

Chun-Ran Chang

List of Publications by Year in descending order

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58
papers

6,624
citations

117625

34
h-index

144013

57
g-index

58
all docs

58
docs citations

58
times ranked

8296
citing authors

#	ARTICLE	IF	CITATIONS
1	Single Carbon Vacancy Traps Atomic Platinum for Hydrogen Evolution Catalysis. <i>Journal of the American Chemical Society</i> , 2022, 144, 2171-2178.	13.7	140
2	Design of Single-Atom and Frustrated-Lewis-Pair dual active sites for direct conversion of CH ₄ and CO ₂ to acetic acid. <i>Journal of Catalysis</i> , 2022, 408, 206-215.	6.2	18
3	Leaching of palladium atoms from small cluster models during Heck reactions – An experimental and theoretical study. <i>Catalysis Communications</i> , 2022, 165, 106441.	3.3	4
4	Perceptions on the treatment of apparent isotope effects during the analyses of reaction rate and mechanism. <i>Physical Chemistry Chemical Physics</i> , 2022, 24, 15182-15194.	2.8	10
5	Synergy of Pd atoms and oxygen vacancies on In ₂ O ₃ for methane conversion under visible light. <i>Nature Communications</i> , 2022, 13, .	12.8	105
6	Boosting the thermal stability and catalytic performance by confining Ag single atom sites over antimony-doped tin oxide via atom trapping. <i>Applied Catalysis B: Environmental</i> , 2021, 283, 119625.	20.2	36
7	Highly efficient ammonia synthesis at low temperature over a Ru-Co catalyst with dual atomically dispersed active centers. <i>Chemical Science</i> , 2021, 12, 7125-7137.	7.4	35
8	Gallium nitride catalyzed the direct hydrogenation of carbon dioxide to dimethyl ether as primary product. <i>Nature Communications</i> , 2021, 12, 2305.	12.8	45
9	Screening silica-confined single-atom catalysts for nonoxidative conversion of methane. <i>Journal of Chemical Physics</i> , 2021, 154, 174706.	3.0	6
10	Hierarchically Hollow MnO ₂ @CeO ₂ Heterostructures for NO Oxidation: Remarkably Promoted Activity and SO ₂ Tolerance. <i>ACS Catalysis</i> , 2021, 11, 10988-10996.	11.2	36
11	Silica-Confined Two-Atom Single-Cluster Catalyst for Direct Nonoxidative Conversion of Methane: A DFT Study. <i>Journal of Physical Chemistry C</i> , 2021, 125, 23212-23218.	3.1	3
12	Theoretical Insights into Dual-Metal-Site Catalysts for the Nonoxidative Coupling of Methane. <i>ACS Catalysis</i> , 2021, 11, 13149-13159.	11.2	16
13	Cooperative Catalysis by Multiple Active Centers in Nonoxidative Conversion of Methane. <i>Journal of Physical Chemistry C</i> , 2020, 124, 13656-13663.	3.1	18
14	Role of surface frustrated Lewis pairs on reduced CeO ₂ (110) in direct conversion of syngas. <i>Chinese Journal of Catalysis</i> , 2020, 41, 1906-1915.	14.0	23
15	An oxygen vacancy-rich two-dimensional Au/TiO ₂ hybrid for synergistically enhanced electrochemical N ₂ activation and reduction. <i>Journal of Materials Chemistry A</i> , 2020, 8, 6586-6596.	10.3	54
16	Single-atom Rh/N-doped carbon electrocatalyst for formic acid oxidation. <i>Nature Nanotechnology</i> , 2020, 15, 390-397.	31.5	420
17	Unravelling the Enigma of Nonoxidative Conversion of Methane on Iron Single-Atom Catalysts. <i>Angewandte Chemie</i> , 2020, 132, 18745-18749.	2.0	12
18	Unravelling the Enigma of Nonoxidative Conversion of Methane on Iron Single-Atom Catalysts. <i>Angewandte Chemie - International Edition</i> , 2020, 59, 18586-18590.	13.8	44

