

Jiang Wu

List of Publications by Year in descending order

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56
papers

1,860
citations

201674

27
h-index

265206

42
g-index

57
all docs

57
docs citations

57
times ranked

823
citing authors

#	ARTICLE	IF	CITATIONS
1	Novel carbon-based sorbents for elemental mercury removal from gas streams: A review. <i>Chemical Engineering Journal</i> , 2020, 391, 123514.	12.7	112
2	Removal of elemental mercury from simulated flue gas by ZSM-5 modified with Mn-Fe mixed oxides. <i>Chemical Engineering Journal</i> , 2019, 375, 121946.	12.7	104
3	Surface defect engineering of Fe-doped Bi ₇ O ₉ I ₃ microflowers for ameliorating charge-carrier separation and molecular oxygen activation. <i>Applied Catalysis B: Environmental</i> , 2021, 284, 119727.	20.2	104
4	Constructing 3D Bi/Bi ₄ O ₅ I ₂ microspheres with rich oxygen vacancies by one-pot solvothermal method for enhancing photocatalytic activity on mercury removal. <i>Chemical Engineering Journal</i> , 2021, 425, 131599.	12.7	93
5	CeO ₂ @ MnO _x /ZSM-5 sorbents for H ₂ S removal at high temperature. <i>Chemical Engineering Journal</i> , 2016, 284, 862-871.	12.7	77
6	Spherical-shaped CuS modified carbon nitride nanosheet for efficient capture of elemental mercury from flue gas at low temperature. <i>Journal of Hazardous Materials</i> , 2021, 415, 125692.	12.4	64
7	Removal of elemental mercury by Ce-Mn co-modified activated carbon catalyst. <i>Catalysis Communications</i> , 2017, 93, 62-66.	3.3	59
8	Effect of Ce and La on the activity of CuO/ZSM-5 and MnO _x /ZSM-5 composites for elemental mercury removal at low temperature. <i>Fuel</i> , 2017, 194, 115-122.	6.4	58
9	High performance of Fe nanoparticles/carbon aerogel sorbents for H ₂ S Removal. <i>Chemical Engineering Journal</i> , 2017, 313, 1051-1060.	12.7	55
10	A review of sorbents for high-temperature hydrogen sulfide removal from hot coal gas. <i>Environmental Chemistry Letters</i> , 2019, 17, 259-276.	16.2	53
11	Elemental Mercury Capture from Simulated Flue Gas by Graphite-Phase Carbon Nitride. <i>Energy & Fuels</i> , 2020, 34, 6851-6861.	5.1	51
12	Bimetallic sulfides ZnIn ₂ S ₄ modified g-C ₃ N ₄ adsorbent with wide temperature range for rapid elemental mercury uptake from coal-fired flue gas. <i>Chemical Engineering Journal</i> , 2021, 426, 131343.	12.7	51
13	Elemental Mercury Removal by MnO ₂ Nanoparticle-Decorated Carbon Nitride Nanosheet. <i>Energy & Fuels</i> , 2019, 33, 3089-3097.	5.1	50
14	Sorbents for hydrogen sulfide capture from biogas at low temperature: a review. <i>Environmental Chemistry Letters</i> , 2020, 18, 113-128.	16.2	49
15	Nanosized ZnIn ₂ S ₄ supported on facet-engineered CeO ₂ nanorods for efficient gaseous elemental mercury immobilization. <i>Journal of Hazardous Materials</i> , 2021, 419, 126436.	12.4	49
16	Defective molybdenum disulfide nanosheet for elemental mercury capture in simulated flue gas. <i>Journal of the Energy Institute</i> , 2021, 94, 120-128.	5.3	44
17	Magnetically recyclable CoS-modified graphitic carbon nitride-based materials for efficient immobilization of gaseous elemental mercury. <i>Fuel</i> , 2022, 326, 125117.	6.4	43
18	Gaseous Mercury Capture by Copper-Activated Nanoporous Carbon Nitride. <i>Energy & Fuels</i> , 2018, 32, 8287-8295.	5.1	42

