

Kohei Kusada

List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/7027363/publications.pdf>

Version: 2024-02-01

70
papers

2,649
citations

236925

25
h-index

189892

50
g-index

76
all docs

76
docs citations

76
times ranked

2596
citing authors

#	ARTICLE	IF	CITATIONS
1	Continuous-flow syntheses of alloy nanoparticles. <i>Materials Horizons</i> , 2022, 9, 547-558.	12.2	17
2	Phase Control of Solid-Solution Nanoparticles beyond the Phase Diagram for Enhanced Catalytic Properties. <i>ACS Materials Au</i> , 2022, 2, 110-116.	6.0	4
3	Carbon-supported WO ₃ –Ru-based catalysts for the selective hydrogenolysis of glycerol to 1,2-propanediol. <i>Catalysis Science and Technology</i> , 2022, 12, 259-272.	4.1	15
4	Noble-Metal High-Entropy-Alloy Nanoparticles: Atomic-Level Insight into the Electronic Structure. <i>Journal of the American Chemical Society</i> , 2022, 144, 3365-3369.	13.7	94
5	Crystal Structure Control of Binary and Ternary Solid-Solution Alloy Nanoparticles with a Face-Centered Cubic or Hexagonal Close-Packed Phase. <i>Journal of the American Chemical Society</i> , 2022, 144, 4224-4232.	13.7	40
6	Quantitative Characterization of the Thermally Driven Alloying State in Ternary Ir–Pd–Ru Nanoparticles. <i>ACS Nano</i> , 2022, 16, 1612-1624.	14.6	5
7	Compositional dependence of structures and hydrogen evolution reaction activity of platinum-group-metal quinary RuRhPdIrPt alloy nanoparticles. <i>Chemical Communications</i> , 2022, 58, 6421-6424.	4.1	5
8	The Effect of Ru Precursor and Support on the Hydrogenation of Aromatic Aldehydes/Ketones to Alcohols. <i>ChemCatChem</i> , 2022, 14, .	3.7	1
9	Continuous-Flow Reactor Synthesis for Homogeneous 1 nm-Sized Extremely Small High-Entropy Alloy Nanoparticles. <i>Journal of the American Chemical Society</i> , 2022, 144, 11525-11529.	13.7	60
10	Boosting reverse water-gas shift reaction activity of Pt nanoparticles through light doping of W. <i>Journal of Materials Chemistry A</i> , 2021, 9, 15613-15617.	10.3	17
11	Efficient overall water splitting in acid with anisotropic metal nanosheets. <i>Nature Communications</i> , 2021, 12, 1145.	12.8	124
12	Enhanced Hydrogenation Catalytic Activity of Ruthenium Nanoparticles by Solid-Solution Alloying with Molybdenum. <i>European Journal of Inorganic Chemistry</i> , 2021, 2021, 1186-1189.	2.0	3
13	Highly Stable and Active Solid-Solution Alloy Three-Way Catalyst by Utilizing Configurational Entropy Effect. <i>Advanced Materials</i> , 2021, 33, e2005206.	21.0	22
14	Phase Control of Noble Monometallic and Alloy Nanomaterials by Chemical Reduction Methods. <i>ChemPlusChem</i> , 2021, 86, 504-519.	2.8	9
15	Cu–Pd–B Alloy Nanoparticles Synthesized by External Boron Doping Method. <i>Chemistry Letters</i> , 2021, 50, 611-614.	1.3	1
16	First Observation of Superconductivity in Molybdenum–Ruthenium–Carbon Alloy Nanoparticles. <i>Chemistry Letters</i> , 2021, 50, 596-598.	1.3	1
17	Investigation of microstructure and hydrogen absorption properties of bulk immiscible AgRh alloy nanoparticles. <i>Journal of Alloys and Compounds</i> , 2021, 869, 159268.	5.5	2
18	Investigation of Local Structure and Enhanced Thermal Stability of Ir-Doped PdRu Nanoparticles for Three-Way Catalytic Applications. <i>Journal of Physical Chemistry C</i> , 2021, 125, 20583-20591.	3.1	3

