

Iso Christl

List of Publications by Year in descending order

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46
papers

2,428
citations

218381

26
h-index

214527

47
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all docs

47
docs citations

47
times ranked

2825
citing authors

#	ARTICLE	IF	CITATIONS
1	Synthetic coprecipitates of exopolysaccharides and ferrihydrite. Part I: Characterization. <i>Geochimica Et Cosmochimica Acta</i> , 2008, 72, 1111-1127.	1.6	165
2	Relating Ion Binding by Fulvic and Humic Acids to Chemical Composition and Molecular Size. 2. Metal Binding. <i>Environmental Science & Technology</i> , 2001, 35, 2512-2517.	4.6	158
3	Chemical Heterogeneity of Organic Soil Colloids Investigated by Scanning Transmission X-ray Microscopy and C-1s NEXAFS Microspectroscopy. <i>Environmental Science & Technology</i> , 2005, 39, 9094-9100.	4.6	147
4	Relating Ion Binding by Fulvic and Humic Acids to Chemical Composition and Molecular Size. 1. Proton Binding. <i>Environmental Science & Technology</i> , 2001, 35, 2505-2511.	4.6	135
5	Interaction of copper and fulvic acid at the hematite-water interface. <i>Geochimica Et Cosmochimica Acta</i> , 2001, 65, 3435-3442.	1.6	120
6	Competitive sorption of carbonate and arsenic to hematite: Combined ATR-FTIR and batch experiments. <i>Journal of Colloid and Interface Science</i> , 2012, 377, 313-321.	5.0	116
7	Competitive sorption of copper and lead at the oxide-water interface: Implications for surface site density. <i>Geochimica Et Cosmochimica Acta</i> , 1999, 63, 2929-2938.	1.6	108
8	C-1s NEXAFS Spectroscopy Reveals Chemical Fractionation of Humic Acid by Cation-Induced Coagulation. <i>Environmental Science & Technology</i> , 2007, 41, 1915-1920.	4.6	97
9	Reduction and Reoxidation of Humic Acid: Influence on Spectroscopic Properties and Proton Binding. <i>Environmental Science & Technology</i> , 2010, 44, 5787-5792.	4.6	95
10	Competitive sorption of protons and metal cations onto kaolinite: experiments and modeling. <i>Journal of Colloid and Interface Science</i> , 2005, 282, 270-282.	5.0	87
11	Effect of Humic and Fulvic Acid Concentrations and Ionic Strength on Copper and Lead Binding. <i>Environmental Science & Technology</i> , 2005, 39, 5319-5326.	4.6	86
12	Cation Binding of Antimicrobial Sulfathiazole to Leonardite Humic Acid. <i>Environmental Science & Technology</i> , 2009, 43, 6632-6638.	4.6	73
13	Polymerization of Silicate on Hematite Surfaces and Its Influence on Arsenic Sorption. <i>Environmental Science & Technology</i> , 2012, 46, 13235-13243.	4.6	71
14	Characterization of dissolved organic matter in anoxic rock extracts and in situ pore water of the Opalinus Clay. <i>Applied Geochemistry</i> , 2007, 22, 2926-2939.	1.4	70
15	Sorption of Cu and Pb to kaolinite-fulvic acid colloids: Assessment of sorbent interactions. <i>Geochimica Et Cosmochimica Acta</i> , 2005, 69, 1675-1686.	1.6	66
16	Reduction and Reoxidation of Humic Acid: Influence on Speciation of Cadmium and Silver. <i>Environmental Science & Technology</i> , 2012, 46, 8808-8816.	4.6	66
17	Copper Redox Transformation and Complexation by Reduced and Oxidized Soil Humic Acid. 1. X-ray Absorption Spectroscopy Study. <i>Environmental Science & Technology</i> , 2013, 47, 10903-10911.	4.6	66
18	Isolation and characterization of dissolved organic matter from the Callovo-Oxfordian formation. <i>Applied Geochemistry</i> , 2007, 22, 1537-1548.	1.4	63