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19	Gaseous mercury removal by graphene-like carbon nitride impregnated with ammonium bromide. <i>Fuel</i> , 2020, 280, 118635.	6.4	41
20	Photocatalytic oxidation removal of elemental mercury from flue gas. A review. <i>Environmental Chemistry Letters</i> , 2020, 18, 417-431.	16.2	40
21	Co ₃ O ₄ /g-C ₃ N ₄ Hybrids for Gas-Phase Hg ⁰ Removal at Low Temperature. <i>Processes</i> , 2019, 7, 279.	2.8	38
22	Graphitic carbon nitride for elemental mercury capture. <i>Materials Letters</i> , 2018, 227, 308-310.	2.6	36
23	CeO ₂ -La ₂ O ₃ /ZSM-5 sorbents for high-temperature H ₂ S removal. <i>Korean Journal of Chemical Engineering</i> , 2016, 33, 1837-1845.	2.7	35
24	La ₂ CuO ₄ /ZSM-5 sorbents for high-temperature desulphurization. <i>Fuel</i> , 2016, 177, 251-259.	6.4	34
25	CuO/g-C ₃ N ₄ nanocomposite for elemental mercury capture at low temperature. <i>Journal of Nanoparticle Research</i> , 2018, 20, 1.	1.9	33
26	Synthesis and photocatalytic performance of CuO-CeO ₂ /Graphene Oxide. <i>Materials Letters</i> , 2016, 185, 503-506.	2.6	30
27	Perovskite LaMnO ₃ /ZSM-5 composites for H ₂ S reactive adsorption at high temperature. <i>Adsorption</i> , 2016, 22, 327-334.	3.0	29
28	BiOI/O ₃ /graphene interfacial heterojunction for enhancing gaseous heavy metal removal. <i>Materials Research Bulletin</i> , 2020, 122, 110620.	5.2	29
29	Cu Nanoparticles Inlaid Mesoporous Carbon Aerogels as a High Performance Desulfurizer. <i>Environmental Science & Technology</i> , 2016, 50, 5370-5378.	10.0	27
30	Elemental mercury capture from industrial gas emissions using sulfides and selenides: a review. <i>Environmental Chemistry Letters</i> , 2021, 19, 1395-1411.	16.2	26
31	Spherical In ₂ S ₃ anchored on g-C ₃ N ₄ nanosheets for efficient elemental mercury removal in the wide temperature range. <i>Chemical Engineering Journal</i> , 2022, 430, 132857.	12.7	25
32	CuO-CeO ₂ /ZSM-5 composites for reactive adsorption of hydrogen sulphide at high temperature. <i>Canadian Journal of Chemical Engineering</i> , 2016, 94, 2276-2281.	1.7	22
33	Insights into the Mechanism of Elemental Mercury Adsorption on Graphitic Carbon Nitride: A Density Functional Theory Study. <i>Energy & Fuels</i> , 2021, 35, 9322-9331.	5.1	21
34	Copper sulfide microsphere for Hg ⁰ capture from flue gas at low temperature. <i>Materials Today Communications</i> , 2020, 25, 101188.	1.9	20
35	Copper Sulfide-Loaded Boron Nitride Nanosheets for Elemental Mercury Removal from Simulated Flue Gas. <i>Energy & Fuels</i> , 2021, 35, 2234-2242.	5.1	19
36	Molten salt synthesis of WS ₂ and MoS ₂ nanosheets toward efficient gaseous elemental mercury capture. <i>Science of the Total Environment</i> , 2022, 824, 153934.	8.0	19