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19	Facet-dependent catalytic activities of Pd/rGO: Exploring dehydrogenation mechanism of dodecahydro-N-ethylcarbazole. <i>Applied Catalysis B: Environmental</i> , 2020, 266, 118658.	20.2	56
20	Gas-Phase Mechanism Study of Methane Nonoxidative Conversion by ReaxFF Method. <i>Wuli Huaxue Xuebao/ Acta Physico - Chimica Sinica</i> , 2020, .	4.9	3
21	Synergy of the catalytic activation on Ni and the CeO ₂ –TiO ₂ /Ce ₂ Ti ₂ O ₇ stoichiometric redox cycle for dramatically enhanced solar fuel production. <i>Energy and Environmental Science</i> , 2019, 12, 767-779.	30.8	90
22	Dynamic Frustrated Lewis Pairs on Ceria for Direct Nonoxidative Coupling of Methane. <i>ACS Catalysis</i> , 2019, 9, 5523-5536.	11.2	54
23	Ethylene-glycol ligand environment facilitates highly efficient hydrogen evolution of Pt/CoP through proton concentration and hydrogen spillover. <i>Energy and Environmental Science</i> , 2019, 12, 2298-2304.	30.8	227
24	Hydrogen activation enabled by the interfacial frustrated Lewis pairs on cobalt borate nanosheets. <i>Journal of Catalysis</i> , 2019, 372, 142-150.	6.2	27
25	Engineering the electronic structure of single atom Ru sites via compressive strain boosts acidic water oxidation electrocatalysis. <i>Nature Catalysis</i> , 2019, 2, 304-313.	34.4	757
26	A DFT study on the enthalpies of thermite reactions and enthalpies of formation of metal composite oxide. <i>Chemical Physics</i> , 2018, 507, 19-27.	1.9	15
27	Theoretical studies on copper-catalyzed arylation of nitrogen heterocycles from benzenediazonium acetate under ligand-free conditions. <i>Journal of Organometallic Chemistry</i> , 2018, 864, 50-57.	1.8	10
28	Understanding All-Solid Frustrated-Lewis-Pair Sites on CeO ₂ from Theoretical Perspectives. <i>ACS Catalysis</i> , 2018, 8, 546-554.	11.2	135
29	In situ encapsulation of iron(0) for solar thermochemical syngas production over iron-based perovskite material. <i>Communications Chemistry</i> , 2018, 1, .	4.5	55
30	Semi-solid and solid frustrated Lewis pair catalysts. <i>Chemical Society Reviews</i> , 2018, 47, 5541-5553.	38.1	102
31	Mechanistic investigations of Co(II)-Catalyzed C-N coupling reactions. <i>Journal of Organometallic Chemistry</i> , 2018, 868, 144-153.	1.8	16
32	Single-Walled Carbon Nanotube Induced Optimized Electron Polarization of Rhodium Nanocrystals To Develop an Interface Catalyst for Highly Efficient Electrocatalysis. <i>ACS Catalysis</i> , 2018, 8, 8092-8099.	11.2	82
33	Solid frustrated-Lewis-pair catalysts constructed by regulations on surface defects of porous nanorods of CeO ₂ . <i>Nature Communications</i> , 2017, 8, 15266.	12.8	272
34	Modulating electronic structure of CoP electrocatalysts towards enhanced hydrogen evolution by Ce chemical doping in both acidic and basic media. <i>Nano Energy</i> , 2017, 38, 290-296.	16.0	212
35	Step-Edge Assisted Direct Linear Alkane Coupling. <i>Chemistry - A European Journal</i> , 2017, 23, 6185-6189.	3.3	26
36	Trends in water-promoted oxygen dissociation on the transition metal surfaces from first principles. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 2364-2371.	2.8	39

