

# Eiki Niwa

## List of Publications by Year in descending order

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#	ARTICLE	IF	CITATIONS
1	Oxide-ion conduction in the Dionâ€“Jacobson phase CsBi <sub>2</sub> Ti <sub>2</sub> NbO <sub>10</sub> â€“. Nature Communications, 2020, 11, 1224.	12.8	50
2	Direct evidence for two-dimensional oxide-ion diffusion in the hexagonal perovskite-related oxide Ba <sub>3</sub> MoNbO <sub>8.5</sub> â€“. Journal of Materials Chemistry A, 2019, 7, 13910-13916.	10.3	44
3	Sintering temperature dependence of conductivity, porosity and specific surface area of LaNi <sub>0.6</sub> Fe <sub>0.4</sub> O <sub>3</sub> ceramics as cathode material for solid oxide fuel cellsâ€“Superiority of Pechini method among various solution mixing processes. Materials Research Bulletin, 2013, 48, 1-6.	5.2	35
4	Discovery of a Rare-Earth-Free Oxide-Ion Conductor Ca <sub>3</sub> Ga <sub>4</sub> O <sub>9</sub> by Screening through Bond Valence-Based Energy Calculations, Synthesis, and Characterization of Structural and Transport Properties. Inorganic Chemistry, 2019, 58, 9460-9468.	4.0	34
5	Conductivity and sintering property of LaNi <sub>1-x</sub> Fe <sub>x</sub> O <sub>3</sub> ceramics prepared by Pechini method. Solid State Ionics, 2011, 201, 87-93.	2.7	30
6	Thermodynamic analyses of structural phase transition of Pr <sub>2</sub> NiO <sub>4</sub> +Î“ involving variation of oxygen content. Thermochimica Acta, 2014, 575, 129-134.	2.7	25
7	Experimental visualization of oxide-ion diffusion paths in pyrochlore-type Yb <sub>2</sub> Ti <sub>2</sub> O <sub>7</sub> . Journal of the Ceramic Society of Japan, 2018, 126, 341-345.	1.1	25
8	Thermal Expansion and Phase Transition Behavior of Al <sub>2-x</sub> M <sub>x</sub> (WO <sub>4</sub> ) <sub>3</sub> (M=Y, Ga and Sc) Ceramics. Journal of the Ceramic Society of Japan, 2007, 115, 176-181.	1.3	24
9	High performance of electroless-plated platinum electrode for electrochemical hydrogen pumps using strontium-zirconate-based proton conductors. Electrochimica Acta, 2008, 53, 8172-8177.	5.2	24
10	Evaluation of reaction kinetics of CO <sub>2</sub> and Li <sub>4</sub> SiO <sub>4</sub> by thermogravimetry under various CO <sub>2</sub> partial pressures. Materials Research Bulletin, 2018, 97, 56-60.	5.2	22
11	Analysis of chemical reaction between Li <sub>4</sub> SiO <sub>4</sub> and CO <sub>2</sub> by thermogravimetry under various CO <sub>2</sub> partial pressuresâ€“Clarification of CO <sub>2</sub> partial pressure and temperature region of CO <sub>2</sub> absorption or desorption. Materials Research Bulletin, 2017, 94, 134-139.	5.2	19
12	Dependence of thermal expansion of LaNi <sub>0.6</sub> Fe <sub>0.4</sub> O <sub>3</sub> â€“ and La <sub>0.6</sub> Sr <sub>0.4</sub> Co <sub>0.2</sub> Fe <sub>0.8</sub> O <sub>3</sub> â€“ on oxygen partial pressure. Solid State Ionics, 2016, 285, 187-194.	2.7	18
13	Discovery of Oxide-Ion Conductors with a New Crystal Structure, BaSc <sub>2</sub> A <sub>3</sub> Si <sub>3</sub> O <sub>10</sub> â€“ (A = Mg, Ca) by Screening Sc-Containing Oxides through the Bond-Valence Method and Experiments. ACS Applied Energy Materials, 2018, 1, 4009-4015.	5.1	17
14	Crystal structure of blue-colored ceria during redox reactions in a hydrogen atmosphere. CrystEngComm, 2018, 20, 155-158.	2.6	16
15	Analysis of structural phase transition of Nd <sub>2</sub> NiO <sub>4</sub> +Î“ by scanning thermal measurement under controlled oxygen partial pressure. Thermochimica Acta, 2011, 523, 46-50.	2.7	15
16	Preparation of Ba <sub>1-x</sub> La <sub>x</sub> FeO <sub>3</sub> â€“ (x = 0.1â€“0.6) with cubic perovskite phase and random distribution of oxide ion vacancy and their electrical conduction property and thermal expansion behavior. Solid State Ionics, 2018, 320, 76-83.	2.7	15
17	A new structure family of oxide-ion conductors Ca <sub>0.8</sub> Y <sub>2.4</sub> Sn <sub>0.8</sub> O <sub>6</sub> discovered by a combined technique of the bond-valence method and experiments. Dalton Transactions, 2018, 47, 7515-7521.	3.3	15
18	Evaluation of Specific Surface Area and Pore Size Distribution of La <sub>0.6</sub> Fe <sub>0.4</sub> O <sub>3</sub> â€“ Ceramics Prepared using Pechini Method by N <sub>2</sub> Adsorption Methodâ€“Optimization of Sintering Temperature as Cathode Material of Solid Oxide Fuel Cells. Journal of the American Ceramic Society, 2012, 95, 3802-3806.	3.8	14

