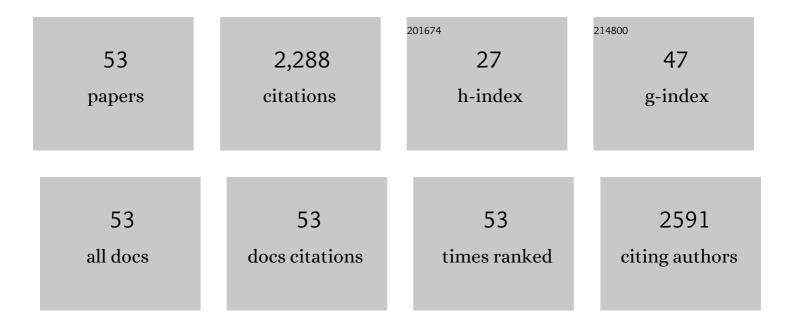


List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	DCU-Net: a dual-channel U-shaped network for image splicing forgery detection. Neural Computing and Applications, 2023, 35, 5015-5031.	5.6	19
2	A technological quality control system for rice supply chain. Food and Energy Security, 2023, 12, .	4.3	7
3	Incorporation of incompatible trace elements into molybdenite: Layered PbS precipitates within molybdenite. American Mineralogist, 2022, 107, 54-64.	1.9	8
4	Transformation of Ordered Albite into Kaolinite: Implication for the "Booklet―Morphology. ACS Earth and Space Chemistry, 2022, 6, 1133-1142.	2.7	3
5	Multiple Growth Mechanisms of 2:1 Type Layered Aluminosilicates during Mineral Transformation. ACS Earth and Space Chemistry, 2022, 6, 1930-1936.	2.7	0
6	Hyperledger Fabric Access Control System for Internet of Things Layer in Blockchain-Based Applications. Entropy, 2021, 23, 1054.	2.2	34
7	Application Research: Big Data in Food Industry. Foods, 2021, 10, 2203.	4.3	16
8	Formation of saponite by hydrothermal alteration of metal oxides: Implication for the rarity of hydrotalcite. American Mineralogist, 2019, 104, 1156-1164.	1.9	6
9	Transformation of boehmite into 2:1 type layered aluminosilicates with different layer charges under hydrothermal conditions. Applied Clay Science, 2019, 181, 105207.	5.2	7
10	Food Safety Supervision System Based on Hierarchical Multi-Domain Blockchain Network. IEEE Access, 2019, 7, 51817-51826.	4.2	62
11	Chemical and structural studies of coexisting 1M- and 2M1-polytypes in synthetic fluorophlogopites and influence of Al on the polytype formation. Physics and Chemistry of Minerals, 2019, 46, 259-270.	0.8	3
12	Hydrothermal transformation of mixed metal oxides and silicate anions to phyllosilicate under highly alkaline conditions. Applied Clay Science, 2018, 156, 224-230.	5.2	5
13	Structural effects on dissolution of silica polymorphs in various solutions. Inorganica Chimica Acta, 2018, 471, 57-65.	2.4	9
14	Adsorption of ammonium by different natural clay minerals: Characterization, kinetics and adsorption isotherms. Applied Clay Science, 2018, 159, 83-93.	5.2	218
15	Conversion of serpentine to smectite under hydrothermal condition: Implication for solid-state transformation. American Mineralogist, 2018, 103, 241-251.	1.9	25
16	Montmorillonite/chitosan nanoparticles as a novel controlled-release topical ophthalmic delivery system for the treatment of glaucoma. International Journal of Nanomedicine, 2018, Volume 13, 3975-3987.	6.7	70
17	Controlled drug delivery for glaucoma therapy using montmorillonite/Eudragit microspheres as an ion-exchange carrier. International Journal of Nanomedicine, 2018, Volume 13, 415-428.	6.7	19
18	Metal occupancy and its influence on thermal stability of synthetic saponites. Applied Clay Science, 2017, 135, 282-288.	5.2	17

