

Robert Pietrzak

List of Publications by Year in descending order

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115
papers

4,606
citations

70961

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all docs

115
docs citations

115
times ranked

4726
citing authors

#	ARTICLE	IF	CITATIONS
1	Removal of Organic Dyes from Aqueous Solutions by Activated Carbons Prepared from Residue of Supercritical Extraction of Marigold. <i>Materials</i> , 2022, 15, 3655.	1.3	9
2	Bioavailability of Hesperidin and Its Aglycone Hesperetin Compounds Found in Citrus Fruits as a Parameter Conditioning the Pro-Health Potential (Neuroprotective and Antidiabetic) <i>Tj ETQq0 0 0 rgBT /Overlock 10.7f 50 6971Td (Activ</i>	10.7	50
3	Removal of NO ₂ from gas stream by activated bio-carbons from physical activation of residue of supercritical extraction of hops. <i>Chemical Engineering Research and Design</i> , 2021, 166, 67-73.	2.7	16
4	Activated Bio-Carbons Prepared from the Residue of Supercritical Extraction of Raw Plants and Their Application for Removal of Nitrogen Dioxide and Hydrogen Sulfide from the Gas Phase. <i>Materials</i> , 2021, 14, 3192.	1.3	10
5	N-doped sawdust-based activated biocarbons prepared by microwave-assisted heat treatment as potential electrode materials for supercapacitors. <i>Journal of Wood Chemistry and Technology</i> , 2021, 41, 307-320.	0.9	6
6	Removal of Heavy Metal Ions from One- and Two-Component Solutions via Adsorption on N-Doped Activated Carbon. <i>Materials</i> , 2021, 14, 7045.	1.3	16
7	Iron(II) Sulfate(VI) from Titania Production as a Raw Material for Preparation of Hydrogen Sulfide Sorbents. <i>Chemical Engineering and Technology</i> , 2020, 43, 104-110.	0.9	1
8	Characterization and application of spherical carbonaceous materials prepared with the use of microwave radiation. <i>Diamond and Related Materials</i> , 2020, 108, 107927.	1.8	4
9	Adsorption of organic and inorganic pollutants on activated bio-carbons prepared by chemical activation of residues of supercritical extraction of raw plants. <i>Chemical Engineering Journal</i> , 2020, 393, 124785.	6.6	49
10	Hay-based activated biochars obtained using two different heating methods as effective low-cost sorbents: Solid surface characteristics, adsorptive properties and aggregation in the mixed Cu(II)/PAM system. <i>Chemosphere</i> , 2020, 250, 126312.	4.2	19
11	Composite sulfur cathode for Li-S batteries comprising hierarchical carbon obtained from waste PET bottles. <i>Synthetic Metals</i> , 2020, 261, 116305.	2.1	19
12	The effect of demineralization on the physicochemical and sorption properties of activated bio-carbons. <i>Adsorption</i> , 2019, 25, 337-343.	1.4	15
13	Comparison of physicochemical, sorption and electrochemical properties of nitrogen-doped activated carbons obtained with the use of microwave and conventional heating. <i>Adsorption</i> , 2019, 25, 405-417.	1.4	7
14	Optimal synthesis of oxidized mesoporous carbons for the adsorption of heavy metal ions. <i>Journal of Molecular Liquids</i> , 2019, 276, 630-637.	2.3	53
15	Removal of rhodamine B from water by modified carbon xerogels. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2018, 543, 109-117.	2.3	62
16	Production of new activated bio-carbons by chemical activation of residue left after supercritical extraction of hops. <i>Environmental Research</i> , 2018, 161, 456-463.	3.7	26
17	Physicochemical characterization of ordered mesoporous carbons functionalized by wet oxidation. <i>Journal of Materials Science</i> , 2018, 53, 5997-6007.	1.7	16
18	Removal of Iron and Copper Ions from the Liquid Phase by Modified Polymeric Membranes. <i>Journal of Polymers and the Environment</i> , 2018, 26, 3237-3242.	2.4	3

