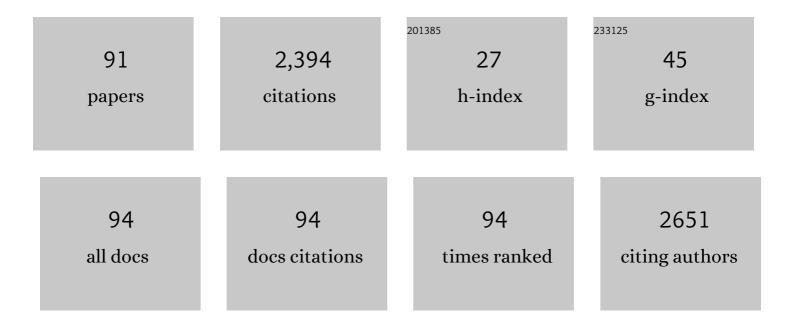
Brian A Powell

List of Publications by Year in descending order

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RDIAN & POWELL

#	Article	IF	CITATIONS
1	Adaptive Interactions between Zinc Oxide Nanoparticles and <i>Chlorella</i> sp Environmental Science & Technology, 2012, 46, 12178-12185.	4.6	139
2	Pu(V)O2+Adsorption and Reduction by Synthetic Magnetite (Fe3O4). Environmental Science & Technology, 2004, 38, 6016-6024.	4.6	129
3	Formation of a Protein Corona on Silver Nanoparticles Mediates Cellular Toxicity via Scavenger Receptors. Toxicological Sciences, 2015, 143, 136-146.	1.4	125
4	Pu(V)O2+Adsorption and Reduction by Synthetic Hematite and Goethite. Environmental Science & Technology, 2005, 39, 2107-2114.	4.6	115
5	Stabilization of Plutonium Nano-Colloids by Epitaxial Distortion on Mineral Surfaces. Environmental Science & Technology, 2011, 45, 2698-2703.	4.6	90
6	Surface Complexation Modeling of Eu(III) and U(VI) Interactions with Graphene Oxide. Environmental Science & Technology, 2016, 50, 1821-1827.	4.6	82
7	Influence of Oxidation States on Plutonium Mobility during Long-Term Transport through an Unsaturated Subsurface Environment. Environmental Science & Technology, 2004, 38, 5053-5058.	4.6	80
8	Plant litter chemistry alters the content and composition of organic carbon associated with soil mineral and aggregate fractions in invaded ecosystems. Global Change Biology, 2017, 23, 4002-4018.	4.2	77
9	Thermodynamics and Electronic Properties of Heterometallic Multinuclear Actinide-Containing Metal–Organic Frameworks with "Structural Memory― Journal of the American Chemical Society, 2019, 141, 11628-11640.	6.6	71
10	Plutonium Oxidation and Subsequent Reduction by Mn(IV) Minerals in Yucca Mountain Tuff. Environmental Science & Technology, 2006, 40, 3508-3514.	4.6	70
11	Np(V) and Pu(V) Ion Exchange and Surface-Mediated Reduction Mechanisms on Montmorillonite. Environmental Science & Technology, 2012, 46, 2692-2698.	4.6	70
12	Elevated concentrations of primordial radionuclides in sediments from the Reedy River and surrounding creeks in Simpsonville, South Carolina. Journal of Environmental Radioactivity, 2007, 94, 121-128.	0.9	61
13	Pu(V) and Pu(IV) Sorption to Montmorillonite. Environmental Science & Technology, 2013, 47, 5146-5153.	4.6	58
14	Evaluation of a Radioiodine Plume Increasing in Concentration at the Savannah River Site. Environmental Science & Technology, 2011, 45, 489-495.	4.6	56
15	Quantitative Measurement of Ligand Exchange on Iron Oxides via Radiolabeled Oleic Acid. Langmuir, 2014, 30, 10918-10925.	1.6	56
16	Influence of Sources on Plutonium Mobility and Oxidation State Transformations in Vadose Zone Sediments. Environmental Science & Chrology, 2007, 41, 7417-7423.	4.6	55
17	A self-consistent model describing the thermodynamics of Eu(III) adsorption onto hematite. Geochimica Et Cosmochimica Acta, 2013, 122, 430-447.	1.6	52
18	Mobilization of actinides by dissolved organic compounds at the Nevada Test Site. Applied Geochemistry, 2011, 26, 308-318.	1.4	51