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19	Phenol and/or Zn2+ adsorption by single- or dual-cation organomontmorillonites. Applied Clay Science, 2017, 140, 1-9.	5.2	33
20	Preparation of functionalized kaolinite/epoxy resin nanocomposites with enhanced thermal properties. Applied Clay Science, 2017, 148, 103-108.	5.2	43
21	An efficient SO2-adsorbent from calcination of natural magnesite. Ceramics International, 2017, 43, 12557-12562.	4.8	23
22	Transformation of halloysite and kaolinite into beidellite under hydrothermal condition. American Mineralogist, 2017, 102, 997-1005.	1.9	20
23	A novel ion-exchange carrier based upon liposome-encapsulated montmorillonite for ophthalmic delivery of betaxolol hydrochloride. International Journal of Nanomedicine, 2017, Volume 12, 1731-1745.	6.7	46
24	Kinetics and thermodynamic analysis of the adsorption of hydroxy-Al cations by montmorillonite. Applied Clay Science, 2016, 129, 79-87.	5.2	14
25	Silylation of saponite with 3-aminopropyltriethoxysilane. Applied Clay Science, 2016, 132-133, 133-139.	5.2	37
26	Preparation and in vitro study of lipid nanoparticles encapsulating drug loaded montmorillonite for ocular delivery. Applied Clay Science, 2016, 119, 277-283.	5.2	37
27	Adsorption of phenol and Cu(II) onto cationic and zwitterionic surfactant modified montmorillonite in single and binary systems. Chemical Engineering Journal, 2016, 283, 880-888.	12.7	112
28	Thermal analysis evidence for the location of zwitterionic surfactant on clay minerals. Applied Clay Science, 2015, 112-113, 62-67.	5.2	27
29	Surface chemistry and reactivity of SiO2 polymorphs: A comparative study on α-quartz and α-cristobalite. Applied Surface Science, 2015, 355, 1161-1167.	6.1	56
30	Multi-Authority Attribute Based Encryption Scheme with Revocation. , 2015, , .		12
31	Aluminum ion occupancy in the structure of synthetic saponites: Effect on crystallinity. American Mineralogist, 2014, 99, 109-116.	1.9	26
32	Investigation of structure and thermal stability of surfactant-modified Al-pillared montmorillonite. Journal of Thermal Analysis and Calorimetry, 2014, 115, 219-225.	3.6	13
33	Tailoring surface properties and structure of layered double hydroxides using silanes with different number of functional groups. Journal of Solid State Chemistry, 2014, 213, 176-181.	2.9	30
34	Restricting layer collapse enhances the adsorption capacity of reduced-charge organoclays. Applied Clay Science, 2014, 88-89, 73-77.	5.2	17
35	Kinetics, isotherm, thermodynamic, and adsorption mechanism studies of La(OH) 3 -modified exfoliated vermiculites as highly efficient phosphate adsorbents. Chemical Engineering Journal, 2014, 236, 191-201.	12.7	256
36	Al13-pillared montmorillonite modified by cationic and zwitterionic surfactants: A comparative study. Applied Clay Science, 2014, 101, 327-334.	5.2	13

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#	Article	IF	CITATIONS
37	From used montmorillonite to carbon monolayer–montmorillonite nanocomposites. Applied Clay Science, 2014, 100, 112-117.	5.2	39
38	Effect of functionalized kaolinite on the curing kinetics of cycloaliphatic epoxy/anhydride system. Applied Clay Science, 2014, 95, 317-322.	5.2	19
39	One-pot synthesis of Fe(III)-coordinated diamino-functionalized mesoporous silica: Effect of functionalization degrees on structures and phosphate adsorption. Microporous and Mesoporous Materials, 2013, 170, 200-210.	4.4	29
40	Studies on self-assembly hydrothermal fabrication and thermal stability of Chromium oxyhydroxide nanomaterials synthesised from Chromium oxide colloids. Journal of Thermal Analysis and Calorimetry, 2013, 111, 329-334.	3.6	1
41	Silylation of montmorillonite surfaces: Dependence on solvent nature. Journal of Colloid and Interface Science, 2013, 391, 16-20.	9.4	59
42	Silylation of clay mineral surfaces. Applied Clay Science, 2013, 71, 15-20.	5.2	134
43	Locking effect: A novel insight in the silylation of montmorillonite surfaces. Materials Chemistry and Physics, 2012, 136, 292-295.	4.0	48
44	Silylation of layered double hydroxides via an induced hydrolysis method. Journal of Materials Chemistry, 2011, 21, 10711.	6.7	37
45	Preparation and characterization of zwitterionic surfactant-modified montmorillonites. Journal of Colloid and Interface Science, 2011, 360, 386-392.	9.4	70
46	Silylation of Layered Double Hydroxides via a Calcinationâ^'Rehydration Route. Langmuir, 2010, 26, 2769-2773.	3.5	30
47	Thermal decomposition of silylated layered double hydroxides. Journal of Thermal Analysis and Calorimetry, 2010, 101, 153-159.	3.6	18
48	Nanomaterials based upon silylated layered double hydroxides. Applied Surface Science, 2009, 255, 4334-4340.	6.1	73
49	Effect of surfactant concentration on the stacking modes of organo-silylated layered double hydroxides. Applied Clay Science, 2009, 45, 262-269.	5.2	46
50	In situ synthesis of surfactant/silane-modified hydrotalcites. Journal of Colloid and Interface Science, 2008, 319, 498-504.	9.4	64
51	A combined study by XRD, FTIR, TG and HRTEM on the structure of delaminated Fe-intercalated/pillared clay. Journal of Colloid and Interface Science, 2008, 324, 142-149.	9.4	167
52	Synthesis and infrared spectroscopic characterization of selected layered double hydroxides containing divalent Ni and Co. Materials Chemistry and Physics, 2008, 112, 869-875.	4.0	43
53	Synthesis and characterization of layered double hydroxides with a high aspect ratio. Journal of Solid State Chemistry, 2006, 179, 708-715.	2.9	68