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19	Synthesis of carbon xerogels modified with amine groups and copper for efficient adsorption of caffeine. <i>Chemical Engineering Journal</i> , 2018, 345, 13-21.	6.6	35
20	Catalytic performance of ordered mesoporous carbons modified with lanthanides in dry methane reforming. <i>Catalysis Today</i> , 2018, 301, 204-216.	2.2	28
21	Coniferous Wood Sawdust-based Activated Carbons as Adsorbents Obtained with the Use of Microwave Radiation. <i>Journal of Wood Chemistry and Technology</i> , 2018, 38, 286-299.	0.9	6
22	Uptake of heavy metal ions from aqueous solutions by sorbents obtained from the spent ion exchange resins. <i>Microporous and Mesoporous Materials</i> , 2017, 244, 127-136.	2.2	49
23	Adsorption of poly(acrylic acid) on the surface of microporous activated carbon obtained from cherry stones. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2017, 514, 137-145.	2.3	34
24	The influence of activation procedure on the physicochemical and sorption properties of activated carbons prepared from pistachio nutshells for removal of NO ₂ /H ₂ S gases and dyes. <i>Journal of Cleaner Production</i> , 2017, 152, 211-222.	4.6	54
25	Thermal and physicochemical properties of phosphorus-containing activated carbons obtained from biomass. <i>Journal of the Taiwan Institute of Chemical Engineers</i> , 2017, 80, 1006-1013.	2.7	32
26	Characterization and application of bio-activated carbons prepared by direct activation of hay with the use of microwave radiation. <i>Powder Technology</i> , 2017, 319, 302-312.	2.1	24
27	Adsorption of dyes on the surface of polymer nanocomposites modified with methylamine and copper(II) chloride. <i>Journal of Colloid and Interface Science</i> , 2017, 504, 549-560.	5.0	33
28	THERMAL STUDY OF ADSORBENTS PREPARED FROM WASTE TYRES. <i>Environmental Engineering and Management Journal</i> , 2017, 16, 439-447.	0.2	2
29	Removal of NO ₂ by carbonaceous adsorbents obtained from residue after supercritical extraction of marigold. <i>Adsorption</i> , 2016, 22, 465-471.	1.4	11
30	Thermal analysis of activated carbon obtained from residue after supercritical extraction of hops. <i>Journal of Thermal Analysis and Calorimetry</i> , 2016, 125, 1199-1204.	2.0	57
31	The use of microwave radiation for obtaining carbonaceous adsorbents from biomass and their use in elimination of inorganic pollutants. <i>Adsorption</i> , 2016, 22, 473-480.	1.4	9
32	Membranes obtained on the basis of cellulose acetate and their use in removal of metal ions from liquid phase. <i>Polish Journal of Chemical Technology</i> , 2016, 18, 104-110.	0.3	6
33	Biomass-derived hierarchical carbon as sulfur cathode stabilizing agent for lithium-sulfur batteries. <i>Solid State Ionics</i> , 2016, 297, 59-63.	1.3	39
34	Preparation and physicochemical characterisation of functionalised multi-walled carbon nanotubes. <i>Adsorption</i> , 2016, 22, 481-488.	1.4	0
35	β-Cyclodextrin complexation as an effective drug delivery system for meropenem. <i>European Journal of Pharmaceutics and Biopharmaceutics</i> , 2016, 99, 24-34.	2.0	44
36	Stability analysis of functionalized mesoporous carbon materials in aqueous solution. <i>Chemical Engineering Journal</i> , 2016, 290, 209-219.	6.6	35