#	Article	IF	CITATIONS
19	Quantitative Measurement of Ligand Exchange with Small-Molecule Ligands on Iron Oxide Nanoparticles via Radioanalytical Techniques. Langmuir, 2016, 32, 13716-13727.	1.6	49
20	Influence of pH on Plutonium Desorption/Solubilization from Sediment. Environmental Science & Technology, 2006, 40, 5937-5942.	4.6	38
21	Phosphorus Stress-Induced Changes in Plant Root Exudation Could Potentially Facilitate Uranium Mobilization from Stable Mineral Forms. Environmental Science & Technology, 2018, 52, 7652-7662.	4.6	38
22	Kinetics of neptunium(V) sorption and desorption on goethite: An experimental and modeling study. Geochimica Et Cosmochimica Acta, 2011, 75, 6584-6599.	1.6	33
23	Observed Changes in the Mechanism and Rates of Pu(V) Reduction on Hematite As a Function of Total Plutonium Concentration. Environmental Science & Technology, 2014, 48, 9255-9262.	4.6	32
24	Effect of Fulvic Acid Surface Coatings on Plutonium Sorption and Desorption Kinetics on Goethite. Environmental Science & Technology, 2015, 49, 2776-2785.	4.6	32
25	Influence of iron redox transformations on plutonium sorption to sediments. Radiochimica Acta, 2010, 98, 685-692.	0.5	31
26	Eleven-Year Field Study of Pu Migration from Pu III, IV, and VI Sources. Environmental Science & Technology, 2006, 40, 443-448.	4.6	30
27	A Pro ₁₀₆ to Ala Substitution is Associated with Resistance to Glyphosate in Annual Bluegrass (<i>Poa annua</i>). Weed Science, 2015, 63, 613-622.	0.8	28
28	Removal capacity and chemical speciation of groundwater iodide (Iâ^') and iodate (IO3â^') sequestered by organoclays and granular activated carbon. Journal of Environmental Radioactivity, 2018, 192, 505-512.	0.9	28
29	Examination of the effect of alpha radiolysis on plutonium(V) sorption to quartz using multiple plutonium isotopes. Journal of Colloid and Interface Science, 2013, 403, 105-112.	5.0	27
30	Complexation of UVIwith 1-Hydroxyethane-1,1-diphosphonic Acid in Acidic to Basic Solutions. Inorganic Chemistry, 2007, 46, 2870-2876.	1.9	26
31	Enthalpy of Uranium Adsorption onto Hematite. Environmental Science & Technology, 2020, 54, 15004-15012.	4.6	26
32	Geochemical controls of iodine uptake and transport in Savannah River Site subsurface sediments. Applied Geochemistry, 2014, 45, 105-113.	1.4	22
33	Pu(V) transport through Savannah River Site soils - an evaluation of a conceptual model of surface- mediated reduction to Pu (IV). Journal of Environmental Radioactivity, 2014, 131, 47-56.	0.9	20
34	Effect of Natural Organic Matter on Plutonium Sorption to Goethite. Environmental Science & Technology, 2017, 51, 699-708.	4.6	20
35	Compounds of Hexavalent Uranium and Dibutylphosphate in Nitric Acid Systems. Solvent Extraction and Ion Exchange, 2003, 21, 347-368.	0.8	19
36	Mobility of radionuclides in soil/groundwater system: Comparing the influence of EDTA and four of its degradation products. Environmental Pollution, 2010, 158, 3077-3084.	3.7	17