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19	Pr/Ba cation-disordered perovskite $\text{Pr}_{2/3}\text{Ba}_{1/3}\text{CoO}_3$ as a new bifunctional electrocatalyst for oxygen reduction and oxygen evolution reactions. Journal of the Ceramic Society of Japan, 2018, 126, 814-819.	1.1	14
20	Dependence of crystal symmetry, electrical conduction property and electronic structure of $\text{LnFeO}_3$ (Ln: La, Pr, Nd, Sm) on kinds of Ln <sup>3+</sup> . Journal of the Ceramic Society of Japan, 2015, 123, 501-506.	1.1	13
21	Crystal structure and electrical conductivity of $\text{Ba}_{1-x}\text{R}_{2x}\text{ZnO}_5$ (R = Sm, Gd). Journal of the Ceramic Society of Japan, 2018, 126, 292-299.	1.1	13
22	Thermal analysis of structural phase transition behavior of $\text{Ln}_2\text{Ni}_{1-x}\text{Cu}_x\text{O}_4$ (Ln = Nd, Pr) under various oxygen partial pressures. Journal of Thermal Analysis and Calorimetry, 2019, 135, 2765-2774.	3.6	13
23	Li vaporization property of two-phase material of $\text{Li}_2\text{TiO}_3$ and $\text{Li}_2\text{SiO}_3$ for tritium breeder. Fusion Engineering and Design, 2015, 98-99, 1859-1863.	1.9	12
24	Oxygen nonstoichiometry and electrical conductivity of $\text{LaNi}_{0.6}\text{Fe}_{0.4}\text{O}_3$ at high temperatures under various oxygen partial pressures. Solid State Ionics, 2015, 274, 119-122.	2.7	10
25	Electrical conduction mechanism of $\text{LaNi}_x\text{Me}_{1-x}\text{O}_3$ (Me = Fe, Mn). Materials Research Bulletin, 2015, 70, 241-247.	5.2	10
26	Dependence of crystal structure, phase transition temperature, chemical state of Fe, oxygen content and electrical conductivity of $\text{Ba}_{2-x}\text{La}_x\text{Fe}_2\text{O}_5$ ( $x = 0.00 \sim 0.15$ ) on La content. Solid State Ionics, 2016, 290, 71-76.	2.7	10
27	Construction of structural phase diagram of $\text{Nd}_2\text{Ni}_1\text{-Cu}_x\text{O}_4$ and effect of crystal structure and phase transition on electrical conduction behavior. Materials Research Bulletin, 2019, 111, 61-69.	5.2	10
28	Chemical state of Fe in $\text{LaNi}_{1-x}\text{Fe}_x\text{O}_3$ and its effect on electrical conduction property. Hyperfine Interactions, 2012, 206, 47-50.	0.5	8
29	Analysis of structural phase transition behavior of $\text{Ln}_2\text{NiO}_4$ (Ln: Nd, Pr) with variation of oxygen content. Solid State Ionics, 2014, 262, 724-727.	2.7	8
30	Sillarsite Aurivillius phase bismuth niobium oxychloride, $\text{Bi}_4\text{Nb}_8\text{O}_{28}\text{Cl}$ , as a new oxide-ion conductor. Journal of Materials Chemistry A, 2022, 10, 2550-2558.	10.3	8
31	$\text{K}_2\text{NiF}_4$ type oxides, $\text{Ln}_2\text{-Sr}_x\text{NiO}_4$ (Ln = La and Pr; $x = 0 \sim 1.4$ ) as an oxygen electrocatalyst for aqueous lithium oxygen rechargeable batteries. Solid State Ionics, 2021, 369, 115708.	2.7	7
32	Prevention of Sulfur Poisoning and Performance Recovery of Sulfur-Poisoned-Anode Electrode by Shifting Anode Electrode Potential. Journal of the Electrochemical Society, 2015, 162, F1107-F1113.	2.9	6
33	High-Temperature Gravimetric Study on the Kinetics of the Formation of $\text{SrTiO}_3$ by Solid State Reaction of $\text{SrCO}_3$ and $\text{TiO}_2$ . ECS Transactions, 2009, 16, 205-210.	0.5	5
34	Conductivities and Seebeck Coefficients of donor-doped- $\text{SrTiO}_3$ Oxide Ceramics. ECS Transactions, 2009, 25, 2631-2638.	0.5	5
35	Thermodynamics and kinetics analyses of high $\text{CO}_2$ absorption properties of $\text{Li}_3\text{NaSiO}_4$ under various $\text{CO}_2$ partial pressures. Dalton Transactions, 2021, 50, 5301-5310.	3.3	5
36	Analysis of thermal stability of $\text{LaNi}_{1-x}\text{Fe}_x\text{O}_3$ ( $x = 0.0, 0.2, 0.4$ ) by thermogravimetry and high-temperature X-ray diffraction under controlled oxygen partial pressures. Journal of Thermal Analysis and Calorimetry, 2016, 123, 1769-1775.	3.6	4

