

Inti Zumeta-DubÃ©

List of Publications by Year in descending order

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37
papers

880
citations

516710

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477307

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38
all docs

38
docs citations

38
times ranked

1509
citing authors

#	ARTICLE	IF	CITATIONS
1	Magnetic properties of Pd@Ag nanoalloys obtained by liquid-assisted mechanochemical pathway. Journal of Physics and Chemistry of Solids, 2022, 161, 110427.	4.0	3
2	Unraveling amazing structural features of a highly efficient α -oxo-Co/phosphate catalyst for water oxidation. Applied Catalysis B: Environmental, 2021, 282, 119549.	20.2	6
3	Strong texture tuning along different crystalline directions in glass-supported CeO ₂ thin films by ultrasonic spray pyrolysis. Scientific Reports, 2021, 11, 2006.	3.3	4
4	New Understanding on an Old Compound: Insights on the Origin of Chain Sequence Defects and Their Impact on the Electronic Structure of AuCN. European Journal of Inorganic Chemistry, 2021, 2021, 3742-3751.	2.0	1
5	Nanostructured CuO film grown from solution by preferential microwave heating of the conducting glass substrate. Materials Letters, 2020, 270, 127687.	2.6	2
6	Production of Methanol from Aqueous CO ₂ by Using Co ₃ O ₄ Nanostructures as Photocatalysts. Journal of Nanomaterials, 2019, 2019, 1-10.	2.7	10
7	Preparation and characterization of (CuInTe ₂) _{1-x} (TaTe) _x solid solutions (0 < x < 1). Journal of Alloys and Compounds, 2018, 747, 176-188.	5.5	2
8	Mechanochemically obtained Pd@Ag nanoalloys. Structural considerations and catalytic activity. Materialia, 2018, 4, 166-174.	2.7	16
9	Facile synthesis of rod-shaped bismuth sulfide@graphene oxide (Bi ₂ S ₃ @GO) composite. Materials Chemistry and Physics, 2018, 219, 376-389.	4.0	16
10	Transformation of Bismuth and Bi ₂ O ₃ Nanoparticles into (BiO) ₂ CO ₃ and (BiO) ₄ (OH) ₂ CO ₃ by Capturing CO ₂ : The Role of Halloysite Nanotubes and Sunlight on the Crystal Shape and Size. Crystal Growth and Design, 2018, 18, 4334-4346.	3.0	20
11	Bismuth Oxide Nanoparticles Partially Substituted with Eu ^{III} , Mn ^{IV} , and Si ^{IV} : Structural, Spectroscopic, and Optical Findings. Inorganic Chemistry, 2017, 56, 3394-3403.	4.0	22
12	Degradation of bis- <i>p</i> -nitrophenyl phosphate using zero-valent iron nanoparticles. Journal of Physics: Conference Series, 2017, 838, 012034.	0.4	1
13	Can Silver Be Alloyed with Bismuth on Nanoscale? An Optical and Structural Approach. Journal of Physical Chemistry C, 2017, 121, 940-949.	3.1	10
14	Combined experimental/theoretical investigation on the interactions of Diuron with a urea-formaldehyde matrix: implications for its use as an intelligent pesticide. Chemical Papers, 2017, 71, 2495-2503.	2.2	3
15	First principle calculations on the adsorption of molecular H ₂ in the largest pore of Co[Fe(CN) ₅ NO] and Ni[Fe(CN) ₅ NO] metal nitroprussides. Effect of the charged cavities on the adsorption and H ₂ -host interactions. Computational Materials Science, 2016, 114, 102-111.	3.0	3
16	Cu ₃ TaSe ₄ and Cu ₃ NbSe ₄ : X-ray diffraction, differential thermal analysis, optical absorption and Raman scattering. Journal of Alloys and Compounds, 2016, 658, 749-756.	5.5	21
17	Kinetic studies of the release profiles of antiepileptic drug released from a nanostructured TiO ₂ matrix.. Journal of Advances in Chemistry, 2016, 12, 4365-4373.	0.1	1
18	Thermoelectric transport properties of CuFeInTe ₃ . Journal of Alloys and Compounds, 2015, 651, 490-496.	5.5	9