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37	PAMAM dendrimer for mitigating humic foulant. RSC Advances, 2012, 2, 7997.	1.7	17
38	Chemical stabilization of chromate in blast furnace slag mixed cementitious materials. Chemosphere, 2015, 138, 247-252.	4.2	17
39	Dissolution and Vertical Transport of Uranium from Stable Mineral Forms by Plants as Influenced by the Co-occurrence of Uranium with Phosphorus. Environmental Science & Technology, 2020, 54, 6602-6609.	4.6	17
40	Evaluation of Surface Sorption Processes Using Spectral Induced Polarization and a ²² Na Tracer. Environmental Science & Technology, 2015, 49, 9866-9873.	4.6	16
41	Effects of Titanium Doping in Titanomagnetite on Neptunium Sorption and Speciation. Environmental Science & Technology, 2016, 50, 1853-1858.	4.6	15
42	ANOMALOUSLY HIGH LEVELS OF URANIUM AND OTHER NATURALLY OCCURRING RADIONUCLIDES IN PRIVATE WELLS IN THE PIEDMONT REGION OF SOUTH CAROLINA. Health Physics, 2005, 88, 248-252.	0.3	14
43	Increase in the reduction potential of uranyl upon interaction with graphene oxide surfaces. Physical Chemistry Chemical Physics, 2018, 20, 1752-1760.	1.3	14
44	Plutonium environmental chemistry: mechanisms for the surface-mediated reduction of Pu(<scp>v</scp> / <scp>vi</scp>). Environmental Sciences: Processes and Impacts, 2018, 20, 1306-1322.	1.7	13
45	Integration of ecosystem science into radioecology: A consensus perspective. Science of the Total Environment, 2020, 740, 140031.	3.9	13
46	Rapid quantification of TBP and TBP degradation product ratios by FTIR-ATR. Journal of Radioanalytical and Nuclear Chemistry, 2013, 296, 859-868.	0.7	12
47	Influence of humic acid on plutonium sorption to gibbsite: Determination of Pu-humic acid complexation constants and ternary sorption studies. Radiochimica Acta, 2014, 102, 629-643.	0.5	12
48	Experimental evidence for ternary colloid-facilitated transport of Th(IV) with hematite (α-Fe2O3) colloids and Suwannee River fulvic acid. Journal of Environmental Radioactivity, 2016, 165, 168-181.	0.9	12
49	The effect of post-synthesis aging on the ligand exchange activity of iron oxide nanoparticles. Journal of Colloid and Interface Science, 2018, 511, 374-382.	5.0	12
50	Plutonium binding affinity to sediments increases with contact time. Chemical Geology, 2019, 505, 100-107.	1.4	12
51	Anion-Exchange Fibers for Improved Sample Loading in Ultra-Trace Analysis of Plutonium by Thermal Ionization Mass Spectrometry. Analytical Chemistry, 2017, 89, 8638-8642.	3.2	11
52	Linear Free Energy Relationship for Actinide Sorption to Graphene Oxide. ACS Applied Materials & Interfaces, 2018, 10, 32086-32092.	4.0	11
53	Stability constant determination of sulfur and selenium amino acids with Cu(II) and Fe(II). Journal of Inorganic Biochemistry, 2019, 195, 20-30.	1.5	11
54	High effectiveness of pure polydopamine in extraction of uranium and plutonium from groundwater and seawater. RSC Advances, 2019, 9, 30052-30063.	1.7	11

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55	THEORETICAL ORGANICALLY BOUND TRITIUM DOSE ESTIMATES. Health Physics, 2004, 86, 183-186.	0.3	10
56	A novel technique for the rapid determination of tributyl phosphate degradation from alkaline hydrolysis in aqueous and organic phases using FTIR–ATR and verification of this technique by gas chromatography. Journal of Radioanalytical and Nuclear Chemistry, 2016, 307, 1891-1899.	0.7	10
57	Uptake of plutonium on a novel thin film for use in spectrometry. Journal of Radioanalytical and Nuclear Chemistry, 2016, 307, 2333-2338.	0.7	10
58	High-Resolution 4D Preclinical Single-Photon Emission Computed Tomography/X-ray Computed Tomography Imaging of Technetium Transport within a Heterogeneous Porous Media. Environmental Science & Technology, 2017, 51, 2864-2870.	4.6	10
59	An ab initio study of the adsorption of Eu3+, Pu3+, Am3+, and Cm3+ hydroxide complexes on hematite (001) surface: Role of magnetism on adsorption. Surface Science, 2017, 664, 120-128.	0.8	10
60	Nuclear Magnetic Resonance Spectroscopy of Aqueous Plutonium(IV) Desferrioxamine B Complexes. European Journal of Inorganic Chemistry, 2014, 2014, 3312-3321.	1.0	9
61	Observations of surface-mediated reduction of Pu(VI) to Pu(IV) on hematite nanoparticles by ATR FT-IR. Radiochimica Acta, 2015, 103, 553-563.	0.5	9
62	Functionalized Polymer Thin Films for Plutonium Capture and Isotopic Screening from Aqueous Sources. Analytical Chemistry, 2020, 92, 5214-5221.	3.2	9
63	Effect of bundling on the π plasmon energy in sub-nanometer single wall carbon nanotubes. Carbon, 2011, 49, 3803-3807.	5.4	8
64	Long-Term Radiostrontium Interactions and Transport through Sediment. Environmental Science & Technology, 2014, 48, 8919-8925.	4.6	8
65	A Novel Method for Tracer Concentration Plutonium(V) Solution Preparation. Analytical Chemistry, 2016, 88, 4196-4199.	3.2	8
66	Uranium Attenuated by a Wetland 50 Years after Release into a Stream. ACS Earth and Space Chemistry, 2020, 4, 1360-1366.	1.2	8
67	Comparison of the surface ion density of silica gel evaluated via spectral induced polarization versus acid–base titration. Journal of Applied Geophysics, 2016, 135, 427-435.	0.9	7
68	Rhenium filament oxidation: Effect on TIMS performance and the roles of carburization and humidity. Talanta, 2017, 168, 183-187.	2.9	7
69	Physical transformations of iron oxide and silver nanoparticles from an intermediate scale field transport study. Journal of Nanoparticle Research, 2014, 16, 1.	0.8	6
70	Effect of equilibration time on Pu desorption from goethite. Radiochimica Acta, 2015, 103, 695-705.	0.5	6
71	Anion-exchange polymer filament coating for ultra-trace isotopic analysis of plutonium by thermal ionization mass spectrometry. Talanta, 2018, 189, 502-508.	2.9	6
72	Mobility of Aqueous and Colloidal Neptunium Species in Field Lysimeter Experiments. Environmental Science & Technology, 2018, 52, 1963-1970.	4.6	5

