

# Xuemei Ren

## List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/6892649/publications.pdf>

Version: 2024-02-01

51  
papers

4,642  
citations

201385

27  
h-index

182168

51  
g-index

52  
all docs

52  
docs citations

52  
times ranked

5472  
citing authors

#	ARTICLE	IF	CITATIONS
1	Few-Layered Graphene Oxide Nanosheets As Superior Sorbents for Heavy Metal Ion Pollution Management. <i>Environmental Science &amp; Technology</i> , 2011, 45, 10454-10462.	4.6	1,594
2	Removal of Pb(II) ions from aqueous solutions on few-layered graphene oxide nanosheets. <i>Dalton Transactions</i> , 2011, 40, 10945.	1.6	488
3	Graphene oxide-iron oxide and reduced graphene oxide-iron oxide hybrid materials for the removal of organic and inorganic pollutants. <i>RSC Advances</i> , 2012, 2, 8821.	1.7	300
4	Comparative study of graphene oxide, activated carbon and carbon nanotubes as adsorbents for copper decontamination. <i>Dalton Transactions</i> , 2013, 42, 5266.	1.6	188
5	Efficient removal of arsenate by versatile magnetic graphene oxide composites. <i>RSC Advances</i> , 2012, 2, 12400.	1.7	169
6	Polyaniline Multiwalled Carbon Nanotube Magnetic Composite Prepared by Plasma-Induced Graft Technique and Its Application for Removal of Aniline and Phenol. <i>Journal of Physical Chemistry C</i> , 2010, 114, 21524-21530.	1.5	161
7	Impact of Al <sub>2</sub> O <sub>3</sub> on the Aggregation and Deposition of Graphene Oxide. <i>Environmental Science &amp; Technology</i> , 2014, 48, 5493-5500.	4.6	144
8	Highly active MnO <sub>2</sub> nanosheet synthesis from graphene oxide templates and their application in efficient oxidative degradation of methylene blue. <i>RSC Advances</i> , 2013, 3, 12909.	1.7	89
9	New Insight into GO, Cadmium(II), Phosphate Interaction and Its Role in GO Colloidal Behavior. <i>Environmental Science &amp; Technology</i> , 2016, 50, 9361-9369.	4.6	85
10	X-ray absorption fine structure study of enhanced sequestration of U(VI) and Se(IV) by montmorillonite decorated with zero-valent iron nanoparticles. <i>Environmental Science: Nano</i> , 2016, 3, 1460-1472.	2.2	85
11	Impact of graphene oxide on the antibacterial activity of antibiotics against bacteria. <i>Environmental Science: Nano</i> , 2017, 4, 1016-1024.	2.2	84
12	Coupling g-C <sub>3</sub> N <sub>4</sub> nanosheets with metal-organic frameworks as 2D/3D composite for the synergetic removal of uranyl ions from aqueous solution. <i>Journal of Colloid and Interface Science</i> , 2019, 550, 117-127.	5.0	84
13	Immobilization of uranium by biomaterial stabilized FeS nanoparticles: Effects of stabilizer and enrichment mechanism. <i>Journal of Hazardous Materials</i> , 2016, 302, 1-9.	6.5	79
14	Adsorption and co-adsorption of graphene oxide and Ni(II) on iron oxides: A spectroscopic and microscopic investigation. <i>Environmental Pollution</i> , 2018, 233, 125-131.	3.7	79
15	Graphene analogues in aquatic environments and porous media: dispersion, aggregation, deposition and transformation. <i>Environmental Science: Nano</i> , 2018, 5, 1298-1340.	2.2	68
16	Insights into key factors controlling GO stability in natural surface waters. <i>Journal of Hazardous Materials</i> , 2017, 335, 56-65.	6.5	64
17	Highly Cation Permselective Metal-Organic Framework Membranes with Leaf-Like Morphology. <i>ChemSusChem</i> , 2019, 12, 2593-2597.	3.6	61
18	Graphene oxide interactions with co-existing heavy metal cations: adsorption, colloidal properties and joint toxicity. <i>Environmental Science: Nano</i> , 2018, 5, 362-371.	2.2	54

