

Wen-Hua Li

List of Publications by Year in descending order

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Version: 2024-02-01

23
papers

1,258
citations

687363

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642732

23
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24
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docs citations

24
times ranked

2482
citing authors

#	ARTICLE	IF	CITATIONS
1	A brief review of hole transporting materials commonly used in perovskite solar cells. <i>Rare Metals</i> , 2021, 40, 2712-2729.	7.1	138
2	Synthesis of hybrid Au-Ag ₂ S-Cu ₂ -xS nanocrystals with disparate interfacial features. <i>Journal of Colloid and Interface Science</i> , 2021, 603, 11-16.	9.4	3
3	A Green Solvent Processable Wide-Bandgap Conjugated Polymer for Organic Solar Cells. <i>Solar Rrl</i> , 2020, 4, 2000547.	5.8	13
4	The preparation of plasmonic Au@SiO ₂ NPs and its application in polymer solar cells. <i>Materials Letters</i> , 2020, 268, 127599.	2.6	7
5	Crosslinked and dopant free hole transport materials for efficient and stable planar perovskite solar cells. <i>Journal of Materials Chemistry A</i> , 2019, 7, 5522-5529.	10.3	41
6	The preparation of Ag ₃ BiBr ₆ films and their preliminary use for solution processed photovoltaics. <i>SN Applied Sciences</i> , 2019, 1, 1.	2.9	5
7	Ladder-like conjugated polymers used as hole-transporting materials for high-efficiency perovskite solar cells. <i>Journal of Materials Chemistry A</i> , 2019, 7, 14473-14477.	10.3	48
8	Improved Efficiency of Perovskite Solar Cells by the Interfacial Modification of the Active Layer. <i>Nanomaterials</i> , 2019, 9, 204.	4.1	12
9	Broadband Absorption Enhancement in Polymer Solar Cells Using Highly Efficient Plasmonic Heterostructured Nanocrystals. <i>ACS Applied Materials & Interfaces</i> , 2018, 10, 30919-30924.	8.0	16
10	A simple strategy to achieve shape control of Au-Cu ₂ -xS colloidal heterostructured nanocrystals and their preliminary use in organic photovoltaics. <i>Nanoscale</i> , 2018, 10, 11745-11749.	5.6	12
11	Molecular "Flower" as the High-Mobility Hole-Transport Material for Perovskite Solar Cells. <i>ACS Applied Materials & Interfaces</i> , 2017, 9, 43855-43860.	8.0	31
12	The Influence of Fluorination on Nano-Scale Phase Separation and Photovoltaic Performance of Small Molecular/PC71BM Blends. <i>Nanomaterials</i> , 2016, 6, 80.	4.1	4
13	Enhancing the power conversion efficiency of polymer solar cells to 9.26% by a synergistic effect of fluoro and carboxylate substitution. <i>Journal of Materials Chemistry A</i> , 2016, 4, 8097-8104.	10.3	39
14	Elimination of the J-V hysteresis of planar perovskite solar cells by interfacial modification with a thermo-cleavable fullerene derivative. <i>Journal of Materials Chemistry A</i> , 2016, 4, 17649-17654.	10.3	24
15	Highly Efficient Planar Perovskite Solar Cells Via Interfacial Modification with Fullerene Derivatives. <i>Small</i> , 2016, 12, 1098-1104.	10.0	107
16	Enhancing the performance of polymer solar cells by tuning the drying process of blend films via changing side chains and using solvent additives. <i>Journal of Materials Chemistry C</i> , 2015, 3, 9670-9677.	5.5	7
17	Enhancing the Photovoltaic Performance by Tuning the Morphology of Polymer:PC ₇₁ BM Blends with a Commercially Available Nucleating Agent. <i>ACS Applied Materials & Interfaces</i> , 2015, 7, 18924-18929.	8.0	8
18	Spin-coated Ag nanoparticles onto ITO substrates for efficient improvement of polymer solar cell performance. <i>Journal of Materials Chemistry C</i> , 2015, 3, 1319-1324.	5.5	10

#	ARTICLE	IF	CITATIONS
19	Cu ₂ HgSnSe ₄ nanoparticles: synthesis and thermoelectric properties. CrystEngComm, 2013, 15, 8966.	2.6	25
20	Metal Ions To Control the Morphology of Semiconductor Nanoparticles: Copper Selenide Nanocubes. Journal of the American Chemical Society, 2013, 135, 4664-4667.	13.7	112
21	Colloidal synthesis and thermoelectric properties of Cu ₂ SnSe ₃ nanocrystals. Journal of Materials Chemistry A, 2013, 1, 1421-1426.	10.3	86
22	CuTe Nanocrystals: Shape and Size Control, Plasmonic Properties, and Use as SERS Probes and Photothermal Agents. Journal of the American Chemical Society, 2013, 135, 7098-7101.	13.7	403
23	Morphology evolution of Cu ₂ xS nanoparticles: from spheres to dodecahedrons. Chemical Communications, 2011, 47, 10332.	4.1	107