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73	Interdisciplinary Round-Robin Test on Molecular Spectroscopy of the U(VI) Acetate System. ACS Omega, 2019, 4, 8167-8177.	1.6	5
74	Sorption and desorption of radiocesium by muscovite separated from the Georgia kaolin. Journal of Environmental Radioactivity, 2020, 211, 106074.	0.9	5
75	Experimental measurements and numerical simulations of the transport and retention of nanocrystal CdSe/ZnS quantum dots in saturated porous media: effects of pH, organic ligand, and natural organic matter. Environmental Science and Pollution Research, 2021, 28, 8050-8073.	2.7	5
76	Ambient aging of rhenium filaments used in thermal ionization mass spectrometry: Growth of oxo-rhenium crystallites and anti-aging strategies. Heliyon, 2017, 3, e00232.	1.4	4
77	Preferential flow systems amended with biogeochemical components: imaging of a two-dimensional study. Hydrology and Earth System Sciences, 2018, 22, 2487-2509.	1.9	4
78	The Uptake and Translocation of 99Tc, 133Cs, 237Np, and 238U Into Andropogon Virginicus With Consideration of Plant Life Stage. Health Physics, 2018, 115, 550-560.	0.3	4
79	Response to Comment on "Enthalpy of Uranium Adsorption onto Hematite― Environmental Science & Technology, 2021, 55, 3444-3446.	4.6	4
80	Influence of the goethite (α-FeOOH) surface on the stability of distorted PuO ₂ and PuO _{2–x} phases. Radiochimica Acta, 2016, 104, 821-841.	0.5	3
81	Postemergence Control and Glyphosate Tolerance of Doveweed (<i>Murdannia nudiflora</i>). Weed Technology, 2017, 31, 582-589.	0.4	3
82	Impact of Natural Organic Matter on Plutonium Vadose Zone Migration from an NH ₄ Pu(V)O ₂ CO ₃ (s) Source. Environmental Science & Technology, 2020, 54, 2688-2697.	4.6	3
83	Mechanisms and kinetics of citrate-promoted dissolution of a uranyl phosphate mineral. Geochimica Et Cosmochimica Acta, 2022, 318, 247-262.	1.6	3
84	Uranium partitioning from contaminated wetland soil to aqueous and suspended iron-floc phases: Implications of dynamic hydrologic conditions on contaminant release. Geochimica Et Cosmochimica Acta, 2022, 318, 292-304.	1.6	3
85	Neptunium(V) sorption to vadose zone sediments: Reversible, not readily reducible, and predictable based on Fe-oxide content. Chemical Geology, 2018, 481, 53-64.	1.4	2
86	Grain boundary facilitated dissolution of nanocrystalline NpO2(s) from legacy waste processing. Environmental Science: Nano, 2020, 7, 2293-2301.	2.2	2
87	Examination of Uranium(VI) Leaching during Ligand Promoted Dissolution of Waste Tank Sludge Surrogates. Separation Science and Technology, 2008, 43, 3798-3812.	1.3	1
88	Effect of 1-Hydroxyethane-1,1-diphosphonic Acid (HEDPA) on Partitioning of Np and Pu to Synthetic Boehmite. Separation Science and Technology, 2010, 45, 721-731.	1.3	1
89	Plutonium Transport in Soil and Plants. Geophysical Monograph Series, 0, , 181-208.	0.1	1
90	One-dimensional Spatial Distributions of Gamma-ray Emitting Contaminants in Field Lysimeters Using a Collimated Gamma-ray Spectroscopy System. Health Physics, 2018, 114, 532-536.	0.3	1

#	Article	IF	CITATIONS
91	Effect of calcination temperature on neptunium dioxide microstructure and dissolution. Environmental Science: Nano, 2020, 7, 3869-3876.	2.2	1