

Gero Frisch

List of Publications by Year in descending order

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Version: 2024-02-01

32
papers

2,210
citations

471371

17
h-index

454834

30
g-index

36
all docs

36
docs citations

36
times ranked

2058
citing authors

#	ARTICLE	IF	CITATIONS
1	Processing of metals and metal oxides using ionic liquids. <i>Green Chemistry</i> , 2011, 13, 471.	4.6	309
2	Electrodeposition of copper composites from deep eutectic solvents based on choline chloride. <i>Physical Chemistry Chemical Physics</i> , 2009, 11, 4269.	1.3	302
3	Electroplating Using Ionic Liquids. <i>Annual Review of Materials Research</i> , 2013, 43, 335-358.	4.3	228
4	The effect of additives on zinc electrodeposition from deep eutectic solvents. <i>Electrochimica Acta</i> , 2011, 56, 5272-5279.	2.6	186
5	EXAFS Study into the Speciation of Metal Salts Dissolved in Ionic Liquids and Deep Eutectic Solvents. <i>Inorganic Chemistry</i> , 2014, 53, 6280-6288.	1.9	170
6	The electrodeposition of silver composites using deep eutectic solvents. <i>Physical Chemistry Chemical Physics</i> , 2012, 14, 2443.	1.3	151
7	Double layer effects on metal nucleation in deep eutectic solvents. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 10224.	1.3	134
8	Speciation, physical and electrolytic properties of eutectic mixtures based on $\text{CrCl}_3 \cdot 6\text{H}_2\text{O}$ and urea. <i>Physical Chemistry Chemical Physics</i> , 2014, 16, 9047.	1.3	123
9	Anodic dissolution of metals in ionic liquids. <i>Progress in Natural Science: Materials International</i> , 2015, 25, 595-602.	1.8	105
10	Metal complexation in ionic liquids. <i>Annual Reports on the Progress of Chemistry Section A</i> , 2008, 104, 21.	0.8	72
11	Electrocatalytic recovery of elements from complex mixtures using deep eutectic solvents. <i>Green Chemistry</i> , 2015, 17, 2172-2179.	4.6	70
12	Leaching and Selective Extraction of Indium and Tin from Zinc Flue Dust Using an Oxalic Acid-Based Deep Eutectic Solvent. <i>ACS Sustainable Chemistry and Engineering</i> , 2019, 7, 5300-5308.	3.2	58
13	Direct extraction of copper from copper sulfide minerals using deep eutectic solvents. <i>Green Chemistry</i> , 2019, 21, 6502-6512.	4.6	57
14	Ionic liquids form ideal solutions. <i>Chemical Communications</i> , 2011, 47, 11876.	2.2	52
15	Voltammetric and spectroscopic study of ferrocene and hexacyanoferrate and the suitability of their redox couples as internal standards in ionic liquids. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 28841-28852.	1.3	39
16	Ligand exchange in ionic systems and its effect on silver nucleation and growth. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 17314.	1.3	29
17	Electrochemistry and speciation of Au^{+} in a deep eutectic solvent: growth and morphology of galvanic immersion coatings. <i>Physical Chemistry Chemical Physics</i> , 2015, 17, 30540-30550.	1.3	20
18	Complexation Equilibria of Indium in Aqueous Chloride, Sulfate and Nitrate Solutions: An Electrochemical Investigation. <i>Journal of Solution Chemistry</i> , 2017, 46, 1928-1940.	0.6	17

#	ARTICLE	IF	CITATIONS
19	Ionic liquid, glass or crystalline solid? Structures and thermal behaviour of $(C_4mim)_2CuCl_3$. Dalton Transactions, 2016, 45, 3327-3333.	1.6	15
20	Alkaline Metal Stannide-Stannates: "Double Salts"™ with Zintl Sn_{44}^{4-} and Stannate SnO_3^{4-} Anions. Zeitschrift Fur Anorganische Und Allgemeine Chemie, 2003, 629, 1661-1672.	0.6	14
21	Investigating the dissolution of iron sulfide and arsenide minerals in deep eutectic solvents. Hydrometallurgy, 2020, 198, 105511.	1.8	13
22	$A_5Fe_3O_6$ (A = Rb, Cs), $Cs[FeO_2]$ und $Cs_8[Fe_2O_7]$: Neue Oxoferrate der schweren Alkalimetalle / $A_5Fe_3O_6$ (A = Rb, Cs), $Cs[FeO_2]$ and $Cs_8[Fe_2O_7]$: New Oxoferrates of the Heavy Alkaline Metals. Zeitschrift Fur Naturforschung - Section B Journal of Chemical Sciences, 2004, 59, 771-781.	0.3	11
23	Neue Orthoferrate von Rubidium und Caesium: $?, ?-Cs_5[FeIII O_4]$ und $A_7I[FeIVO_4][FeVO_4]$ (AI = Rb, Cs). Zeitschrift Fur Anorganische Und Allgemeine Chemie, 2005, 631, 507-517.	0.6	8
24	A Low-Cost Al-Graphite Battery with Urea and Acetamide-Based Electrolytes. ChemElectroChem, 2021, 8, 1988-1992.	1.7	7
25	Synthesis and stability of single-phase chalcopyrite " a potential reference material for key investigations in chemistry and metallurgical engineering. RSC Advances, 2021, 11, 3153-3161.	1.7	6
26	$Cs_6[Fe_2O_6]$ and $Rb_4[Fe_2O_5]$: New oxoferrates(III) with edge sharing FeO_4 tetrahedra. Zeitschrift Fur Kristallographie - Crystalline Materials, 2005, 220, 135-141.	0.4	5
27	A particle-based approach to predict the success and selectivity of leaching processes using ethaline - Comparison of simulated and experimental results. Hydrometallurgy, 2022, 211, 105869.	1.8	3
28	Quantifying indium with ion chromatography in hydro- and biohydrometallurgical leaching solutions. Journal of Separation Science, 2019, 42, 2517-2522.	1.3	2
29	Symmetry relationships between the crystal structures of chalcopyrite and its derivatives " a systematic approach to inform XRD analysis of Cu_2FeS_5 phases. Zeitschrift Fur Anorganische Und Allgemeine Chemie, 0, , .	0.6	1
30	Alkaline Metal Stannide-Stannates: "œDouble Salts"™with Zintl Sn_{4-4} and Stannate SnO_{4-3} Anions.. ChemInform, 2003, 34, no.	0.1	0
31	New Orthoferrates of Rubidium and Cesium: $?, ?-Cs_5[FeIII O_4]$ and $A_5I[FeIVO_4][FeVO_4]$ (AI: Rb, Cs).. ChemInform, 2005, 36, no.	0.1	0
32	A Low-Cost Al-Graphite Battery with Urea and Acetamide-Based Electrolytes. ChemElectroChem, 2021, 8, 1928-1928.	1.7	0