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37	Toxic gases removal onto activated carbons obtained from hay with the use of microwave radiation. <i>Chemical Engineering Research and Design</i> , 2016, 109, 346-353.	2.7	10
38	The effect of surface modification of mesoporous carbons on Auramine-O dye removal from water. <i>Adsorption</i> , 2016, 22, 531-540.	1.4	27
39	Physicochemical and adsorption properties of carbonaceous sorbents prepared by activation of tropical fruit skins with potassium carbonate. <i>Materials and Design</i> , 2016, 90, 579-585.	3.3	69
40	Comparison of adsorption properties of $Mg_xO_y \cdot SiO_2$ and $Zn_xO_y \cdot SiO_2$ in the mixed oxide-poly(vinyl alcohol) system. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2016, 492, 12-18.	2.3	17
41	Production of activated carbons from biodegradable waste materials as an alternative way of their utilisation. <i>Adsorption</i> , 2016, 22, 489-502.	1.4	31
42	Complex of Rutin with β -Cyclodextrin as Potential Delivery System. <i>PLoS ONE</i> , 2015, 10, e0120858.	1.1	50
43	Textural, surface, thermal and sorption properties of the functionalized activated carbons and carbon nanotubes. <i>Polish Journal of Chemical Technology</i> , 2015, 17, 120-127.	0.3	8
44	Sorption Properties of Carbonaceous Adsorbents Obtained by Pyrolysis and Activation of Pistachio Nut Shells. <i>Adsorption Science and Technology</i> , 2015, 33, 581-586.	1.5	17
45	The use of microwave radiation for obtaining activated carbons enriched in nitrogen. <i>Powder Technology</i> , 2015, 273, 71-75.	2.1	27
46	Equilibrium and kinetic studies of chromotrope 2R adsorption onto ordered mesoporous carbons modified with lanthanum. <i>Chemical Engineering Journal</i> , 2015, 270, 140-149.	6.6	37
47	The use of microwave radiation for obtaining activated carbons from sawdust and their potential application in removal of NO_2 and H_2S . <i>Chemical Engineering Journal</i> , 2015, 269, 352-358.	6.6	73
48	Removal of tartrazine from aqueous solution by carbon nanotubes decorated with silver nanoparticles. <i>Catalysis Today</i> , 2015, 249, 259-264.	2.2	57
49	Ordered mesoporous silica modified with lanthanum for ibuprofen loading and release behaviour. <i>European Journal of Pharmaceutics and Biopharmaceutics</i> , 2015, 94, 550-558.	2.0	49
50	Ordered mesoporous carbons modified with cerium as effective adsorbents for azo dyes removal. <i>Separation and Purification Technology</i> , 2015, 154, 236-245.	3.9	62
51	Comparison of physicochemical and sorption properties of activated carbons prepared by physical and chemical activation of cherry stones. <i>Powder Technology</i> , 2015, 269, 312-319.	2.1	124
52	Nitrogen-enriched activated carbons prepared by the activation of coniferous tree sawdust and their application in the removal of Nitrogen dioxide. <i>International Journal of Environmental Science and Technology</i> , 2015, 12, 2233-2244.	1.8	25
53	Comparison of the effects of different chemical activation methods on properties of carbonaceous adsorbents obtained from cherry stones. <i>Chemical Engineering Research and Design</i> , 2014, 92, 1187-1191.	2.7	49
54	Adsorption of l-phenylalanine on ordered mesoporous carbons prepared by hard template method. <i>Journal of the Taiwan Institute of Chemical Engineers</i> , 2014, 45, 347-353.	2.7	29

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55	Hydrogen sulphide removal on carbonaceous adsorbents prepared from coffee industry waste materials. <i>Chemical Engineering Journal</i> , 2014, 248, 208-215.	6.6	68
56	Synthesis of polymer membranes of different porosity and their application for phenol removal from liquid phase. <i>Korean Journal of Chemical Engineering</i> , 2014, 31, 304-309.	1.2	5
57	Mesoporous carbons modified with lanthanum(III) chloride for methyl orange adsorption. <i>Chemical Engineering Journal</i> , 2014, 247, 258-264.	6.6	114
58	Physicochemical and sorption properties of multi-walled carbon nanotubes decorated with silver nanoparticles. <i>Chemical Engineering Journal</i> , 2014, 250, 295-302.	6.6	11
59	Comparison of ordered mesoporous materials sorption properties towards amino acids. <i>Adsorption</i> , 2013, 19, 581-588.	1.4	22
60	On competitive uptake of SO ₂ and CO ₂ from air by porous carbon containing CaO and MgO. <i>Chemical Engineering Journal</i> , 2013, 226, 348-356.	6.6	34
61	Sorption properties of activated carbons obtained from corn cobs by chemical and physical activation. <i>Adsorption</i> , 2013, 19, 273-281.	1.4	71
62	NO ₂ removal on adsorbents prepared from coffee industry waste materials. <i>Adsorption</i> , 2013, 19, 521-528.	1.4	33
63	Porous carbon material containing CaO for acidic gas capture: Preparation and properties. <i>Journal of Hazardous Materials</i> , 2013, 263, 353-360.	6.5	22
64	MgO/CaO-loaded porous carbons for carbon dioxide capture. <i>Journal of Thermal Analysis and Calorimetry</i> , 2013, 111, 357-364.	2.0	33
65	Adsorption of l-phenylalanine onto mesoporous silica. <i>Materials Chemistry and Physics</i> , 2013, 142, 586-593.	2.0	35
66	The effect of chemical activation method on properties of activated carbons obtained from pine cones. <i>Open Chemistry</i> , 2013, 11, 78-85.	1.0	15
67	Synergy effect in the photocatalytic degradation of methylene blue on a suspended mixture of TiO ₂ and N-containing carbons. <i>Carbon</i> , 2013, 54, 460-471.	5.4	48
68	The influence of pore generating agent on the efficiency of copper and iron ions removal from liquid phase by polyethersulfone membranes. <i>Chemical Engineering Journal</i> , 2013, 228, 449-454.	6.6	12
69	MgO/CaO-Loaded Activated Carbon for Carbon Dioxide Capture: Practical Aspects of Use. <i>Industrial & Engineering Chemistry Research</i> , 2013, 52, 6669-6677.	1.8	26
70	In vitro release of l-phenylalanine from ordered mesoporous materials. <i>Microporous and Mesoporous Materials</i> , 2013, 177, 32-36.	2.2	18
71	Copper ions removal from liquid phase by polyethersulfone (PES) membranes functionalized by introduction of carbonaceous materials. <i>Chemical Engineering Journal</i> , 2013, 215-216, 216-221.	6.6	29
72	Metals Ions Removal by Polymer Membranes of Different Porosity. <i>Scientific World Journal</i> , The, 2013, 2013, 1-7.	0.8	11