#	ARTICLE	IF	CITATIONS
19	Mutual effects of copper and phosphate on their interaction with $\hat{I}^3$ -Al <sub>2</sub> O <sub>3</sub> : Combined batch macroscopic experiments with DFT calculations. <i>Journal of Hazardous Materials</i> , 2012, 237-238, 199-208.	6.5	53
20	Macroscopic and spectroscopic insights into the mutual interaction of graphene oxide, Cu(II), and Mg/Al layered double hydroxides. <i>Chemical Engineering Journal</i> , 2017, 313, 527-534.	6.6	51
21	Exploring the Aggregation Mechanism of Graphene Oxide in the Presence of Radioactive Elements: Experimental and Theoretical Studies. <i>Environmental Science &amp; Technology</i> , 2018, 52, 12208-12215.	4.6	49
22	Plasma Induced Multiwalled Carbon Nanotube Grafted with 2-vinylpyridine for Preconcentration of Pb(II) from Aqueous Solutions. <i>Plasma Processes and Polymers</i> , 2011, 8, 589-598.	1.6	41
23	Three dimensional flower-like magnetic polyethyleneimine@MoS <sub>2</sub> composites for highly efficient removal of Cr(VI) and Pb(II) ions. <i>Journal of Colloid and Interface Science</i> , 2020, 580, 550-560.	5.0	40
24	Theoretical investigation of uranyl ion adsorption on hydroxylated $\hat{I}^3$ -Al <sub>2</sub> O <sub>3</sub> surfaces. <i>RSC Advances</i> , 2013, 3, 19551.	1.7	37
25	Reductive immobilization of uranium by PAAM-Fe <sub>3</sub> O <sub>4</sub> magnetic composites. <i>Environmental Science: Water Research and Technology</i> , 2015, 1, 169-176.	1.2	36
26	Design of Chitosan-Grafted Carbon Nanotubes: Evaluation of How the -OH Functional Group Affects Cs <sup>+</sup> Adsorption. <i>Marine Drugs</i> , 2015, 13, 3116-3131.	2.2	32
27	Polyamidoxime functionalized with phosphate groups by plasma technique for effective U(VI) adsorption. <i>Journal of Industrial and Engineering Chemistry</i> , 2018, 67, 380-387.	2.9	27
28	Environmental fate and risk of ultraviolet- and visible-light-transformed graphene oxide: A comparative study. <i>Environmental Pollution</i> , 2019, 251, 821-829.	3.7	27
29	Highly selective enrichment of radioactive cesium from solution by using zinc hexacyanoferrate(III)-functionalized magnetic bentonite. <i>Journal of Colloid and Interface Science</i> , 2020, 580, 171-179.	5.0	27
30	Controlled synthesized natroalunite microtubes applied for cadmium(II) and phosphate co-removal. <i>Journal of Hazardous Materials</i> , 2016, 314, 249-259.	6.5	26
31	Efficient removal of Cd(II) by core-shell Fe <sub>3</sub> O <sub>4</sub> @polydopamine microspheres from aqueous solution. <i>Journal of Molecular Liquids</i> , 2019, 295, 111724.	2.3	26
32	A comprehensive review on emerging natural and tailored materials for chromium-contaminated water treatment and environmental remediation. <i>Journal of Environmental Chemical Engineering</i> , 2022, 10, 107325.	3.3	26
33	Co-sequestration of Zn(II) and phosphate by $\hat{I}^3$ -Al <sub>2</sub> O <sub>3</sub> : From macroscopic to microscopic investigation. <i>Journal of Hazardous Materials</i> , 2015, 297, 134-145.	6.5	22
34	Colloidal properties and stability of UV-transformed graphene oxide in aqueous solutions: The role of disorder degree. <i>Journal of Hazardous Materials</i> , 2020, 382, 121097.	6.5	22
35	The influence of dissolved Si on Ni precipitate formation at the kaolinite water interface: Kinetics, DRS and EXAFS analysis. <i>Chemosphere</i> , 2017, 173, 135-142.	4.2	21
36	Solvent-free engineering of FeO/Fe <sub>3</sub> C nanoparticles encased in nitrogen-doped carbon nanoshell materials for highly efficient removal of uranyl ions from acidic solution. <i>Journal of Colloid and Interface Science</i> , 2020, 575, 16-23.	5.0	21