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19	Clarithromycin and Tetracycline Binding to Soil Humic Acid in the Absence and Presence of Calcium. <i>Environmental Science & Technology</i> , 2016, 50, 9933-9942.	4.6	51
20	Aggregation Kinetics of Kaolinite-Fulvic Acid Colloids as Affected by the Sorption of Cu and Pb. <i>Environmental Science & Technology</i> , 2005, 39, 807-813.	4.6	50
21	Chemical composition of aquatic dissolved organic matter in five boreal forest catchments sampled in spring and fall seasons. <i>Biogeochemistry</i> , 2006, 80, 263-275.	1.7	49
22	Title is missing!. <i>Hyperfine Interactions</i> , 2001, 136, 73-95.	0.2	41
23	Climate-sensitive ecosystem carbon dynamics along the soil chronosequence of the Damma glacier forefield, Switzerland. <i>Global Change Biology</i> , 2012, 18, 1941-1955.	4.2	38
24	Ionic strength- and pH-dependence of calcium binding by terrestrial humic acids. <i>Environmental Chemistry</i> , 2012, 9, 89.	0.7	36
25	Copper Redox Transformation and Complexation by Reduced and Oxidized Soil Humic Acid. 2. Potentiometric Titrations and Dialysis Cell Experiments. <i>Environmental Science & Technology</i> , 2013, 47, 10912-10921.	4.6	35
26	Nitrogen and phosphorus availability at early stages of soil development in the Damma glacier forefield, Switzerland; implications for establishment of N ₂ -fixing plants. <i>Plant and Soil</i> , 2016, 404, 251-261.	1.8	29
27	Interactions of ferrous iron with clay mineral surfaces during sorption and subsequent oxidation. <i>Environmental Sciences: Processes and Impacts</i> , 2020, 22, 1355-1367.	1.7	25
28	Reductive solubilization of arsenic in a mining-impacted river floodplain: Influence of soil properties and temperature. <i>Environmental Pollution</i> , 2017, 231, 722-731.	3.7	24
29	Aerobic Reduction of Chromium(VI) by <i>Pseudomonas corrugata</i> : Influence of Metabolism and Fate of Reduced Chromium. <i>Geomicrobiology Journal</i> , 2012, 29, 173-185.	1.0	22
30	Zinc solubility in tropical paddy soils: A multi-chemical extraction technique study. <i>Geoderma</i> , 2017, 301, 1-10.	2.3	22
31	Soil-to-plant transfer of arsenic and phosphorus along a contamination gradient in the mining-impacted Ogosta River floodplain. <i>Science of the Total Environment</i> , 2016, 572, 742-754.	3.9	21
32	Two-year and multi-site field trials to evaluate soil amendments for controlling cadmium accumulation in rice grain. <i>Environmental Pollution</i> , 2021, 289, 117918.	3.7	20
33	Copper complexation of methanobactin isolated from <i>Methylosinus trichosporium</i> OB3b: pH-dependent speciation and modeling. <i>Journal of Inorganic Biochemistry</i> , 2012, 116, 55-62.	1.5	19
34	Competitive ligand exchange between Cu-humic acid complexes and methanobactin. <i>Geobiology</i> , 2013, 11, 44-54.	1.1	18
35	Proton and Trivalent Metal Cation Binding by Dissolved Organic Matter in the Opalinus Clay and the Callovo-Oxfordian Formation. <i>Environmental Science & Technology</i> , 2008, 42, 5985-5991.	4.6	17
36	Biochar as possible long-term soil amendment for phytostabilisation of TE-contaminated soils. <i>Environmental Science and Pollution Research</i> , 2016, 23, 17449-17458.	2.7	17

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37	Sulfur amendments to soil decrease inorganic arsenic accumulation in rice grain under flooded and nonflooded conditions: Insights from temporal dynamics of porewater chemistry and solid-phase arsenic solubility. <i>Science of the Total Environment</i> , 2021, 779, 146352.	3.9	16
38	The Effect of Aeration on Mn(II) Sorbed to Clay Minerals and Its Impact on Cd Retention. <i>Environmental Science & Technology</i> , 2021, 55, 1650-1658.	4.6	16
39	Surface precipitation of Mn ²⁺ on clay minerals enhances Cd ²⁺ sorption under anoxic conditions. <i>Environmental Sciences: Processes and Impacts</i> , 2020, 22, 1654-1665.	1.7	15
40	Isolation and purification of Cu-free methanobactin from <i>Methylosinus trichosporium</i> OB3b. <i>Geochemical Transactions</i> , 2011, 12, 2.	1.8	13
41	Effect of NOM on copper sulfide nanoparticle growth, stability, and oxidative dissolution. <i>Environmental Science: Nano</i> , 2020, 7, 1163-1178.	2.2	11
42	Effects of natural organic matter (NOM), metal-to-sulfide ratio and Mn ²⁺ on cadmium sulfide nanoparticle growth and colloidal stability. <i>Environmental Science: Nano</i> , 2020, 7, 3385-3404.	2.2	7
43	Magnesium binding by terrestrial humic acids. <i>Environmental Chemistry</i> , 2018, 15, 317.	0.7	6
44	Copper mobilisation from Cu sulphide minerals by methanobactin: Effect of pH, oxygen and natural organic matter. <i>Geobiology</i> , 2022, 20, 690-706.	1.1	5
45	Exploring Key Soil Parameters Relevant to Arsenic and Cadmium Accumulation in Rice Grain in Southern China. <i>Soil Systems</i> , 2022, 6, 36.	1.0	4
46	Effect of extreme metal(loid) concentrations on prokaryotic community structure in floodplain soils contaminated with mine waste. <i>Applied Soil Ecology</i> , 2019, 144, 182-195.	2.1	2