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73	Thermal analysis of activated carbons modified with silver metavanadate. <i>Thermochimica Acta</i> , 2012, 541, 42-48.	1.2	11
74	MgO-loaded porous carbon for carbon dioxide sorption: Study on cyclic sorptionâ€“regeneration. <i>International Journal of Greenhouse Gas Control</i> , 2012, 10, 164-168.	2.3	8
75	Nitrogen-Doped Carbonaceous Materials for Removal of Phenol from Aqueous Solutions. <i>Scientific World Journal</i> , The, 2012, 2012, 1-8.	0.8	8
76	NO ₂ removal by adsorbents prepared from waste paper sludge. <i>Chemical Engineering Journal</i> , 2012, 183, 278-283.	6.6	16
77	The influence of silver on the physicochemical and catalytic properties of activated carbons. <i>Chemical Engineering Journal</i> , 2012, 189-190, 422-430.	6.6	14
78	NO ₂ removal on adsorbents obtained by pyrolysis and physical activation of corrugated cardboard. <i>Chemical Engineering Journal</i> , 2012, 195-196, 7-14.	6.6	45
79	Effect of ammoxidation of activated carbons obtained from sub-bituminous coal on their NO ₂ sorption capacity under dry conditions. <i>Chemical Engineering Journal</i> , 2011, 166, 1039-1043.	6.6	48
80	Adsorbents obtained from waste tires for NO ₂ removal under dry conditions at room temperature. <i>Chemical Engineering Journal</i> , 2011, 170, 202-208.	6.6	50
81	Active carbons prepared by chemical activation of plum stones and their application in removal of NO ₂ . <i>Journal of Hazardous Materials</i> , 2010, 181, 1088-1094.	6.5	92
82	Nitrogen-enriched bituminous coal-based active carbons as materials for supercapacitors. <i>Fuel</i> , 2010, 89, 3457-3467.	3.4	60
83	Carbonaceous adsorbents prepared by physical activation of pine sawdust and their application for removal of NO ₂ in dry and wet conditions. <i>Bioresource Technology</i> , 2010, 101, 5802-5807.	4.8	48
84	Sorption properties of active carbons obtained from walnut shells by chemical and physical activation. <i>Catalysis Today</i> , 2010, 150, 107-114.	2.2	96
85	Effect of activation method on the physicochemical properties and NO ₂ removal abilities of sorbents obtained from plum stones (<i>Prunus domestica</i>). <i>Chemical Engineering Journal</i> , 2010, 162, 723-729.	6.6	28
86	Sawdust pellets from coniferous species as adsorbents for NO ₂ removal. <i>Bioresource Technology</i> , 2010, 101, 907-913.	4.8	33
87	X-ray Photoelectron Spectroscopy Study of Nitrogen-Enriched Active Carbons Obtained by Ammoxidation and Chemical Activation of Brown and Bituminous Coals. <i>Energy & Fuels</i> , 2010, 24, 1197-1206.	2.5	73
88	Interactions of NO ₂ and NO with Carbonaceous Adsorbents Containing Silver Nanoparticles. <i>Langmuir</i> , 2010, 26, 9457-9464.	1.6	29
89	XPS study and physico-chemical properties of nitrogen-enriched microporous activated carbon from high volatile bituminous coal. <i>Fuel</i> , 2009, 88, 1871-1877.	3.4	245
90	The influence of oxidation with nitric acid on the preparation and properties of active carbon enriched in nitrogen. <i>Applied Surface Science</i> , 2009, 255, 3586-3593.	3.1	17