#	ARTICLE	IF	CITATIONS
37	Investigation of radionuclide $^{60}\text{Co}(\text{II})$ binding to $\text{TiO}_2$ by batch technique, surface complexation model and DFT calculations. <i>Science China Chemistry</i> , 2012, 55, 1752-1759.	4.2	17
38	Colloidal Behaviors of Two-Dimensional Titanium Carbide in Natural Surface Waters: The Role of Solution Chemistry. <i>Environmental Science &amp; Technology</i> , 2020, 54, 3353-3362.	4.6	17
39	Transformation details of poly(acrylonitrile) to poly(amidoxime) during the amidoximation process. <i>RSC Advances</i> , 2021, 11, 1909-1915.	1.7	17
40	Poly(amidoxime) functionalized $\text{MoS}_2$ for efficient adsorption of uranium(VI) in aqueous solutions. <i>Journal of Radioanalytical and Nuclear Chemistry</i> , 2019, 319, 379-386.	0.7	16
41	Retention of $\text{Pb}(\text{II})$ by a Low-Cost Magnetic Composite Prepared by Environmentally-Friendly Plasma Technique. <i>Separation Science and Technology</i> , 2013, 48, 1211-1219.	1.3	14
42	Sequestration and speciation of $\text{Eu}(\text{III})$ on gamma alumina: role of temperature and contact order. <i>Environmental Sciences: Processes and Impacts</i> , 2015, 17, 1904-1914.	1.7	14
43	Nanocomposites of polyaniline functionalized graphene oxide: synthesis and application as a novel platform for removal of $\text{Cd}(\text{II})$ , $\text{Eu}(\text{III})$ , $\text{Th}(\text{IV})$ and $\text{U}(\text{VI})$ in water. <i>Journal of Radioanalytical and Nuclear Chemistry</i> , 2018, 315, 509-522.	0.7	13
44	Influence of pH, soil humic acid, ionic strength and temperature on sorption of $\text{U}(\text{VI})$ onto attapulgite. <i>Journal of Radioanalytical and Nuclear Chemistry</i> , 2018, 316, 981-991.	0.7	13
45	Insight into UV-induced simultaneous photocatalytic degradation of $\text{Ti}_3\text{C}_2\text{T}_x$ MXene and reduction of $\text{U}(\text{VI})$ . <i>Journal of Hazardous Materials</i> , 2022, 430, 128377.	6.5	13
46	A carboxymethyl cellulose modified magnetic bentonite composite for efficient enrichment of radionuclides. <i>RSC Advances</i> , 2016, 6, 65136-65145.	1.7	12
47	Characterization of $\text{Fe}(\text{III})$ -saturated montmorillonite and evaluation its sorption behavior for $\text{U}(\text{VI})$ . <i>Radiochimica Acta</i> , 2016, 104, 481-490.	0.5	12
48	Macroscopic and microscopic insight into the mutual effects of europium(III) and phosphate on their interaction with graphene oxide. <i>RSC Advances</i> , 2016, 6, 85046-85057.	1.7	10
49	Facile Synthesis and Characterization of Chrysotile Nanotubes and Their Application for Lead(II) Removal from Aqueous Solution. <i>Separation Science and Technology</i> , 2015, 50, 700-709.	1.3	5
50	Kinetic and thermodynamic studies on the interaction of europium(III) and phosphate with $\text{Al}_2\text{O}_3$ . <i>Journal of Radioanalytical and Nuclear Chemistry</i> , 2017, 311, 395-408.	0.7	3
51	Corrigendum to: Effect of humic acid, fulvic acid, pH, ionic strength and temperature on $^{63}\text{Ni}(\text{II})$ sorption to $\text{MnO}_2$ . <i>Radiochimica Acta</i> , 2020, 108, 591-591.	0.5	0