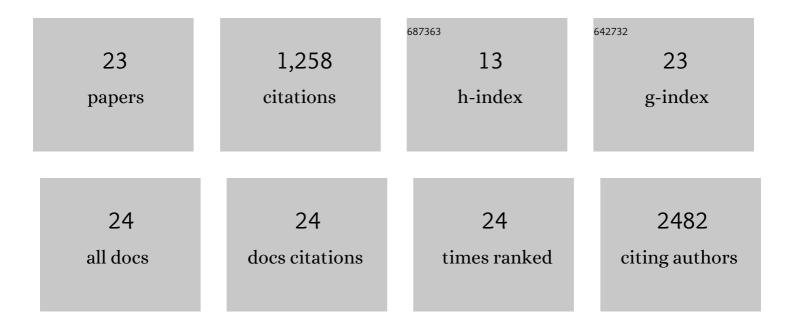
Wen-Hua Li

List of Publications by Year in descending order

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<u> Μεν-Ημα Γι</u>

#	Article	IF	CITATIONS
1	A brief review of hole transporting materials commonly used in perovskite solar cells. Rare Metals, 2021, 40, 2712-2729.	7.1	138
2	Synthesis of hybrid Au-Ag2S-Cu2-xS nanocrystals with disparate interfacial features. Journal of Colloid and Interface Science, 2021, 603, 11-16.	9.4	3
3	A Green Solvent Processable Wideâ€Bandgap Conjugated Polymer for Organic Solar Cells. Solar Rrl, 2020, 4, 2000547.	5.8	13
4	The preparation of plasmonic Au@SiO2 NPs and its application in polymer solar cells. Materials Letters, 2020, 268, 127599.	2.6	7
5	Crosslinked and dopant free hole transport materials for efficient and stable planar perovskite solar cells. Journal of Materials Chemistry A, 2019, 7, 5522-5529.	10.3	41
6	The preparation of Ag3BiBr6 films and their preliminary use for solution processed photovoltaics. SN Applied Sciences, 2019, 1, 1.	2.9	5
7	Ladder-like conjugated polymers used as hole-transporting materials for high-efficiency perovskite solar cells. Journal of Materials Chemistry A, 2019, 7, 14473-14477.	10.3	48
8	Improved Efficiency of Perovskite Solar Cells by the Interfacial Modification of the Active Layer. Nanomaterials, 2019, 9, 204.	4.1	12
9	Broadband Absorption Enhancement in Polymer Solar Cells Using Highly Efficient Plasmonic Heterostructured Nanocrystals. ACS Applied Materials & Interfaces, 2018, 10, 30919-30924.	8.0	16
10	A simple strategy to achieve shape control of Au-Cu2â^'xS colloidal heterostructured nanocrystals and their preliminary use in organic photovoltaics. Nanoscale, 2018, 10, 11745-11749.	5.6	12
11	Molecular "Flower―as the High-Mobility Hole-Transport Material for Perovskite Solar Cells. ACS Applied Materials & Interfaces, 2017, 9, 43855-43860.	8.0	31
12	The Influence of Fluorination on Nano-Scale Phase Separation and Photovoltaic Performance of Small Molecular/PC71BM Blends. Nanomaterials, 2016, 6, 80.	4.1	4
13	Enhancing the power conversion efficiency of polymer solar cells to 9.26% by a synergistic effect of fluoro and carboxylate substitution. Journal of Materials Chemistry A, 2016, 4, 8097-8104.	10.3	39
14	Elimination of the J–V hysteresis of planar perovskite solar cells by interfacial modification with a thermo-cleavable fullerene derivative. Journal of Materials Chemistry A, 2016, 4, 17649-17654.	10.3	24
15	Highly Efficient Planar Perovskite Solar Cells Via Interfacial Modification with Fullerene Derivatives. Small, 2016, 12, 1098-1104.	10.0	107
16	Enhancing the performance of polymer solar cells by tuning the drying process of blend films via changing side chains and using solvent additives. Journal of Materials Chemistry C, 2015, 3, 9670-9677.	5.5	7
17	Enhancing the Photovoltaic Performance by Tuning the Morphology of Polymer:PC ₇₁ BM Blends with a Commercially Available Nucleating Agent. ACS Applied Materials & Interfaces, 2015, 7, 18924-18929.	8.0	8
18	Spin-coated Ag nanoparticles onto ITO substrates for efficient improvement of polymer solar cell performance. Journal of Materials Chemistry C, 2015, 3, 1319-1324.	5.5	10

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#	Article	IF	CITATIONS
19	Cu2HgSnSe4 nanoparticles: synthesis and thermoelectric properties. CrystEngComm, 2013, 15, 8966.	2.6	25
20	Metal Ions To Control the Morphology of Semiconductor Nanoparticles: Copper Selenide Nanocubes. Journal of the American Chemical Society, 2013, 135, 4664-4667.	13.7	112
21	Colloidal synthesis and thermoelectric properties of Cu ₂ SnSe ₃ nanocrystals. Journal of Materials Chemistry A, 2013, 1, 1421-1426.	10.3	86
22	CuTe Nanocrystals: Shape and Size Control, Plasmonic Properties, and Use as SERS Probes and Photothermal Agents. Journal of the American Chemical Society, 2013, 135, 7098-7101.	13.7	403
23	Morphology evolution of Cu2â^'xS nanoparticles: from spheres to dodecahedrons. Chemical Communications, 2011, 47, 10332.	4.1	107