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91	Active Carbons Obtained from Bituminous Coal for NO ₂ Removal under Dry and Wet Conditions at Room Temperature. Energy & Fuels, 2009, 23, 3617-3624.	2.5	30
92	Influence of the Precursor Metamorphism Degree on Preparation of Nitrogen-enriched Activated Carbons by Ammoxidation and Chemical Activation of Coals. Energy & Fuels, 2009, 23, 2205-2212.	2.5	42
93	Interactions of NO ₂ with sewage sludge based composite adsorbents. Journal of Hazardous Materials, 2008, 154, 946-953.	6.5	35
94	Capacitance behaviour of brown coal based active carbon modified through chemical reaction with urea. Electrochimica Acta, 2008, 53, 5469-5475.	2.6	130
95	Siberian anthracite as a precursor material for microporous activated carbons. Fuel, 2008, 87, 2037-2040.	3.4	33
96	Comparison of Physicochemical Properties of Nitrogen-enriched Activated Carbons Prepared by Physical and Chemical Activation of Brown Coal. Energy & Fuels, 2008, 22, 4133-4138.	2.5	41
97	Reactive Adsorption of NO ₂ at Dry Conditions on Sewage Sludge-Derived Materials. Environmental Science & Technology, 2007, 41, 7516-7522.	4.6	30
98	Role of Graphite Oxide (GO) and Polyaniline (PANI) in NO ₂ Reduction on GO-PANI Composites. Industrial & Engineering Chemistry Research, 2007, 46, 6925-6935.	1.8	53
99	Activated carbons modified with sewage sludge derived phase and their application in the process of NO ₂ removal. Carbon, 2007, 45, 2537-2546.	5.4	65
100	Microporous activated carbons from ammoxidised anthracite and their capacitance behaviours. Fuel, 2007, 86, 1086-1092.	3.4	52
101	XPS study of pyrite-free coals subjected to different oxidizing agents. Fuel, 2007, 86, 2616-2624.	3.4	114
102	Preparation of modified active carbon from brown coal by ammoxidation. Fuel Processing Technology, 2007, 88, 409-415.	3.7	58
103	Microwave saturation of EPR spectra of oxidised coal. Open Chemistry, 2007, 5, 330-340.	1.0	9
104	Preparation of Nitrogen-Enriched Activated Carbons from Brown Coal. Energy & Fuels, 2006, 20, 1275-1280.	2.5	115
105	Capacitance properties of multi-walled carbon nanotubes modified by activation and ammoxidation. Carbon, 2006, 44, 2368-2375.	5.4	115
106	The influence of oxidation with air in comparison to oxygen in sodium carbonate solution on the surface composition of coals of different ranks. Fuel, 2006, 85, 1016-1023.	3.4	51
107	The influence of oxidation with HNO ₃ on the surface composition of high-sulphur coals: XPS study. Fuel Processing Technology, 2006, 87, 1021-1029.	3.7	94
108	The effect of flame coal oxidation on the solid and soluble products of its extraction. Open Chemistry, 2005, 3, 852-865.	1.0	0

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109	Studies of the soluble part of oxidised coals. <i>Open Chemistry</i> , 2004, 2, 278-289.	1.0	2
110	Thermal analysis of oxidised coals. <i>Thermochimica Acta</i> , 2004, 419, 247-251.	1.2	45
111	The Comparison of Oxygen and Sulfur Species Formed by Coal Oxidation with O ₂ /Na ₂ CO ₃ or Peroxyacetic Acid Solution. <i>XPS Studies. Energy & Fuels</i> , 2004, 18, 804-809.	2.5	27
112	Low temperature oxidation of coals of different rank and different sulphur content. <i>Fuel</i> , 2003, 82, 705-713.	3.4	82
113	X-ray photoelectron spectroscopy study of oxidized coals with different sulphur content. <i>Fuel Processing Technology</i> , 2002, 77-78, 1-7.	3.7	140
114	TPR study of sulphur in coals subjected to mild oxidation. Part 1. Demineralised coals. <i>Fuel</i> , 2002, 81, 2397-2405.	3.4	38
115	Oxidation of demineralized coal and coal free of pyrite examined by EPR spectroscopy. <i>Fuel</i> , 2002, 81, 1925-1931.	3.4	56