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37	Low Temperature Preparation of $\text{LaNi}_{1-x}\text{Fe}_x\text{O}_3$ as New Cathode Material for SOFC - Advantage of Liquid Phase Mixing Method -. ECS Transactions, 2011, 35, 1935-1943.	0.5	3
38	Preparation of Structural Phase Diagram of $\text{Ln}_{2-x}\text{Ni}_{1-x}\text{Cu}_x\text{O}_{4+\delta}$ (Ln=La, Pr, Nd) Tj ETQq 0,0 0 rgBJ /Overlock Transactions, 2017, 78, 613-622.	0.5	3
39	Crystal structure, thermal expansion and electrical conduction behavior of $\text{PrNi}_{1-x}\text{Fe}_x\text{O}_{3-\delta}$ at high temperature. Journal of the Ceramic Society of Japan, 2017, 125, 227-235.	1.1	3
40	Effect of chemical state and occupation site of RE (RE = Yb, Y, Eu, Sm, Nd) on crystal structure and optical property of $\text{BaCe}_{1-x}\text{RE}_x\text{O}_{3-\delta}$ Analyses of origin of peculiar crystal structure and property of $\text{BaCe}_{1-x}\text{Nd}_x\text{O}_{3-\delta}$ . Materials Research Bulletin, 2017, 87, 6-13.	5.2	2
41	Preparation of Structural Phase Diagram of $\text{Nd}_2\text{Ni}_{1-x}\text{Cu}_x\text{O}_{4+\delta}$ As New Cathode Materials “ Clarification of Existence of Miscibility Gap. ECS Transactions, 2017, 78, 603-612.	0.5	1
42	Thermodynamic analyses of the orthorhombic-to-tetragonal phase transition in $\text{Pr}_{2-x}\text{Nd}_x\text{NiO}_{4+\delta}$ under controlled oxygen partial pressures. Dalton Transactions, 2020, 49, 11931-11941.	3.3	1
43	Dense-film preparation of zirconium oxide by self-oxidation in air. Fusion Engineering and Design, 2021, 171, 112793.	1.9	0