## Sankara Sarma V Tatiparti

List of Publications by Year in descending order

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758635 713013 36 491 12 21 citations g-index h-index papers 36 36 36 483 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	Modeling and numerical simulation of a 5Âkg LaNi5-based hydrogen storage reactor with internal conical fins. International Journal of Hydrogen Energy, 2020, 45, 8794-8809.	3.8	66
2	Tension–compression asymmetry in an extruded Mg alloy AM30: Temperature and strain rate effects. Materials Science & Engineering A: Structural Materials: Properties, Microstructure and Processing, 2013, 572, 8-18.	2.6	49
3	Mg–C Interaction Induced Hydrogen Uptake and Enhanced Hydrogen Release Kinetics in MgH <sub>2</sub> -rGO Nanocomposites. Journal of Physical Chemistry C, 2018, 122, 22389-22396.	1.5	40
4	Nano-structured palladium impregnate graphitic carbon nitride composite for efficient hydrogen gasÂsensing. International Journal of Hydrogen Energy, 2020, 45, 10623-10636.	3.8	36
5	Synthesis, characterization and photocatalytic activity evaluation of TiO2 – ZnO nanocomposites: Elucidating effect of varying Ti:Zn molar ratio. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2019, 565, 47-58.	2.3	34
6	Electrodeposition of Al–Mg Alloy Powders. Journal of the Electrochemical Society, 2008, 155, D363.	1.3	27
7	Electrode and symmetric supercapacitor device performance of boronâ€incorporated reduced graphene oxide synthesized by electrochemical exfoliation. Energy Storage, 2020, 2, e134.	2.3	21
8	Potentiostatic versus galvanostatic electrodeposition of nanocrystalline Al–Mg alloy powders. Journal of Solid State Electrochemistry, 2012, 16, 1255-1262.	1.2	15
9	Strategies for scaling-up LaNi5-based hydrogen storage system with internal conical fins and cooling tubes. International Journal of Hydrogen Energy, 2021, 46, 19031-19045.	3.8	15
10	The Formation of Morphologies and Microstructures in Electrodeposited Nanocrystalline Al–Mg Alloy Powders. Journal of the Electrochemical Society, 2010, 157, E167.	1.3	14
11	The dehydrogenation mechanism during the incubation period in nanocrystalline MgH <sub>2</sub> . Physical Chemistry Chemical Physics, 2017, 19, 6677-6687.	1.3	14
12	Electrochemical Behavior of Cobalt Oxide/Boron-Incorporated Reduced Graphene Oxide Nanocomposite Electrode for Supercapacitor Applications. Journal of Materials Engineering and Performance, 2020, 29, 6535-6549.	1.2	14
13	Contributions of multiple phenomena towards hydrogenation: A case of Mg. International Journal of Hydrogen Energy, 2015, 40, 13518-13529.	3.8	13
14	An understanding of the electrodeposition process of Al–Mg alloys using an organometallic-based electrolyte. Journal of Applied Electrochemistry, 2010, 40, 2091-2098.	1.5	11
15	Annealing response of AA5182 deformed in plane strain and equibiaxial strain paths. Philosophical Magazine, 2013, 93, 2613-2629.	0.7	11
16	Selective Removal of Photocatalytically Active Anatase TiO <sub>2</sub> Phase from Mixedâ€Phase TiO <sub>2</sub> â€ZnO Nanocomposites: Impact on Physicochemical Properties and Photocatalytic Activity. Energy and Environmental Materials, 2020, 3, 548-559.	7.3	11
17	Transition from interfacial to diffusional growth during hydrogenation of Mg. Materials Letters, 2015, 161, 271-274.	1.3	10
18	Boron from net charge acceptor to donor and its effect on hydrogen uptake by novel Mg-B-electrochemically synthesized reduced graphene oxide. Scientific Reports, 2021, 11, 10995.	1.6	10

#	Article	IF	Citations
19	Effect of calcination temperature on the microstructure and electronic properties of TiO2–ZnO nanocomposites and implications on photocatalytic activity. Applied Nanoscience (Switzerland), 2018, 8, 915-930.	1.6	8
20	Hydrogen Sorption Mechanism of Magnesium (Hydride). Materials Today: Proceedings, 2018, 5, 23235-23241.	0.9	8
21	Instantaneous-Progressive nucleation and growth of palladium during electrodeposition. Results in Surfaces and Interfaces, 2022, 6, 100044.	1.0	8
22	Synergetic effect of C and Ni on hydrogen release from Mg–Ni-electrochemically synthesized reduced graphene oxide based hydride. Sustainable Energy and Fuels, 2021, 5, 4414-4424.	2.5	7
23	Preferred orientation and shape of electrodeposited nanocrystalline Al–Mg alloy dendrites. Materials Letters, 2011, 65, 1915-1918.	1.3	6
24	Internal structure of the electrodeposited nanocrystalline Al–Mg alloy dendrites. Materials Letters, 2011, 65, 2413-2415.	1.3	6
25	Evolution of Morphology and Microstructure in Electrodeposited Nanocrystalline Al–Mg Alloy Dendrites. Metals, 2011, 1, 3-15.	1.0	5
26	Substrate effect on electrodeposited nanocrystalline Al–Mg alloy powders. Materials Letters, 2011, 65, 1859-1861.	1.3	5
27	On the parameters of Johnson-Mehl-Avrami-Kolmogorov equation for the hydride growth mechanisms: A case of MgH2. Journal of Alloys and Compounds, 2018, 742, 1002-1005.	2.8	5
28	Stoichiometry–grain size-specific capacitance interrelationships in nickel oxide. RSC Advances, 2022, 12, 8333-8344.	1.7	5
29	Extended solubility in the electrodeposited nanocrystalline Al–Mg alloy dendrites. Materials Letters, 2011, 65, 3173-3175.	1.3	4
30	Banded structure of the electrodeposited nanocrystalline Al–Mg alloy dendrites. Materials Letters, 2011, 65, 3262-3264.	1.3	4
31	Nanostructure stabilization in electrodeposited Al–Mg dendrites. Journal of Alloys and Compounds, 2017, 694, 632-635.	2.8	4
32	Anomalous Al–Mg Electrodeposition Using an Organometallic-Based Electrolyte. Journal of the Electrochemical Society, 2016, 163, D722-D727.	1.3	2
33	Influence of surface condition on the current densities rendering nucleation loop during cyclic voltammetry for electrodeposition of Pd thin films. Surfaces and Interfaces, 2020, 20, 100525.	1.5	2
34	Synthesis, structural and morphological property of a novel Pd/g-CN nano composite for gas sensing application. IOP Conference Series: Materials Science and Engineering, 2019, 499, 012003.	0.3	1
35	Effect of Mg shell on MgH2 dehydrogenation by morphological and mathematical analysis. Nanomaterials and Energy, 2019, 8, 186-195.	0.1	O
36	On the Nucleation Loop in Cyclic Voltammetry for Electrodeposition of Pd Thin Films. ECS Meeting Abstracts, 2020, MA2020-01, 1189-1189.	0.0	0