#	ARTICLE	IF	CITATIONS
19	First synthesis of air-stable NiZn homogeneous alloy nanoparticles through chemical reduction. <i>Materials Advances</i> , 2021, 2, 684-687.	5.4	1
20	Ni@onion-like carbon and Co@amorphous carbon: control of carbon structures by metal ion species in MOFs. <i>Chemical Communications</i> , 2021, 57, 5897-5900.	4.1	4
21	Nonequilibrium Flow-Synthesis of Solid-Solution Alloy Nanoparticles: From Immiscible Binary to High-Entropy Alloys. <i>Journal of Physical Chemistry C</i> , 2021, 125, 458-463.	3.1	18
22	Quantum Size Effect Probed by NMR Measurements. <i>Creative Economy</i> , 2021, , 215-230.	0.1	0
23	Total x-ray scattering setup for crystalline particles at SPring-8 BL15XU NIMS beamline. <i>Review of Scientific Instruments</i> , 2021, 92, 113905.	1.3	0
24	Discovery of face-centred cubic Os nanoparticles. <i>Chemical Communications</i> , 2020, 56, 372-374.	4.1	20
25	Rational Synthesis for a Noble Metal Carbide. <i>Journal of the American Chemical Society</i> , 2020, 142, 1247-1253.	13.7	15
26	New Aspects of Platinum Group Metal-Based Solid-Solution Alloy Nanoparticles: Binary to High-Entropy Alloys. <i>Chemistry - A European Journal</i> , 2020, 26, 5105-5130.	3.3	41
27	Significant Enhancement of Hydrogen Evolution Reaction Activity by Negatively Charged Pt through Light Doping of W. <i>Journal of the American Chemical Society</i> , 2020, 142, 17250-17254.	13.7	103
28	On the electronic structure and hydrogen evolution reaction activity of platinum group metal-based high-entropy-alloy nanoparticles. <i>Chemical Science</i> , 2020, 11, 12731-12736.	7.4	142
29	Platinum-Group-Metal High-Entropy-Alloy Nanoparticles. <i>Journal of the American Chemical Society</i> , 2020, 142, 13833-13838.	13.7	223
30	Crystalline to amorphous transformation in solid-solution alloy nanoparticles induced by boron doping. <i>Chemical Communications</i> , 2020, 56, 12941-12944.	4.1	8
31	Statistical Evaluation of the Solid-Solution State in Ternary Nanoalloys. <i>Journal of Physical Chemistry C</i> , 2020, 124, 21843-21852.	3.1	2
32	Synthesis of Mo and Ru solid-solution alloy NPs and their hydrogen evolution reaction activity. <i>Chemical Communications</i> , 2020, 56, 14475-14478.	4.1	23
33	Frontispiece: New Aspects of Platinum Group Metal-Based Solid-Solution Alloy Nanoparticles: Binary to High-Entropy Alloys. <i>Chemistry - A European Journal</i> , 2020, 26, .	3.3	0
34	Hydrogen absorption and desorption on Rh nanoparticles revealed by <i>in situ</i> dispersive X-ray absorption fine structure spectroscopy. <i>RSC Advances</i> , 2020, 10, 19751-19758.	3.6	0
35	Chemoselective hydrogenation of heteroarenes and arenes by Pd-Ru-PVP under mild conditions. <i>RSC Advances</i> , 2020, 10, 44191-44195.	3.6	11
36	Magnetic-Field Dependence of Novel Gap Behavior Related to the Quantum-Size Effect. <i>Journal of the Physical Society of Japan</i> , 2020, 89, 095002.	1.6	2

#	ARTICLE	IF	CITATIONS
37	Investigation of selective chemisorption of fcc and hcp Ru nanoparticles using X-ray photoelectron spectroscopy analysis. <i>Journal of Catalysis</i> , 2019, 380, 247-253.	6.2	5
38	Observation of the Formation Processes of Hexagonal Close-packed and Face-centered Cubic Ru Nanoparticles. <i>Chemistry Letters</i> , 2019, 48, 1062-1064.	1.3	11
39	Correlation between the electronic/local structure and CO-oxidation activity of Pd _x Ru _{1-x} alloy nanoparticles. <i>Nanoscale Advances</i> , 2019, 1, 546-553.	4.6	12
40	Emergence of high ORR activity through controlling local density-of-states by alloying immiscible Au and Ir. <i>Chemical Science</i> , 2019, 10, 652-656.	7.4	50
41	Solid-solution alloy nanoparticles of a combination of immiscible Au and Ru with a large gap of reduction potential and their enhanced oxygen evolution reaction performance. <i>Chemical Science</i> , 2019, 10, 5133-5137.	7.4	48
42	Hydrogen in Palladium and Storage Properties of Related Nanomaterials: Size, Shape, Alloying, and Metal-Organic Framework Coating Effects. <i>ChemPhysChem</i> , 2019, 20, 1158-1176.	2.1	80
43	Synchrotron-radiation-based Mössbauer absorption spectroscopy with high resonant energy nuclides. <i>Hyperfine Interactions</i> , 2019, 240, 1.	0.5	2
44	Solid-solution Alloy Nanoparticles of the Immiscible Iridium-Copper System with a Wide Composition Range for Enhanced Electrocatalytic Applications. <i>Angewandte Chemie</i> , 2018, 130, 4595-4599.	2.0	13
45	Selective control of fcc and hcp crystal structures in Au-Ru solid-solution alloy nanoparticles. <i>Nature Communications</i> , 2018, 9, 510.	12.8	90
46	Solid-solution Alloy Nanoparticles of the Immiscible Iridium-Copper System with a Wide Composition Range for Enhanced Electrocatalytic Applications. <i>Angewandte Chemie - International Edition</i> , 2018, 57, 4505-4509.	13.8	86
47	Size effects on rhodium nanoparticles related to hydrogen-storage capability. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 15183-15191.	2.8	11
48	Crystal Structure-dependent Thermal Stability and Catalytic Performance of AuRu ₃ Solid-solution Alloy Nanoparticles. <i>Chemistry Letters</i> , 2018, 47, 559-561.	1.3	8
49	Electronic Structure Evolution with Composition Alteration of Rh _x Cu _y Alloy Nanoparticles. <i>Scientific Reports</i> , 2017, 7, 41264.	3.3	12
50	First-Principles Calculation, Synthesis, and Catalytic Properties of Rh-Cu Alloy Nanoparticles. <i>Chemistry - A European Journal</i> , 2017, 23, 57-60.	3.3	26
51	Stacking fault density and bond orientational order of fcc ruthenium nanoparticles. <i>Applied Physics Letters</i> , 2017, 111, 253101.	3.3	8
52	Facile Synthesis of Size-controlled Rh Nanoparticles via Microwave-assisted Alcohol Reduction and Their Catalysis of CO Oxidation. <i>Chemistry Letters</i> , 2017, 46, 1254-1257.	1.3	16
53	A Route for Phase Control in Metal Nanoparticles: A Potential Strategy to Create Advanced Materials. <i>Advanced Materials</i> , 2016, 28, 1129-1142.	21.0	72
54	Metal Nanoparticles: A Route for Phase Control in Metal Nanoparticles: A Potential Strategy to Create Advanced Materials (Adv. Mater. 6/2016). <i>Advanced Materials</i> , 2016, 28, 978-978.	21.0	0

