, 2006, 123-126, 303-311.	14.7	54

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#	Article	IF	CITATIONS
19	Ion Pair Formation in Water. Association Constants of Bolaform, Bisquaternary Ammonium, Electrolytes by Chemical Trapping. Journal of Physical Chemistry B, 2005, 109, 23629-23637.	2.6	28
20	Origin of the Sphere-to-Rod Transition in Cationic Micelles with Aromatic Counterions:  Specific Ion Hydration in the Interfacial Region Matters. Langmuir, 2005, 21, 562-568.	3.5	71
21	Determining Partition Constants of Polar Organic Molecules between the Oil/Interfacial and Water/Interfacial Regions in Emulsions:Â A Combined Electrochemical and Spectrometric Method. Langmuir, 2004, 20, 3047-3055.	3.5	44
22	Concentration of Urea in Interfacial Regions of Aqueous Cationic, Anionic, and Zwitterionic Micelles Determined by Chemical Trapping. Langmuir, 2003, 19, 9179-9190.	3.5	39
23	Kinetic Method for Determining Antioxidant Distributions in Model Food Emulsions:  Distribution Constants of t-Butylhydroquinone in Mixtures of Octane, Water, and a Nonionic Emulsifier. Journal of Agricultural and Food Chemistry, 2002, 50, 3328-3336.	5.2	25
24	Estimating Concentrations of Condensed Counterions around a Polyelectrolyte by Chemical Trapping. ACS Symposium Series, 2002, , 184-199.	0.5	2
25	Arenediazonium Salts:Â New Probes of the Interfacial Compositions of Association Colloids. 6. Relationships between Interfacial Counterion and Water Concentrations and Surfactant Headgroup Size, Sphere-to-Rod Transitions, and Chemical Reactivity in Cationic Micellesâ€. Langmuir, 2000, 16, 59-71.	3.5	137
26	Rates and pH-dependent product distributions of the CuCl2-catalyzed dediazoniation ofp-nitrobenzenediazonium Âŧetrafluoroborate in aqueous acid. Journal of Physical Organic Chemistry, 1999, 12, 130-140.	1.9	29
27	Arenediazonium Salts:Â New Probes of the Interfacial Compositions of Association Colloids. 5.1â^4Determination of Hydration Numbers and Radial Distributions of Terminal Hydroxyl Groups in Mixed Nonionic CmEnMicelles by Chemical Trapping. Langmuir, 1999, 15, 326-336.	3.5	31
28	Determination of Interfacial Co-ion Concentration in Ionic Micelles by Chemical Trapping:  Halide Concentration at the Interface of Sodium Dodecyl Sulfate Micelles. Langmuir, 1997, 13, 5032-5035.	3.5	26
29	New Method for Estimating the Degree of Ionization and Counterion Selectivity of Cetyltrimethylammonium Halide Micelles:  Chemical Trapping of Free Counterions by a Water Soluble Arenediazonium Ion. Langmuir, 1997, 13, 647-652.	3.5	62
30	Micellar catalysis, a useful misnomer. Current Opinion in Colloid and Interface Science, 1997, 2, 622-628.	7.4	108
31	Arenediazonium Salts:Â New Probes of the Interfacial Compositions of Association Colloids. 4.1-3Estimation of the Hydration Numbers of Aqueous Hexaethylene Glycol Monododecyl Ether, C12E6, Micelles by Chemical Trappingâ€. Langmuir, 1996, 12, 2425-2432.	3.5	82
32	Arenediazonium Salts: New Probes of the Interfacial Compositions of Association Colloids. 3. Distributions of Butanol, Hexanol, and Water in Four-Component Cationic Microemulsions. Journal of the American Chemical Society, 1994, 116, 11779-11786.	13.7	36
33	Arenediazonium salts: new probes of the interfacial compositions of association colloids. 2. Binding constants of butanol and hexanol in aqueous three-component cetyltrimethylammonium bromide microemulsions. Journal of the American Chemical Society, 1993, 115, 8362-8367.	13.7	52
34	Arenediazonium salts: new probes of the interfacial compositions of association colloids. 1. Basic approach, methods, and illustrative applications. Journal of the American Chemical Society, 1993, 115, 8351-8361.	13.7	126
35	Simultaneous determination of counterion, alcohol, and water concentrations at a three-component microemulsion interface using product distributions from a dediazoniation reaction. Journal of the American Chemical Society, 1991, 113, 5052-5053.	13.7	46
36	Anomalous salt effects on a micellar-mediated reaction of bromide ion. Journal of Physical Organic Chemistry, 1990, 3, 239-247.	1.9	8

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37	A new method for estimating counter-ion selectivity of a cationic association colloid: Trapping of interfacial chloride and bromide counter-ions by reaction with micellar bound aryldiazonium salts. Colloids and Surfaces, 1990, 48, 123-137.	0.9	47
38	Quantitative treatment of benzimidazole deprotonation equilibria in aqueous micellar solutions of cetyltrimethylammonium ion (CTAX, X- = Cl-, Br-, and NO3-) surfactants. 1. Variable surfactant concentration. The Journal of Physical Chemistry, 1985, 89, 5107-5113.	2.9	106