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19	First Order Raman Scattering in Bulk Bi ₂ S ₃ and Quantum Dots: Reconsidering Controversial Interpretations. <i>Journal of Physical Chemistry C</i> , 2014, 118, 30244-30252.	3.1	66
20	TiO ₂ Sensitization with Bi ₂ S ₃ Quantum Dots: The Inconvenience of Sodium Ions in the Deposition Procedure. <i>Journal of Physical Chemistry C</i> , 2014, 118, 11495-11504.	3.1	72
21	Easy Synthesis of High-Purity BiFeO ₃ Nanoparticles: New Insights Derived from the Structural, Optical, and Magnetic Characterization. <i>Inorganic Chemistry</i> , 2013, 52, 10306-10317.	4.0	105
22	Bismuth oxide aqueous colloidal nanoparticles inhibit <i>Candida albicans</i> growth and biofilm formation. <i>International Journal of Nanomedicine</i> , 2013, 8, 1645.	6.7	59
23	Stabilization of Strong Quantum Confined Colloidal Bismuth Nanoparticles, One-Pot Synthesized at Room Conditions. <i>Journal of Physical Chemistry C</i> , 2012, 116, 14717-14727.	3.1	52
24	Synthesis of TiO ₂ Nanoparticles with Narrow Size Distribution and Their Evaluation in the Photocatalytic Oxidative Degradation of Bis(4-nitrophenyl) Phosphate. <i>Journal of Physical Chemistry C</i> , 2010, 114, 11381-11389.	3.1	14
25	TiO ₂ films obtained by microwave-activated chemical-bath deposition used to improve TiO ₂ -conducting glass contact. <i>Solar Energy Materials and Solar Cells</i> , 2009, 93, 1728-1732.	6.2	32
26	TiO ₂ –CuO three-dimensional heterostructure obtained using short time photochemical deposition of copper oxide inside a porous nanocrystalline TiO ₂ layer. <i>Microporous and Mesoporous Materials</i> , 2008, 109, 560-566.	4.4	10
27	New low-temperature preparation method of the TiO ₂ porous photoelectrode for dye-sensitized solar cells using UV irradiation. <i>Journal of Photochemistry and Photobiology A: Chemistry</i> , 2005, 175, 165-171.	3.9	106
28	Nanocrystalline TiO ₂ photosensitized with natural polymers with enhanced efficiency from 400 to 600nm. <i>Solar Energy Materials and Solar Cells</i> , 2005, 85, 359-369.	6.2	38
29	Preparation of photoelectrodes with spectral response in the visible without applied bias based on photochemically deposited copper oxide inside a porous titanium dioxide film. <i>Thin Solid Films</i> , 2005, 489, 50-55.	1.8	22
30	Photovoltaic behavior of structures based on nanocrystalline semiconductor oxides. <i>Physica Status Solidi (B): Basic Research</i> , 2005, 242, 1807-1811.	1.5	2
31	Two-layer TiO ₂ nanostructured photoelectrode with underlying film obtained by microwave-activated chemical bath deposition (MW-CBD). <i>Semiconductor Science and Technology</i> , 2004, 19, L52-L55.	2.0	10
32	The role of conducting-oxide-substrate type and morphology in TiO ₂ films grown by microwave chemical bath deposition (MW-CBD) and their photovoltaic characteristics. <i>Journal of Crystal Growth</i> , 2004, 262, 366-374.	1.5	20
33	Rutherford backscattering spectrometry analysis of TiO ₂ thin films. <i>Materials Characterization</i> , 2003, 50, 155-160.	4.4	5
34	Comparative study of nanocrystalline TiO ₂ photoelectrodes based on characteristics of nanopowder used. <i>Solar Energy Materials and Solar Cells</i> , 2003, 76, 15-24.	6.2	39
35	Role of the conducting layer substrate on TiO ₂ nucleation when using microwave activated chemical bath deposition. <i>Semiconductor Science and Technology</i> , 2002, 17, 1218-1222.	2.0	15
36	Structural analysis of TiO ₂ films grown using microwave-activated chemical bath deposition. <i>Thin Solid Films</i> , 2002, 419, 65-68.	1.8	17

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37	TiO ₂ thin film deposition from solution using microwave heating. Thin Solid Films, 2000, 365, 12-18.	1.8	46