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37	Design of N-Coordinated Dual-Metal Sites: A Stable and Active Pt-Free Catalyst for Acidic Oxygen Reduction Reaction. <i>Journal of the American Chemical Society</i> , 2017, 139, 17281-17284.	13.7	1,220
38	The promotional role of water in heterogeneous catalysis: mechanism insights from computational modeling. <i>Wiley Interdisciplinary Reviews: Computational Molecular Science</i> , 2016, 6, 679-693.	14.6	58
39	Mechanistic Insights on Ternary Ni ₂ xCo _x P for Hydrogen Evolution and Their Hybrids with Graphene as Highly Efficient and Robust Catalysts for Overall Water Splitting. <i>Advanced Functional Materials</i> , 2016, 26, 6785-6796.	14.9	500
40	Inside Cover Image, Volume 6, Issue 6. <i>Wiley Interdisciplinary Reviews: Computational Molecular Science</i> , 2016, 6, ii.	14.6	0
41	Mechanistic Insights into Propene Epoxidation with O ₂ /H ₂ O Mixture on Au ₇ /Al ₂ O ₃ : A Hydroperoxyl Pathway from ab Initio Molecular Dynamics Simulations. <i>ACS Catalysis</i> , 2016, 6, 2525-2535.	11.2	70
42	High Catalytic Activity and Chemoselectivity of Sub-nanometric Pd Clusters on Porous Nanorods of CeO ₂ for Hydrogenation of Nitroarenes. <i>Journal of the American Chemical Society</i> , 2016, 138, 2629-2637.	13.7	387
43	Theoretical Studies on the Synergetic Effects of Au-Pd Bimetallic Catalysts in the Selective Oxidation of Methanol. <i>Journal of Physical Chemistry C</i> , 2015, 119, 16072-16081.	3.1	45
44	A theoretical study on the catalytic role of water in methanol steam reforming on PdZn(111). <i>Catalysis Science and Technology</i> , 2015, 5, 2935-2944.	4.1	37
45	Visible-Light-Activated Suzuki-Miyaura Coupling Reactions of Aryl Chlorides over the Multifunctional Pd/Au/Porous Nanorods of CeO ₂ Catalysts. <i>ACS Catalysis</i> , 2015, 5, 6481-6488.	11.2	126
46	Hydrogenation of molecular oxygen to hydroperoxyl: An alternative pathway for O ₂ activation on nanogold catalysts. <i>Nano Research</i> , 2015, 8, 3737-3748.	10.4	34
47	Ultrathin rhodium nanosheets. <i>Nature Communications</i> , 2014, 5, 3093.	12.8	428
48	A Water-Promoted Mechanism of Alcohol Oxidation on a Au(111) Surface: Understanding the Catalytic Behavior of Bulk Gold. <i>ACS Catalysis</i> , 2013, 3, 1693-1699.	11.2	118
49	Nitric acid catalyzed hydrolysis of SO ₃ in the formation of sulfuric acid: A theoretical study. <i>Chemical Physics Letters</i> , 2013, 581, 26-29.	2.6	37
50	Theoretical Studies on Gas-Phase Reactions of Sulfuric Acid Catalyzed Hydrolysis of Formaldehyde and Formaldehyde with Sulfuric Acid and H ₂ SO ₄ ·H ₂ O Complex. <i>Journal of Physical Chemistry A</i> , 2013, 117, 5106-5116.	2.5	69
51	Theoretical study on the leaching of palladium in a CO atmosphere. <i>Catalysis Science and Technology</i> , 2012, 2, 2238.	4.1	18
52	Theoretical investigations of the catalytic role of water in propene epoxidation on gold nanoclusters: A hydroperoxyl-mediated pathway. <i>Nano Research</i> , 2011, 4, 131-142.	10.4	98
53	Structural Characterization and Thermal Behavior of 1-Amino-1-methylamino-2,2-dinitroethylene. <i>Chinese Journal of Chemistry</i> , 2010, 28, 583-588.	4.9	15
54	Non-isothermal Decomposition Kinetics, Specific Heat Capacity and Adiabatic Time-to-explosion of 1-Amino-1-hydrazino-2,2-dinitroethylene (AHDNE). <i>Chinese Journal of Chemistry</i> , 2009, 27, 665-671.	4.9	18

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55	Preparation, Crystal Structure and Theoretical Calculation of G(FOX-7). Chinese Journal of Chemistry, 2008, 26, 495-499.	4.9	30
56	Molecular Structure, Theoretical Calculation and Thermal Behavior of DAG(NTO). Chinese Journal of Chemistry, 2008, 26, 1549-1554.	4.9	5
57	Thermal behavior, specific heat capacity and adiabatic time-to-explosion of G(FOX-7). Journal of Hazardous Materials, 2008, 158, 333-339.	12.4	76
58	Molecular structure, theoretical calculation and thermal behavior of 2-(1,1-dinitromethylene)-1,3-diazepentane. Journal of Molecular Structure, 2008, 891, 340-345.	3.6	32