#	ARTICLE	IF	CITATIONS
55	Recent progress in the structure control of Pd–Ru bimetallic nanomaterials. <i>Science and Technology of Advanced Materials</i> , 2016, 17, 583-596.	6.1	49
56	Size dependence of structural parameters in fcc and hcp Ru nanoparticles, revealed by Rietveld refinement analysis of high-energy X-ray diffraction data. <i>Scientific Reports</i> , 2016, 6, 31400.	3.3	50
57	Origin of the catalytic activity of face-centered-cubic ruthenium nanoparticles determined from an atomic-scale structure. <i>Physical Chemistry Chemical Physics</i> , 2016, 18, 30622-30629.	2.8	39
58	A Synthetic Pseudo-Rh: NO _x Reduction Activity and Electronic Structure of Pd–Ru Solid-solution Alloy Nanoparticles. <i>Scientific Reports</i> , 2016, 6, 28265.	3.3	44
59	Structural studies of metal nanoparticles using high-energy x-ray diffraction. <i>AIP Conference Proceedings</i> , 2016, , .	0.4	5
60	Dual Lewis Acidic/Basic Pd _{0.5} Ru _{0.5} –Poly(N-vinyl-2-pyrrolidone) Alloyed Nanoparticle: Outstanding Catalytic Activity and Selectivity in Suzuki–Miyaura Cross-Coupling Reaction. <i>ChemCatChem</i> , 2015, 7, 3887-3894.	3.7	25
61	Creation of Novel Solid-Solution Alloy Nanoparticles on the Basis of Density-of-States Engineering by Interelement Fusion. <i>Accounts of Chemical Research</i> , 2015, 48, 1551-1559.	15.6	107
62	Hybrid materials of Ni NP@MOF prepared by a simple synthetic method. <i>Chemical Communications</i> , 2015, 51, 12463-12466.	4.1	70
63	The valence band structure of Ag _x Rh _{1-x} alloy nanoparticles. <i>Applied Physics Letters</i> , 2014, 105, .	3.3	27
64	Systematic Study of the Hydrogen Storage Properties and the CO-oxidizing Abilities of Solid Solution Alloy Nanoparticles in an Immiscible Pd–Ru System. <i>Springer Theses</i> , 2014, , 29-57.	0.1	0
65	Solid Solution Alloy Nanoparticles of Immiscible Pd and Ru Elements Neighboring on Rh: Changeover of the Thermodynamic Behavior for Hydrogen Storage and Enhanced CO-Oxidizing Ability. <i>Journal of the American Chemical Society</i> , 2014, 136, 1864-1871.	13.7	229
66	Changeover of the Thermodynamic Behavior for Hydrogen Storage in Rh with Increasing Nanoparticle Size. <i>Springer Theses</i> , 2014, , 69-76.	0.1	0
67	Discovery of Face-Centered-Cubic Ruthenium Nanoparticles: Facile Size-Controlled Synthesis Using the Chemical Reduction Method. <i>Journal of the American Chemical Society</i> , 2013, 135, 5493-5496.	13.7	290
68	Changeover of the Thermodynamic Behavior for Hydrogen Storage in Rh with Increasing Nanoparticle Size. <i>Chemistry Letters</i> , 2013, 42, 55-56.	1.3	10
69	Hydrogen-Storage Properties of Solid-Solution Alloys of Immiscible Neighboring Elements with Pd. <i>Journal of the American Chemical Society</i> , 2010, 132, 15896-15898.	13.7	112
70	Enhancing Hydrogen Storage Capacity of Pd Nanoparticles by Sandwiching between Inorganic Nanosheets. <i>Zeitschrift Fur Anorganische Und Allgemeine Chemie</i> , 0, , .	1.2	0