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37	Fractal characterization of graphene oxide nanosheet. <i>Materials Letters</i> , 2018, 220, 40-43.	2.6	15
38	Salt-Assisted Synthesis of Rod-Like Bi ₂ S ₃ Single Crystals for Gas-Phase Elemental Mercury Removal. <i>Energy & Fuels</i> , 2022, 36, 2591-2599.	5.1	15
39	Graphitic Carbon Nitride for Gaseous Mercury Emission Control: A Review. <i>Energy & Fuels</i> , 2022, 36, 4297-4313.	5.1	15
40	Elemental mercury capture by graphene-analogous carbon nitride anchored with copper sulfide. <i>Chemical Engineering Journal</i> , 2021, 417, 127931.	12.7	14
41	Gaseous Elemental Mercury Capture by Magnetic FeS ₂ Nanorods Synthesized via a Molten Salt Method. <i>ACS Applied Nano Materials</i> , 2022, 5, 2626-2635.	5.0	14
42	CuS-Doped Ti ₃ C ₂ MXene Nanosheets for Highly Efficient Adsorption of Elemental Mercury in Flue Gas. <i>Energy & Fuels</i> , 2022, 36, 2503-2514.	5.1	13
43	Gaseous mercury removal using biogenic porous silica modified with potassium bromide. <i>Journal of the Energy Institute</i> , 2021, 99, 161-169.	5.3	12
44	Molten salt shielded preparation of rice straw biochars doped by copper sulfide for elemental mercury capture. <i>Journal of the Energy Institute</i> , 2022, 102, 176-183.	5.3	12
45	Coordinative sulfur site over flower-structured MoS ₂ for efficient elemental mercury uptake from coal-fired flue gas. <i>Chemical Engineering Journal</i> , 2022, 434, 134649.	12.7	10
46	HONEYCOMB-LIKE MESOPOROUS g-C ₃ N ₄ FOR ELEMENTAL MERCURY REMOVAL FROM SIMULATED FLUE GAS. <i>Surface Review and Letters</i> , 2020, 27, 2050017.	1.1	8
47	Gaseous mercury capture using iodine-modified carbon nitride derived from guanidine hydrochloride. <i>Chemical Physics Letters</i> , 2022, 793, 139171.	2.6	8
48	ZnS-modified carbon nitride nanosheet with enhanced performance of elemental Hg removal: An experimental and density functional theory study. <i>Korean Journal of Chemical Engineering</i> , 2022, 39, 1641-1650.	2.7	8
49	KINETIC BEHAVIOR OF ELEMENTAL MERCURY SORPTION ON CERIUM- AND LANTHANUM-BASED COMPOSITE OXIDES. <i>Surface Review and Letters</i> , 2019, 26, 1850141.	1.1	7
50	Tuning sulfur vacancies in CoS ₂ via a molten salt approach for promoted mercury vapor adsorption. <i>Chemical Engineering Journal</i> , 2022, 450, 137956.	12.7	7
51	TEXTURE AND STRUCTURE VARIATION OF PEROVSKITE LaFeO ₃ /ZSM-5 DURING HIGH-TEMPERATURE DESULFURIZATION. <i>Surface Review and Letters</i> , 2020, 27, 1950151.	1.1	6
52	Rod-Shaped Bi ₂ S ₃ Supported on Flaky Carbon Nitride for Effective Removal of Elemental Mercury in Flue Gas. <i>Energy & Fuels</i> , 2021, 35, 14634-14646.	5.1	6
53	Flue gas mercury removal using WS ₂ -doped carbon nitride via physical mixing. <i>Chemical Physics</i> , 2022, 562, 111643.	1.9	4
54	Fabrication and Fractality of Fe ₂ O ₃ -CeO ₂ /ZSM-5 Composites for High-Temperature Desulfurization. <i>Colloids and Interfaces</i> , 2017, 1, 10.	2.1	1

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55	Rare-Earth Oxide Desulfurizers. Energy and Environment Research in China, 2018, , 55-95.	1.1	0
56	Nano Elemental Metal Desulfurizers. Energy and Environment Research in China, 2018, , 97-120.	1.1	0